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Development of new syntheses of condensed aromatic compounds Vývoj nových syntéz kondenzovaných aromatických sloučenin

PhD Thesis

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Prague, 2018

## Prohlášení

Prohlašuji, že jsem tuto práci vypracoval samostatně pod vedením školitele prof. RNDr. Martina Kotory, CSc. a řádně ocitoval všechny použité prameny. Dále prohlašuji, že jsem tuto práci ani její část nepředložil k získání stejného nebo jiného akademického titulu.

V Praze dne

## Acknowledgment

I would like to thank Prof. RNDr. Martin Kotora, CSc. for his expert guidance, encouragement and support during all stages of the work and on this dissertation.

Besides, I thank Doc. RNDr. Jîri Mosinger, PhD for the measurement of photophysical properties, Doc. RNDr. Robert Gyepes, PhD for computational calculations, RNDr. Ivana Císařová, CSc for conducting X-ray analysis, Dr. Martin Popr and Ing. Adam Málek for IR measurement, and Carlos V. Melo, PhD for his consultation during the writing process. Besides, I want to thank Prof. Yasuhiro Uozumi for his invitation and supervision during a research project at the IMS in Okazaki, Japan.

I would also like to thank my colleagues for their advices and good working atmosphere in the laboratory and my wife, my family and friends for their support by any means and encouragement during my studies.

The work was financed by support from the Czech Science Foundation (Project 1817823S) and Charles University Grant Agency (Project 243-250104).

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## 1. Introduction

Since the first proposal of "spiro-concept" for organic material science by Salbeck et al. in the 1990 s, ${ }^{1}$ the following systematic studies have shown that compounds possessing spiro functionality are key substances for application in organic functional devices, such as organic semi-conductors, in monolayers, in light-emitting diodes (OLEDs) and in dyesensitized solar cells. ${ }^{2}$ The spiro concept is based on the connection of two or more more aromatic $\pi$-systems via an $\mathrm{sp}^{3}$-hybridized atom (a typical example is 9,9 -spirobifluorene, Figure 1).


Figure 1. 9,9'-Spirobifluorene SBF

The spiro compounds have improved morphological and thermal stability in comparison with the individual parent $\pi$-systems. The perpendicular arrangement efficiently suppresses unwanted excimer (excited dimer) formation reactions, extended $\pi$-systems increase the solubility in organic non-polar solvents and stability of the molecule in excited state, leading to high-emission color stability. ${ }^{1 \mathrm{~b}, 2 \mathrm{a}}$ In particular, the latter property is one of the key characteristics for continuous light-emitting devices. In the last decade, many studies on photophysical stability, electronic properties and applications of electroluminescent layers based on spiro compounds have been reported. ${ }^{2 b, 3}$ Furthermore, several reports predict that such organic light-emitting diodes (OLEDs), phosphorescent organic light-emitting diodes (PhOLEDs), ${ }^{3 \mathrm{a}}$ white organic light-emitting diodes (WOLEDs) ${ }^{4}$ and other devices will become the future of illumination systems. ${ }^{5}$

Substituted 9, ${ }^{\prime}$-spirobifluorenes and dispiroindenofluorenes constitute a class of polycyclic aromatic hydrocarbon spiro compounds (Figure 2) that significantly contribute to the aforementioned areas because of their unique physical and chemical properties. ${ }^{1-6}$ The thesis will deal with development of new synthetic pathways for preparation of various 1,2,3,4-tetrasubstituted spirobifluorenes (SBFs), dispiroindeno[2,1-c]fluorenes (DS-IFs) and their derivatives as well as with possibilities for tuning of their photophysical properties.



Figure 2. 1,2,3,4-Tetrasubstituted SBF and DS-IF compounds in comparison

## 2. Literature Review

### 2.1. Synthesis of fluorene compounds

Polycyclic aromatic hydrocarbons (PAHs) are one of the most extensively studied organic substances for their unique properties and broad applications in many areas of chemistry, e.g. materials science. For the synthesis of fluorenes, mainly two disconnection approaches $A$ and $B$ have been used, leading to suitably substituted diphenyl methane compounds (Scheme 1, Disconnection A) or biphenyls as starting materials (Disconnection B). Ring-closing reactions can be performed under either transition metal-, Lewis acid-, or base-mediated or -catalyzed conditions.


Scheme 1. General disconnection approaches A and B of fluorenes

### 2.1.1. Acid- and base-catalyzed synthesis

One of the most used synthetic strategies, based on intramolecular Friedel-Crafts reaction starting from biphenyl or benzophenone precursors, was developed in the 1970s. ${ }^{7}$ At present, a number of synthetic strategies utilizing substituted (1,1'-biphenyl)-2-carboxylic acids I and their derivatives (e.g. amides) are available (Scheme 2). As typical examples may serve reactions carried out in the presence of thionyl chloride $\left(\mathrm{SOCl}_{2}\right),{ }^{8, \mathrm{~b}}$ polyphosphoric acid $(\mathrm{PPA}),{ }^{8 \mathrm{c}}$ methanesulfonic acid $(\mathrm{MsOH})^{8 \mathrm{~d}}$ and trifluoroacetic anhydride $\left(\left(\mathrm{CF}_{3} \mathrm{CO}\right)_{2} \mathrm{O}\right)^{8 \mathrm{e}}$ producing substituted fluorenones II. In addition, mixtures of bis(pyridine)iodonium(I) tetrafluoro-borate/ $\mathrm{HBF}_{4} \quad\left(\mathrm{IPy}_{2} \mathrm{BF}_{4} / \mathrm{HBF}_{4}\right)^{8 \mathrm{ff}}$ or tetraethylammonium bromide/potassium persulfate (TEAB/ $\left.\mathrm{K}_{2} \mathrm{~S}_{2} \mathrm{O}_{8}\right)^{8 g}$ were also successfully used to perform ring closing FriedelCrafts reaction of (1,1'-biphenyl)-2-carbaldehydes to fluorenones II.


Scheme 2. Friedel-Crafts reactions of I to fluorenes II

Recently, $\mathrm{AlCl}_{3}$-catalyzed reaction of differently substituted (1,1'-biphenyl)-2carbaldehydes III in the presence of sulfonamide has been reported. It demonstrated that 9aminofluorenes IV can be easily accessed in up to $88 \%$ (Scheme 3). ${ }^{9}$


Scheme 3. $\mathrm{AlCl}_{3}$-catalyzed condensation cyclization cascade from III to fluorenones II

In 2006, Rueping et al. introduced Friedel-Crafts benzylation of substituted 2phenylbenzyl alcohols $\mathbf{V}$ with $\mathrm{Bi}(\mathrm{OTf})_{3}$ as Lewis acid to promote cyclization to fluorenes $\mathbf{V I}$ in up to $95 \%$ yield (Scheme 4). ${ }^{10}$


Scheme 4. $\mathrm{Bi}\left(\mathrm{OTf}_{3}\right)_{3}$-promoted intramolecular benzylation reaction of $\mathbf{V}$ to fluorenes $\mathbf{V I}$

In addition to Lewis acids, base-mediated reactions have also been used to synthesize compounds with the fluorene scaffold. This method is based on the use of a sterically hindered base (LDA) in a combination with a tertiary amide VII, which undergoes cyclization to provide benzofluorenone VIII. However, this method suffers from a setback such as a limited functional group tolerance (Scheme 5). ${ }^{11}$


Scheme 5. LDA-mediated synthesis of benzofluorenes VIII

Another base-promoted synthetic approach was developed by Studer et al. They explored radical cross-dehydrogenative coupling via base-mediated homolytic aromatic substitution. They prepared fluorenones II in up to $84 \%$ from (1,1'-biphenyl)-2-carbaldehydes III using $t$ - BuOOH as oxidant. $\mathrm{FeCp}_{2}$ was the most effective radical chain initiator (Scheme 6). ${ }^{12}$


Scheme 6. Base-promoted radical cross-dehydrogenative coupling in (1,1'-biphenyl)-2carbaldehydes III to fluorenones II

The syntheses of symmetrically octa- and nonamethylated fluorenes were studied as well, particularly by O'Hare ${ }^{13 \mathrm{a}}$ and Tilley ${ }^{13 \mathrm{~b}}$. Both synthetic procedures were based on the use of 1,2,3,4-tetramethylbenzene IX as the starting material. In the first case, O'Hare used highly alkaline conditions for cyclization of the monobrominated benzophenone $\mathbf{X}$ to fluorenone XII in $53 \%$ isolated yield. In the second approach, Tilley used electrophilic aromatic substitution and cyclization of XI with paraformaldehyde, producing the target fluorene XIII in $80 \%$ yield (Scheme 7).


Scheme 7. Synthesis of octamethylfluorenone XII and octamethylfluorene XIII

### 2.1.2. Transition metal-catalyzed synthetic approaches

In addition to acid- and base-catalyzed synthetic approaches, many examples of transition metal-catalyzed reactions have emerged during the past several decades; however, many of them suffer from similar drawbacks and limitations as the aforementioned strategies. ${ }^{14}$ As a typical starting material used in syntheses of functionalized fluorenes serve substituted benzophenones, diarylmethanes, or biaryl compounds. In addition to these precursors, also new methods based on utilization of various alkenes, alkynes and their derivatives for annulation have appeared.

### 2.1.2.1. Benzophenones/diarylmethanes as precursors

In the early 1980s, Ames and Opalko described the first pioneering studies on intramolecular dehydrogenative coupling of halogenated diaryl derivatives using catalytic $\mathrm{Pd}(\mathrm{II})$-acetate under basic conditions, albeit with many side products. ${ }^{15}$ As a representative of Rh-complex catalyzed formation of fluorenones, Blum et al. developed conversion of benzoic anhydride into fluorenones catalyzed by $\mathrm{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}$ in the late 1960s. ${ }^{16}$ More recently, numerous procedures based on intramolecular cyclization of substituted benzophenones XIV into target fluorenones II catalyzed by Pd -, Ag - or Cu -compounds have been reported (Scheme 8).


XIV, $R=$ alkyl, aryl,
alkoxy, halide
$A^{1} \mathrm{Y}=\mathrm{H} \quad \mathrm{Pd}(\mathrm{OAc})_{2}(5 \mathrm{~mol} \%), \mathrm{Ag}_{2} \mathrm{O}, \mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{TFA}, 140^{\circ} \mathrm{C}, 24 \mathrm{~h}$, up to $91 \%$.
$A^{2} \mathrm{Y}=\mathrm{H} \quad \mathrm{Pd}(\mathrm{OAc})_{2}(10 \mathrm{~mol} \%), \mathrm{Ag}_{2} \mathrm{O}, \mathrm{TFA}, 130^{\circ} \mathrm{C}, 24 \mathrm{~h}$, up to $90 \%$.
$A^{3} \mathrm{Y}=\mathrm{H} \quad \mathrm{Pd}(\mathrm{II})$ on $\mathrm{MgLa}(5 \mathrm{~mol} \% \mathrm{Pd}), \mathrm{Ag}_{2} \mathrm{O}$, TFA: $\mathrm{H}_{2} \mathrm{O}, 130^{\circ} \mathrm{C}, 24 \mathrm{~h}$, up to $76 \%$.
$B^{1} \mathrm{Y}=\mathrm{Br}, \mathrm{I} \quad \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(6 \mathrm{~mol} \%), \mathrm{KOAc}, \mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{DMA}, 110^{\circ} \mathrm{C}, 23 \mathrm{~h}$, up to $96 \%$.
$B^{2} \mathrm{Y}=\mathrm{Br}, \mathrm{I} \quad \mathrm{Cu}(\mathrm{I})(15 \mathrm{~mol} \%), \mathrm{K}_{3} \mathrm{PO}_{4}, \mathrm{DMF}, 160^{\circ} \mathrm{C}, \mathrm{MW}$, up to $90 \%$.
C $\mathrm{Y}=\mathrm{COOH} \mathrm{AgOAc}(20 \mathrm{~mol} \%), \mathrm{K}_{2} \mathrm{~S}_{2} \mathrm{O}_{8}, \mathrm{MeCN}, 130^{\circ} \mathrm{C}, \mathrm{MW}, 1 \mathrm{~h}$, up to $84 \%$.
Scheme 8. Synthetic pathways to fluorenones II from substituted benzophenones XIV

Shi ${ }^{17 \mathrm{a}}$ and Cheng ${ }^{17 \mathrm{~b}}$ independently and concurrently developed Pd-catalyzed oxidative dehydrogenative dual C-H-functionalization of benzophenones in 2012 (Scheme 8, approach $\boldsymbol{A}^{1}$ and $\boldsymbol{A}^{2}$ ). Recently, a heterogeneous catalytic dual C-H activation system ( Pd on MgLa ) have been also developed (Approach $\boldsymbol{A}^{3}$ ). ${ }^{17 \mathrm{c}}$ Direct intramolecular arylation of substituted halogenated benzophenones was carried out under Pd- ${ }^{17 \mathrm{~d}}$ or Cu -catalysis ${ }^{17 \mathrm{e}}$ (Approach $\boldsymbol{B}^{1}$ and $\boldsymbol{B}^{\mathbf{2}}$ ). In addition to these approaches, radical decarboxylative C-H arylation under Ag-catalysis was reported as well (Approach C). ${ }^{17 \mathrm{f}}$

Several other diarylmethane derivatives were applied in syntheses of fluorene compounds. Miura et al. presented Rh- and Ir-catalyzed dehydrogenative cyclization via double C-H bond cleavage of $\mathbf{X V}$ to form fluorene $\mathbf{X V I}$ in the presence of Cu salts (Scheme 9). ${ }^{18}$


A $\mathrm{Y}=\mathrm{NH}_{2} \quad[\mathrm{RhCl}(\mathrm{cod})]_{2}(2 \mathrm{~mol} \% \mathrm{Rh}), \mathrm{Cu}(\mathrm{OAc})_{2}(2 \mathrm{eq})$, xylene, up to $98 \%$.
B $\mathrm{Y}=\mathrm{OH}, \mathrm{COOH} \quad\left[\mathrm{Cp}^{*} \mathrm{IrCl}_{2}\right]_{2},(2 \mathrm{~mol} \% \mathrm{Ir}), \mathrm{Cu}(\mathrm{OAc})_{2}(2 \mathrm{eq})$, dodecane, up to $82 \%$.
Scheme 9. Rh- and Ir-catalyzed synthesis of fluorene XVI from diarylmethanes XV

Another example of syntheses of compounds with the fluorene scaffold from diaryl ether, benzophenone, or diarylmethane derivatives consists of a Ag-catalyzed Pschorr-type radical cyclization. It furnished the target benzofurane, fluorenone, and fluorene compounds in up to $77 \%$ isolated yields (Scheme 10). ${ }^{19}$

$\mathrm{R}=$ alkoxy, ester, halide
$\mathrm{Y}=\mathrm{O}, \mathrm{CH}_{2}, \mathrm{CHOH}, \mathrm{CO}$
Scheme 10. Ag-catalyzed synthesis fluorenes XVIII from diarylethers, benzophenones, or diarylmethanes

In addition to these approaches, a two-step Pd-complex catalyzed procedure starting from substituted bromobenzaldehydes and arylboronic acids was developed. In a one-pot reaction various arylboronic acids XIX were added to the carbonyl group of 2bromobenzaldehydes $\mathbf{X X}$ initially forming intermediate benzylic alcohols and, upon heating, the intermediate underwent cyclization via a $\mathrm{C}-\mathrm{H}$ activation and oxidation sequence, producing the target fluorenone compounds II in up to $90 \%$ (Scheme 11). ${ }^{20}$

$\mathrm{R}=$ alkyl, alkoxy, halide
Scheme 11. One-step synthesis of 2-bromobenzaldehydes XIX with arylboronic acids XX

### 2.1.2.2. Syntheses starting from 1,1'-biphenyls and tandem reactions

### 2.1.2.2.1. 1,1’-Biphenyls as precursors

The second approach, using transition metal-catalyzed reactions to provide fluorene and fluorenone compounds, is based on the use of appropriately substituted biphenyls. One of the first reports on the synthesis of fluorenes based on the use of transition metal compounds, albeit a stoichiometric one, appeared in the late 1980s. It was based on reactions of $2,2^{\prime}$ dilithiobiphenyls XXI with $\mathrm{PPh}_{3}\left(\mathrm{CO}_{4}\right) \mathrm{ReBr}$ that served as the CO source. The corresponding substituted fluorenones II were obtained in up to $68 \%$ isolated yields (Scheme 12, A). ${ }^{21 a}$ In 2000, Larock reported a Pd-complex catalyzed cyclocarbonylation of 2-halo-1,1'-biphenyls XXII with CO (1 atm) to synthesize fluorenones II in up to quantitative yields (Scheme 12, B). ${ }^{2 \mathrm{~b}, \mathrm{c}}$


Scheme 12 Synthetic procedure for fluorenones II from (A) dilithiobiphenyl XXI or (B) 2-halo-1,1'-biphenyls XXII

Currently, many approaches to fluorenes are based on Pd-complex catalyzed C-H bond activation and functionalization. As typical examples may serve transformations of 2-(halomethyl)-1,1'-biphenyls XXIII or 2-alkynyl-1,1'-biphenyls XXV to fluorenes XXIV (Scheme 13, A) ${ }^{22 \mathrm{a}}$ and XXVI (Scheme $13, \boldsymbol{B}$ ). ${ }^{22 \mathrm{~b}}$ The reactions proceed via an intramolecular dehydrogenative cyclization with good-to-excellent yields in all reactions. In addition, a Pdcatalyzed decarboxylative cyclization of (1,1'-biphenyl)-2-carboxylic acids XXVII using $t$-butyl isocyanide as the carbonyl source was also explored and it gave rise to substituted fluorenones II (Scheme 13, C). ${ }^{22 \mathrm{c}}$

A


XXIII, R = alkyl, aryl, alkoxy, halide

B


XXV, R = alkyl, aryl, alkoxy, halide

XXIV, up to $99 \%$



XXVI, up to $98 \%$


Scheme 13. Pd-catalyzed synthesis of fluorenones II

Although Pd-complex catalyzed methods have been the most commonly used for the synthesis of fluorene derivatives, also Rh-complex catalyzed procedures have been developed as well. These methods have found its application either in intramolecular insertion of carbenes in biaryldiazoacetates XXVIII to form fluorenyl carboxylates XXIX (Scheme 14, A, the patent did not show any specific catalytic system) $)^{23 a}$ or in intramolecular acylation of biarylcarboxylic acids XXVII to form fluorenones II (Scheme 14, B). ${ }^{23 \mathrm{~b}}$






Scheme 14. Rh-catalyzed synthesis of fluorene-9-carboxylates XXIX and fluorenones II

In 2014, the groups of Echavarren and Ye independently published two different Aucatalyzed synthetic procedures to form fluorenes. Echavarren showed that a Au-complex catalyzed retro-Buchner reaction of 2-(2,4,6-cycloheptatrien-1-yl)-1,1'-biphenyls XXX provided fluorenes XXIV (Scheme 15, A). ${ }^{24 \mathrm{a}}$ Ye et al. exploited a Au-complex intermolecular oxidative cyclization 2-alkynyl-1,1'-biphenyls XXXI in the presence of $N$-oxides to fluorene-9-carboxylic acid amides XXXII (Scheme 15, B). ${ }^{24 \mathrm{~b}}$

A


B


Scheme 15. Au-catalyzed formation of XXIV and fluorenylcaboxylic acid amides XXXII
The last examples of transition metal-catalyzed or -promoted strategies for the synthesis of fluorenes from the parent biaryl precursors are Mo- and Nb -catalyzed reactions.

Double activation of C-F and C-H bonds in substituted 2-trifluoromethylbiphenyls XXXIII with low valent Nb -compounds (prepared by reduction with $\mathrm{LiAlH}_{4}$ ) was used for the synthesis of fluorenes XXIV in up to $92 \%$ yield (Scheme $16, \boldsymbol{A}$ ). ${ }^{25 a}$ In the area of Mocatalyzed reactions, nanostructured $\mathrm{MoO}_{3}$ was used to initiate intramolecular benzylation (Friedel-Crafts reaction) of [(1,1'-biphenyl)-2-yl]methanol XXXIV forming fluorene XXIV in $88 \%$ yield (Scheme 16, B). ${ }^{25 b}$


Scheme 16. Nb- and Mo-catalyzed formation of fluorenes XXIV

### 2.1.2.2.2. Tandem syntheses via biaryl intermediates

Development of tandem procedures for preparation of fluorenes via biphenyl intermediates is highly desirable, because they shorten the reaction sequence and avoid the need for the synthesis of the intermediate. The tandem procedures comprise two synthetic steps based on two reaction. In most cases the first step is a cross-coupling reaction that provides the intermediate biphenyls. These then, without isolation, undergo the second step: a ring-closing reaction.

In 2006, Hu et al. published a series of Pd-complex catalyzed reactions of variously substituted 1,2-dihaloarenes or tosylated haloarenes XXXV with sterically hindered Grignard reagents XXXVI forming the biphenyl intermediates. The subsequent cyclization proceeded via the C-H bond activation of the methyl group followed by cross coupling producing the desired fluorenes XXIV (Scheme 17). ${ }^{26}$


$$
\begin{aligned}
& \mathrm{Y}^{1}, \mathrm{Y}^{2}=\mathrm{Cl}, \mathrm{Br}, \mathrm{I}, \mathrm{OTs} \\
& \mathrm{R}=\text { alkyl, alkoxy }
\end{aligned}
$$

Scheme 17. Tandem Pd-catalyzed synthesis of fluorenes XXIV

In addition to Grignard reagents, various arylboronic acids were used in Suzuki crosscoupling with substituted haloarenes. Tandem Pd-catalyzed Suzuki cross-coupling of substituted 2-methylphenylboronic acids XX with 1,2-dihaloarenes XXXVII followed by cyclization through the $\mathrm{C}-\mathrm{H}$ bond activation was used for synthesis of fluorenes XXIV (Scheme 18, A). ${ }^{27 \mathrm{a}}$ A similar approach was also used in the reaction of 1,2-dihaloarenes XXXVII and arylboronic acids XX under CO atmosphere in Pd-catalyzed synthesis to provide fluorenones II (Scheme 18, B). ${ }^{27 \mathrm{~b}}$ Alternatively, Suzuki cross-coupling of 2bromobenzaldehydes XIX and 2-bromophenylboronic acids XX giving rise to intermediate 2'-bromo-( 1,1 '-biphenyl)-2-carbaldehydes was also used as the first step. It was then followed by intramolecular addition of arylpalladium species to the carbonyl group forming fluorenones II (Scheme 18, C) ${ }^{27 c}$


$X^{1}, X^{2}=C l, B r, I$
$\mathrm{R}=$ alkyl, aryl, alkoxy, halide


Scheme 18. Tandem Pd-catalyzed procedures for synthesis of various fluorenes XXIV and fluorenones II

In addition to this approach, 2-halobenzaldehydes have been often used for the $\mathrm{Pd}-$ catalyzed cross-coupling step with other substrates, e.g. silanes and boronates. In 2005, Larock et al. reported a Pd-catalyzed tandem annulation of 2-halobenzaldehydes XIX and 2trimethylsilylaryl triflates XXXVIII to furnish fluorenones II in up to $75 \%$ yields. In this approach, the intermediate aryne was formed in situ and it was subsequently added to arylpalladium species, forming a new arylpalladium species. This intermediate was then intramolecularly added to the carbonyl group giving rise to fluorenones II (Scheme 19, $\boldsymbol{A}) .{ }^{28 a, b}$ Another tandem cyclization based on Pd-catalyzed cross-coupling of 2bromobenzaldehyde XIX and potassium 3-hydroxyphenyl trifluoroborate XXXIX followed by cyclization to fluorenones II was reported (Scheme 19, B). ${ }^{28 \mathrm{c}}$


Scheme 19. Pd-catalyzed method for synthesis of fluorenones II

In 2016, You et al. successfully conducted, for the first time, chelation-assisted C-H activation of aromatic carboxylic acids. They used Pd-catalyzed ortho-selective oxidative C-H cross-coupling of benzoic acids XL with arenes XLI, followed by intramolecular FriedelCrafts acylation to furnish target fluorenones II (Scheme 20). ${ }^{29}$


Scheme 20. Pd-catalyzed C-H cross-coupling of carboxylic acids XL followed by FriedelCrafts reaction to fluorenes II

Finally, other examples of tandem Pd-catalyzed cross-coupling and cyclization procedures involved the use of oximes, nitriles or benzyl amines. Cheng ${ }^{30 \mathrm{a}, \mathrm{b}}$ and $\mathrm{Shi}^{30 \mathrm{c}}$ reported a Pd-catalyzed functionalization of ortho-methyl oximes XLII with arylboronic acids XX to biphenyl aldoximes, followed by cyclization (Scheme 21, A). The subsequent acidic hydrolysis yielded fluorenones II. This approach was similar to the previous mechanistic studies by Larock et al. ${ }^{30 \mathrm{~d}}$ The second approach used Pd-catalyzed remote nitrile-
directed dual C-H activation of substituted benzonitriles XLIII and subsequent reaction with aryl iodides XXIV (Scheme 21, B). ${ }^{30 e, f}$ Interestingly, even reactions of simple benzylamines XXXV with aryl iodides XXXIV were suitable for tandem Pd-catalyzed synthesis of fluorenones II (Scheme 21, C). ${ }^{30 \mathrm{~g}}$


$$
\begin{aligned}
& \text { R = alkyl, alkoxy, aryl } \\
& \text { ester, halide }
\end{aligned}
$$



$$
\begin{gathered}
\mathrm{R}=\mathrm{alkyl}, \text { alkoxy, } \\
\text { ester, halide }
\end{gathered}
$$



Scheme 21. Syntheses of fluorenones II by using Pd-catalyzed tandem reaction

### 2.1.2.3. Alkene and alkyne derivatives as precursors

Following two previously described transition metal-catalyzed approaches that were based on the use of benzophenone/diarylmethane derivatives (Section 2.1.2.1) or various substituted biphenyl precursors (Section 2.1.2.2), other synthetic approaches using various substituted alkenes and alkynes as starting materials have also emerged in the last decades.

