

Abstract

Three compounds were synthesized in this bachelor thesis– hydroxy(phenyl)methandiyl-bis(phosphonic) acid – disodium salt ($\text{Na}_2\text{H}_2\text{L}^1$), 1-hydroxy-2-phenylethan-1,1-diyl-bis(phosphonic) acid – disodium salt ($\text{Na}_2\text{H}_2\text{L}^2$) and tetraethyl hepta-1,6-diyne-4,4-diyl-bis(phosphonate) (L^4).

The next part of the thesis is focused on sorption towards to powdered titanium dioxide surface. Following compounds were used for these sorption experiments: $\text{Na}_2\text{H}_2\text{L}^1$, $\text{Na}_2\text{H}_2\text{L}^2$ and 1-hydroxy-2-(4-nitrophenyl)ethan-1,1-diyl-bis(phosphonic) acid (H_4L^3).

The experimental results show that compounds $\text{Na}_2\text{H}_2\text{L}^1$ and $\text{Na}_2\text{H}_2\text{L}^2$ have the ability to solubilize titanium dioxide into solution - for this reason there is no possibility to use UV-Vis spectroscopy to evaluate their sorption. In the case of H_4L^3 there are no solubilizing properties. The compound interacts intensively with titanium dioxide surface and forms monomolecular layer.

Keywords: Bis(phosphonates), sorption, titanium dioxide.