The first synthetic strategy is based on the annulation of enynes and alkynes. In the first method fluorenes XLVIII were synthesized under Au-catalysis from 1,3-enynes XLVI and propargyl esters XLVII. The reaction was initiated by intermolecular cyclopropanation of enynes via gold carbenoids, which were formed by rearrangement of propargyl esters. The rearrangement started with coordination of the cationic gold catalyst followed by 5-endo-dig cyclization that lead to formation of tertiary carbocation. Then, after several other skeletal rearrangements, cyclopropyl ring opening and nucleophilic substitution followed forming the target fluorene XLVIII (Scheme 22, A). ${ }^{31 a}$ The second approach, using appropriately substituted enynes and alkynes as starting material, was reported by Liang et al. They successfully applied $\mathrm{BiCl}_{3}$-catalyzed reaction of ( $Z$ )-pent-2-en-yl acetates XLIX with ethynylarenes $\mathbf{L}$ via tandem isomerization/Friedel-Crafts intermolecular electrophilic reaction/cycloisomerization/aromatization sequence to form substituted fluorenes LI (Scheme $22, B) .{ }^{3 \mathrm{~b}, \mathrm{c}}$

 $\mathrm{R}=$ alkyl, alkoxy, aryl, halide

Scheme 22. Au- and Bi-catalyzed enyne and alkyne-based annulation reactions to fluorenes XLVIII and LI

In 2013, two new similar approaches were reported for the synthesis of fluorenes using reactions of dienes with alkynes. They involved either inter- or intramolecular reaction pathways. The intermolecular approach to fluorenones II was based on the following reaction sequence: Co-catalyzed Diels-Alder reaction of aryl-substituted propiolates LII and appropriately substituted dienes LIII, followed by DDQ oxidation providing in situ a biphenyl intermediate, followed by acidic Friedel-Crafts cyclization to form target fluorenone II (Scheme 23, A) ${ }^{32 a}$ The intramolecular pathway utilized $\mathrm{TiCl}_{4}$-mediated cyclization of substituted hept-4,6-diene-1-yn-3-ols LIV. The propargylic alcohol was converted in the presence of $\mathrm{TiCl}_{4}$ into the diene-allenylic cation intermediate through Meyer-Schuster rearrangement, which underwent electrocyclization and isomerization, thereby forming the target fluorenes $\mathbf{L V}$ (Scheme 23, B). ${ }^{32 b}$


Scheme 23. Conversion of alkynes and dienes to fluorenes II and LV

Lastly, a class of not fully conjugated fluorenes, i.e. compounds with incomplete aromatic systems, can also be assembled using unsaturated molecular systems. Radical cyclization of 1,6-enynes LVI to unsaturated compounds with the fluorene scaffold LVII was conducted using either $\mathrm{AgSCF}_{3}{ }^{33 \mathrm{a}}$ or $\mathrm{Ag}(\mathrm{OAc})$ in the presence of diphenylphosphine oxide (Scheme 24). ${ }^{33 \mathrm{~b}}$ It underwent a regioselective radical intramolecular addition to the alkyne moiety through the same mechanistic pathway. The addition yielded an alkenyl radical intermediate, which subsequently formed the secondary radical by 6 -exo-trig ring closing. Intramolecular addition of this intermediate to the aromatic ring formed the target compounds LVII.


Conditions for:
$\mathrm{Z}=\cdot \mathrm{SCF}_{3}: \mathrm{AgSCF}_{3}(1.5 \mathrm{eq}), \mathrm{K}_{2} \mathrm{~S}_{2} \mathrm{O}_{8}(3 \mathrm{eq}), \mathrm{HMPA}(0.5 \mathrm{eq})$, terpyridine ( $10 \mathrm{~mol} \%$ ), MeCN/DMF, $80^{\circ} \mathrm{C}$, up to $87 \%$.
$\mathrm{Z}=\cdot \mathrm{POR}_{2}: \operatorname{AgOAc}(10 \mathrm{~mol} \%), \mathrm{HPOR}_{2}, \mathrm{Zn}\left(\mathrm{NO}_{3}\right)_{2}(2 \mathrm{eq}), \mathrm{MeCN}, 10{ }^{\circ} \mathrm{C}$, up to $85 \%$.
Scheme 24. Ag-promoted radical formation of non-fully aromatic fluorenes XLVII

In addition to the Ag-mediated radical approach, also Au-catalysis and, very recently, Cu -mediated cyclization of 1,6-diynes were also reported. Au-catalysis of substituted, nonactivated diynes LVIII promoted intramolecular [3+2] cycloaddition via a Au-(I)-containing $\pi$-alkyne moiety to form a vinylgold intermediate. Subsequently, this intermediate underwent either 5-exo-dig or Nazarov cyclization, thus forming the fluorenes LIX (Scheme 25, A) . ${ }^{34 \mathrm{a}}$ In the second approach, treatment of substituted 1,6-diynes LVIII with Cu and sodium sulfonate led to the formation of two new C-C bonds and the assembly of fluorenes $\mathbf{L X}$. The reaction involved tandem nucleophilic addition of sulfonate and the subsequent cascade cyclization via an ionic mechanism (Scheme 25, B). ${ }^{34 b}$
A


$\mathrm{Y}=\mathrm{O}, \mathrm{CH}_{2}, \mathrm{NTs}$,
$\mathrm{C}(\mathrm{COOMe})_{2}$
R = alkyl, alkoxy
B

$\mathrm{Y}=\mathrm{O}, \mathrm{CH}_{2}, \mathrm{NTs}$
$\mathrm{C}(\mathrm{COOMe})_{2}$
R = alkyl, alkoxy,
amino, halide

Scheme 25. Au-catalyzed and Cu-mediated annulations to of LVIII to LIX and LX

### 2.1.3. 1,2,3,4-Tetrasubstituted fluorenes

Synthesis of 1,2,3,4-tetrasubstituted fluorenes is quite challenging due to high level of substitution. In this respect only a handful of procedures have been developed so far. One of the first approaches to 1,2,3,4-tetrasubstituted fluorenes was developed in 2009 by Takahashi et al. and was based on Lewis-acid mediated rearrangement of homo-substituted bis(indenyl)zirconacyclopentadienes, prepared by reductive dimerization of the corresponding alkynes with reduced titanocene. The zirconacyclopentadienes were treated with an excess of $\mathrm{TiCl}_{4}$ that promoted the skeletal rearrangement, formally addition of the ziconacyclopentadiene moiety to the indenyl ligand, resulting in the formation of 1,2,3,4tetrasubstituted fluorenes XXIV in up to $90 \%$ yield (Scheme 26). ${ }^{35}$


Scheme 26. Zr-catalyzed formation of 1,2,3,4-tetrasubstituted fluorenes

The second approach, for the preparation of the same tetrasubstituted fluorene scaffold was based on the use of Dewar benzenes. The Dewar benzenes were prepared by a reaction of methyl phenylpropynoate LXI with tetraalkylcyclobutadiene- $\mathrm{AlCl}_{3}$ complexes LXII. Then the ester moiety was hydrolyzed to free carboxylic acid and photochemically rearranged to substituted (1,1'-biphenyl)-2-carboxylic acids. The treatment of the acids with thionyl chloride furnished the 1,2,3,4-tetrasubstituted fluorenones II (Scheme 27). ${ }^{8 a}$


Scheme 27. Formation of a 1,2,3,4-tetrasubstituted fluorenones II from substituted cyclobutadienes

Since the above mentioned methods to 1,2,3,4-tetrasubstituted fluorene and fluorenones have limitations with respect to substituents R , development of other methods was desirable.

As a part of my previous project I studied a new reaction pathway for the synthesis of 1,2,3,4-tetrasubstituted fluorenes based on a transition metal catalyzed [2+2+2] cyclotrimerization of symmetrically substituted diynes with alkynes to provide various alkyl and aryl substituted 9 H -fluorenes in up to $93 \%$ isolated yield (see section 4.1.1). ${ }^{36}$

Soon after publishing of my study, the group of Ratovelomanana-Vidal et al. reported in 2016 modified [2+2+2] cyclotrimerization reaction of diynones LXIII with alkynes to furnish fluorenones II by using Ru-catalysis under simple conditions (solvent-free, no additional ligands, no additives) in up to $84 \%$ yield (Scheme 28). ${ }^{37}$


Scheme 28. Ru-catalyzed route to unsymmetrically substituted fluorenones II

### 2.2. Synthesis of indeno[ $\mathbf{2 , 1 - c}]$ fluorene compounds

In contrast to the vast research on syntheses of fluorenes conducted thus far, few synthetic strategies for indeno[2,1-c]fluorenes have been developed or are available. In fact, only a handful of studies have been published after the pioneering work of Ginsburg, Chardonnens and Stanfield, in the 1960s. They synthesized indeno[2,1-c]fluorenes via Lewis acid-assisted annulations of 3,3'-bi-(indenyl-cyclohepta-2,7-dienyls) or phenylcinnamalones. ${ }^{38}$ The family of indenofluorenes consist of five different constitutional isomers, ${ }^{39}$ chemistry and synthesis of which have been studied to different extents. These studies have mostly focused on the synthesis of indeno[1,2-a]fluorene, indeno[2,1-a]fluorene, indeno[1,2-b]fluorene and indeno[2,1-b]fluorene isomers, overlooking the indeno[2,1$c$ ]fluorene isomer (Figure 3).


11,12-dihydroindeno
[2,1-a]fluorene


6,12-dihydroindeno
[1,2-b]fluorene


5,7-dihydroindeno
[2,1-b]fluorene


7,12-dihydroindeno [1,2-a]fluorene


5,8-dihydroindeno [2,1-c]fluorene

Figure 3. Isomers of the dihydroindenofluorene family

Furthermore, almost all known examples of the reported synthesis of indeno[2,1c]fluorenes are based on either intramolecular Friedel-Crafts reaction (Scheme 29, Disconnection approach A), on intermolecular Diels-Alder reaction (Disconnection approach B) or on a combination of both.


Scheme 29. General disconnection approaches A and B for indeno[2,1-c]fluorenes

The most intramolecular Friedel-Crafts reactions (Disconnection A) use substituted terphenyls bearing diester groups LXIV and acids, e.g. polyphosphoric acid (PPA), ${ }^{6 \mathrm{~b}} \mathrm{FeCl}_{3}{ }^{6 \mathrm{c}}$ or $\mathrm{H}_{2} \mathrm{SO}_{4}$, ${ }^{6 \mathrm{~d}}$ to catalyze the double Friedel-Crafts reaction to furnish indeno[2,1c]fluorendiones LXV (Scheme 30, A). Moreover, a similar Friedel-Crafts reaction for the assembly of LXVII was also reported by Poriel et al., in 2015. They started from substituted 9,9'-spirobifluorene LXVI bearing 2-ethyl benzoate functionality at position 4 of the core structure and used MsOH to mediate the cyclization (Scheme 30, B). ${ }^{6 \mathrm{a}}$


Scheme 30. Intramolecular Friedel-Crafts reaction for synthesis of indeno[2,1-c]fluorene

## LXVII

In 2014, Chen et al. presented another synthetic approach for the preparation of substituted indeno[2,1-c]fluorenes based on intermolecular Diels-Alder reaction. The reaction included Diels-Alder reaction of diene LXVIII with an excess of maleic anhydride. The subsequent aromatization with $\mathrm{Pb}(\mathrm{OAc})_{4}$ led to the formation of indeno[2,1-c]fluorene LXIX in a high yield of $90 \%$ (Scheme 31). ${ }^{6 e}$


LXVIII
 $100^{\circ} \mathrm{C}, 4 \mathrm{~h}$


LXIX, 90\%

Scheme 31. Intermolecular Diels-Alder reaction to indeno[2,1-c]fluorene LXIX

Another strategy combining the two above mentioned synthetic approaches was also developed. The synthesis included intermolecular Diels-Alder reaction of indanocyclone LXX with 9-ethynyl-9-fluorenol LXXI, which led to the formation of a mixture of fluorenone LXXII and its structural isomer LXXII-a in a 1:1 ratio. Friedel-Crafts alkylation took place in the presence of $\mathrm{HOAc} /$ catalytic amount of HCl mixture under reflux and produced monospirobifluorene-indeno[2,1-c]fluorene LXXIII in a high yield of $96 \%$ (Scheme 32). ${ }^{6 f}$




Scheme 32. Diels-Alder/intramolecular Friedel-Crafts reaction to indeno[2,1-c]fluorene LXXIII

An unusual synthetic strategy, significantly differing from pathways $\boldsymbol{A}$ and $\boldsymbol{B}$, was developed in the group of Tessier et al. in 1991. This approach consists of lithium-induced cyclization of tribenzocyclotriyne LXXIV. It was shown that combining 4 equivalents lithium with tribenzocyclotriyne LXXIV in THF under anaerobic conditions yielded dianion LXXV. Its methanolysis gave rise to the target indeno[2,1-c]fluorene LXXVI (Scheme 33). ${ }^{40}$


Scheme 33. Synthesis of indeno[2,1-c]fluorene LXXVI from tribenzocyclotriyne LXXIV

Using the parent indeno[2,1-c]fluorene-5,8-dione LXXVII, Haley et al. synthesized, for the first time, fully conjugated and antiaromatic indeno[2,1-c]fluorenes LXXVIII and LXXIX (Scheme 34). ${ }^{6 g, 41}$ In their pioneering study, they found that indeno[2,1-c]fluorenes LXXVIII have high electron affinities and small HOMO-LUMO energy gaps, thus suggesting that these modified indeno[2,1-c]fluorenes LXXVIII and LXXIX are attractive candidates for applications in organic electronic devices.


Scheme 34. Synthesis antiaromatic indeno[2,1-c]fluorene LXXVIII and LXXIX

All these synthetic approaches show that the preparation of indeno[2,1-c]fluorenes is still challenging task. With respect to the above said, it is possible to constitute that there has not been developed a general synthetic strategy for preparation of regioselectively substituted indeno[2,1-c]fluorenes yet. Moreover, the preparation of specifically substituted indeno[2,1$c]$ fluorenes is still challenging and requires specific methods. Therefore the area of synthesis of variously substituted fluorenes or indenofluorenes remains highly attractive and interesting field of organic synthesis.

## 3. Aim of the Work

As shown in the previous chapter, scientific literature describes numerous pathways for the synthesis of compounds containing the fluorene skeleton and the corresponding 9, ${ }^{9}$ ' spirobifluorenes (SBFs). Nevertheless, a possibility of developing of a general method for the regioselective synthesis of unsymmetrically substituted fluorenes and their properties has not been explored thus far. Therefore, the aims of this thesis were the following:
a) To develop a synthetic pathway for compounds with the fluorene scaffold using catalytic $[2+2+2]$ cyclotrimerization of symmetrically or unsymmetrically substituted diynol precursors ( $\mathrm{R}^{1}$ and $\mathrm{R}^{2}$ ) with alkynes to selectively form 1,2,3,4-tetrasubstituted fluorenols and their SBF derivatives (Scheme 35).


Scheme 35. Proposed synthetic pathway for SBF using transition metal-catalyzed [2+2+2] cyclotrimerization
b) To assess the effect of different alkyl, aromatic and extended $\pi$-aromatic hydrocarbon substituents ( $\mathrm{R}^{1}$ and $\mathrm{R}^{2}$ ) at positions 1 and 4 on the photophysical properties of the corresponding SBFs.
c) To extend the SBF synthetic method to the preparation of dispiroindeno[2,1-c] fluorenes (DS-IFs). They can be considered as extended SBF derivatives (Scheme 36).




Scheme 36. Proposed pathway of synthesis of DS-IFs using the synthetic approach developed
d) To assess substituent effects $\left(\mathrm{R}^{1}\right.$ and $\mathrm{R}^{2}$ or substituents $\mathrm{R}^{3}$ and $\left.\mathrm{R}^{4}\right)$ on the photophysical properties of DS-IFs (Scheme 36), as performed for the SBFs.
e) To develop a synthesis route to a new class of [7]-helical DS-IFs by using the cyclotrimerization approach (Scheme 37).


Scheme 37. Proposed synthetic strategy for $(P)$ - and ( $M$ )-[7]-helical DS-IFs

## 4. Results and Discussion

### 4.1. 9,9'-Spirobifluorenes (SBFs)

### 4.1.1. Study for the synthesis of $\mathbf{9 H}$-fluorene

In my previous study, I was able to find suitable reaction conditions for a transition metal catalyzed $[2+2+2]$ cyclotrimerization of substituted diynes with alkynes to provide 1,2,3,4-tetrasubstituted 9 H -fluorenes. ${ }^{36}$

These fluorenes were synthesized using two sets of symmetrically substituted diynes. The diynes were prepared by one-step Negishi coupling of the corresponding alkynylzinc halides with 2-iodobenzyl bromide using a Pd catalyst. ${ }^{42}$ Then, different catalytic systems for $[2+2+2]$ cyclotrimerization containing $\mathrm{Co}, \mathrm{Ru}, \mathrm{Ni}$ and Rh complexes were studied. Excluding the Ru-catalysts, all other catalysts gave rise to the desired $9 H$-fluorenes. The rhodium complex (Wilkinson's catalyst, $\mathrm{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}, 10 \mathrm{~mol} \%$ ) proved to be the most powerful catalyst for $[2+2+2]$ cyclotrimerization of the diynes with alkynes to provide various alkyl and aryl substituted 9 H -fluorene derivatives in up to $93 \%$ isolated yield (Scheme 38). The substituent screening was then performed. Various symmetrically $\left(\mathrm{R}^{2}=\mathrm{R}^{3}=\mathrm{Me}, \mathrm{Et}, n-\mathrm{Pr}\right.$, $\left.\mathrm{CH}_{2} \mathrm{OH}\right)$ and unsymmetrically $\left(\mathrm{R}^{2}=\mathrm{H}, \mathrm{R}^{3}=\mathrm{Ph}, \mathrm{Fc}\right)$ substituted alkynes were applied in the cyclotrimerization reaction. The results showed that 3-hexyne $\left(R^{2}=R^{3}=E t\right)$ was the most suitable reaction partner for Rh-catalyzed cyclotrimerization. ${ }^{36}$


Scheme 38. General synthetic approach for 9 H -fluorenes

These compounds were originally synthesized to serve as $\eta^{5}$ - or $\eta^{6}$ - ligands for several transition metals, because unsymmetrically substituted 9 H -fluorenes are potential ligands for asymmetrical catalysis. ${ }^{43}$ However, all attempts to form complexes of 1,2,3,4-tetrasubstituted 9 H -fluorenes with transition metals, such as Mn and Fe , failed.

### 4.1.2. Study for the synthesis of $\mathbf{1 , 2 , 3 , 4}$-tetrasubstituted fluorenols

As shown above, the transition metal catalyzed reaction of diynes and alkynes was successfully used to synthesize tetrasubstituted 9 H -fluorenes, but no further applications of these compounds were found. Therefore, the focus of this project shifted towards preparing the structurally related $1,2,3,4$-tetrasubstituted fluorenols.

In contrast to $9 H$-fluorenes, fluorenols have the hydroxyl functional group at position 9. The presence of the hydroxyl group enables fluorenols to undergo conversion into the corresponding $9,9^{\prime}$-spirobifluorenes (SBFs). Moreover, I hypothesized that selective substitution at positions 1-4 of the SBFs framework could be used to tune their properties based on a recent research study on 2 - and 4 -substituted $\mathrm{SBFs},{ }^{44}$ although a systematic study of substituent effects on SBFs has not been reported thus far. Therefore, I aimed to develop a general synthetic strategy for the regioselective synthesis of variously substituted SBFs using the cyclotrimerization method previously developed for the synthesis of tetrasubstituted 9 H fluorenes according to the reaction sequence below (Scheme 39). ${ }^{36}$


Scheme 39. General synthethic route for 1,2,3,4-tetrasubstituted fluorenols

Thus, the goal of this study was to assess whether the previously developed reaction pathway is also applicable for the synthesis of 1,2,3,4-tetrasubstituted fluorenols. For this purpose, the following objectives were considered:
a) to find suitable diynol compounds for $[2+2+2]$ cyclotrimerization reaction,
b) to study applicable catalytic systems for the $[2+2+2]$ cyclotrimerization reaction,
c) to test different alkynes in the $[2+2+2]$ cyclotrimerization reaction, and
d) to assess the effect of hydroxyl group protection.
a) Either 2-bromobenzaldehyde or 2-iodobenzaldehyde could have been chosen as commercially available starting material; however, the former was used because of its lower price. In the first step, Sonogashira coupling of 2-bromobenzaldehyde with phenylacetylene using $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(5 \mathrm{~mol} \%)$ and $\mathrm{CuI}(10 \mathrm{~mol} \%)$ with triethylamine as base in THF at $80^{\circ} \mathrm{C}$ for 3 h yielded 1a in $95 \%$ isolated yield. The subsequent alkynylation with lithium phenylacetylide in THF at $-78^{\circ} \mathrm{C}$ gave rise to the symmetrically substituted diynol 2a in $86 \%$ isolated yield ( $82 \%$ after two steps, Scheme 40).


Scheme 40. Synthesis of diynol 2a
b) Initially, diynol 2a was applied as the model compound for the $[2+2+2]$ cyclotrimerization study with different alkynes in the presence of various catalytic systems. The first task was to find suitable catalytic reaction conditions for the $[2+2+2]$ cyclotrimerization of 2a with alkynes. For this purpose, reactions of diynol 2a and 3-hexyne 3b, as a representative of internal alkynes, were performed using various transition metalbased ( $\mathrm{Co}, \mathrm{Ru}, \mathrm{Ni}, \mathrm{Rh}$ ) catalytic systems (Table 1). Similar to the study of 9 H -fluorene synthesis, the highest yield (76\%) of the desired fluorenol 4ab (Entry 1) was found when using the Wilkinson's catalyst $\left(\mathrm{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}\right)$. The other catalytic systems, consisting of $\mathrm{CpCo}(\mathrm{CO})_{2}$ and $\mathrm{Ni}(\operatorname{cod})_{2} / 2 \mathrm{PPh}_{3}$, also furnished fluorenol 4ab, albeit in lower yields of $11 \%$ and $19 \%$, respectively (Entries 2 and 3). No product was formed only in the case of $\mathrm{Cp} * \mathrm{RuCl}(\mathrm{cod})$, and the starting material was fully recovered. Subsequent attempts to change the reaction conditions also failed (Entries 4 and 5).

Table 1. Transition metal complex-catalyzed cyclotrimerization reactions of 2a with 3b

${ }^{\text {a }}$ Isolated yields.
c) At this stage, the substituent effect of different symmetrically substituted internal alkynes $\mathbf{3}$ on cyclotrimerization with 2a was tested. Therefore, alkynes were classified according to Charton steric parameters (upsilon steric parameter $v_{\text {ef }}$ ) of the R substituents. Alkynes with small substituents such as $\mathbf{3 a}\left(\mathrm{R}=\mathrm{Me}, v_{\mathrm{ef}}=0.52\right), \mathbf{3 b}\left(\mathrm{R}=\mathrm{Et}, v_{\mathrm{ef}}=0.56\right)$ and 3d $\left(\mathrm{R}=\mathrm{CH}_{2} \mathrm{OH}, v_{\mathrm{ef}}=0.53\right)$, alkyne $\mathbf{3 c}$ with a medium size substituent $\left(\mathrm{R}=n-\mathrm{Pr}, v_{\mathrm{ef}}=0.68\right)$ and alkynes $\mathbf{3 e}-\mathbf{3 g}$ with sterically more demanding aryl substituents $\left(\mathrm{R}=\mathrm{Ph}, \mathrm{v}_{\text {min }}=0.57, \mathrm{v}_{\max }\right.$ $=2.15$ ) were chosen. ${ }^{45}$ Interestingly, the product was formed only when using alkyl substituted alkynes ( $\mathbf{3 a - 3 d}$ ). In detail, fluorenol $\mathbf{4 a a}$ was synthesized in cyclotrimerization reaction with 2-butyne 3a, albeit in a low yield (25\%; Table 2, Entry 1). The low yield most likely resulted from the high volatility of the alkyne ( $27{ }^{\circ} \mathrm{C}$ boiling point). Thus, the concentration of the alkyne in the liquid phase was presumably low because the reaction was performed at $90^{\circ} \mathrm{C}$. Increasing the quantity of 3a to 5 or 10 equivalents in a closed system failed to give fluorenol 4aa in higher yields. The reactions of 3-hexyne and 4-octyne provided products $\mathbf{4 a b}$ and $\mathbf{4 a c}$ in $76 \%$ and $24 \%$ yield, respectively (Entries 2 and 3). Gratifyingly, the cyclotrimerization with 1,4-butynediol 3d gave fluorenol 4ad in $76 \%$ yield (Entry 4). However, the reactions of $\mathbf{2 a}$ with arylalkynes $\mathbf{3 e - 3 g}$ unexpectedly did not give arylsubstituted fluorenols 4ae-4ag, most likely due to steric hindrance (Entries 5-7). The comparison of the reactions of $\mathbf{2 a}$ with 3-hexyne $\mathbf{3 b}$ or with 4 -octyne $\mathbf{3 c}$ shows that steric effects $\left(\mathbf{3 b}, v_{\text {ef }}=0.56 ; \mathbf{3 c}, v_{\text {ef }}=0.68\right)$ decrease the product yield from 76 to $24 \%$ (Entries 2 and 3).

Table 2. Rh-catalyzed cyclotrimerization reactions of 2a with alkynes $\mathbf{3}$

${ }^{\mathrm{a}}$ Isolated yields.
d) Lastly, the effect of the hydroxyl group protection at diynol 2a for the Rh-catalyzed cyclotrimerization with various alkynes was assessed. The protection of the free hydroxyl group at 2a was achieved using TBSCl and imidazole/DMAP at $0{ }^{\circ} \mathrm{C}$. The product $\mathbf{2 a - T B S}$ was obtained in a surprisingly low yield of $53 \%$ after isolation (Scheme 41). All attempts to improve these yields by changing reaction conditions such as time (up to 24h), temperature (up to $50^{\circ} \mathrm{C}$ ) and additives (bases) failed. The TBS group was chosen as the protecting group because it can be selectively removed with fluoride based reagents under mild reaction conditions.


Scheme 41. Synthesis of protected diynol 2a-TBS

The protected diynol 2a-TBS was then subjected to Rh-catalyzed cyclotrimerization using different alkyl and aryl substituted internal alkynes, such as 2-butyne (3a), 3-hexyne (3b), 4-octyne (3c), 1,4-but-2-ynediol (3d), diphenylethyne (3e), di(4-tolyl)ethyne (3f), and bis(4-trifluoromethylphenyl)ethyne (3g) (Table 3).

The cyclotrimerization reaction of 2a-TBS with alkyl substituted alkynes 3a-d yielded 1,2,3,4-tetrasubstituted fluorenols 4a-d-TBS (Entries 1-4). The reaction with 3a gave 4aaTBS ( $\mathrm{R}=\mathrm{Me}$, Entry 1 ) in $\mathbf{1 9 \%}$ yield, with 3b it furnished 4ab-TBS ( $\mathrm{R}=\mathrm{Et}$, Entry 2 ) in $71 \%$ yield, with 3c it afforded 4ac-TBS (R = n-Pr, Entry 3 ) in $6 \%$ yield, and with 3d 4ad-TBS (R $=\mathrm{CH}_{2} \mathrm{OH}$, Entry 4) was formed in $82 \%$ yield. In all cases, except 4ad-TBS, the yields of products were slightly lower than those of the reaction with the unprotected diynol 2a (see Table 2). No cyclotrimerization occurred and no products were detected in the reactions with aryl-substituted alkynes $\mathbf{3 e - g}$ (Entries 5-7).

Table 3. Results from Rh-catalyzed cyclotrimerization reactions of 2a-TBS with alkynes $\mathbf{3}$


These results showed no significant effect of the hydroxyl group protection on the outcome of the cyclotrimerization reaction. Therefore, unprotected diynols were used in subsequent experiments. The results showed that 3-hexyne 3b was the most suitable internal alkyne. Hence, this hexyne 3b was selected for all further reactions. But-2-yne-1,4-diol 3d yielded the corresponding fluorenols 4ad and 4ad-TBS in the same or in even higher yields than those of 3-hexyne $\mathbf{3 b}$, but the presence of two additional hydroxyl group could affect the subsequent steps of the synthesis.

### 4.1.3 Synthesis of $\mathbf{1 , 2 , 3 , 4}$-tetrasubstituted SBFs and analysis of the effect of the substituents at positions 1 and 4 on its photophysical properties

All synthesized fluorenols can be converted into the corresponding SBFs. The properties of the SBFs may be tuned by selective substitution at positions 1-4, yet no systematic study on the effect of the substituent on the photophysical properties of SBFs has been conducted thus far, as previously stated in section 4.1.2. This is particularly relevant when considering that SBFs have useful properties in organic optoelectronics, such as thermal and morphological stability, and their photophysical properties depend on their substituents.

Initially, in addition to the previously obtained diynol 2a, three other diynols 2b-d with different combinations of substituents at the ends of the triple bond in symmetrical and unsymmetrical alignment were prepared. The synthesis started with Sonogashira coupling of 2-bromobenzaldehyde with pent-1-yne or phenylacetylene $\left(\mathrm{Pd}_{( }\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}\right.$ ( $\left.5 \mathrm{~mol} \%\right), \mathrm{CuI}(10$ $\mathrm{mol} \%$ ), $\mathrm{NEt}_{3}, \mathrm{THF}, 80^{\circ} \mathrm{C}, 3 \mathrm{~h}$ ) producing $\mathbf{1 a}$ and $\mathbf{1 b}$ in 95 and $92 \%$ isolated yields, respectively. Their subsequent alkynylation with pent-1-yn-1-yllithium or lithium phenylacetylide in THF at $-78{ }^{\circ} \mathrm{C}$ gave rise to the set of $\mathbf{2 a - d}$, again in very high isolated yields ranging from 86 to $96 \%$.

Diynols 2a-d were then cyclotrimerized with 3-hexyne 3b in the presence of Wilkinson's catalyst ( $10 \mathrm{~mol} \%$ ) under the previously used conditions (toluene, $90^{\circ} \mathrm{C}, 16 \mathrm{~h}$ ). Fluorenols 4ab, 4bb, 4cb and $\mathbf{4 d b}$ were obtained in $38-76 \%$ isolated yields (Scheme 42). In some cases the formation of homo-coupled product from diynols $\mathbf{2 a} \mathbf{a} \mathbf{d}$ was observed as minor side products (3-7\%). Attempts to overcome the homocoupling reaction by slow addition of diynols $\mathbf{2 a} \mathbf{- d}$ into the reaction mixture and to increase the amount of the alkyne $\mathbf{3 b}$ to 5 or even 10 equivalents failed.


Scheme 42. Synthesis of unsymmetrically 1,2,3,4-tetrasubstituted fluorenols 4

Then, fluorenols $\mathbf{4 a b}, \mathbf{4 b b}, \mathbf{4 c b}$ and $\mathbf{4 d b}$ were converted into the corresponding SBFs. Initially, fluorenols 4 were oxidized with pyridinium chlorochromate (PCC) in dichloromethane at $25^{\circ} \mathrm{C}$ to the corresponding fluorenones $\mathbf{5 a}$-d. Fluorenones $\mathbf{5 a}, \mathbf{5 b}, \mathbf{5 c}$, and 5d were isolated in $89,94,93$ and $90 \%$ yields, respectively. Subsequently, fluorenones $\mathbf{5}$ were converted into the desired SBFs 6a-d using a two-step approach ${ }^{46}$ consisting of 2lithiobiphenyl addition to 5a-d that was followed by treatment with a mixture of acetic acid containing a catalytic amount of concentrated HCl at $110^{\circ} \mathrm{C}$. SBFs $\mathbf{6 a}, \mathbf{6 b}, \mathbf{6 c}$, and $\mathbf{6 d}$ were obtained in $66,74,68$, and $83 \%$ yields. In summary, regioselectively 1,2,3,4-tetrasubstituted SBFs 6a-d were obtained, through the five-step synthetic pathway using the Rh-catalyzed cyclotrimerization as the key step, in overall yields ranging from 24 to $37 \%$ (Scheme 43, overall yields are in parenthesis). The structures of $\mathbf{6 a}$ and $\mathbf{6 c}$ were unequivocally confirmed by a single-crystal X-ray structure analysis (Figures 4 and 5).


4ab, $R^{1}=P h, \quad R^{2}=P h$
4bb, $\mathrm{R}^{1}=n-\mathrm{Pr}, \mathrm{R}^{2}=\mathrm{Ph}$
4cb, $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=n-\mathrm{Pr}$
$\mathbf{4 d b}, \mathbf{R}^{1}=n-\mathrm{Pr}, \mathrm{R}^{2}=n-\mathrm{Pr}$

5a, 89\%
6a, 66\% (37\%)
5b, $93 \%$
5c, 94\%
5d, 90\%
6b, 74\% (30\%)
6c, 68\% (35\%)
6d, 83\% (24\%)


Scheme 43. Synthesis of 1,2,3,4-tetrasubstituted SBFs 6a-d


Figure 4. Platon plot of $\mathbf{6 a}$ showing displacement ellipsoids at a $30 \%$ probability level


Figure 5. Platon plot of $\mathbf{6 c}$ showing displacement ellipsoids at a $30 \%$ probability level

After preparing the set of SBFs 6a-d, their photophysical properties were studied. Initially, the UV/Vis absorption spectra were recorded. Samples were measured in $10^{-6} \mathrm{M}$ solutions in cyclohexane and the relative quantum yields $\Phi_{\text {s }}$ were determined using 9,10diphenylanthracene (DPA) as the standard. To calculate the relative quantum yields $\Phi_{s}$, DPA and the sample molecules must have the same absorption value at a certain wavelength. This specific wavelength, the isobestic point, is used as excitation wavelength $\lambda_{\text {exc }}$ in the further measurement of fluorescence emission spectroscopy. In this case the comparison of absorption spectra show the isobestic points at 242 and 243 nm , respectively and were used as $\lambda_{\text {exc }}$ for the emission spectroscopy.

The analysis of the fluorescence emission spectral data (Figure 6 and Table 4) showed that:
a) Compounds $\mathbf{6 a}$ and $\mathbf{6 c}$ with the phenyl substituents at position 4 show broad emission peaks at $330 / 333 \mathrm{~nm}$ after excitation at 242 nm and 243 nm , respectively.
b) Compounds 6b and 6d with the $n$-propyl substituent at position 4 have narrower emission patterns with emission maxima at 315 nm and smaller emission peaks at 328 nm after excitation at 243 nm and 242 nm , repectively.
c) The quantum yields of all four compounds, namely $\mathbf{6 a}\left(\Phi_{\mathrm{s}}=0.62\right), \mathbf{6 b}\left(\Phi_{\mathrm{s}}=0.72\right), \mathbf{6 c}$ $\left(\Phi_{\mathrm{s}}=0.48\right)$ and $\mathbf{6 d}\left(\Phi_{\mathrm{s}}=0.87\right)$, have similar values. These results show that substituted SBFs have good to high quantum yields.


6a



6c


6d

Figure 6. Absorption and emission spectral data of $\mathbf{6 a - d}$ in $10^{-6} \mathrm{M}$ samples

Table 4. Photophysical properties of $\mathbf{6 a - d}\left(10^{-6} \mathrm{M}\right.$ in cyclohexane)

| Sample | $\lambda_{\text {exc }} / \mathrm{nm}$ | $\lambda_{\mathrm{em}}^{\mathrm{max}} / \mathrm{nm}$ | $\Phi_{\mathrm{s}}{ }^{\text {a }}$ |
| :---: | :---: | :---: | :---: |
| $\mathbf{6 a}$ | 242 | 333 | 0.62 |
| $\mathbf{6 b}$ | 243 | $315 / 328$ | 0.72 |
| $\mathbf{6 c}$ | 243 | 330 | 0.48 |
| $\mathbf{6 d}$ | 242 | $315 / 328$ | 0.87 |

${ }^{\text {a }}$ Quantum yields were calculated relative to DPA $\left(\Phi_{r}=0.97\right)$.

These observations and comparisons led to the following conclusions:
a) Substituents at position 4 strongly affect the photophysical properties of SBFs. A significant difference was observed in the red shift of the fluorescence emission spectra caused by these substituents. The presence of the phenyl substituent at position 4 of $\mathbf{6 a}$ and $\mathbf{6 c}$ ( $\lambda_{\max }=333$ and 330 nm ) red-shifted the emission maxima by more than 15 nm in comparison with the propyl substituted SBFs $\mathbf{6 b}$ and $\mathbf{6 d}$ ( $\lambda_{\text {max }}=315 \mathrm{~nm}$ for both).
b) The well-resolved profile with distinct peaks in the emission spectra of $\mathbf{6 b}$ and $\mathbf{6 d}$ match the spectra of similar compounds in the same solvent, such as the parent $\operatorname{SBF}$ ( $\lambda_{\mathrm{em}}=$ $310,323 \mathrm{~nm}){ }^{44 \mathrm{a}}$ 2-phenyl-SBF $\left(\lambda_{\mathrm{em}}=335,351 \mathrm{~nm}\right)$ and 2-pyrimidyl-SBF $\left(\lambda_{\mathrm{em}}=338,355\right.$
$\mathrm{nm}) .{ }^{44 \mathrm{~b}}$ The broad emission band of $\mathbf{6 a}$ and $\mathbf{6 c}$ may be a characteristic property of aromatic substitution at position 4 of SBFs.
c) The measured emission spectra of $\mathbf{6 a} / \mathbf{6 c}$ and $\mathbf{6 b} / \mathbf{6 d}$ show that the substituent at position 1 of the SBF has negligible, if at all any, effect on the fluorescence emission shift or on the quantum yield. The quantum yields for all measured SBFs 6 ( $\Phi_{\mathrm{s}}=0.48-0.87$ ) are also in accordance with previously reported results. ${ }^{44}$

In summary, I was able to synthesize selectively 1,2,3,4-tetrasubstituted SBFs in a five-step approach from commercially available compounds. The Rh-catalyzed $[2+2+2]$ cyclotrimerization approach developed in this study was used as the key step. Furthermore, a preliminary analysis of the effect of the substituent at positions 1 and 4 on the photophysical properties of SBFs was conducted. The results indicate a connection between the aromatic substituents at position 4 of the SBF and a red shift of the fluorescence emission peak towards the visible light region. Therefore, substitution at this particular position is an interesting option for further synthetic modification.

### 4.1.4. Synthesis of SBFs bearing different substituents at position 4

In the previous chapter, I showed that 4 -aryl substituents of SBFs red shift the fluorescence emission wavelength of these compounds. Thus, the next step of this study on the modulation of photophysical properties of SBF derivatives was to use different substituents at position 4 . This approach was divided into three stages:
a) to synthesize 4-substituted SBFs with 4-substituted phenyl groups bearing electrondonating and -withdrawing groups,
b) to synthesize the SBFs with extended aromatic moieties at position 4,
c) to attach fused $\pi$-aromatic hydrocarbons (PAHs) at position 4 of SBFs.

### 4.1.4.1. Synthesis of 4 -substituted phenyl groups at position 4 of SBFs

In the first stage, a set of 4 -substituted SBFs bearing 4 -substituted phenyl groups was prepared. The 4 -substituted phenyl groups encompassed moieties bearing electron-donating and -withdrawing functional groups. Because of structural similarity with compounds in section 4.1.3, the same five-step synthetic approach was applied. ${ }^{47}$

In the first step, 3-alkynylated benzaldehydes 7a-g were synthesized by using Sonogashira coupling $\left(\mathrm{Pd}_{( }\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}(5 \mathrm{~mol} \%), \mathrm{CuI}(10 \mathrm{~mol} \%), \mathrm{NEt}_{3}, \mathrm{THF}, 8{ }^{\circ} \mathrm{C}, 3 \mathrm{~h}\right)$ of 2bromobenzaldehyde with various 4 -substituted phenylacetylenes having various electrondonating ( $\mathrm{OMe}, \mathrm{Me}, \mathrm{H}$ ) and -withdrawing ( $\mathrm{Ph}, \mathrm{Cl}, \mathrm{CF}_{3}, \mathrm{CN}$ ) groups. The corresponding alkynylbenzaldehydes $\mathbf{7 a - g}$ were obtained in good isolated yields in the range of 80 to $96 \%$. The subsequent alkynylation of $7 \mathbf{a - g}$ with lithium $p$-tolylacetylide at $-78^{\circ} \mathrm{C}$ in THF gave rise to diynols 8a-g in 84 to $96 \%$ yields (Scheme 44).

$\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}$ (5 mol\%)
Cul ( $10 \mathrm{~mol} \%$ )
$\mathrm{NEt}_{3}, \mathrm{THF}, 80^{\circ} \mathrm{C}$, 3 h

7a $\quad \mathrm{R}=\mathrm{C}_{6} \mathrm{H}_{4}-\mathrm{OMe}, 96 \%$
8a, 84\%
7b $\quad R=\mathrm{C}_{6} \mathrm{H}_{4}-\mathrm{Me}, \quad 96 \%$
8b, 91\%
7c, $\quad \mathrm{R}=\mathrm{C}_{6} \mathrm{H}_{4}-\mathrm{Ph}, \quad 92 \%$
8c, 90\%
7d, $R=\mathrm{C}_{6} \mathrm{H}_{4}-\mathrm{H}, \quad 97 \%$
8d, 93\%
7e, $\quad R=\mathrm{C}_{6} \mathrm{H}_{4}-\mathrm{Cl}, \quad 90 \%$
8e, 89\%
7f, $\quad R=\mathrm{C}_{6} \mathrm{H}_{4}-\mathrm{CF}_{3}, \quad 80 \%$
8f, $90 \%$
8g, 94\%

Scheme 44. Synthesis of substituted diynols 8a-g

In the third step, I applied the previously developed Wilkinson's complex catalyzed
 section 4.1.3) to mediate reactions of substituted diynols with small internal alkynes. As already mentioned, 3-hexyne 3b was chosen as the most suitable compound for the cyclotrimerization reactions with diynols. The catalytic [ $2+2+2]$ cyclotrimerization of diynols 8a-g with 3b proceeded to provide the desired fluorenols 9a-g in the range of 30 to $64 \%$ isolated yields (Scheme 45). These isolated yields matched the previously reported values obtained for $\mathbf{4 a b}, \mathbf{4 b b}, \mathbf{4 c b}$ and $\mathbf{4 d b}$, which were ranging from 38 to $76 \%$ (section 4.1.3.). Similar to the previous results also in this case the homo-coupling product of diynols 8a-g was observed as minor side product ( $2-5 \%$ ) and the same attempts (slow addition of diynol and increase of alkyne concentration, section 4.1.3) failed to suppress this unwanted side reaction.

In the next step, fluorenols 9a-g were oxidized with PCC in dichloromethane at $25^{\circ} \mathrm{C}$ uneventfully to fluorenones 10a-g in good to high isolated yields (61-86\%). Finally, the fluorenones 10a-g were converted to SBFs 11a-f by reaction with 2-lithiobiphenyl that was followed by treatment with a mixture of acetic acid containing a catalytic amount of concentrated HCl at $110{ }^{\circ} \mathrm{C}$. The final SBFs 11a-f were obtained in high isolated yields ranging from $63 \%$ to $90 \%$. The overall yields from the commercially available starting material after five steps were in a reasonable range from 19 to $34 \%$ (Scheme 45, overall yields in parenthesis). Only in the case of $\mathbf{1 1 g}(\mathrm{R}=\mathrm{CN})$, 2-lithiobiphenyl added preferentially to the more reactive cyano group, thereby leading to the imine product. Changes in reaction conditions to suppress imine formation and to favor addition to the carbonyl group were not
met with success. The structures of the SBFs 11a, 11c and 11e were unequivocally confirmed by a single-crystal X-ray analysis (Figures 7-9).


11a, 83\% (25\%)
11b, $74 \%$ (32\%)
11c, 63\% (21\%)
11d, 73\% (34\%)
11e, 82\% (25\%)
11f, $90 \%$ (19\%)

Scheme 45. Synthesis of 4-aryl substituted SBFs 11a-f


Figure 7. Platon plot of 11a showing displacement ellipsoids at a $50 \%$ probability level


Figure 8. Platon plot of 11c showing displacement ellipsoids at a $50 \%$ probability level


Figure 9. Platon plot of 11e showing displacement ellipsoids at a $50 \%$ probability level

The X-ray diffraction data of all three SBFs showed large dihedral angles (almost perpendicular) between the SBF and the para-decorated aryl substituents at position 4. In detail, for 11a the angles between C4-C5-C6 and C37-C38-C42 were $86^{\circ}$, for 11c the angles between C8-C9-C44 and C10-C11-C47 were $94^{\circ}$ and for 11e the angles between C20-C24C 25 and C26-C27-C31 were $83^{\circ}$. This high rotation may be caused by the ethyl substituent at position 3 of the SBFs.

The almost perpendicular arrangement between the SBFs and the substituents at position 4 may affect the electron distribution within the molecule. This can in particular influence and lower the ability for a strong performance in photophysical properties, especially after excitation. Nevertheless, based on results obtained for 4-aryl-substituted $\mathrm{SBFs}^{2,44,48}$ and our previous results regarding 1,2,3,4-tetrasubstituted $\mathrm{SBFs}^{36}$ these compounds presumably undergo planarization by stabilization of the excited state, which then improves the alignment and interaction of extended $\pi$-systems at the central carbon atoms. ${ }^{44 a, 49}$

Besides, also the aromatic substituent at position 1 of the SBF showed large dihedral angles. But in contrast to position 4, substituents at position 1 of the SBF are not influencing the photophysical properties at all, as found in the previous study (see section 4.1.3).

### 4.1.4.2. Synthesis of extended aromatic substituents at position 4 of SBFs

In the second stage, besides the effect of electron-donating and -withdrawing groups at the 4-substituted phenyl group of the SBFs at position 4, the effect on aromatic systems, such as biphenyl or diphenylethynyl moieties at the same position of the SBFs, was studied. In order to attach these substituents at position 4, 4-bromophenyl-substituted SBF 11h was deemed the most suitable key intermediate to synthesize the desired 4 -substituted SBF derivatives.

For the synthesis of the key compound $\mathbf{1 1 h}$, the above mentioned five-step synthetic approach was modified. Specifically, to avoid competing side reactions in the Pd-catalyzed Sonogashira cross-coupling of 2-bromobenzaldehyde and 1-bromo-4-ethynylbenzene, the approach to diynol $\mathbf{8 h}$ had to be changed. First, 2-ethynylbenzaldehyde was allowed to react with lithium $p$-tolylacetylide at $-78{ }^{\circ} \mathrm{C}$ in THF to furnish diynol 12a. 12a was then reacted with 1-bromo-4-iodobenzene under Sonogashira coupling conditions $\left(\mathrm{Pd}_{( }\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}(5 \mathrm{~mol} \%)\right.$, $\mathrm{CuI}(10 \mathrm{~mol} \%), \mathrm{NEt}_{3}, \mathrm{THF}, 8{ }^{\circ} \mathrm{C}, 3 \mathrm{~h}$ ) to give diynol $\mathbf{8 h}$ that was isolated in excellent yield of $96 \%$. Cyclotrimerization using $\mathrm{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}(10 \mathrm{~mol} \%)$ as the catalyst in toluene for 16 h at 90 ${ }^{\circ} \mathrm{C}$ gave rise to fluorenol $\mathbf{9 h}$ in an unsatisfactory isolated yield of $26 \%$ (Scheme 46, Conditions A). This rather low yield combined with a high catalytic loading led to optimization of the reaction conditions. An extensive screening of cyclotrimerization reactions with Wilkinson's catalyst under various conditions was carried out. It included different solvents, catalytic loadings, reaction temperatures, time and additives (Tables 5-11). After a lot of experiments, suitable catalytic conditions were found, which consisted of using the Wilkinson`s catalyst $\left(\mathrm{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}, 3 \mathrm{~mol} \%\right)$ and the additive $\mathrm{Ag}_{2} \mathrm{CO}_{3}(6 \mathrm{~mol} \%)$ in THF under microwave irradiation (MW, $1.5 \mathrm{~h}, 180^{\circ} \mathrm{C}$ ) (Conditions B). In this case the isolated yield of $\mathbf{9 h}$ almost doubled to $57 \%$ (Table 11, Entry 4).


Scheme 46. Synthesis of 4-bromophenyl substituted fluorenol 9h

Table 5. Reaction condition parameters screened for Rh-catalyzed cyclotrimerization of 9h


| Solvent | Catalyst (mol\%) | $\mathrm{T}\left({ }^{\circ} \mathrm{C}\right)$ | time ( h$)$ | additive $(6 \mathrm{~mol} \%)$ |
| :--- | :---: | :---: | :---: | :--- |
| toluene | 10 | 50 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ |
| DMF | 5 | 70 | 48 | $\mathrm{AgSbF}_{6}$ |
| $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 4 | 80 | 96 | $\mathrm{AgNO}_{3}$ |
| $\mathrm{Et}_{2} \mathrm{O}$ | 3 | 110 | $1, \mathrm{MW}$ | $\mathrm{AgCOOCF}_{3}$ |
| THF | 2 | $150, \mathrm{MW}$ | $1.5, \mathrm{MW}$ | $\mathrm{AgOTf}^{3}$ |
|  | 1 | $170, \mathrm{MW}$ |  | $\mathrm{Ag}_{3} \mathrm{PO}_{4}$ |
|  |  | $180, \mathrm{MW}$ |  | $\mathrm{AgBF}_{4}$ |
|  |  | $190, \mathrm{MW}$ |  | $\mathrm{AgCOOCH}_{3}$ |
|  |  |  |  | $\mathrm{AgClO}_{4}$ |
|  |  |  |  | $\mathrm{Ag}_{2} \mathrm{SO}_{4}$ |

Table 6. Variation of the catalyst loading and solvents

| Entry | Solvent | Catalyst $(\mathrm{mol} \%)$ | $\mathrm{T}\left({ }^{\circ} \mathrm{C}\right)$ | $\mathrm{t}(\mathrm{h})$ | Additive $(6 \mathrm{~mol} \%)$ | Recovered $\mathbf{8 h}(\%)^{\mathrm{a}}$ | 9h, Yield $(\%)^{\mathrm{a}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | toluene | 5 | 90 | 16 |  |  | 30 |
| 2 | toluene | 5 | 90 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 32 |  |
| 3 | DMF | 5 | 90 | 16 |  |  | 8 |
| 4 | DMF | 5 | 90 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ |  | 15 |
| 5 | toluene | 4 | 90 | 16 |  |  | 29 |
| 6 | toluene | 4 | 90 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ |  | 36 |
| 7 | toluene | 3 | 90 | 16 |  | 25 | 11 |
| 8 | toluene | 3 | 90 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 28 | 28 |
| 9 | toluene | 2 | 90 | 16 |  | 41 | 15 |
| 10 | toluene | 2 | 90 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 14 | 19 |
| 11 | toluene | 2 | 90 | 96 |  | 8 | 21 |
| 12 | toluene | 2 | 90 | 96 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 23 | 16 |
| 13 | toluene | 1 | 90 | 96 |  | 20 | 18 |
| 14 | toluene | 1 | 90 | 96 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ |  |  |

${ }^{\text {a }}$ Isolated yields

Table 7. Variation of additives

| Entry | Solvent | Catalyst (mol\%) | $\mathrm{T}\left({ }^{\circ} \mathrm{C}\right)$ | $\mathrm{t}(\mathrm{h})$ | Additive $(6 \mathrm{~mol} \%)$ | Recovered $\mathbf{8 h}(\%)^{\mathrm{a}}$ | 9h, Yield $(\%)^{\mathrm{a}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | toluene | 4 | 90 | 16 | $\mathrm{AgSbF}_{6}$ |  | 9 |
| 2 | toluene | 4 | 90 | 16 | $\mathrm{AgNO}_{3}$ |  | 37 |
| 3 | toluene | 4 | 90 | 16 | $\mathrm{AgOOCCF}_{3}$ | 13 | 21 |
| 4 | toluene | 4 | 90 | 16 | $\mathrm{AgOTf}^{2}$ |  | 0 |
| 5 | toluene | 4 | 90 | 16 | $\mathrm{Ag}_{3} \mathrm{PO}_{4}$ | 16 | 21 |
| 6 | toluene | 4 | 90 | 16 | $\mathrm{AgBF}_{4}$ |  | 6 |
| 7 | toluene | 4 | 90 | 16 | $\mathrm{AgOOCCH}_{3}$ |  | 19 |
| 8 | toluene | 4 | 90 | 16 | $\mathrm{AgClO}_{4}$ | 31 | 7 |
| 9 | toluene | 4 | 90 | 16 | $\mathrm{Ag}_{2} \mathrm{SO}_{4}$ |  | 31 |

${ }^{\text {a }}$ Isolated yields

Table 8. Variation of solvent

| Entry | Solvent | Catalyst (mol\%) | $\mathrm{T}\left({ }^{\circ} \mathrm{C}\right)$ | $\mathrm{t}(\mathrm{h})$ | Additive $(6 \mathrm{~mol} \%)$ | Recovered $\mathbf{8 h}(\%)^{\mathrm{a}}$ | 9h, Yield (\%) ${ }^{\mathrm{a}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 4 | 90 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 5 | 30 |
| 2 | $\mathrm{Et}_{2} \mathrm{O}$ | 4 | 90 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 10 | 32 |
| 3 | THF | 4 | 90 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ |  | 41 |

${ }^{a}$ Isolated yields

Table 9. Cyclotrimerization in THF in the presence of different additives

| Entry | Solvent | Catalyst (mol\%) | $\mathrm{T}\left({ }^{\circ} \mathrm{C}\right)$ | $\mathrm{t}(\mathrm{h})$ | Additive (6 mol\%) | Recovered 8h (\%) ${ }^{\mathrm{a}}$ | 9h, Yield (\%) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | THF | 4 | 90 | 16 | - |  | 48 |
| 2 | THF | 4 | 90 | 16 | AgNO $_{3}$ | 38 | 29 |
| 2 | THF | 4 | 90 | 16 | AgOTf |  | 21 |
| 4 | THF | 4 | 90 | 16 | AgOOCCF $_{3}$ | 93 | 5 |

${ }^{\mathrm{a}}$ Isolated yields

Table 10. Effect of temperature

| Entry | Solvent | Catalyst (mol\%) | T ( ${ }^{\circ} \mathrm{C}$ ) | t (h) | Additive(6 mol\%) | Recovered 8h (\%) ${ }^{\text {a }}$ | 9h, Yield (\%) ${ }^{\text {a }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | THF | 4 | 50 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 67 | 15 |
| 2 | THF | 4 | 50 | 16 | - | 81 | 11 |
| 3 | THF | 4 | 70 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 29 | 26 |
| 4 | THF | 4 | 70 | 16 | - | 31 | 29 |
| 5 | THF | 4 | 80 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ |  | 42 |
| 6 | THF | 4 | 80 | 16 | - |  | 41 |
| 7 | THF | 4 | 110 | 16 | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ |  | 35 |
| 8 | THF | 4 | 110 | 16 | - |  | 33 |
| 9 | THF | 4 | 150 | 1, MW | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 10 | 37 |
| 10 | THF | 4 | 150 | 1, MW | - |  | 34 |

${ }^{\text {a }}$ Isolated yields

Table 10 (continued). Effect of temperature

| Entry | Solvent | Catalyst (mol\%) | $\mathrm{T}\left({ }^{\circ} \mathrm{C}\right)$ | $\mathrm{t}(\mathrm{h})$ | Additive $(6 \mathrm{~mol} \%)$ | Recovered $\mathbf{8 h}(\%)^{\mathrm{a}}$ | 9h, Yield $(\%)^{\mathrm{a}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 11 | THF | 4 | 150 | $1.5, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ |  | 41 |
| 12 | THF | 4 | 150 | $1.5, \mathrm{MW}$ | - | 29 |  |
| 13 | THF | 4 | 180 | $1, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ |  | 51 |
| 14 | THF | 4 | 180 | $1, \mathrm{MW}$ | - | 6 | 40 |
| 15 | THF | 4 | 180 | $2, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 32 |  |
| 16 | THF | 4 | 180 | $1, \mathrm{MW}$ | $\mathrm{AgOTf}^{2}$ | 3 |  |
| 17 | THF | 4 | 180 | $1, \mathrm{MW}$ | $\mathrm{AgSbF}_{6}$ | 4 |  |
| 18 | THF | 4 | 180 | $1, \mathrm{MW}$ | $\mathrm{AgNO}_{3}$ | 32 |  |
| 19 | THF | 4 | 180 | $1, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{SO}_{4}$ | 6 |  |

${ }^{a}$ Isolated yields

Table 11. Variation of the catalyst loading and temperature

| Entry | Solvent | Catalyst (mol\%) | $\mathrm{T}\left({ }^{\circ} \mathrm{C}\right)$ | $\mathrm{t}(\mathrm{h})$ | Additive $(6 \mathrm{~mol} \%)$ | Recovered $\mathbf{8 h}(\%)^{\mathrm{a}}$ | 9h, Yield (\%) ${ }^{\mathrm{a}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | THF | 10 | 180 | $1, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 47 |  |
| 2 | THF | 5 | 180 | $1, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 43 |  |
| 3 | THF | 3 | 180 | $1, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 49 |  |
| 4 | THF | 3 | 180 | $1.5, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 57 |  |
| 5 | THF | 2 | 180 | $1, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 32 |  |
| 6 | THF | 4 | 170 | $1, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 51 |  |
| 7 | THF | 3 | 170 | $1, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 37 |  |
| 8 | THF | 2 | 170 | $1, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 27 |  |
| 9 | THF | 4 | 190 | $1, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 43 |  |
| 10 | THF | 3 | 190 | $1, \mathrm{MW}$ | $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ | 40 |  |

${ }^{\text {a }}$ Isolated yields

Having found suitable conditions for preparation of fluorenol $\mathbf{9 h}$ in a good yield under catalytic conditions B, the next step was conducted. Fluorenol $9 \mathbf{h}$ was subjected to oxidation with PCC in dichloromethane at $25^{\circ} \mathrm{C}$ to give fluorenone $\mathbf{1 0 h}$ in a nice $93 \%$ isolated yield. Its subsequent reaction $\mathbf{1 0 h}$ with 2-lithiobiphenyl followed by treatment with a mixture of acetic acid and a catalytic amount of concentrated HCl at $110{ }^{\circ} \mathrm{C}$ gave $\mathrm{SBF} \mathbf{1 1 h}$ isolated in $74 \%$ yield. Thus SBF 11h was formed in 36\% overall yield after five steps (Scheme 47, the overall yield is in parenthesis).


Scheme 47. Synthesis of 4-bromophenyl substituted SBF 11h

SBF 11h was then transformed by using a Pd-catalyzed Suzuki cross coupling reaction into 13a and 13b. The reactions with (4-methoxyphenyl)boronic acid or (4-(trifluoromethyl) phenyl)boronic acid were catalyzed by $\mathrm{Pd}(\mathrm{OAc})_{2}$ in the presence of $\mathrm{K}_{2} \mathrm{CO}_{3}$ in a mixture of $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ at $80^{\circ} \mathrm{C}$. The desired SBF 13a with electron-donating ( $\mathrm{R}=\mathrm{OMe}$ ) and SBF 13b with electron-withdrawing ( $\mathrm{R}=\mathrm{CF}_{3}$ ) substituents were obtained in 62 and $58 \%$ isolated yields, respectively (Scheme 48).


Scheme 48. Synthetic procedure for Pd-catalyzed Suzuki reactions to 13a and 13b

A similar approach was used for the synthesis of SBFs 14a-c bearing 4 -substituted diphenylethynyl substituents at position 4 of the SBF framework. In this instance Sonogashira cross coupling of $\mathbf{1 1 h}$ with 4 -substituted phenylacetylenes was used. First, the reaction of $\mathbf{1 1 h}$ with 4-ethynylanisole $(\mathrm{R}=\mathrm{OMe})$ was carried out in the presence of $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(5 \mathrm{~mol} \%)$, $\mathrm{CuI}(10 \mathrm{~mol} \%)$ and diisopropylamine as a base under microwave irradiation at $180^{\circ} \mathrm{C}$ to give $\mathbf{1 4 a}$ in $54 \%$ isolated yield. The reactions of $\mathbf{1 1 h}$ with phenylacetylene $(\mathrm{R}=\mathrm{H})$ or 1-ethynyl-4(trifluoromethyl)benzene ( $\mathrm{R}=\mathrm{CF}_{3}$ ) proceeded already at $80^{\circ} \mathrm{C}$, albeit under slightly different reaction conditions. Specifically, $\mathrm{SBF} \mathbf{1 4 b}(\mathrm{R}=\mathrm{H})$ was synthesized in $72 \%$ yield by using PEPPSI $^{\text {TM }}-\operatorname{IPr}(5 \mathrm{~mol} \%), \mathrm{CuI}(10 \mathrm{~mol})$ and triethylamine in THF after 3 h , whereas SBF 14c $\left(\mathrm{R}=\mathrm{CF}_{3}\right)$ was formed $57 \%$ isolated yield by using $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(5 \mathrm{~mol} \%), \mathrm{CuI}(10 \mathrm{~mol} \%)$ and triethylamine as base in DMF after 16h reaction time.(Scheme 49).

a) $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(5 \mathrm{~mol} \%)$, $\mathrm{Cul}(10 \mathrm{~mol} \%)$, diisopropylamine, $\mathrm{MW}, 180^{\circ} \mathrm{C}, 1 \mathrm{~h}$.
b) PEPPSI ${ }^{\mathrm{TM}}-\operatorname{IPr}(5 \mathrm{~mol} \%)$, $\mathrm{Cul}(10 \mathrm{~mol} \%)$, triethylamine, $\mathrm{THF}, 80^{\circ} \mathrm{C}, 3 \mathrm{~h}$.
c) $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(5 \mathrm{~mol} \%)$, $\mathrm{Cul}(10 \mathrm{~mol} \%)$, triethylamine, $\mathrm{DMF}, 80^{\circ} \mathrm{C}, 16 \mathrm{~h}$.

Scheme 49. Synthetic procedure for Pd-catalyzed Sonogashira reactions to 14a-c

The structures of 13a and $\mathbf{1 4 b}$ were unequivocally confirmed by a single-crystal X-ray analysis (Figures 10 and 11), and the diffraction data showed a similar behavior when analyzing the alignment of the aryl substituent at position 4, as previously observed in SBF 11c. In the case of 13a the twisting angle between the parent fluorene plane $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 6$ and the phenyl moiety C37-C38-C42, which is connected to the SBF, had the angle of $83^{\circ}$ and in the case of $\mathbf{1 4 b}$ this angle was $75^{\circ}$. This may be explained by the same effect as previously described when discussing the anomaly in deformation of SBF 11a, 11c and 11e.


Figure 10. Platon plot of 13a show displacement ellipsoids at a $50 \%$ probability level


Figure 11. Platon plot of $\mathbf{1 4 b}$ show displacement ellipsoids at a $50 \%$ probability level

SBFs 14 contain the internal triple bond. In general, alkynes are suitable reaction partners in the reaction with $6,9-\left(\mathrm{Me}_{2} \mathrm{~S}\right)$-arachno- $\mathrm{B}_{10} \mathrm{H}_{12}$ to form the corresponding 1,2carboranes. ${ }^{50}$ Thus SBFs $\mathbf{1 4}$ can be suitable substrates to form the corresponding carborane and the carborane itself might be an interesting target to study its photophysical properties. The previous studies have shown that 1,2-diaryl substituted carboranes exhibit interesting absorption and fluorescence emission properties due to their unique "face-on" conformation of the aryl substituents. ${ }^{51}$ Accordingly, $\mathbf{1 4 a}$ was converted with 6,9-( $\left.\mathrm{Me}_{2} \mathrm{~S}\right)$-arachno- $\mathrm{B}_{10} \mathrm{H}_{12}$ in toluene at $120{ }^{\circ} \mathrm{C}$ for 16 h into carborane-14a containing the SBF moiety in $60 \%$ isolated yield (Scheme 50). Its structure was unequivocally confirmed by a single crystal X-ray analysis and showed indeed the "face-on" conformation of the 1,2 -aryl substituents of the carborane structure (Figure 12).


Scheme 50. Conversion of 14a into carborane containing SBF carborane-14a


Figure 12. Platon plot of carborane-14a show displacement ellipsoids at a $50 \%$ probability level

### 4.1.4.3. Synthesis of fused $\pi$-aromatic hydrocarbons at position 4 of SBFs

Fused polyaromatic hydrocarbons (PAHs) represent another class of substituents that can be attached at position 4 of SBFs. PAHs, such as naphthalene, anthracene, phenanthrene or pyrene, have been studied since the 1970s and have shown interesting photophysical properties such as the fluorescence emission wavelength in the $350-420 \mathrm{~nm}$ range and reasonable quantum yields of $\Phi=0.18-0.72$ in $\mathrm{H}_{2} \mathrm{O}$ or ethanol. ${ }^{52}$ Thus, SBFs having the PAH based substituents at position 4 seemed to be promising moieties to tune light emission wavelength and quantum yields.

In the first step, 2-ethynylbenzaldehyde was alkynylated with pent-1-yn-1-yllithium in THF at $-78{ }^{\circ} \mathrm{C}$ to form diynol $\mathbf{1 2 b}$ possessing the terminal triple bond in $95 \%$ isolated yield. The choice for the $n$-propyl substituent at position 1 was based on the fact that it has a negligible effect, if any, on the photophysical properties of the SBF at all (Section 4.1.3). In the next step, 12b was converted to diynols 15a-d with halo-PAHs (the use of ethynylsubstituted PAHs was also considered, but the above mentioned approach was chosen because of commercial availability of the halo-PAHs). The diynols 15a-d were formed by Sonogashira coupling $\left(\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(5 \mathrm{~mol} \%), \mathrm{CuI}(10 \mathrm{~mol} \%), \mathrm{NEt}_{3}, \mathrm{THF}, 80^{\circ} \mathrm{C}, 3 \mathrm{~h}\right)$ of $\mathbf{1 2 b}$ with 2 iodonaphthalene, 9-iodoanthracene, 9-iodophenanthrene and 1-iodopyrene, respectively, in good isolated yields ranging from 41 to $71 \%$. The yields were lower in comparison to the previous results of Sonogashira coupling with phenyl halides used for preparation of 7 (see section 4.1.4.1). This might be explained by a lower reactivity of diynol 12b compared to aryl acetylenes, where the yields were ranging from 80 to $96 \%$.

With diynols 15a-d in hand, cyclotrimerization reactions were performed. All were reacted with 3-hexyne 3b under the developed catalytic condition $\left(\mathrm{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}(3 \mathrm{~mol} \%)\right.$, $\mathrm{Ag}_{2} \mathrm{CO}_{3}(6 \mathrm{~mol} \%)$, THF, $180{ }^{\circ} \mathrm{C}$, MW) into fluorenols 16a-d in good isolated yields ranging from 45 to $59 \%$, which were similar or even higher than previously observed results for $9 \mathrm{a}-\mathrm{g}$ ( $30-61 \%$, section 4.1.4.1). The subsequent oxidation with PCC in dichloromethane at $25^{\circ} \mathrm{C}$ gave rise to fluorenones 17a-d uneventfully in very high yields ranging from 83 to $91 \%$. In the last step, reactions of 2-lithiobiphenyl with fluorenones $\mathbf{1 7 a} \mathbf{- d}$, followed by treatment with a mixture of acetic acid and a catalytic amount of concentrated HCl at $110^{\circ} \mathrm{C}$ furnished SBFs 18a-d with fused $\pi$-aromatic hydrocarbons at position 4 in high yields ranging from 51 to $91 \%$. The overall yields were again in an acceptable range from 9 to $26 \%$ (Scheme 51, the overall yields are in parenthesis).


12b


15a, $\mathrm{PAH}=2$-naphthyl, $\quad 57 \% \quad$ 16a, 45\% 16c, 48\%
15b, PAH = 9-phenanthryl, 71\%
15c, $\mathrm{PAH}=9$-anthranyl, $\quad 41 \%$
15d, PAH = 2-pyrenyl, $\quad 62 \%$
16b, 59\% 16d, 54\%


17a, 91\% 17c, 83\% 17b, 86\% 17d, 84\%

18


18a, $57 \%$ (10\%)


18b, $91 \%$ (26\%)


18c, 51\% (9\%)


18d, $84 \%$ (20\%)

Scheme 51. Synthetic route of 4-PAH substituted SBF 18a-d


Figure 13. Platon plot of $\mathbf{1 8 b}$ showing displacement ellipsoids at a $50 \%$ probability level

The structure of 18b was confirmed by a single-crystal X-ray analysis (Figure 13) and showed an almost right angle of $91.8^{\circ}$ between the plane of the attached phenanthryl moiety C34-C33-C46 and the fluorene part of the SBF with C4-C5-C6. Thus, also in this case, the molecular twist findings match the results previously observed for SBFs 11, 13 and $\mathbf{1 4}$.

### 4.1.5. Photophysical properties of $\mathbf{4}$-substituted SBFs

Until now, three different sets of 4 -aryl substituted SBFs were prepared. They consisted of:
a) 11a-f with electron-donating and -withdrawing groups at the 4 -aryl substituent,
b) 13a and 13b with 4-substituted biphenyl moieties and 14a-c, with 4-substituted 4-(2-(phen-1'-yl)ethynylphenyl moieties, one example of carborane-14a and,
c) $\quad$ SBFs 18a-d with the fused $\pi$-aromatic hydrocarbons at position 4.

All photophysical properties were measured under the same conditions, including a sample concentration of $10^{-6} \mathrm{M}$ in cyclohexane, and the relative quantum yields $\Phi_{\mathrm{s}}$ were calculated with DPA as the reference at the isobestic point.

The first set, 4-aryl substituted SBFs 11a-f was subjected to photophysical properties. After the absorption spectra were measured, the isobestic points of SBFs 11a-f with the reference DPA (see section 4.1.3) were determined and were in the range of $\lambda=240-243 \mathrm{~nm}$. These specific isobestic points then were used as excitation wavelengths $\lambda_{\text {exc }}$ for the fluorescence emission spectroscopy and the results from the emission data are shown in Figures 14 and 15 and in Table 12.


11a


11d


11b


11e


11c

$11 f$



Figure 14. Absorption spectra (left) and normalized and corrected emission spectral data of 11 in $10^{-6} \mathrm{M}$ samples (right)


Figure 15. Plot of emission wavelength versus $\sigma_{p a r a}$ Hammett constants

Table 12. Photophysical properties ( $10^{-6} \mathrm{M}$ in cyclohexane) of $\mathbf{1 1}$ and $\sigma_{p a r a}$ Hammett constants

| Sample | R | $\lambda_{\text {exc }} / \mathrm{nm}$ | $\lambda_{\mathrm{em}}^{\max / n m}$ | $\Phi_{\mathrm{s}}{ }^{\text {a }}$ | $\sigma_{\text {para }}$ |
| :---: | :--- | :---: | :---: | :---: | :--- |
| 11a | OMe | 241 | 330 | 0.50 | -0.268 |
| 11b | Me | 242 | 331 | 0.61 | -0.17 |
| 11c | Ph | 243 | 367 | 0.87 | -0.01 |
| 11d | H | 241 | 333 | 0.66 | 0 |
| 11e | Cl | 241 | 344 | 0.61 | 0.227 |
| 11f | $\mathrm{CF}_{3}$ | 240 | 361 | 0.64 | 0.54 |

${ }^{\text {a }}$ Quantum yields were calculated relative to DPA ( $\Phi_{\mathrm{r}}=0.97$ ).

The results showed that indeed electron-donating and -withdrawing groups had an effect on the emission wavelength maxima. Moreover there was a reasonable correlation of the emission wavelength maxima with the $\sigma_{p a r a}$ Hammett constants (Figure 15). ${ }^{53}$ The obtained data indicate that the emission maxima were red-shifted in the presence of electrondeficient para-groups. This was clearly demonstrated by the difference of more than 30 nm between 11a ( $\lambda_{\mathrm{em}}=330 \mathrm{~nm}, \sigma_{\text {para }}=-0.268$ ) and $11 \mathrm{f}\left(\lambda_{\mathrm{em}}=361 \mathrm{~nm}, \sigma_{\text {para }}=0.54\right)$. The quantum yields of SBFs 11 were in general high ( $\Phi_{\mathrm{s}}=0.50-0.87$ ) and match those of other 4-aryl substituted SBF derivatives. ${ }^{2,3 a, 8 d, 44,48}$ Notwithstanding, it should be noted that, in comparison with the parent 4-phenyl-substituted $\operatorname{SBF}\left(\lambda_{\mathrm{em}}=358 \mathrm{~nm}, \Phi_{\mathrm{s}}=0.40\right)^{2 \mathrm{a}}$ and with 11d, which also have 4-phenyl substituent, the quantum yield, $\Phi_{\mathrm{s}}=0.66$, was even better. Surprisingly, the difference of 25 nm in the emission maximum between 4-phenyl-substituted SBF and 11d ( $\lambda_{\mathrm{em}}=333 \mathrm{~nm}$ ) was rather high.

Additionally, the only exception to the linear correlation between $\sigma_{p a r a}$ and the emission wavelength maximum was observed in 11c ( $\sigma_{\text {para }}=-0.01$ ). The effect of one additional $\pi$-system on the SBF may explain the red-shifted emission wavelength ( $\lambda_{\mathrm{em}}=367$ nm ) and the very high quantum yield, $\Phi_{\mathrm{s}}=0.87$.

After these results, samples 13a, 13b and 14a-c were also analyzed and compared with 11c because the structure of 11c is similar to that of the extended biaryl substituents at position 4 of SBFs without electron-donating or -withdrawing groups. The photophysical properties of $\mathbf{1 3}$ and $\mathbf{1 4}$ were measured under the same conditions and using DPA as reference as well. However, in contrast to the parent biphenyl substituted SBF 11c, neither the biphenyl substituted derivatives SBF $\mathbf{1 3}$ nor the diphenylethynyl substituted derivatives SBF $\mathbf{1 4}$ showed a significant red shift. Only the relative quantum yields $\Phi_{\mathrm{s}}$ increased (Figure 16 and Table 13). A detailed analysis of the results shows that the emissiom wavelength maxima $\lambda_{\mathrm{em}}$ of all samples ranged from 366 to 370 nm , thus matching the $\lambda_{\mathrm{em}}$ measured for $\mathbf{1 1 c}$ ( $\lambda_{\mathrm{em}}=367$ nm ). The measured quantum yields $\Phi_{\mathrm{s}}$ ranged from 0.65 , for $\mathbf{1 3 a}$, to 1.00 , for $\mathbf{1 4 c}$ with the electron-withdrawing $\mathrm{CF}_{3}$ substituent (Table 13), thereby showing that they are some of the most powerful 4 -substituted SBFs prepared thus far regarding quantum yields. ${ }^{2 a, 3 a, 8 d, 44,48,49}$

The analysis of carborane-14a showed no fluorescence emission after excitation in the range of $\lambda_{\text {exc }}=250-330 \mathrm{~nm}$, although the "face-on" conformation was confirmed based on X-ray data.





Figure 16. Absorption spectra (left) and normalized and corrected emission spectral data of 13 and $\mathbf{1 4}$ in $10^{-6} \mathrm{M}$ samples (right)

Table 13. Photophysical properties of $\mathbf{1 3 a} \mathbf{- b}$ and $\mathbf{1 4 a - c}\left(10^{-6} \mathrm{M}\right.$ in cyclohexane)

| Sample | R | $\lambda_{\text {exc }} / n m$ | $\lambda_{\mathrm{em}}^{\text {max } / n m}$ | $\Phi_{\mathrm{s}}{ }^{\text {a }}$ |
| :---: | :--- | :---: | :---: | :---: |
| 13a | OMe | 254 | 366 | 0.65 |
| 13b | $\mathrm{CF}_{3}$ | 251 | 369 | 0.99 |
| 14a | $\mathrm{OMe}^{2}$ | 253 | 367 | 0.82 |
| 14b | H | 254 | 369 | 0.79 |
| 14c | $\mathrm{CF}_{3}$ | 248 | 370 | 1.00 |

[^0]The photophysical properties of the last set with PAH-substituted SBFs 18a-d, were also analyzed, and the results are shown in Figure 17. Although all four samples exhibit absorption spectra in the range of $\lambda_{\text {abs }}=250-330 \mathrm{~nm}$, only $\mathbf{1 8 b}-\mathbf{d}$ also showed red-shifted emission ( $\lambda_{\mathrm{em}}=353-429 \mathrm{~nm}$ ). The emission quantum yields were low for $\mathbf{1 8 b}\left(\Phi_{\mathrm{s}}=0.12\right.$, Table 14) and $\mathbf{1 8 d}\left(\Phi_{\mathrm{s}}=0.11\right)$, but very high in the case of the anthranyl-substituted SBF 18c ( $\Phi_{\mathrm{s}}=0.80$ ). It is noteworthy that the fluorescence emission spectra of 18c and 18d show emission patterns similar to those of the parent $\pi$-aromatic hydrocarbons anthracene and pyrene, albeit red-shifted by almost 30 nm and 10 nm , respectively. ${ }^{54,55}$


18a


18b


18c


18d



Figure 17. Absorption spectra (left) and normalized and corrected emission spectral data of 18 in $10^{-6} \mathrm{M}$ samples (right)

Table 14. Photophysical properties for $\mathbf{1 8 b} \mathbf{- d}\left(10^{-6} \mathrm{M}\right.$ in cyclohexane)

| Sample | R | $\lambda_{\text {exc }} / \mathrm{nm}$ | $\lambda_{\mathrm{em}}^{\max } / \mathrm{nm}$ | $\Phi_{\mathrm{s}}{ }^{a}$ |
| :---: | :--- | :---: | :---: | :---: |
| 18b | $\mathrm{Ar}=9$-phenanthryl | 267 | 353,370 | 0.12 |
| 18c | $\mathrm{Ar}=$ 9-anthranyl | 270 | 405,429 | 0.80 |
| 18d | $\mathrm{Ar}=$ 2-pyrenyl | 267 | 378,389 | 0.11 |

${ }^{\text {a }}$ Quantum yields were calculated relative to DPA ( $\Phi_{\mathrm{r}}=0.97$ ).

### 4.1.6. Summary and outlook of $\mathbf{4}$-substituted SBFs

In the first part of the study, 19 different substituted SBFs at position 4 were synthesized. These substituents included various electron-donating and -withdrawing groups, aromatic substituents and even fused $\pi$-extended aromatic hydrocarbons (PAHs). The results showed the effect of different substituents on the photophysical properties of SBFs and indicate the existence of $\pi$-conjugation breaking between the 4 -substituent and the fluorene, based on X-ray data. Indeed, the X-ray data of all analyzed substituted SBFs reveal a large and, in most cases, almost perpendicular arrangement between the aforementioned fragments. This unique property disrupts the electronic coupling between them, ${ }^{48}$ which is presumably restored, at least partly, by planarization of the molecular structure during the stabilization of the excited state. ${ }^{44 a, 49}$ This characteristic of steric hindrance may be used to tune the $\pi$ conjugation intensity in materials science for applications in electronic devices. Because of the spiro-framework, these 4-substituted SBFs also exhibit superior thermal and morphological stability, ${ }^{1}$ which are key properties for a broad range of applications. ${ }^{2}$ Although the effect on 4 -substituted SBFs must be further studied, the results already show a high potential for key contributions to the fields of materials science and to organic electronics.

### 4.2. Dispiroindeno[2,1-c]fluorenes (DS-IFs)

### 4.2.1. Retrosynthesis of indeno[2,1-c]fluorenes

In the previous chapters, a new synthetic approach to 1,2,3,4-tetrasubstituted SBFs was described and their photophysical properties were studied (Section 4.1). In this chapter, the synthesis of structurally related indenofluorenes (IF), dihydroindenoflurenes and dispiroindenofluorenes was examined (DS-IF). DS-IFs, due to their structural similarity with SBFs, belong to the same class of condensed polyaromatic hydrocarbons. Similarly to SBFs, the DS-IFs are potential candidates for several applications, specifically in organic electronics, dyes and photovoltaic cells. ${ }^{6 a-\mathrm{g}, 41}$

The indenofluorene framework is composed of the fluorene unit and the indene unit. The indene unit can be attached to different edges of the fluorene scaffold leading to 5membered family of the corresponding regioisomers: 7,12-dihydroindeno[1,2-a]fluorene, 11,12-dihydroindeno[2,1-a]fluorene, 6,12-dihydroindeno[1,2-b]fluorene, 5,7-dihydroindeno[2,1-b]fluorene and 5,8-dihydroindeno[2,1-c]fluorene (Figure 18). ${ }^{39}$


11,12-dihydroindeno [2,1-a]fluorene


6,12-dihydroindeno [1,2-b]fluorene


5,7-dihydroindeno [2,1-b]fluorene


7,12-dihydroindeno
[1,2-a]fluorene


5,8-dihydroindeno [2,1-c]fluorene

Figure 18. Isomers of the dihydroindenofluorene family

Synthetic approaches for the synthesis of the first four regioisomeric dihydroindenofluorenes have been extensively studied, yet only few reports on the synthesis of the last regioisomer - indeno[2,1-c]fluorene - are available. Furthermore, almost all known examples of the synthesis of indeno[2,1-c]fluorenes are based on intramolecular Friedel-Crafts-type cyclizations (Scheme 52, Disconnection approach A), on intermolecular Diels-

Alder cyclization (Disconnection approach B) or on a combination of both approaches (for further details, please see: 2.1. Synthesis of indeno[2,1-c]fluorene).


Scheme 52. General disconnection approaches A and B for indeno[2,1-c]fluorenes

The limited approaches for the synthesis of indeno[2,1-c]fluorenes narrow the choices of suitably substituted starting material. Moreover, no general approach for the selective synthesis of the substituted indenofluorenes and structure-property analysis has been developed thus far. However, the indeno[2,1,-c]fluorenes can be seen as more structurally complex congener of fluorenes. Thus, the previously developed synthetic approach to 1,2,3,4tetrasubstituted SBFs based on catalytic $[2+2+2]$ cyclotrimerization could also be used for preparation indeno[2,1-c]fluorenes (IFs) and their dispiro derivatives (DS-IFs).

The retrosynthetic approach indicates that the DS-IFs could be formed from diketones that can be obtained from substituted indeno[2,1-c]fluorenediols by oxidation. The diols can be formed in a transition metal-catalyzed intramolecular [2+2+2] cyclotrimerization as the key step from substituted triyndiols. The triyndiols can be formed using the alkynylation protocol with lithium acetylides from dialdehydes that in turn can be obtained by Sonogashira coupling, as shown in the previous chapters (Scheme 53). ${ }^{36,47}$ The desired DS-IFs can then also be studied upon their photophysical properties.



Scheme 53. Retrosynthetic approach for substituted indeno[2,1-c]fluorene-5,8-diols

### 4.2.2. Synthesis of substituted DS-IFs

### 4.2.2.1. Synthesis of symmetrically substituted DS-IFs

Preparation of triyndiols 19 consisted of 2 steps. The first step was the Sonogashira coupling of commercially available 2-bromobenzaldehyde and 2-ethynylbenzaldehyde using $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(5 \mathrm{~mol} \%)$ and $\mathrm{CuI}(10 \mathrm{~mol} \%)$ with triethylamine as a base in THF at $80^{\circ} \mathrm{C}$ for 3 h furnishing dialdehyde 19 in $96 \%$ isolated yield. In the second step, twofold alkynylation of the dialdehyde with various 4 -substituted lithium phenylacetylides $(\mathrm{R}=\mathrm{MeO}-, \mathrm{Me}-, \mathrm{Ph}-, \mathrm{Br}-$, $\mathrm{CF}_{3}$-) or other acetylides ( $\mathrm{R}=$ thienyl-, $n$-Pr-, TMS-, 1-naphthyl-) at $-78{ }^{\circ} \mathrm{C}$ gave rise to nine symmetrically substituted triyndiols $\mathbf{2 0}$ in very high isolated yields ranging from 77 to $95 \%$ (Scheme 54).



20
20a, $\mathrm{R}=4-\mathrm{MeOC}_{6} \mathrm{H}_{4}-94 \%$ 20f, $\mathrm{R}=2$-thienyl- $88 \%$
20b, $\mathrm{R}=4-\mathrm{MeC}_{6} \mathrm{H}_{4}-\quad 89 \% \quad 20 \mathrm{~g}, \mathrm{R}=n-\mathrm{Pr}-\quad 95 \%$
20c, $\mathrm{R}=\mathrm{Ph}-\quad 84 \%$ 20h, $\mathrm{R}=\mathrm{TMS} / \mathrm{H} \quad 84 \%$
20d, $\mathrm{R}=4-\mathrm{BrC}_{6} \mathrm{H}_{4}-\quad 77 \%$ 20i, $\mathrm{R}=1$-naphthyl- $85 \%$
20e, $\mathrm{R}=4-\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}-\quad 86 \%$
Scheme 54. Synthesis of symmetrically substituted triyndiols 20a-i

Since the cyclotrimerization is the key step of DS-IF formation, the rhodium-complex catalyzed $[2+2+2]$ cyclotrimerization to form indeno[2,1-c]fluorene-5,8-diols 21 in the third step was studied. Prior to the experiments, DFT calculations (B3LYP/6-311+G(d,p) at $\mathrm{T}=$ 453.15 K ) were done to determine the free energy $\Delta G$ of the conversion of both cis and trans stereoisomers of triyndiols 20 into indeno[2,1-c]fluorene-5,8-diols 21. The $\Delta G$ values of $-381.78 \mathrm{~kJ} / \mathrm{mol}$ (cis-isomer) and $-378.04 \mathrm{~kJ} / \mathrm{mol}$ (trans-isomer) showed that the conversion is energetically favorable despite the high torsional strain caused by steric hindrance in all five ortho-fused rings. Encouraged by the favorable computational results, experiments ensued.

Regarding intramolecular $[2+2+2]$ cyclotrimerization reaction conditions of 20, the ones previously used, based on Wilkinson's catalysts ( $3 \mathrm{~mol} \%$ ) and $\mathrm{Ag}_{2} \mathrm{CO}_{3}(6 \mathrm{~mol} \%$ ) as an additive in THF under microwave irradiation $\left(1.5 \mathrm{~h}, 180^{\circ} \mathrm{C}\right)$, were applied. Gratifyingly, the desired indeno[2,1-c]fluorene-5,8-diols 21 were formed in very high isolated yields (73-96\%) as inseparable mixtures of diastereoisomers. In contrast to intermolecular cyclotrimerizations, no side reactions, such as homo-coupling (Section 4.1), occured in this step. In the cases of triyndiols 20a-c, 20e and 20 f a full conversion of the starting material into the target indeno[2,1-c]fluorene-5,8-diols was obtained. In the cases of compounds 21d with 4bromoaryl substituted and 21i with naphthyl substituted indeno[2,1-c]fluorene-5,8-diols the yields were slightly lower ( $84 \%$ and $83 \%$, respectively). The lowest yield of $73 \%$ was observed in the case of $n$-propyl substituted indeno[2,1-c]fluorene-5,8-diol 21g. The only exception among the symmetrically substituted indeno[2,1-c]fluorene-5,8-diols was the monosilylated diol $\mathbf{2 1 h}$. The monosilylated diol $\mathbf{2 1 h}$ was obtained after work-up of the reaction mixture as the result of protodesilylation. Similar mono-desilylation reactions have already been observed and explained by the effect of sterically driven protosilylation during the purification process on silica gel. ${ }^{56}$

Then, 2-step conversion of indeno[2,1-c]fluorene-5,8-diols into DS-IFs consisting of oxidation and the final formation of the target product ensued. In the fourth step, diols 21a-i were subjected to oxidation with PCC in dichloromethane at $25^{\circ} \mathrm{C}$ forming the corresponding indeno[2,1-c]fluorene-5,8-diones 22a-i in almost quantitative isolated yields ranging from 83 to $96 \%$. In the final step, the reaction of indeno[2,1-c]fluorene-5,8-diones 22a-i with two equivalents of 2-lithiobiphenyl, followed by acidic cyclization using acetic acid and HCl at $110{ }^{\circ} \mathrm{C}$ furnished symmetrically substituted 5,8-dispiroindeno[2,1-c]fluorenes (DS-IFs) 23a-i in very good yields (52-83\%, Scheme 55). The overall yields of the successfully applied fivestep synthesis of symmetrically substituted DS-IFs 23a-i ranged from 28 to 60\% (the overall yields are specified in parenthesis).


Scheme 55. Synthetic pathway of symmetrically substituted DS-IFs 23

The structure of DS-IF 23b was unequivocally determined by a single crystal X-ray structure analysis and showed a high degree of molecular twist in the altering ortho-fused five- and six-membered ring structure (Figure 19).


Figure 19. Platon plot of 23b showing displacement ellipsoids at a $50 \%$ probability level

The derived dihedral angles add up to a high twisting angle of the backbone in which the dihedral angle of the fused ring system 1-2-3 was $15.5^{\circ}$, the angle of the fused ring system 2-3-4 was $18.3^{\circ}$ and the dihedral angle of the ring system 3-4-5 was $3.2^{\circ}$ (Figure 20). The total value of $32.4^{\circ}$ was almost three times higher than the $13^{\circ}$ backbone twisting of the parent 5,8-dispiroindeno[2,1-c]fluorene, as reported by Poriel et.al. ${ }^{6 a}$ Furthermore, the aromatic substituents at positions 6 and 7 are almost perpendicular to ring 3, as shown for SBF derivatives.


Figure 20.Visualization of DS-IF backbone twisting

### 4.2.2.2. Synthesis of unsymmetrically substituted DS-IFs

In the previous chapter, the symmetrically substituted SD-IFs $\mathbf{2 3}$ were synthesized in good overall yields. In addition to symmetrically substituted $\mathbf{2 3}$, it is assumed that the same methodology can be applied for synthesis of unsymmetrically substituted DS-IF derivatives. As far as properties of the unsymmetrically substituted DS-IFs are concerned, it is expected that they should exhibit interesting photophysical properties. For example, the interplay of electron-donating and -withdrawing groups within one molecule could result in dramatical change of their fluorescence emission spectra. With this goal in mind, strategies were developed for:
a) synthesis of DS-IF with different substituents at positions 6 and 7 and
b) synthesis of DS-IF with different substituents at positions 3 and 10 .
a) Synthesis of DS-IF with different substituents at positions 6 and 7

In the first step, unsymmetrically substituted indeno[2,1-c]fluorene-5,8-diols at positions 6 and 7 were prepared. For this purpose, alkyne 24 and diyne 25 were synthesized using alkynylation reaction of the commercially available starting materials. More specifically, alkyne 24 was formed by alkynylation reaction of 2-iodobenzaldehyde with lithium (4-trifluoromethylphenyl)acetylide at $-78{ }^{\circ} \mathrm{C}$ in THF for 3 h in $83 \%$ isolated yield. Diynol 25 was also synthesized by alkynylation reaction of 2-ethynylbenzaldehyde with lithium (4-methoxyphenyl)acetylide at $-78{ }^{\circ} \mathrm{C}$ in THF for 3 h in $94 \%$ isolated yield. Compounds 24 and 25 were then connected using the Sonogashira cross-coupling reaction $\left(\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(5 \mathrm{~mol} \%), \mathrm{CuI}(10 \mathrm{~mol} \%), \mathrm{NEt}_{3}\right.$, THF, $\left.80{ }^{\circ} \mathrm{C}, 3 \mathrm{~h}\right)$ to form triyndiol 26 in $91 \%$ isolated yield (Scheme 56). Gratifyingly, triyndiol 26 was cyclotrimerized in the presence of Wilkinson's catalyst under the previously studied conditions $\left(\mathrm{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}(3 \mathrm{~mol} \%), \mathrm{Ag}_{2} \mathrm{CO}_{3}\right.$ ( $6 \mathrm{~mol} \%$ ), THF, $180{ }^{\circ} \mathrm{C}$, MW, 1.5h) into indeno[2,1-c]fluorene-5,8-diol 27 in a very high yield of $84 \%$. Subsequently, oxidation to the corresponding indeno[2,1-c]fluorene-5,8-dione 28 with PCC in dichloromethane at $25^{\circ} \mathrm{C}$ also furnished 28 in a high isolated yield of $80 \%$. Lastly, the unsymmetrically substituted DS-IF $\mathbf{2 9}$ was synthesized by reaction of $\mathbf{2 8}$ with 2 equivalents of 2-lithiobiphenyl, followed by treatment with a mixture of acetic acid and a catalytic amount of concentrated HCl at $110{ }^{\circ} \mathrm{C}$ in a $65 \%$ isolated yield. The five-step synthetic pathway allowed synthesizing 29 in the overall yield of $33 \%$ (shown in parenthesis).



25, 94\%
$\mathrm{Ag}_{2} \mathrm{CO}_{3},(6 \mathrm{~mol} \%)$
$\mathrm{THF}, 180^{\circ} \mathrm{C}, 1.5 \mathrm{~h}, \mathrm{MW}$


29, 65\% (33\%)

1. $2-\mathrm{LiC}_{6} \mathrm{H}_{4}-\mathrm{Ph}, \mathrm{THF}$
2. $12 \mathrm{M} \mathrm{HCl}, \mathrm{AcOH}$ $110^{\circ} \mathrm{C}, 3 \mathrm{~h}$

28, 80\%


27, 84\%

Scheme 56. Synthesis of unsymmetrically substituted DS-IF 29 (Pathway A)
b) Synthesis of DS-IF with different substituents at positions 3 and 10 .

The synthesis started with the preparation of substituted 3-ethynylbenzaldehyde S1 and bromobenzaldehyde $\mathbf{S 2}$. 3-Ethynylbenzaldehyde $\mathbf{S} 1$ was formed by Sonogashira reaction of 2-bromo-5-methoxybenzaldehyde with trimethylsilylacetylene using $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}$ (5 $\mathrm{mol} \%$ ) and $\mathrm{CuI}(10 \mathrm{~mol} \%)$ with triethylamine as base in THF at $80^{\circ} \mathrm{C}$ for 3 h . After filtration and evaporation of the solvents, the crude was subjected to deprotection of the TMS group under basic conditions using $\mathrm{K}_{2} \mathrm{CO}_{3}$ in a mixture of $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}$ at $0{ }^{\circ} \mathrm{C}$. It furnished $\mathbf{S 1}$ with terminal triple bond in $96 \%$ isolated yield. ${ }^{57}$ Bromobenzaldehyde $\mathbf{S 2}$ was formed by metallation of 4-trifluoromethylbromobenzene with LDA at $-78{ }^{\circ} \mathrm{C}$ for 30 min , then addition of DMF followed and the reaction mixture was stirred additionally at $25^{\circ} \mathrm{C}$ for 1 h yielding $\mathbf{S} \mathbf{2}$ in $89 \%$ yield after acidic work-up and isolation. ${ }^{58}$ Sonogashira reaction of S1 with S2 $\left(\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(5 \mathrm{~mol} \%), \mathrm{CuI}(10 \mathrm{~mol} \%), \mathrm{NEt}_{3}, \mathrm{THF}, 80^{\circ} \mathrm{C}, 3 \mathrm{~h}\right)$ gave rise to dialdehyde $\mathbf{3 0}$ in $72 \%$ isolated yield. The substituted dialdehyde $\mathbf{3 0}$ was alkynylated with lithium $p$ tolylacetylide at $-78{ }^{\circ} \mathrm{C}$ in THF into triyndiol $\mathbf{3 1}$ in $75 \%$ isolated yield (Scheme 57).


Scheme 57. Synthesis of unsymmetrically substituted triyndiol 31

Next, the intramolecular Rh-catalyzed cyclotrimerization was attempted. The previously used reaction conditions (Wilkinson's catalyst ( $3 \mathrm{~mol} \%$ ), $\mathrm{Ag}_{2} \mathrm{CO}_{3}$ ( $6 \mathrm{~mol} \%$ ), THF, $180{ }^{\circ} \mathrm{C}$, MW, 1.5 h ) were applied and triyndiol 31 was converted into the desired unsymmetrically substituted indeno[2,1-c]fluorene-5,8-diol 32, albeit in a slightly lower isolated yield of $60 \%$. To understand why the yield of the reaction was somewhat lower than in the case of indeno[2,1-c]fluorene-5,8-diols 21a-i, which showed almost full conversion (Section 4.2.2.1, Scheme 55), additional experiments were carried out. The reaction conditions, such as reaction time ( $30 \mathrm{~min}-1.5 \mathrm{~h}$ ) and temperature $\left(100-180{ }^{\circ} \mathrm{C}\right)$ were changed. Although in all experiments triyndiol $\mathbf{3 1}$ was fully consumed, judged by the disappearance of the starting material (TLC analysis), no change in yields of 32 were observed. Partial decomposition of $\mathbf{3 1}$ during the course of the reaction to undefined products cannot be excluded. Oxidation of $\mathbf{3 2}$ with PCC in dichloromethane at $25^{\circ} \mathrm{C}$ furnished the corresponding unsymmetrically substituted indeno[2,1-c]fluorene-5,8-dione 32 in a high isolated yield of $87 \%$. In the last step, reaction of 32 with 2 equivalents of 2 -lithiobiphenyl, followed by treatment with a mixture of acetic acid and a catalytic amount of concentrated HCl at $110^{\circ} \mathrm{C}$ provided the unsymmetrically substituted DS-IF 34 in $84 \%$ isolated yield. The overall yield of the six-step reaction procedure was $21 \%$ (Scheme 58).



32, 60\%

33, 87\%


34, 84\% (21\%)

Scheme 58. Synthesis of unsymmetrically substituted DS-IF 34 (Pathway B)

### 4.2.3. Preparation of [7]-helical DS-IFs

Until now, a set of nine symmetrically substituted DS-IFs, 23, and two unsymmetrically substituted DS-IFs 29 bearing substituents at positions 6 and 7 and $\mathbf{3 4}$ at positions 3 and 10 were prepared. In this step the developed synthetic method for the preparation of DS-IFs was applied to DS-IFs with more than five ortho-fused rings. Although the synthetic strategy was the same, now 1-bromo-2-naphthaldehyde was used as the starting material to synthesize DS-IFs possessing the [7]- helical DS-IF scaffold. Such compounds are structurally related to [7]-helical fluorenes and their heterocyclic analogs. ${ }^{59}$

The reaction sequence started with Sonogashira coupling of 1-bromo-2naphthaldehyde with ethynyltrimethylsilane and was performed using $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}$ ( $5 \mathrm{~mol} \%$ ) and $\mathrm{CuI}(10 \mathrm{~mol} \%)$ with triethylamine as a base in THF at $80^{\circ} \mathrm{C}$ for 3 h . The product was not isolated, but was instead filtered, and the solvents were removed. The subsequent removal of the TMS protection group under basic conditions with $\mathrm{MeOH} / \mathrm{K}_{2} \mathrm{CO}_{3}$ at $0{ }^{\circ} \mathrm{C}$ furnished 1-ethynyl-2-naphthaldehyde 36 in almost quantitative yield of $96 \%$. In the following step, the second Sonogashira reaction of 1-bromo-2-naphthaldehyde $\left(\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}\right.$ ( $\left.5 \mathrm{~mol} \%\right), \mathrm{CuI}(10$ $\mathrm{mol} \%$ ), $\mathrm{NEt}_{3}$, THF, $80{ }^{\circ} \mathrm{C}, 3 \mathrm{~h}$ ) with 36 gave rise to dialdehyde 37, again, in an almost quantitative isolated yield (96\%). In the third step, the subsequent alkynylation of 37 with different 4 -substituted lithium phenylacetylides ( $\mathrm{R}=\mathrm{MeO}-, \mathrm{H}-, \mathrm{CF}_{3}$-) in THF at $-78{ }^{\circ} \mathrm{C}$ provided three symmetrically substituted triyndiols 38a-c in very high isolated yields ranging from 76 to $94 \%$.

With triyndiols 38a-c in hand, catalytic cyclotrimerization reactions followed. As in the previous cases, the $[2+2+2]$ cyclotrimerization, catalyzed by $\mathrm{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}(3 \mathrm{~mol} \%)$ with $\mathrm{Ag}_{2} \mathrm{CO}_{3}(6 \mathrm{~mol} \%)$ as additive in THF under MW irradiation at $180^{\circ} \mathrm{C}$, gave the desired [7]helical indenofluorenediols 39a-c in almost quantitative isolated yields (94-96\%) as racemic mixtures of $(M)$ - and $(P)$-helical enantiomers. Gratifyingly, these examples show the potential and efficiency of the Rh-catalyzed $[2+2+2]$ cyclotrimerization. In particular, the excellent yield of the synthesis of the highly twisted helical systems (39a-c), consisting of all orthofused six- and five-membered rings, highlights the importance of the intramolecular [2+2+2] cyclotrimerization key step. In this study, no attempts were performed to synthesize enantiomerically enriched molecules $\mathbf{3 9}$ or to separate the racemic mixtures into individual enantiomers.

Finally, the [7]-helical indenofluorenediols 39a-c were converted into [7]-helical DSIFs by oxidation and formation of the dispiroderivatives (the final products). For this purpose,

39a-c were oxidized with PCC in dichloromethane at $25{ }^{\circ} \mathrm{C}$ forming [7]-helical indenofluorenediones 40a-c in high isolated yields ranging from 82 to $95 \%$. Then, the twostep conversion based on the reaction of 40a-c with 2-lithiobiphenyl and followed by treatment with a mixture of acetic acid and a catalytic amount of concentrated HCl at $110{ }^{\circ} \mathrm{C}$ furnished [7]-helical DS-IFs 41a-c in high yields ranging from 76 to $79 \%$.

In summary, three [7]-helical DS-IFs 41a-c were formed in six steps from the commercially available starting materials, the overall yields were in a very good range of 4856\% (Scheme 59, overall yields are shown in parenthesis).


Scheme 59. Synthesis of extended [7]-helical DS-IF 41

The helical structures of 40a and 41a were unequivocally confirmed by a single crystal X-ray analysis (41a, Figure 21). The data show dihedral angles of $7.4^{\circ}$ for the ring system 1 -$2-3,14.8^{\circ}$ for 2-3-4, $24.0^{\circ}$ for 3-4-5, $23.3^{\circ}$ for 4-5-6 and 2.0 for 5-6-7 (Figure 22). The total sum of $64.1^{\circ}$ reflects the high degree of molecular twist. The molecular twisting angles of structurally related [7]-helical compounds range from 82.8 to $97.8^{\circ} .{ }^{59}, \mathrm{~b}$ Thus, the degree of molecular twist in [7]-helical DS-IF 41a is lower, most likely due to the presence of two cyclopentadiene rings.


Figure 21. Platon plot of 41a showing displacement ellipsoids at a $50 \%$ probability level



Figure 22. Colorized visualization of (P)-[7]-helical DS-IF 41a backbone twisting

### 4.2.4. Analysis of photophysical properties

### 4.2.4.1. Photophysical properties of DS-IFs 23, 29 and 34

In the previous sections, a set of nine symmetrically substituted DS-IFs 23a-i, two unsymmetrically substituted DS-IFs 29 and $\mathbf{3 4}$ and three [7]-helical DS-IFs 41a-c were synthesized.

Firstly, the analysis of the absorption spectral data with DPA as the standard reference was conducted for the symmetrically substituted DS-IF 23a-i. The data of the absorption spectroscopy show that the isobestic point for the further calculation of relative quantum yields of DS-IFs 23a-i was best in the range from 264 to 271 nm . The subsequent fluorescence emission data after excitation at corresponding $\lambda_{\text {exc }}=264-271 \mathrm{~nm}$ showed a maximum in the very narrow range of $359-380 \mathrm{~nm}$ (Figure 23) and quantum yields ranging from $\Phi_{s}=0.09-0.70$ (Table 15) for all DS-IFs, except 23h with TMS-substituent, which showed no emission in this area.



23e

23f




Figure 23. Absorption and normalized emission spectral data of symmetrically substituted DS-IFs 23 in $10^{-6} \mathrm{M}$

Table 15. Photophysical properties for 23a-i $\left(10^{-6} \mathrm{M}\right.$ in cyclohexane)

| Sample | R | $\lambda_{\mathrm{exc}} / \mathrm{nm}$ | $\lambda_{\mathrm{em}}^{\max } / \mathrm{nm}$ | $\Phi_{\mathrm{s}}{ }^{\text {a }}$ |
| :---: | :--- | :--- | :--- | :--- |
| 23a | $\mathrm{R}=4-\mathrm{MeOC}_{6} \mathrm{H}_{4}-$ | 264 | 370,380 | 0.31 |
| 23b | $\mathrm{R}=4-\mathrm{MeC}_{6} \mathrm{H}_{4^{-}}$ | 266 | 366 | 0.09 |
| 23c | $\mathrm{R}=\mathrm{Ph}-$ | 266 | 366 | 0.67 |
| 23d | $\mathrm{R}=4-\mathrm{BrC}_{6} \mathrm{H}_{4}-$ | 269 | 365 | 0.33 |
| 23e | $\mathrm{R}=4-\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}-$ | 269 | 364 | 0.70 |
| 23f | $\mathrm{R}=2-$-thienyl- | 271 | 366,380 | 0.20 |
| 23g | $\mathrm{R}=n-\mathrm{Pr}-$ | 269 | 351,359 | 0.63 |
| 23h | $\mathrm{R}=\mathrm{TMS} / \mathrm{H}-$ | 266 | - | 0.00 |
| 23i | $\mathrm{R}=$ 1-naphthyl- | 264 | 369 | 0.28 |

${ }^{\text {a }}$ Quantum yields were calculated relative to DPA ( $\Phi_{\mathrm{r}}=0.97$ ).

The unsymmetrically substituted DS-IFs 29 and $\mathbf{3 4}$ showed, after excitation at the isobestic point $\lambda_{\text {exc }}=272 \mathrm{~nm}$ (for 29) and $\lambda_{\text {exc }}=268 \mathrm{~nm}$ (for 34), fluorescence emission maxima at $\lambda_{\mathrm{em}}=370 \mathrm{~nm}$ and 389 nm (Figure 24) and quantum yields $\Phi_{\mathrm{s}}=0.54$ and 0.60 , respectively (Table 16).


29


34



Figure 24. Absorption and normalized emission spectral data of unsymmetrically substituted DS-IF 29 and 34

Table 16. Photophysical properties for 29 and $\mathbf{3 4}\left(10^{-6} \mathrm{M}\right.$ in cyclohexane)

| Sample | $\lambda_{\text {exc }} / \mathrm{nm}$ | $\lambda_{\mathrm{em}}^{\max } / \mathrm{nm}$ | $\Phi_{\mathrm{s}}{ }^{\text {a }}$ |
| :---: | :---: | :--- | :---: |
| 29 | 272 | 370 | 0.54 |
| 34 | 268 | 389 | 0.60 |

${ }^{\text {a }}$ Quantum yields were calculated relative to DPA ( $\Phi_{\mathrm{r}}=0.97$ ).

Comparative analysis with parent 5,8-dispiroindeno[2,1-c]fluorene showed that: ${ }^{6 a}$

1) Instead of two distinct emission peaks, all substituted DS-IFs show only one maximum emission peak, which overlaps a second emission peak in some cases.
2) All symmetrically substituted compounds $\mathbf{2 3}$ show the emission maximum $\lambda_{\mathrm{em}}=359-$ 380 nm red-shifted by 1-22 nm in relation to the unsubstituted DS-IF $\left(\lambda_{\mathrm{em}}=358 \mathrm{~nm}\right)$.
3) Electron-rich substituents apparently enhanced the red shift of the emission maxima, particularly 4-methoxyphenyl and thienyl substituents (23a and 23f, both $\lambda_{\mathrm{em}}=380 \mathrm{~nm}$ ) at positions 6 and 7 of the DS-IF. They both show a 22 nm shift in relation to the parent DS-IF.
4) The effect of the electron-donating and -withdrawing substituents directly attached to the DS-IF ( $\mathbf{3 4}, \lambda_{\mathrm{em}}=389 \mathrm{~nm}$, substituents at positions 3 and 10) is stronger than that of 4-phenyl-substituted moieties at positions 6 and 7 of the DS-IF 29 ( $\lambda_{\mathrm{em}}=370 \mathrm{~nm}$ ). This may be explained by the virtually perpendicular arrangement of the DS-IF and its substituents at positions 6 and 7, as shown by a X-ray single-crystal analysis of 23b (for details, please see: 4.2.2.1). More specifically, this molecular twist may lead to weaker interactions between $\pi$ systems, thereby decreasing the effect of substituents on the photophysical properties of the DS-IFs. Nevertheless, partial restoration of the planar arrangement presumably occurs in the excited state. The actual effect of twisted $\pi$-systems and their interactions with parent DS-IFs is not yet fully clarified, thus requiring further studies. ${ }^{2 \mathrm{a}, 3 \mathrm{a}, 8 \mathrm{~d}, 44,48,49}$

### 4.2.4.2. Photophysical properties of [7]-helical DS-IFs 41

Similarly to the analysis of DS-IFs 23, 29 and 34 above (Section 4.2.4.1), the photophysical properties of [7]-helical DS-IFs 41 were also analyzed. After excitation at the isobestic point $\lambda_{\text {exc }}=269 \mathrm{~nm}$ for all three compounds, the fluorescence emission data of 41a-c showed red-shifted emission in the range of $\lambda_{\mathrm{em}}=398-428 \mathrm{~nm}$ (Figure 25) with excellent quantum yields ranging from $\Phi_{\mathrm{s}}=0.75-0.88$ (Table 17).


41a



41b

Figure 25. Absorption and emission spectral data of symmetrically substituted [7]-helical DSIF 41 in $10^{-6} \mathrm{M}$ in cyclohexane

Table 17. Photophysical properties for 41a-c ( $10^{-6} \mathrm{M}$ in cyclohexane)

| Sample | $\mathrm{R} \quad \lambda$ | $\lambda_{\text {exc }} / \mathrm{nm}$ | $\lambda_{\text {em }}^{\text {max }} / \mathrm{nm}$ | $\Phi_{s}{ }^{\text {a }}$ |
| :---: | :---: | :---: | :---: | :---: |
| 41a | $\mathrm{R}=4-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ | - 269 | 418, 428 | 0.75 |
| 41b | $\mathrm{R}=\mathrm{Ph}-$ | 269 | 398, 428 | 0.87 |
| 41c | $\mathrm{R}=4-\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}-$ | 269 | 408, 423 | 0.88 |

In comparison with the 5 -ring containing DS-IFs 23a $\left(\mathrm{R}=\mathrm{OMe}, \lambda_{\mathrm{em}}=380 \mathrm{~nm}\right) \mathbf{2 3 c}(\mathrm{R}$ $\left.=\mathrm{H}, \lambda_{\mathrm{em}}=366 \mathrm{~nm}\right)$ and 23e $\left(\mathrm{R}=\mathrm{CF}_{3}, \lambda_{\mathrm{em}}=364 \mathrm{~nm}\right)$, the [7]-helical DSF-IFs 41a $(\mathrm{R}=\mathrm{OMe}$, $\left.\lambda_{\mathrm{em}}=428 \mathrm{~nm}\right), 41 \mathrm{~b}\left(\mathrm{R}=\mathrm{H}, \lambda_{\mathrm{em}}=428 \mathrm{~nm}\right)$ and $41 \mathrm{c}\left(\mathrm{R}=\mathrm{CF}_{3}, \lambda_{\mathrm{em}}=423 \mathrm{~nm}\right)$ show a shift in emission maxima by approximately 60 nm . This red shift of emission wavelength is cleary affected by the modulated $\pi$-extended structure. It is noteworthy that the compounds 41 showed the highest quantum yields ( $\Phi_{\mathrm{s}}=0.75-0.88$ ), whereas other [7]-helical molecules have quantum yields $\Phi=0.30-0.40 .{ }^{59}$

In summary, it was possible to apply the previously developed reaction sequence for the preparation of 1,2,3,4-tetrasubstituted SBFs to synthesize a set of substituted DS-IFs. Thus, for the first time, 12 molecules were prepared containing symmetrically and unsymmetrically substituted DS-IFs and three molecules of [7]-helical DS-IFs. Furthermore, these $\pi$-extended [7]-helical DS-IFs 41 show very high quantum yields and fluorescence emission maxima in the blue region of visible light. Therefore, these DS-IFs could be, similarly to the SBFs, good candidates for applications materials science, such as in organic electronics and light-emitting devices (OLEDs), in dyes or even photovoltaic cells. ${ }^{6-\mathrm{g}, 41}$

## 5. Conclusion

In this study a new synthetic method based on a Rh-complex catalyzed [2+2+2] cyclotrimerization of compounds with the triple bonds for preparation of fluorenes and its congeners was presented. The method was applied in syntheses of (a) selectively 1,2,3,4tetrasubstituted SBFs, (b) symmetrically and unsymmetrically substituted DS-IFs, and (c) [7]helical DS-IFs. This general reaction pathway consists of five or six steps and its length depended only on the availability of the starting materials.
a) In total, 19 different 1,2,3,4-tetrasubstituted SBFs with various substituents at position 4 were prepared. These substituents included electron-donating and -withdrawing groups, aromatic substituents and $\pi$-extended aromatic hydrocarbon moieties. The key step was the catalytic cyclotrimerization of diynols with internal alkynes (the best results were obtained with 3-hexyne). All cyclotrimerization products were isolated in yields ranging from 45 to $76 \%$. Additionally, an extensive screening of cyclotrimerization reactions with Wilkinson's catalyst under various reaction conditions (temperature, time, solvent and additives) was carried out as well. 1,2,3,4-Tetrasubstituted SBFs were synthesized in good overall yields (ranging from 9 to $34 \%$ ) from the commercially available starting material (Scheme 52). The photophysical properties of all SBFs were analyzed by absorption and emission spectroscopy. Fluorescence emission maxima $\lambda_{\mathrm{em}}$ were in the range of $315-389 \mathrm{~nm}$ with quantum yields $\Phi_{\mathrm{s}}$ up to 1.00 .


Conditions $\mathrm{A}: \mathrm{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}(10 \mathrm{~mol} \%)$, toluene, $90^{\circ} \mathrm{C}$, 16 h .
Conditions B: $\mathrm{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}(3 \mathrm{~mol} \%), \mathrm{Ag}_{2} \mathrm{CO}_{3}(6 \mathrm{~mol} \%)$, THF, MW, $180^{\circ} \mathrm{C}, 1.5 \mathrm{~h}$.
Scheme 52. Synthetic pathway for SBFs using Rh-catalyzed $[2+2+2]$ cyclotrimerization
b) $\quad 9$ symmetrically and 2 unsymmetrically substituted DS-IFs at positions 6 and 7 were synthesized bearing various electron-donating and -withdrawing groups by using the same strategy. The Rh-complex catalyzed $[2+2+2]$ cyclotrimerization of the prepared triyndiols gave rise to indeno[2,1-c]fluorene-5,8-diols that were isolated in yields ranging from 73 to $96 \%$ for symmetrically (Scheme 53) and from 60 to $84 \%$ for unsymmetrically substituted derivatives (Scheme 54). Subsequently, the corresponding DS-IFs were prepared from indeno[2,1-c]fluorene-5,8-diols in overall yields ranging from 21 to $60 \%$. The photophysical properties of all DS-IFs were analyzed by absorption and emission spectroscopy. Fluorescence emission maxima were in the range of $\lambda_{\mathrm{em}}=351-389 \mathrm{~nm}$ and quantum yields $\Phi_{\mathrm{s}}$ were up to 0.70 .


Scheme 53. Synthetic pathway for symmetrically substituted DS-IF using Rh-catalyzed $[2+2+2]$ cyclotrimerization


Scheme 54. Synthetic pathway for unsymmetrically substituted DS-IF using Rh-catalyzed $[2+2+2]$ cyclotrimerization
c) Finally, a set of three [7]-helical DS-IFs was prepared by using the same synthetic pathway from various commercially available starting compounds. The Rh-complex catalyzed $[2+2+2]$ cyclotrimerization with triyndiols possessing the naphthalene units provided [7]helical indenofluorenediols that were isolated in yields ranging from 94 to $96 \%$. Subsequently, the corresponding [7]-helical DS-IFs were formed from [7]-helical indenofluorenediols with overall yields ranging from 48 to $56 \%$ (Scheme 55). The photophysical properties of [7]-helical DS-IFs were again analyzed by absorption and emission spectroscopy. Fluorescence emission maxima were in the range of $\lambda_{\mathrm{em}}=423-428 \mathrm{~nm}$ with quantum yields $\Phi_{\mathrm{s}}$ up to 0.80 . These molecules have one of the highest measured quantum yields among the class of helicene like compounds recorded thus far.


Scheme 55. Synthetic pathway for [7]-helical DS-IFs using Rh-catalyzed [2+2+2] cyclotrimerization

In summary, this short and high-yielding five- or six-step synthetic pathway opens several possibilities for the preparation of new and selectively substituted SBFs and DS-IFs at different positions. The Rh-catalyzed $[2+2+2]$ cyclotrimerization emerged as the key step of this synthesis using the Wilkinson's catalyst $\left.\left(\mathrm{RhClPPh}_{3}\right)_{3}\right)$ as the most powerful catalyst. The systematic study provide useful guidelines for the future tuning and design of substituted 9,9'spirobifluorenes and dispiroindeno[2,1-c]fluorenes with predefined properties and their application in materials science.

## 6. Experimental part <br> 6.1. General

All reagents were commercially available and obtained from Sigma-Aldrich, Acros Organics, and Strem Chemicals companies. Solvents were purified and dried by distillation: tetrahydrofuran (THF) and toluene from sodium/benzophenone, dichloromethane from calcium hydride. Other solvents and all reagents were used without further purification. All reactions were performed under argon atmosphere unless otherwise noted. Chromatography was performed on Merck Silica gel 60. Thin layer chromatography was performed on Merck silica gel $60 \mathrm{~F}_{254}$ coated aluminium sheets. The ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra were recorded on a Bruker AVANCE III Spectrometer $\left({ }^{1} \mathrm{H}\right.$ at 600 MHz and ${ }^{13} \mathrm{C}$ at 150 MHz$)$ or on Bruker AVANCE III Spectrometer ( ${ }^{1} \mathrm{H}$ at 400 MHz and ${ }^{13} \mathrm{C}$ at 100 MHz ) as solutions in $\mathrm{CDCl}_{3}$, chemical shifts are given in $\delta$-scale $\left({ }^{1} \mathrm{H}\right.$ NMR spectra were referenced to residual peak of $\mathrm{CDCl}_{3}$ at $\delta 7.26,{ }^{13} \mathrm{C}$ NMR spectra to $\mathrm{CDCl}_{3}$ at $\delta 77.00$ ), coupling constants $J$ are given in Hz . The following abbreviations are used to describe spin multiplicity: $\mathrm{s}=$ singlet, $\mathrm{d}=$ doublet, $\mathrm{t}=$ triplet, $\mathrm{q}=$ quartet, $\mathrm{m}=$ multiplet, $\mathrm{dd}=$ doublet of doublet, $\mathrm{dt}=$ doublet of triplet, $\mathrm{dq}=$ doublet of quartet, $\mathrm{tt}=$ triplet of triplet. Assignments of proton resonances were confirmed, when possible, by known literature data. The IR spectra were recorded on a Bruker IFS 55 spectrometer in KBr tablets and are reported in wave numbers $\left(\mathrm{cm}^{-1}\right)$. In case of liquids or oils the compounds were dissolved in acetone, mixed with KBr and the acetone evaporated. The MS spectra were recorded on a VG-Analytical ZAB-SEQ device. All melting points are uncorrected and were determined on a Kofler apparatus.

### 6.2. Synthesis of 9,9 -spirobifluorene scaffolds

### 6.2.1. Synthesis of starting material 2,8 and 12

A: General procedure for Sonogashira reaction ${ }^{60}$ (preparation of 1 and 7). 2Bromobenzaldehyde ( $230 \mu \mathrm{~L}, 2.0 \mathrm{mmol}$ ), $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(69 \mathrm{mg}, 0.1 \mathrm{mmol}, 5 \mathrm{~mol} \%$ ) and CuI ( $38 \mathrm{mg}, 0.2 \mathrm{mmol}, 10 \mathrm{~mol} \%$ ) were dissolved in a Schlenk flask with reflux condensor in triethylamine ( 7 mL ) and THF ( 7 mL ), afterwards alkyne compound ( 2.4 mmol ) was added and the reaction was stirred 3 h under reflux. The reaction mixture was cooled down, filtered off with celite/silica and washed with diethyl ether. The organic fraction was concentrated on vacuum evaporator. Column chromatography of the residue on silica gel (hexanes/EtOAc) provided products.

2-(Phenylethynyl)benzaldehyde (1a). With phenylacetylene ( $260 \mu \mathrm{~L}, 2.4 \mathrm{mmol}$ ) following
 the general procedure A. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $399 \mathrm{mg}(96 \%)$ of the title compound as a yellow oil: ${ }^{1} \mathrm{H}$ NMR ( 600 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 10.66(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.95(\mathrm{ddd}, J=7.8,1.4,0.6 \mathrm{~Hz}, 1 \mathrm{H})$, 7.65 (ddd, $J=7.8,1.3,0.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.59(\mathrm{dt}, J=7.4,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.56-7.58$ (m, 2H), 7.44-7.47 (m, 1H), 7.38-7.40 (m, 3H); ${ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 191.7,135.8$, $133.8,133.2,131.7,129.1,128.6,128.5,127.2,126.9,122.3,96.3,84.9$. The spectral data were in accordance with previously published results. ${ }^{45}$

2-(Pent-1-yn-1-yl)benzaldehyde (1b). With 2-bromobenzaldehyde ( $500 \mu \mathrm{~L}, 4.0 \mathrm{mmol}$ ) and
 1-pentyne $(600 \mu \mathrm{~L}, 6.0 \mathrm{mmol})$ following the general procedure A. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 631 mg ( $92 \%$ ) of the title compound as a yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 10.54(\mathrm{~d}, J=0.9 \mathrm{~Hz}$, $1 \mathrm{H}), 7.87(\mathrm{~m}, 1 \mathrm{H}), 7.49(\mathrm{~m}, 2 \mathrm{H}), 7.35-7.38(\mathrm{~m}, 1 \mathrm{H}), 2.45(\mathrm{t}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 1.65(\mathrm{~m}, 2 \mathrm{H})$, $1.06(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 192.1,135.9$, 133.6, 133.3, 127.9, $127.8,126.9,98.0,76.4,22.0,21.5,13.6$. The spectral data were in accordance with previously published results. ${ }^{45}$

2-((4-Methoxyphenyl)ethynyl)benzaldehyde (7a). With 4-ethynylanisole (311 $\mu \mathrm{L}, 2.4$ mmol ) following the general procedure A. Column chromatography (10/1
 hexanes/EtOAc) yielded 450 mg ( $96 \%$ ) of the title compound as yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 10.65(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.93$ (ddd, $J$ $=7.8,1.3,0.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.61(\mathrm{ddd}, J=7.8,1.3,0.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.57(\mathrm{dt}, J=$ $7.3,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.51(\mathrm{~m}, 2 \mathrm{H}), 7.43(\mathrm{~m}, 1 \mathrm{H}), 6.91(\mathrm{~m}, 2 \mathrm{H}), 3.85(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 150 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 191.9,160.2,135.7,133.8,133.2,133.0,128.2,127.4,127.2,114.4,114.2,96.6$, 83.8, 55.4. The spectral data were in accordance with previously published results. ${ }^{61}$

2-(p-Tolylethynyl)benzaldehyde (7b). With p-tolylacetylene ( $305 \mu \mathrm{~L}, 2.4 \mathrm{mmol}$ ) following
 the general procedure A . Column chromatography ( $10 / 1$ hexanes/EtOAc) yielded 392 mg ( $89 \%$ ) of the title compound as pale yellow solid: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 10.66(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.95(\mathrm{ddd}, J=7.8,1.4,0.5$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 7.63 (ddd, $J=7.8,1.3,0.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.58(\mathrm{dt}, J=7.4,1.4 \mathrm{~Hz}, 1 \mathrm{H})$, 7.47 (m, 2H), $7.44(\mathrm{~m}, 1 \mathrm{H}), 7.19(\mathrm{~m}, 2 \mathrm{H}), 2.39(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 191.8$, 139.4, 135.7, 133.7, 133.1, 131.6, 129.3, 128.4, 127.17, 127.15, 119.2, 96.6, 84.3, 21.6. The spectral data were in accordance with previously published results. ${ }^{62}$

2-([1,1'-Biphenyl]-4-ylethynyl)benzaldehyde (7c). With 4-ethynylbiphenyl ( $428 \mathrm{mg}, 2.4$
 mmol ) following the general procedure A. Column chromatography (10/1 hexanes/EtOAc) yielded 520 mg ( $92 \%$ ) of the title compound as a light yellow solid: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 10.68(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.96$ (dd, $J=7.8,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.59-7.86(\mathrm{~m}, 8 \mathrm{H}), 7.47(\mathrm{~m}, 3 \mathrm{H}), 7.38(\mathrm{~m}, 1 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 191.7,141.8,140.1,135.8,133.8,133.2,132.1,128.9,128.6$, $127.8,127.3,127.2,127.1,126.9,121.2,96.3,85.6$. The spectral data were in accordance with previously published results. ${ }^{63}$

2-(Phenylethynyl)benzaldehyde (7d). With phenylacetylene ( $260 \mu \mathrm{~L}, 2.4 \mathrm{mmol}$ ) following
 the general procedure A. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $399 \mathrm{mg}(96 \%)$ of the title compound as a yellow oil: ${ }^{1} \mathrm{H}$ NMR ( 600 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 10.66(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.95(\mathrm{ddd}, J=7.8,1.4,0.6 \mathrm{~Hz}, 1 \mathrm{H})$, 7.65 (ddd, $J=7.8,1.3,0.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.59(\mathrm{dt}, J=7.4,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.56-7.58$
( $\mathrm{m}, 2 \mathrm{H}$ ), 7.44-7.47 (m, 1H), 7.38-7.40 (m, 3H); ${ }^{13} \mathrm{C}$ NMR ( $\left.150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 191.7, 135.8, $133.8,133.2,131.7,129.1,128.6,128.5,127.2,126.9,122.3,96.3,84.9$. The spectral data were in accordance with previously published results. ${ }^{60}$

2-((4-Chlorophenyl)ethynyl)benzaldehyde (7e). With 1-chloro-4-ethynylbenzene ( 238 mg ,
 2.4 mmol ) following the general procedure A . Column chromatography ( $10 / 1$ hexanes/EtOAc) yielded $431 \mathrm{mg}(90 \%)$ of the title compound as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 10.62(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.95$ (ddd, $J=7.8,1.4,0.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.64$ (ddd, $J=7.8,1.3,0.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.60(\mathrm{dt}$, $J=7.3,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.46-7.51(\mathrm{~m}, 3 \mathrm{H}), 7.36-7.38(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 191.5, 133.8, 133.2, 132.9, 128.9, 128.8, 127.5, 126.4, 120.8, 95.1, 85.9. The spectral data were in accordance with previously published results. ${ }^{64}$

2-((4-(Trifluoromethyl)phenyl)ethynyl)benzaldehyde (7f). With 1-ethynyltrifluortoluene
 ( $392 \mu \mathrm{~L}, 2.4 \mathrm{mmol}$ ) following the general procedure A. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $442 \mathrm{mg}(80 \%)$ of the title compound as a light brown solid: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 10.62(\mathrm{~d}$, $J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.97(\mathrm{ddd}, J=7.8,1.4,0.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.64-7.68(\mathrm{~m}, 5 \mathrm{H})$, $7.59(\mathrm{dt}, J=7.4,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 191.2,136.0$, $133.8,133.4,131.9,130.8\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=32.8 \mathrm{~Hz}\right), 129.2,127.6,126.1,125.8,125.4\left(\mathrm{q},{ }^{3} J_{\mathrm{C}-\mathrm{F}}=\right.$ $3.8 \mathrm{~Hz}), 121.1-126.5\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=272.3 \mathrm{~Hz}\right), 94.5,87.2$. The spectral data were in accordance with previously published results. ${ }^{64}$

4-((2-Formylphenyl)ethynyl)benzonitrile (7g). With 4-ethynylbenzonitrile ( $305 \mathrm{mg}, 2.4$
 mmol ) following the general procedure A. Column chromatography (10/1 hexanes/EtOAc) yielded 392 mg ( $84 \%$ ) of the title compound as a light yellow solid: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 10.59(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.97$ (ddd, $J=7.8,1.3,0.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.65-7.69(\mathrm{~m}, 5 \mathrm{H}), 7.63(\mathrm{dt}, J=7.3,1.4 \mathrm{~Hz}$, $1 \mathrm{H}), 7.53(\mathrm{~m}, 1 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 191.0,136.0,133.9,133.5,132.19,132.16$, $129.5,127.9,127.2,125.3,118.3,112.3,94.1,89.1$. The spectral data were in accordance with previously published results. ${ }^{65}$

B: General procedure for alkynylation reaction ${ }^{60}$ (preparation of 2, 8, 12). $n$ - BuLi $1.6 \mathrm{M}(1.8 \mathrm{~mL}, 2.7 \mathrm{mmol})$ was added dropwise to a solution of 1-ethynyl-4-p-toluene ( $340 \mu \mathrm{~L}$, $2.7 \mathrm{mmol})$ in anhydrous THF $(15 \mathrm{~mL})$, at $-78^{\circ} \mathrm{C}$. After 30 min of stirring, starting material $\mathbf{S}$ ( 1.78 mmol ) in THF ( 5 mL ) was added and the reaction mixture was stirred for 5 min at -78 ${ }^{\circ} \mathrm{C}$, afterwards it was allowed to warm up to ambient temperature and the reaction was stirred 4 h. Then $\mathrm{NH}_{4} \mathrm{Cl}$ aq was used to quench the reaction mixture, it was extracted with diethyl ether ( $3 \times 15 \mathrm{~mL}$ ), the combined organic fractions were washed with a saturated solution of $\mathrm{NaHCO}_{3}$ and brine, the organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$, filtered, and concentrated under reduced pressure. Column chromatography of the residue on silica gel (hexanes/EtOAc) provided products.

3-Phenyl-1-(2-(phenylethynyl)phenyl)prop-2-yn-1-ol (2a). With 1a ( $0.64 \mathrm{~g}, 2.8 \mathrm{mmol}$ ) and
 phenylacetylene ( 3.3 mmol ) following the general procedure B. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 0.73 g ( $86 \%$ ) of the title compound as pale yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.81(\mathrm{~d}, J=7.8$ $\mathrm{Hz}, 1 \mathrm{H}), 7.59-7.61(\mathrm{~m}, 3 \mathrm{H}), 7.48(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.42(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.35-7.38$ (m, $4 \mathrm{H}), 7.30-7.33(\mathrm{~m}, 3 \mathrm{H}), 6.19(\mathrm{~s}, 1 \mathrm{H}), 2.89(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.2$, $132.5,131.7,131.6,128.9,128.6,128.5,128.4,128.3,128.2,126.8,122.8,122.5,121.4,95.0$, $88.3,86.6,86.5,63.7$. The spectral data were in accordance with previously published results. ${ }^{65}$
(3-Phenyl-1-(2-(phenylethynyl)phenyl)prop-2-yn-1-yl)oxy-tert-butyldimethylsilane (2a-
 TBS). $2 \mathbf{2 a}(0.87 \mathrm{~g}, 3.3 \mathrm{mmol}$ ) was dissolved in anhydrous DMF ( 30 mL ) and cooled to $0^{\circ} \mathrm{C}$. Imidazole ( $0.34 \mathrm{~g}, 4.9 \mathrm{mmol}$ ), DMAP ( $60.5 \mathrm{mg}, 0.50 \mathrm{mmol}$ ) and afterwards TBDMSCl $(0.59 \mathrm{~g}, 3.9 \mathrm{mmol})$ were added and the reaction was stirred for 30 min at $0{ }^{\circ} \mathrm{C}$. Afterwards the reaction mixture was allowed to reach $25^{\circ} \mathrm{C}$ and stirred for 6 h . The reaction mixture was quenched with a saturated solution of ammonium chloride and extracted by diethyl ether ( $3 \times 20 \mathrm{~mL}$ ), combined organic fractions were washed with brine, dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and concentrated on vacuum evaporator. The residue was purified by column chromatography on silica gel (10/1 hexanes/EtOAc) to yield $0.61 \mathrm{~g}(53 \%)$ of the title compound as a light brown oil: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.82(\mathrm{~m}, 1 \mathrm{H}), 7.55-7.58(\mathrm{~m}, 2 \mathrm{H}), 7.54(\mathrm{~m}, 1 \mathrm{H}), 7.39-7.43(\mathrm{~m}, 3 \mathrm{H}), 7.35-$ $7.38(\mathrm{~m}, 3 \mathrm{H}), 7.32(\mathrm{dt}, J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.25-7.28(\mathrm{~m}, 3 \mathrm{H}), 6.21(\mathrm{~s}, 1 \mathrm{H}), 0.97(\mathrm{~s}, 9 \mathrm{H})$, $0.26(\mathrm{~s}, 3 \mathrm{H}), 0.20(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $\left.150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 143.6,132.0,131.6,131.5,128.8$,
128.42, 128.38, 128.13, 128.10, 127.6, 126.5, 123.3, 123.1, 120.8, 94.6, 90.1, 87.1, 85.0, 63.6, 25.8, 18.4, -4.4, -4.8; IR (KBr) $v_{\max } 3058,3028,2956,2926,2854,1598,1488,1473,1443$, 1323, 1251, 1111, 1075, 866, 842, $752 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{29} \mathrm{H}_{30} \mathrm{OSiNa}(\mathrm{M}+\mathrm{Na})$ calcd: 445.19581, found: $445.19583 ; R_{f}(10 / 1$ hexanes $/$ EtOAc $)=0.67$ (silica gel plate $)$.

1-(2-(Phenylethynyl)phenyl)hex-2-yn-1-ol (2b). With 1a ( $90 \mathrm{mg}, 0.44 \mathrm{mmol}$ ) and 1-pentyne
 (65 $\mu \mathrm{L}, 0.65 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 115 mg ( $95 \%$ ) of the title compound as a light yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.74(\mathrm{~m}, 1 \mathrm{H})$, $7.55-7.57(\mathrm{~m}, 3 \mathrm{H}), 7.39(\mathrm{dt}, J=7.5,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.35-7.37(\mathrm{~m}, 3 \mathrm{H}), 7.31(\mathrm{dt}, J=7.6,1.4 \mathrm{~Hz}$, $1 \mathrm{H}), 5.96(\mathrm{~m}, 1 \mathrm{H}), 2.62(\mathrm{~d}, J=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.25(\mathrm{dt}, J=7.1,2.1 \mathrm{~Hz}, 2 \mathrm{H}), 1.56(\mathrm{~m}, 2 \mathrm{H}), 0.98$ $(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.9,132.4,131.6,128.8,128.6,128.4$, $128.1,126.6,122.9,121.3,94.7,87.5,86.7,79.5,63.4,22.0,20.8,13.5$; IR (KBr) $v_{\max } 3554$, 3327, 3052, 3031, 2962, 2929, 2866, 2836, 2280, 2223, 1598, 1494, 1446, 1278, 1183, 1132, 1036, 994, $758 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{20} \mathrm{H}_{18} \mathrm{O}(\mathrm{M}+\mathrm{Cl})$ calcd: 274.1358, found: 274.1360; $R_{f}$ $(1 / 1$ hexanes $/ E t O A c)=0.52($ silica gel plate $)$.

1-(2-(Pent-1-yn-1-yl)phenyl)-3-phenylprop-2-yn-1-ol (2c). With 1b (172 mg, 1.0 mmol )
 and phenylacetylene ( $134 \mu \mathrm{~L}, 1.2 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 265 mg ( $96 \%$ ) of the title compound as a yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.73$ (dd, $J=$ $7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.47$ (m, 2H), 7.45 (dd, $J=7.5,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.34(\mathrm{dt}, J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H})$, $7.30-7.32(\mathrm{~m}, 3 \mathrm{H}), 7.28(\mathrm{dt}, J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.07(\mathrm{~d}, J=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.84(\mathrm{~d}, J=4.7 \mathrm{~Hz}$, $1 \mathrm{H}), 2.47(\mathrm{t}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.86(\mathrm{~m}, 2 \mathrm{H}), 1.08(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 150 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 142.1,132.6,131.7,128.4,128.22,128.18,128.1,126.7,122.6,122.2,96.3,88.3$, $86.4,78.7,63.8,22.1,21.6,13.6$. The spectral data were in accordance with previously published results. ${ }^{67}$

1-(2-(Pent-1-yn-1-yl)phenyl)hex-2-yn-1-ol (2d). With $\mathbf{1 b}$ ( $172 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) and 1-pentyne
 $(150 \mu \mathrm{~L}, 1.5 \mathrm{mmol})$ following the general procedure B. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 223 mg ( $93 \%$ ) of the title compound as a slight yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.67(\mathrm{~d}, J=$ $7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.41(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.31(\mathrm{dt}, J=7.6,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.24(\mathrm{dt}, J=7.6,1.3 \mathrm{~Hz}$, $1 \mathrm{H}), 5.85(\mathrm{~m}, 1 \mathrm{H}), 2.63(\mathrm{~d}, J=5.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.44(\mathrm{t}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.25(\mathrm{dt}, J=7.0,2.0 \mathrm{~Hz}$,
$2 \mathrm{H}), 1.67(\mathrm{~m}, 2 \mathrm{H}), 1.58(\mathrm{~m}, 2 \mathrm{H}), 1.07(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.00(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.7,132.5,128.1,128.0,126.5,122.1,96.0,87.4,79.46,78.1,63.5$, $22.2,22.0,21.6,20.97,13.6,13.5$. The spectral data were in accordance with previously published results. ${ }^{60}$

1-(2-((4-Methoxyphenyl)ethynyl)phenyl)-3-(p-tolyl)prop-2-yn-1-ol (8a). With 7a (450 mg,

1.9 mmol ) following the general procedure B. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 560 mg ( $84 \%$ ) of the title compound as pale yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.78(\mathrm{dd}, J=7.7,1.3 \mathrm{~Hz}$, $1 \mathrm{H}), 7.55(\mathrm{dd}, J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.51(\mathrm{~m}, 2 \mathrm{H}), 7.39(\mathrm{dt}, J=7.6,1.3$ $\mathrm{Hz}, 1 \mathrm{H}), 7.35(\mathrm{~m}, 2 \mathrm{H}), 7.33(\mathrm{dt}, J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.10(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.88(\mathrm{~m}, 2 \mathrm{H})$, $6.15(\mathrm{~d}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.83(\mathrm{~s}, 3 \mathrm{H}), 2.84(\mathrm{~d}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.34(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 150 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 159.9,142.1,138.6,133.1,132.3,131.6,129.0,128.5,128.2,126.8,121.8$, $119.4,114.9,114.1,95.1,87.6,86.6,85.4,63.9,55.3,21.5$. The spectral data were in accordance with previously published results. ${ }^{67}$

3-(p-Tolyl)-1-(2-(p-tolylethynyl)phenyl)prop-2-yn-1-ol (8b). With 7b (392 mg, 1.78 mmol )
 following the general procedure B. Column chromatography (5/1 hexanes/EtOAc) yielded $545 \mathrm{mg}(91 \%)$ of the title compound as pale yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.78(\mathrm{dd}, J=7.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.57(\mathrm{dd}, J=7.6,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.46(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.40(\mathrm{dt}, J=7.6$, $1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.35(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.33(\mathrm{dt}, J=7.5,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.16(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H})$, $7.10(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.15(\mathrm{~d}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.77(\mathrm{~d}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.38(\mathrm{~s}, 3 \mathrm{H}), 2.34$ (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.3,138.8,138.6,132.4,131.7,131.5,129.2,129.0$, 128.7, 128.2, 126.8, 121.7, 119.7, 119.4, 95.3, 87.6, 86.7, 86.0, 63.9, 21.53, 21.46. The spectral data were in accordance with previously published results. ${ }^{66}$

1-(2-([1,1'-Biphenyl]-4-yl-ethynyl)phenyl)-3-(p-tolyl)prop-2-yn-1-ol (8c). With 7c (540
 $\mathrm{mg}, \quad 1.8 \mathrm{mmol}$ ) following the general procedure B . Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $650 \mathrm{mg}(90 \%)$ of the title compound as a light yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.80$ (ddd, $J=7.8,0.9,0.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.60-7.66(\mathrm{~m}, 7 \mathrm{H}), 7.47(\mathrm{~m}, 2 \mathrm{H}), 7.42(\mathrm{dt}, J=$
$7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.34-7.40(\mathrm{~m}, 4 \mathrm{H}), 7.10(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.18(\mathrm{~d}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.78$ $(\mathrm{d}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.34(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.4,141.3,140.2,138.6$,
$132.5,132.0,131.7,129.0,128.9,128.9,128.3,127.7,127.1,127.0,126.8,121.7,121.5$, 119.4, 95.0, 87.6, 87.3, 86.7, 63.9, 21.5; IR (KBr) $v_{\max } 3548,3342,3060,3028,2947,2920$, $2866,2211,1595,1512,1494,1446,1404,1374,1275,1224,1183,1108,1027,1009,967$, 842, 815, 764, $698 \mathrm{~cm}^{-1}$; $\operatorname{HRMS}(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{30} \mathrm{H}_{22} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 421.15629 found: 421.15630; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.43$ (silica gel plate) .

1-(2-(Phenylethynyl)phenyl)-3-(p-tolyl)prop-2-yn-1-ol (8d). With 7d (396 mg, 1.95 mmol )
 following the general procedure B. Column chromatography (5/1 hexanes/EtOAc) yielded 590 mg ( $93 \%$ ) of the title compound as pale yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.80(\mathrm{dd}, J=7.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.57-7.60 $(\mathrm{m}, 3 \mathrm{H}), 7.42(\mathrm{dt}, J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.33-7.37(\mathrm{~m}, 6 \mathrm{H}), 7.10(\mathrm{~d}, J=7.8$ $\mathrm{Hz}, 2 \mathrm{H}), 6.16(\mathrm{~d}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.76(\mathrm{~d}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.34(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( 150 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 142.4,138.6,132.5,131.64,131.58,129.0,128.9,128.6,128.4,128.3,126.8,122.8$, $121.5,119.4,95.0,87.5,86.7,86.6,63.8,21.5$. The spectral data were in accordance with previously published results. ${ }^{66}$

1-(2-((4-Chlorophenyl)ethynyl)phenyl)-3-(p-tolyl)prop-2-yn-1-ol (8e). With 7e (434 mg,

1.8 mmol ) following the general procedure B. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 570 mg ( $89 \%$ ) of the title compound as colorless solid: mp (decomp) $126-131{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 7.78 (dd, $J=7.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.56(\mathrm{dd}, J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.48(\mathrm{~m}, 2 \mathrm{H})$, 7.42 (dt, $J=7.7,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.32-7.36(\mathrm{~m}, 5 \mathrm{H}), 7.10(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.12(\mathrm{~s}, 1 \mathrm{H}), 2.67$ $(\mathrm{s}, 1 \mathrm{H}), 2.34(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.4,138.7,134.6,132.8,132.5,131.6$, 129.1, 129.0, 128.8, 128.3, 126.8, 121.3, 121.2, 119.3, 93.8, 87.6, 87.5, 86.8, 63.8, 21.5; IR (KBr) $v_{\max } 3276,3189,3066,3034,2989,2914,2864,2851,2217,1568,1512,1488,1470$, 1392, 1284, 1219, 1186, 1090, 1027, 1015, 970, 830, 818, $749 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{24} \mathrm{H}_{17} \mathrm{OClNa}(\mathrm{M}+\mathrm{Na})$ calcd: 379.08601 , found: $379.08618 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.45$ (silica gel plate).

3-(p-Tolyl)-1-(2-((4-(trifluoromethyl)phenyl)ethynyl)phenyl)prop-2-yn-1-ol (8f). With 7f
 ( $442 \mathrm{mg}, 1.6 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $564 \mathrm{mg}(90 \%)$ of the title compound as colorless solid: mp (decomp) 133-138 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 600
$\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.81(\mathrm{dd}, J=7.8,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.66(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.60(\mathrm{~m}, 3 \mathrm{H}), 7.45$ $(\mathrm{dt}, J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.36(\mathrm{dt}, J=7.6,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.33(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.10(\mathrm{~d}, J=$ $7.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.13(\mathrm{~d}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.63(\mathrm{~d}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.34(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 150 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.6,138.8,132.7,131.8,131.6,130.2\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=32.7 \mathrm{~Hz}\right), 129.5,129.0$, $128.4,126.8,126.7,125.3\left(\mathrm{q},{ }^{3} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}\right), 121.2-126.7\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=272.3 \mathrm{~Hz}\right), 120.8,119.3$, 93.4, 89.0, 87.5, 86.8, 63.7, 21.5; IR (KBr) $v_{\max } 3273,3180,3066,3028,2992,2926,2875$, $2223,1610,1509,1476,1404,1332,1165,1108,1066,1030,967,842,815,749 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{25} \mathrm{H}_{18} \mathrm{OF}_{3}(\mathrm{M}+\mathrm{H})$ calcd: 391.13043, found: 391.13058; $R_{f}(2 / 1$ hexanes $/$ EtOAc $)=0.47$ (silica gel plate) .

4-((2-(1-Hydroxy-3-(p-tolyl)prop-2-yn-1-yl)phenyl)ethynyl)benzonitrile (8g). With 7g
 ( $392 \mathrm{mg}, 1.7 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 556 mg ( $94 \%$ ) of the title compound as light brown solid: mp (decomp) $151-156{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 600 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.81(\mathrm{dd}, J=7.8,0.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.63(\mathrm{~s}, 4 \mathrm{H}), 7.58(\mathrm{dd}, J=$ $7.6,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.46(\mathrm{dt}, J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.37(\mathrm{dt}, J=7.6,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.32(\mathrm{~m}, 2 \mathrm{H})$, $7.10(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.10(\mathrm{~d}, J=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.62(\mathrm{~d}, J=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.34(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.7,138.9,132.7$, 132.1, 132.0, 131.6, 129.8, 129.1, 128.4, $127.8,126.9,120.5,119.2,118.4,111.8,93.1,91.0,87.4,86.9,63.6,21.5$; IR (KBr) $v_{\max }$ 3279, 3324, 3192, 3060, 3028, 2992, 2914, 2881, 2857, 2226, 1598, 1504, 1473, 1455, 1407, $1290,1183,1120,1105,1027,973,839,818,752 \mathrm{~cm}^{-1}$; HRMS ( $m / z$ ) for $\mathrm{C}_{25} \mathrm{H}_{17} \mathrm{NO}\left(\mathrm{M}^{+}\right)$ calcd: 347.1310 , found: $347.1311 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.38$ (silica gel plate) .

1-(2-Ethynylphenyl)-3-(p-tolyl)prop-2-yn-1-ol (12a). With 2-ethynylbenzaldehyde ( 650 mg ,
 5.0 mmol ) following the general procedure B. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $1.19 \mathrm{~g}(96 \%)$ of the title compound as light brown oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.79(\mathrm{dd}, J=7.8,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.54(\mathrm{dd}, J$ $=7.7,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.43(\mathrm{dt}, J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.36(\mathrm{~m}, 2 \mathrm{H}), 7.31(\mathrm{dt}, J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H})$, $7.11(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.12(\mathrm{~d}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.41(\mathrm{~s}, 1 \mathrm{H}), 2.74(\mathrm{~d}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.35$ (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.9,138.7,133.2,131.6,129.4,129.0,128.3,126.9$, $120.4,119.3,87.3,86.8,82.7,81.0,63.5,21.5$; IR (KBr) $v_{\max } 3536,3291,3063,3031,2914$, 2863, 2226, 1509, 1476, 1443, 1383, 1266, 1207, 1180, 1096, 1021, 967, 821, $761 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{18} \mathrm{H}_{14} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 269.09369 , found: 269.09360; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.43$ (silica gel plate) .

Preparation of 1-(2-Ethynylphenyl)hex-2-yn-1-ol (12b). With 2-ethynylbenzaldehyde (1.3
 $\mathrm{g}, 10.0 \mathrm{mmol}$ ) and 1-pentyne ( $1.5 \mathrm{~mL}, 15 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 1.8 g ( $91 \%$ ) of the title compound as light brown oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.72(\mathrm{dd}, J=7.8,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{dd}, J=7.6,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.40(\mathrm{dt}, J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H})$, $7.29(\mathrm{dt}, J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.88(\mathrm{~s}, 1 \mathrm{H}), 3.37(\mathrm{~s}, 1 \mathrm{H}), 2.49(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.25(\mathrm{dt}, J=$ $7.1,2.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.56(\mathrm{~m}, 2 \mathrm{H}), 0.99(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 143.5$, 133.1, 129.4, 128.1, 126.7, 120.3, 87.7, 82.5, 81.1, 79.2, 63.1, 22.0, 20.8, 13.5; IR (KBr) $v_{\max }$ 3545, 3288, 3069, 3028, 2962, 2932, 2905, 2869, 2836, 2283, 2223, 1482, 1452, 1383, 1335, 1275, 1201, 1132, 1036, 991, $758 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{O}\left(\mathrm{M}^{+}\right)$calcd: 198.1045, found: 198.1044; $R_{f}(5 / 1$ hexanes $/ E t O A c)=0.25($ silica gel plate $)$.

1-(2-((4-Bromophenyl)ethynyl)phenyl)-3-(p-tolyl)prop-2-yn-1-ol (8h). 12a (736 mg, 4.6
 mmol ), 1-bromo-4-iodobenzene ( $1.1 \mathrm{~g}, 3.6 \mathrm{mmol}$ ), $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(104 \mathrm{mg}$, $0.15 \mathrm{mmol}, 5 \mathrm{~mol} \%$ ) and $\mathrm{CuI}(57 \mathrm{mg}, 0.3 \mathrm{mmol}, 10 \mathrm{~mol} \%)$ were dissolved in a flask in diisopropylamine ( 30 mL ) and the reaction was stirred 16 h at $50^{\circ} \mathrm{C}$. The reaction mixture was cooled down, filtered off with celite/silica and washed with diethyl ether. The organic fraction was concentrated on vacuum evaporator. Column chromatography of the residue on silica gel ( $2 / 1$ hexanes/EtOAc) yielded 1.15 g ( $95 \%$ ) of the title compound as brown solid: mp (decomp) 127-132 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\mathrm{CDCl}_{3}$ ) $\delta 7.79(\mathrm{dd}, J=7.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.56(\mathrm{dd}, J=7.5,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.47-7.50(\mathrm{~m}, 2 \mathrm{H})$, 7.40-7.44 (m, 3H), 7.32-7.36 (m, 3H), 7.09 (d, J=7.9 Hz, 2H), 6.12 (d, J=4.9 Hz, 1H), 2.74 $(\mathrm{d}, J=4.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.34(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.4,138.7,133.0,132.5$, 131.7, 131.6, 129.1, 129.0, 128.3, 126.8, 122.9, 121.8, 121.1, 119.3, 93.9, 87.8, 87.5, 86.7, 63.7, 21.5; IR (KBr) $v_{\max } 3548,3342,3060,3022,2914,28632223,1506,1488,1446,1389$, 1314, 1269, 1189, 1099, 1069, 1024, 1012, 961, 821, $758,525 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{24} \mathrm{H}_{17} \mathrm{OBrNa}(\mathrm{M}+\mathrm{Na})$ calcd: 423.03550, found: 423.03553; $R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=0.46$ (silica gel plate).

1-(2-(Naphthalen-2-ylethynyl)phenyl)hex-2-yn-1-ol (15a). With 12b and 2-
 bromonaphthalene ( $397 \mathrm{mg}, 2.0 \mathrm{mmol}$ ) following the general procedure A. Column chromatography ( $3 / 1$ hexanes/EtOAc) yielded 434 mg ( $57 \%$ ) of the title compound as a light brown oil: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $8.08(\mathrm{~s}, 1 \mathrm{H}), 7.82(\mathrm{~m}, 3 \mathrm{H}), 7.75(\mathrm{dd}, J=7.8,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.59(\mathrm{~m}, 2 \mathrm{H})$, $7.50(\mathrm{~m}, 2 \mathrm{H}), 7.41(\mathrm{dt}, J=7.5,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.34(\mathrm{dt}, J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.02(\mathrm{~s}, 1 \mathrm{H}), 2.64$ $(\mathrm{s}, 1 \mathrm{H}), 2.27(\mathrm{dt}, J=7.1,2.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.56(\mathrm{~m}, 2 \mathrm{H}), 0.99(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.9,135.2133 .0,132.4,131.5,128.9,128.2,128.12,128.08,127.80$, $127.78,126.8,126.7,126.6,121.4,120.2,95.20,87.56,87.01,79.49,63.5,22.0,20.9,13.5$; IR (KBr) $v_{\max } 3368,3055,2962,2929,2869,2220,1601,1503,1482,1467,1434,1272$, 1183, 1135, 1096, 1030, 997, 955, 896, 857, 815, 749, $\mathrm{cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{24} \mathrm{H}_{20} \mathrm{O}\left(\mathrm{M}^{+}\right)$ calcd: 324.1514 found: $324.1515 ; R_{f}(2 / 1$ hexanes/EtOAc $)=0.44$ (silica gel plate) .

1-(2-(Phenanthren-9-ylethynyl)phenyl)hex-2-yn-1-ol (15b). With 12b and 2-bromo-
 phenanthrene ( $40 \mathrm{mg}, 0.2 \mathrm{mmol}$ ) following the general procedure A . Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 53 mg ( $71 \%$ ) of the title compound as a light brown oil: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 8.72 (dd, $J=5.9,3.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.67(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 8.60(\mathrm{~m}, 1 \mathrm{H}), 8.12$ (s, 1H), $7.88(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.81(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.67-7.74(\mathrm{~m}$, $4 \mathrm{H}), 7.62(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.45(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.38(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.12(\mathrm{~s}, 1 \mathrm{H})$, $2.64(\mathrm{~s}, 1 \mathrm{H}), 2.28(\mathrm{t}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 1.58(\mathrm{~m}, 2 \mathrm{H}), 0.97(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.9,132.7,132.2,131.2,131.0,130.4,130.1,129.0,128.6,128.2,127.6$, 127.2, 127.1, 127.01, 126.97, 126.6, 122.8, 122.6, 121.5, 119.4, 93.2, 91.1, 87.7, 79.6, 63.5, 22.0, 20.9, 13.5; IR (KBr) $v_{\max } 3545,3354,3058,2962,2932,2902,2869,2833,2280,2214$, $1598,1479,1449,1377,1278,1245,1180,1126,1096,1039,1000,955,893,755,722 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{28} \mathrm{H}_{22} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 397.15629 found: 397.15631; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.42($ silica gel plate $)$.

1-(2-(Anthracen-9-ylethynyl)phenyl)hex-2-yn-1-ol (15c). With 12b and 2-
 bromonaphthalene ( $257 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) following the general procedure A . Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $153 \mathrm{mg}(41 \%)$ of the title compound as a light brown oil: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.70$ (dd, $J=8.7,0.9 \mathrm{~Hz}, 2 \mathrm{H}), 8.45(\mathrm{~s}, 1 \mathrm{H}), 8.01(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.86(\mathrm{dd}, J$ $=7.7,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.81(\mathrm{dd}, J=7.5,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.62(\mathrm{~m}, 2 \mathrm{H}), 7.52(\mathrm{~m}, 2 \mathrm{H}), 7.47(\mathrm{dt}, J=$
$7.5,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.43(\mathrm{dt}, J=7.5,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.21(\mathrm{~s}, 1 \mathrm{H}), 2.64(\mathrm{~s}, 1 \mathrm{H}), 2.27(\mathrm{dt}, J=7.1,2.1$ $\mathrm{Hz}, 2 \mathrm{H}$ ), $1.54(\mathrm{~m}, 2 \mathrm{H}), 0.95(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.6,132.8$, 132.7, 131.1, 129.0, 128.7, 128.2, 128.1, 126.8, 126.7, 126.5, 125.7, 121.8, 97.9, 91.7, 87.8, 79.8, 63.5, 22.0, 20.9, 13.5; IR (KBr) $v_{\max } 3304,3053,2962,2932,2902,2870,2224,2187$, $1621,1597,1483,1436,1412,1362,1259,1183,1096,1034,1013,999,888,845,756,739$, $693 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / z)$ for $\mathrm{C}_{28} \mathrm{H}_{22} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 397.25629 found: 397.25602; $R_{f}(2 / 1$ hexanes $/$ EtOAc $)=0.41$ (silica gel plate $)$.

1-(2-(Pyren-2-ylethynyl)phenyl)hex-2-yn-1-ol (15d). With 12b and 2-bromopyrene (281
 $\mathrm{mg}, \quad 1.0 \mathrm{mmol}$ ) following the general procedure A . Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $246 \mathrm{mg}(62 \%)$ of the title compound as a light brown oil: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.68(\mathrm{~d}$, $J=9.1 \mathrm{~Hz}, 1 \mathrm{H}), 8.16-8.22(\mathrm{~m}, 4 \mathrm{H}), 8.00-8.12(\mathrm{~m}, 4 \mathrm{H}), 7.83(\mathrm{dd}, J=$ $7.7,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.74(\mathrm{dd}, J=7.5,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.46(\mathrm{dt}, J=7.5,1.5$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 7.40 (dt, $J=7.5,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.19(\mathrm{~s}, 1 \mathrm{H}), 2.73(\mathrm{~s}, 1 \mathrm{H}), 2.30(\mathrm{dt}, J=7.1,2.1 \mathrm{~Hz}$, $2 \mathrm{H}), 1.57(\mathrm{~m}, 2 \mathrm{H}), 0.99(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.8,132.6$, $131.9,131.4,131.2,131.0,129.7,128.9,128.5,128.2,128.2,127.2,126.6,126.2,125.7$, $125.6,125.4,124.5,124.4,124.2,121.7,117.4,94.1,92.3,87.7,79.7,63.6,22.0,20.9,13.5$; IR (KBr) $v_{\max } 3545,3291,3044,2961,2932,2902,2871,2206,1927,1600,1508,1486$, 1448, 1435, 1379, 1275, 1242, 1184, 1133, 1033, 998, 848, 757, $717 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{30} \mathrm{H}_{22} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 421.15629 found: $421.15618 ; R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=0.40$ (silica gel plate).

### 6.2.2. Cyclotrimerization reaction for 4,9 and 16

$\mathbf{C l}^{\mathbf{1}}$ : General procedure for cyclotrimerization with Wilkinson's catalyst $\mathbf{R h C l}\left(\mathbf{P P h}_{3}\right)_{3}$ under thermal conditions (Preparation of 4). ${ }^{68} \mathrm{~A}$ dry flask was charged with starting diynol ( 0.5 mmol ), alkyne $3(1.0 \mathrm{mmol})$ and dissolved under argon atmosphere in toluene ( 15 mL ). After addition of Wilkinson's catalyst ( $46 \mathrm{mg}, 0.05 \mathrm{mmol}, 10 \mathrm{~mol} \%$ ) the reaction mixture was heated up to $90^{\circ} \mathrm{C}$ for 16 h . The reaction mixture was cooled down to room temperature and the solvent was evaporated under reduced pressure. Column chromatography of the residue on silica gel yielded products.

C: General procedure for cyclotrimerization with Wilkinson's catalyst $\mathbf{~} \mathbf{~ h C l}\left(\mathbf{P P h}_{3}\right)_{\mathbf{3}}$ under MW irradiation (Preparation of 9 and 16). A dry microwave vial was charged with starting diynol ( 0.5 mmol ), 3-hexyne 3b ( $115 \mu \mathrm{~L}, 1.0 \mathrm{mmol}$ ) and dissolved under argon atmosphere in THF ( 15 mL ). After addition of Wilkinson's catalyst ( $23 \mathrm{mg}, 0.015 \mathrm{mmol}, 3$ $\mathrm{mol} \%$ ) and the additive $\mathrm{Ag}_{2} \mathrm{CO}_{3}(14 \mathrm{mg}, 0.05 \mathrm{mmol}, 10 \mathrm{~mol} \%)$ the reaction mixture was sealed and heated up to $180^{\circ} \mathrm{C}$ for 1.5 h in a microwave reactor. The reaction mixture was cooled down to room temperature and the solvent was evaporated under reduced pressure. Column chromatography of the residue on silica gel yielded products.

2,3-Dimethyl-1,4-diphenyl-9H-fluoren-9-ol (4aa). With $\mathbf{2 a}(30.8 \mathrm{mg}, 0.1 \mathrm{mmol})$ and 2-
 butyne $3 \mathrm{a}(14.8 \mu \mathrm{~L}, 0.2 \mathrm{mmol})$ following the general procedure $\mathrm{C}^{1}$. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 9 mg ( $25 \%$ ) of the title compound as a slightly yellow solid: mp (decomp) 132-137 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.48-7.56(\mathrm{~m}, 5 \mathrm{H}), 7.39-7.46(\mathrm{~m}, 4 \mathrm{H}), 7.30-7.34(\mathrm{~m}, 2 \mathrm{H}), 7.11(\mathrm{dt}, J=$ $7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{dt}, J=7.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.10(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.53(\mathrm{~d}, J=5.2 \mathrm{~Hz}$, $1 \mathrm{H}), 2.10(\mathrm{~s}, 3 \mathrm{H}), 2.08(\mathrm{~s}, 3 \mathrm{H}), 1.63(\mathrm{~d}, J=5.2 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $145.3,141.2,140.6,140.5,139.6,138.7,136.5,136.2,135.4,134.2,130.0,129.3,129.2$, $129.02,128.96,128.5,128.42,128.41,127.44,127.37,126.8,124.7,122.5,73.9,17.40$, 17.38; IR (KBr) $v_{\max } 3563,3559,3052,3025,2926,2863,1601,1571,1497,1440,1359$, 1296, 1180, 1099, 1060, 1027, 740, $701 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{27} \mathrm{H}_{22} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 385.15629 , found: $385.15633 ; R_{f}(5 / 1$ hexanes $/ \mathrm{EtOAc})=0.46$ (silica gel plate) .

2,3-Diethyl-1,4-diphenyl-9H-fluoren-9-ol (4ab). With 2a ( $30.8 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and 3-hexyne
 3b $(22.7 \mu \mathrm{~L}, 0.2 \mathrm{mmol})$ following the general procedure $\mathrm{C}^{1}$. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 29 mg ( $76 \%$ ) of the title compound as a slightly yellow solid: mp (decomp) $146-151{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.49-7.55(\mathrm{~m}, 5 \mathrm{H}), 7.44-7.48(\mathrm{~m}, 2 \mathrm{H}), 7.35-7.45(\mathrm{~m}, 4 \mathrm{H}), 7.09(\mathrm{dt}, J=$ $7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.93(\mathrm{dt}, J=7.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.94(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.44(\mathrm{~s}, 1 \mathrm{H}), 2.64$ (dq, $J=15.0,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.59(\mathrm{dq}, J=15.0,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.56(\mathrm{dq}, J=15.0,7.5 \mathrm{~Hz}, 1 \mathrm{H})$, $2.54(\mathrm{dq}, J=15.0,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.62(\mathrm{~s}, 1 \mathrm{H}), 1.03(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.02(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 145.2,141.6,141.4,140.5,140.3,139.8,139.4,139.1,136.8$, $135.8,129.9,129.5,129.4,129.0,128.82,128.80,128.7,128.4,128.3,127.44,127.35,126.7$, $124.6,122.5,73.9,22.9,22.8,15.82,15.78$; IR (KBr) $v_{\max } 3566,3058,3022,2965,2935$, $2872,1604,1470,1443,1377,1227,1072,1030,761,698 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{29} \mathrm{H}_{26} \mathrm{ONa}$ $(\mathrm{M}+\mathrm{Na})$ calcd: 413.18759 , found: $413.18784 ; R_{f}(5 / 1$ hexanes $/ \mathrm{EtOAc})=0.48$ (silica gel plate).

1,4-Diphenyl-2,3-dipropyl-9H-fluoren-9-ol (4ac). With $\mathbf{4 a}(30.8 \mathrm{mg}, 0.1 \mathrm{mmol})$ and 4 -
 octyne $3 \mathbf{c}(28.6 \mu \mathrm{~L}, 0.2 \mathrm{mmol})$ following the general procedure $\mathrm{C}^{1}$. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 11 mg ( $26 \%$ ) of the title compound as a slightly yellow solid: mp (decomp) 153-158 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.48-7.54(\mathrm{~m}, 5 \mathrm{H}), 7.43-7.46(\mathrm{~m}, 2 \mathrm{H}), 7.39-7.41(\mathrm{~m}, 2 \mathrm{H}), 7.33-7.37(\mathrm{~m}$, $2 \mathrm{H}), 7.09(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.93(\mathrm{dt}, J=7.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.95(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H})$, $5.45(\mathrm{~d}, J=4.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.49-2.54(\mathrm{~m}, 1 \mathrm{H}), 2.42-2.46(\mathrm{~m}, 2 \mathrm{H}), 2.37-2.41(\mathrm{~m}, 1 \mathrm{H}), 1.62(\mathrm{~d}, J$ $=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.40-1.47(\mathrm{~m}, 4 \mathrm{H}), 0.774(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 0.772(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 145.2,141.4,140.6,140.5,140.3,139.4,139.0,138.7,136.8$, $135.7,129.8,129.4,129.0,128.8,128.74,128.69,128.4,128.3,127.4,127.3,126.7,124.6$, $122.5,73.9,32.4,32.3,24.9,24.8,14.8,14.7$; IR (KBr) $v_{\max } 3569,3554,3446,3055,2956$, 2932, 2866, 1601, 1574, 1464, 1440, 1380, 1024, 749, $704 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{31} \mathrm{H}_{30} \mathrm{ONa}$ $(\mathrm{M}+\mathrm{Na})$ calcd: 441.21889, found: 441.21886; $R_{f}(5 / 1$ hexanes/EtOAc) $=0.50$ (silica gel plate).
(9-Hydroxy-1,4-diphenyl-9H-fluorene-2,3-diyl)dimethanol (4ad). With 2a (30.8 mg, 0.1
 $\mathrm{mmol})$ and 2-butyne-1,4-diol $\mathbf{3 d}(8.6 \mathrm{mg}, 0.1 \mathrm{mmol})$ following the general procedure $\mathrm{C}^{1}$. Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 34 mg ( $86 \%$ ) of the title compound as a slightly yellow solid: mp (decomp) 139$144{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.44-7.53(\mathrm{~m}, 8 \mathrm{H}), 7.41(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.34(\mathrm{~m}$, $2 \mathrm{H}), 7.16(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.98(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.11(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.56(\mathrm{~s}, 2 \mathrm{H})$, $4.57(\mathrm{~m}, 4 \mathrm{H}), 3.14(\mathrm{~s}, 2 \mathrm{H}, \mathrm{OH}), 1.83(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 145.5,143.8$, 139.8, 139.7, 139.5, 138.6, 138.1, 137.9, 137.4, 137.3, 130.0, 129.3, 129.2, 129.07, 129.06, 128.96, 128.63, 128.59, 128.4, 127.93, 127.90, 127.6, 124.8, 123.0, 73.8, 60.0, 59.9; IR (KBr) $v_{\max } 3569,3327,3055,3022,2926,2887,1503,1473,1443,1347,1296,1111,1009,917$, 860, 752, $707 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{27} \mathrm{H}_{22} \mathrm{O}_{3} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 417.14612, found: 417.14604; $R_{f}(1 / 1$ hexanes $/ E t O A c)=0.49$ (silica gel plate $)$.
(2,3-Dimethyl-1,4-diphenyl-9H-fluoren-9-yl)oxy-tert-butyl-dimethylsilane (4aa-TBS).


With 2a-TBS ( $42.3 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and 2-butyne 3a ( $14.8 \mu \mathrm{~L}, 0.2 \mathrm{mmol}$ ) following the general procedure $\mathrm{C}^{1}$. Column chromatography ( $30 / 1$ hexanes/EtOAc) yielded $9 \mathrm{mg}(19 \%)$ of the title compound as a pale yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.52-7.55(\mathrm{~m}, 2 \mathrm{H}), 7.40-7.50(\mathrm{~m}, 5 \mathrm{H}), 7.33-7.38(\mathrm{~m}, 3 \mathrm{H})$, $7.28-7.30(\mathrm{~m}, 1 \mathrm{H}), 7.05(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{dt}, J=7.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.05(\mathrm{~d}, J=$ $7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.80(\mathrm{~s}, 1 \mathrm{H}), 2.09(\mathrm{~s}, 3 \mathrm{H}), 2.07(\mathrm{~s}, 3 \mathrm{H}), 0.63(\mathrm{~s}, 9 \mathrm{H}),-0.32(\mathrm{~s}, 3 \mathrm{H}),-0.66(\mathrm{~s}, 3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 145.7,141.1,141.0,140.9,140.0,139.4,136.19,136.17$, 135.6, 134.4, 131.2, 129.4, 129.3, 129.1, 128.9, 128.8, 128.4, 128.3, 128.0, 127.3, 126.8, $126.2,125.8,122.4,74.9,26.0,18.1,17.8,17.6,-3.2,-3.8$; IR (KBr) $v_{\max } 3058,3028,2956$, 2929, 2893, 2854, 1718, 1601, 1473, 1437, 1257, 1105, 1072, $833 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{33} \mathrm{H}_{36} \mathrm{OSiNa}(\mathrm{M}+\mathrm{Na})$ calcd: 499.24276, found: 499.24283; $R_{f}(20 / 1$ hexanes $/ \mathrm{EtOAc})=0.49$ (silica gel plate).
(2,3-Diethyl-1,4-diphenyl-9H-fluoren-9-yl)oxy-tert-butyl-dimethylsilane (4ab-TBS). With $\underset{\mathrm{Ph}}{\text { отвдмs }} \mathbf{2 a - T B S}(42.3 \mathrm{mg}, 0.1 \mathrm{mmol})$ and 3-hexyne 3b $(22.7 \mu \mathrm{~L}, 0.2 \mathrm{mmol})$ following the general procedure $\mathrm{C}^{1}$. Column chromatography (30/1 hexanes/EtOAc) yielded 36 mg ( $71 \%$ ) of the title compound as a pale yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.50-7.56(\mathrm{~m}, 4 \mathrm{H}), 7.44-7.47(\mathrm{~m}, 2 \mathrm{H}), 7.36-7.40$ $(\mathrm{m}, 4 \mathrm{H}), 7.30-7.32(\mathrm{~m}, 1 \mathrm{H}), 7.04(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.91(\mathrm{dt}, J=7.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.92$ (d, $J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.66(\mathrm{~s}, 1 \mathrm{H}), 2.69(\mathrm{dq}, J=15.0,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.61(\mathrm{dq}, J=15.0,7.5 \mathrm{~Hz}$,
$1 \mathrm{H}), 2.57(\mathrm{dq}, J=15.0,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.50(\mathrm{dq}, J=15.0,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.03(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H})$, $0.98(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.69(\mathrm{~s}, 9 \mathrm{H}),-0.29(\mathrm{~s}, 3 \mathrm{H}),-0.66(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 150 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 145.7,141.6,141.4,141.0,140.6,140.2,139.81,139.75,136.5,135.9,130.8,129.7$, 129.4, 129.2, 128.9, 128.7, 128.3, 128.2, 127.8, 127.3, 126.8, 126.2, 125.8, 122.4, 74.9, 26.1, 22.9, 22.8, 18.2, 15.9, 15.7, -3.2, -3.8; IR (KBr) $v_{\max } 3055,3022,2968,2929,2851,1718$, 1601, 1559, 1473, 1440, 1257, 1105, 1078, $836 \mathrm{~cm}^{-1} ; \operatorname{HRMS}(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{35} \mathrm{H}_{41} \mathrm{OSi}(\mathrm{M}+\mathrm{H})$ calcd: 505.2927 , found: $505.2923 ; R_{f}(20 / 1$ hexanes $/ E t O A c)=0.51$ (silica gel plate).
(1,4-Diphenyl-2,3-dipropyl-9H-fluoren-9-yl)oxy-tert-butyl-dimethylsilane (4ac-TBS).


With 2a-TBS ( $42.3 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and 4-octyne 3c ( $28.6 \mu \mathrm{~L}, 0.2 \mathrm{mmol}$ ) following the general procedure $\mathrm{C}^{1}$. Column chromatography (30/1 hexanes/EtOAc) yielded 30 mg ( $56 \%$ ) of the title compound as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.47-7.54(\mathrm{~m}, 4 \mathrm{H}), 7.42-7.45(\mathrm{~m}, 2 \mathrm{H}), 7.33-7.39(\mathrm{~m}, 4 \mathrm{H})$, $7.28(\mathrm{~m}, 1 \mathrm{H}), 7.03(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.90(\mathrm{dt}, J=7.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.91(\mathrm{~d}, J=7.6 \mathrm{~Hz}$, $1 \mathrm{H}), 5.65(\mathrm{~s}, 1 \mathrm{H}), 2.54-2.59(\mathrm{~m}, 1 \mathrm{H}), 2.40-2.50(\mathrm{~m}, 2 \mathrm{H}), 2.33-2.38(\mathrm{~m}, 1 \mathrm{H}), 1.32-1.46(\mathrm{~m}$, $4 \mathrm{H}), 0.77(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 0.75(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 0.67(\mathrm{~s}, 9 \mathrm{H}),-0.31(\mathrm{~s}, 3 \mathrm{H}),-0.68(\mathrm{~s}$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 145.7,141.3,141.1,140.6,140.5,139.83,139.81,139.1$, 136.5, 135.8, 130.8, 129.7, 129.3, 129.2, 128.8, 128.6, 128.3, 128.1, 127.8, 127.2, 126.7, $126.2,125.8,122.4,74.9,32.4,32.2,26.1,24.9,24.7,18.2,14.7,14.6,-3.2,-3.8$; IR (KBr) $v_{\max } 3058,3022,2953,2932,2854,1598,1500,1476,1440,1317,1257,1108,1063,833 \mathrm{~cm}^{-}$ ${ }^{1}$; HRMS $(m / z)$ for $\mathrm{C}_{37} \mathrm{H}_{44} \mathrm{OSiNa}(\mathrm{M}+\mathrm{Na})$ calcd: 555.30536, found: 555.30535; $R_{f}(20 / 1$ hexanes $/ E t O A c)=0.53($ silica gel plate $)$.
[9-((tert-Butyldimethylsilyl)oxy]-1,4-diphenyl-9H-fluorene-2,3-diyl)dimethanol
(4ad-
 TBS). With 2a-TBS ( $84 \mathrm{mg}, 0.2 \mathrm{mmol}$ ) and 2-butyne-1,4-diol 3d ( 26 mg , 0.3 mmol ) following the general procedure $\mathrm{C}^{1}$. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $83 \mathrm{mg}(82 \%)$ of the title compound as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.51-7.57(\mathrm{~m}, 4 \mathrm{H}), 7.44-7.47(\mathrm{~m}, 3 \mathrm{H}), 7.35-7.42$ (m, 4H), 7.12 (dt, $J=7.4,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.96(\mathrm{dt}, J=7.4,0.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.07(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H})$, $5.84(\mathrm{~s}, 1 \mathrm{H}), 4.67(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.62(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.59(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H})$, $4.49(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.00(\mathrm{bs}, 2 \mathrm{H}), 0.63(\mathrm{~s}, 9 \mathrm{H}),-0.32(\mathrm{~s}, 3 \mathrm{H}),-0.65(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 145.8,143.8,140.5,140.1,139.7,138.9,138.3,138.2,137.7,137.1$, $131.0,129.4,129.3,129.1,129.0,128.8,128.5,128.4,128.2,127.9,127.4,127.0,125.9$, $123.0,74.8,60.2,60.1,25.9,18.1,-3.2,-3.8$; IR (KBr) $v_{\max } 3318,3052,3022,2950,2932$,

2893, 2854, 1601, 1476, 1446, 1317, 1254, 1213, 1135, 1111, 1072, 1009, 833, 776, $701 \mathrm{~cm}^{-}$ ${ }^{1}$; HRMS $(m / z)$ for $\mathrm{C}_{33} \mathrm{H}_{36} \mathrm{O}_{3} \mathrm{SiNa}(\mathrm{M}+\mathrm{Na})$ calcd: 531.23259, found: 531.23254; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.21($ silica gel plate $)$.

2,3-Diethyl-1,4-diphenyl-9H-fluoren-9-ol (4ab). With 2a ( $30.8 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and 3-hexyne
 3b $(22.7 \mu \mathrm{~L}, 0.2 \mathrm{mmol})$ following the general procedure $\mathrm{C}^{1}$. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 29 mg ( $76 \%$ ) of the title compound as a slightly yellow solid: mp (decomp) $146-151{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 600 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.49-7.55(\mathrm{~m}, 5 \mathrm{H}), 7.44-7.48(\mathrm{~m}, 2 \mathrm{H}), 7.35-7.45(\mathrm{~m}, 4 \mathrm{H}), 7.09$ (dt, $J=7.4$, $1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.93(\mathrm{dt}, J=7.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.94(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.44(\mathrm{~s}, 1 \mathrm{H}), 2.64(\mathrm{dq}, J=$ $15.0,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.59(\mathrm{dq}, J=15.0,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.56(\mathrm{dq}, J=15.0,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.54(\mathrm{dq}, J$ $=15.0,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.62(\mathrm{~s}, 1 \mathrm{H}), 1.03(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.02(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 145.2,141.6,141.4,140.5,140.3,139.8,139.4,139.1,136.8,135.8$, $129.9,129.5,129.4,129.0,128.82,128.80,128.7,128.4,128.3,127.44,127.35,126.7,124.6$, 122.5, 73.9, 22.9, 22.8, 15.82, 15.78; IR (KBr) $v_{\max } 3566,3058,3022,2965,2935,2872$, 1604, 1470, 1443, 1377, 1227, 1072, 1030, 761, $698 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{29} \mathrm{H}_{26} \mathrm{ONa}(\mathrm{M}+$ $\mathrm{Na})$ calcd: 413.18759 , found: $413.18784 ; R_{f}(5 / 1$ hexanes $/ E t O A c)=0.48$ (silica gel plate).

2,3-Diethyl-4-phenyl-1-propyl-9H-fluoren-9-ol (4bb). With $\mathbf{2 b}$ ( $100 \mathrm{mg}, 0.4 \mathrm{mmol}$ ) and 3${ }^{\mathrm{OH}}{ }^{n-\mathrm{Pr}}$ hexyne $\mathbf{3 b}(90 \mu \mathrm{~L}, 0.8 \mathrm{mmol})$ following the general procedure $\mathrm{C}^{1}$. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 65 mg ( $48 \%$ ) of the title compound as a light brown solid: mp (decomp) $151-156{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 600 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.53(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.45-7.51(\mathrm{~m}, 3 \mathrm{H}), 7.31(\mathrm{~m}, 1 \mathrm{H}), 7.28(\mathrm{~m}, 1 \mathrm{H}), 7.12$ $(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}, 1 \mathrm{H})$, ), $6.92(\mathrm{dt}, J=7.4,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.88(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.64(\mathrm{~d}, J=$ $10.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.10-3.15(\mathrm{~m}, 1 \mathrm{H}), 2.81-2.84(\mathrm{~m}, 1 \mathrm{H}), 2.78(\mathrm{~m}, 2 \mathrm{H}), 2.51(\mathrm{~m}, 2 \mathrm{H}), 1.71-1.78$ $(\mathrm{m}, 2 \mathrm{H}), 1.67(\mathrm{~d}, J=10.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.26(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.14(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 0.99(\mathrm{t}, J$ $=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 146.0,141.8,141.0,140.6,140.5,140.3,138.9$, 135.9, 135.3, 129.6, 129.5, 128.7, 128.5, 127.2, 126.6, 124.5, 122.4, 74.6, 31.8, 25.0, 22.8, $21.8,16.0,15.8,15.1$; IR (KBr) $v_{\max } 3536,3392,3055,3019,2962,2932,2869,1601,1494$, 1464, 1437, 1377, 1281, 1189, 1066, 1039, 758, $704 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{29} \mathrm{H}_{26} \mathrm{ONa}(\mathrm{M}+$ Na ) calcd: 379.20324 , found: $379.20331 ; R_{f}(5 / 1$ hexanes $/ \mathrm{EtOAc})=0.56$ (silica gel plate) .

2,3-Diethyl-1-phenyl-4-propyl-9H-fluoren-9-ol (4cb). With $\mathbf{2 c}$ ( $137 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and 3-
 hexyne $\mathbf{3 b}(120 \mu \mathrm{~L}, 1.0 \mathrm{mmol})$ following the general procedure $\mathrm{C}^{1}$. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 108 mg ( $62 \%$ ) of the title compound as a light brown solid: mp (decomp) 137-142 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR (600 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.70(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.47-7.52(\mathrm{~m}, 3 \mathrm{H}), 7.43(\mathrm{dt}, J=7.5,1.4 \mathrm{~Hz}, 1 \mathrm{H})$, 7.39-7.42 (m, 1H), 7.34-7.39 (m, 2H), 7.25 (dt, $J=7.4,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.39$ (d, $J=4.9 \mathrm{~Hz}, 1 \mathrm{H})$, $3.03(\mathrm{~m}, 2 \mathrm{H}), 2.81(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.54-2.60(\mathrm{~m}, 1 \mathrm{H}), 2.42-2.48(\mathrm{~m}, 1 \mathrm{H}), 1.73-1.80(\mathrm{~m}$, $2 \mathrm{H}), 1.60(\mathrm{~d}, J=4.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.27(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}), 1.20(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 1.00(\mathrm{t}, J=7.5$ $\mathrm{Hz}, 3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 145.5,142.1,141.4,140.8,139.9,139.6,137.5$, $136.3,135.8,130.0,129.0,128.9,128.2,127.3,126.6,125.0,122.8,73.8,31.7,23.3,23.0$, $21.9,16.0,15.8,14.7$; IR (KBr) $v_{\max } 3566,3391,3055,3022,2962,2926,2869,1607,1464$, 1443, 1374, 1260, 1198, 1075, 1033, 761, $704 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{26} \mathrm{H}_{28} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 379.20324 , found: $379.20328 ; R_{f}(5 / 1$ hexanes $/ E t O A c)=0.51$ (silica gel plate).

2,3-Diethyl-1,4-dipropyl-9H-fluoren-9-ol (4db). With 2d (120 mg, 0.5 mmol ) and 3-hexyne
 3b $(120 \mu \mathrm{~L}, 1.0 \mathrm{mmol})$ following the general procedure $\mathrm{C}^{1}$. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 61 mg ( $38 \%$ ) of the title compound as a slightly yellow solid: mp $122-127{ }^{\circ} \mathrm{C}$ (decomp); ${ }^{1} \mathrm{H}$ NMR ( 600 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.65(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.61(\mathrm{~m}, 1 \mathrm{H}), 7.38(\mathrm{dt}, J=7.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.28(\mathrm{dt}$, $J=7.4,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.57(\mathrm{~d}, J=10.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.04-3.09(\mathrm{~m}, 1 \mathrm{H}), 2.95(\mathrm{~m}, 2 \mathrm{H}), 2.71-2.78$ (m, 5H), 1.65-1.73 (m, 4H), $1.64(\mathrm{~d}, J=10.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.23(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.22(\mathrm{t}, J=7.5$ $\mathrm{Hz}, 3 \mathrm{H}), 1.17(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 1.11(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR $\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $146.4,141.8,141.7,140.9,140.5,137.2,136.0,134.6,129.0,126.5,124.9,122.7,74.5,31.8$, $31.6,25.0,23.3,21.9,21.8,16.0,15.9,15.1,14.6$; IR (KBr) $v_{\max } 3518,3392,3069,3043$, 2959, 2926,2896, 2869, 1607, 1464, 1431, 1377, 1302, 1245, 1192, 1069, 1003, 764, $737 \mathrm{~cm}^{-}$ ${ }^{1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 345.21889, found: 345.21890; $R_{f}(5 / 1$ hexanes $/$ EtOAc) $=0.63$ (silica gel plate) .

2,3-Diethyl-4-(4-methoxyphenyl)-1-(p-tolyl)-9H-fluoren-9-ol (9a). With 8a (176 mg, 0.5
 mmol ) following the general procedure C. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $103 \mathrm{mg}(47 \%)$ of the title compound as a brown oil: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.40(\mathrm{~m}, 1 \mathrm{H}), 7.25-7.36(\mathrm{~m}, 6 \mathrm{H}), 7.10(\mathrm{dt}, J=$ $7.4,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.06-7.08(\mathrm{~m}, 2 \mathrm{H}), 6.97(\mathrm{dt}, J=7.4,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.07(\mathrm{~d}, J$ $=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.45(\mathrm{~d}, J=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}), 2.60-2.66(\mathrm{~m}, 1 \mathrm{H}), 2.55$ $(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.50-2.53(\mathrm{~m}, 1 \mathrm{H}), 2.46(\mathrm{~s}, 3 \mathrm{H}), 1.69(\mathrm{~d}, J=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.03(\mathrm{t}, J=7.5$ $\mathrm{Hz}, 3 \mathrm{H}), 1.02(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 158.8,145.2,142.0,141.5$, 140.7, 139.9, 138.9, 137.0, 136.30, 136.27, 136.1, 132.5, 130.47, 130.45, 129.8, 129.6, 129.0, 128.6, 128.4, 126.7, 124.6, 122.6, 114.2, 114.1, 73.9, 55.3, 22.83, 22.81, 21.3, 15.9, 15.8; IR (KBr) $v_{\max } 3566,3452,3069,3040,3022,2968,2929,2869,2839,1607,1571,1515,1467$, $1428,1374,1284,1245,1177,1105,1033,854,824,764,746 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{31} \mathrm{H}_{30} \mathrm{O}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 457.21380, found: 457.21381; $R_{f}(5 / 1$ hexanes $/ \mathrm{EtOAc})=0.36$ (silica gel plate).

2,3-Diethyl-1,4-di-p-tolyl-9H-fluoren-9-ol (9b). With $\mathbf{8 b}$ ( $168 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) following the
 general procedure C. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $133 \mathrm{mg}(64 \%)$ of the title compound as a light brown viscous oil. ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.40(\mathrm{~d}, J=7.4, \mathrm{~Hz}, 1 \mathrm{H}), 7.30-7.36(\mathrm{~m}, 6 \mathrm{H}), 7.23-7.27$ $(\mathrm{m}, 2 \mathrm{H}), 7.10(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{dt}, J=7.6,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.02(\mathrm{~d}$, $J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.44(\mathrm{~d}, J=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.60-2.66(\mathrm{~m}, 1 \mathrm{H}), 2.55(\mathrm{q}, J=7.5$ $\mathrm{Hz}, 2 \mathrm{H}), 2.50-2.53(\mathrm{~m}, 1 \mathrm{H}), 2.51(\mathrm{~s}, 3 \mathrm{H}), 2.47(\mathrm{~s}, 3 \mathrm{H}), 1.68(\mathrm{~d}, J=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.03(\mathrm{t}, J=$ $7.5 \mathrm{~Hz}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 145.1,141.7,141.5,140.7,139.9,138.9,137.2$, $137.0,136.8,136.7,136.3,135.8,129.8$, 129.7, 129.5, 129.4, 129.3, 129.2, 129.0, 128.7, $128.3,126.6,124.6,122.6,73.9,22.84,22.79,21.4,21.3,15.9,15.8$; IR (KBr) $v_{\max } 3425$, 3058, 3019, 2968, 2926, 2890, 2869, 1598, 1556, 1470, 1437, 1374, 1290, 1195, 1177, 1072, 1024, 734, 758, $701 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{31} \mathrm{H}_{30} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 441.21889, found: $441.21893 ; R_{f}(5 / 1$ hexanes $/ E t O A c)=0.46$ (silica gel plate) .

4-([1,1'-Biphenyl]-4-yl)-2,3-diethyl-1-(p-tolyl)-9H-fluoren-9-ol (9c). With 8c (199.1 mg,
 0.5 mmol ) following the general procedure C. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $115 \mathrm{mg}(48 \%)$ of the title compound as a viscous brown oil: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.78-7.82(\mathrm{~m}, 4 \mathrm{H}), 7.52$ (m, 2H), 7.40-7.47 (m, 4H), 7.32-7.38 (m, 4H), $7.11(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}$, $1 \mathrm{H}), 6.94(\mathrm{dt}, J=7.4,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.11(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.47(\mathrm{~d}, J=4.7$ $\mathrm{Hz}, 1 \mathrm{H}), 2.63-2.69(\mathrm{~m}, 1 \mathrm{H}), 2.60(\mathrm{q}, ~ J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.51-2.57(\mathrm{~m}, 1 \mathrm{H}), 2.48(\mathrm{~s}, 3 \mathrm{H}), 1.71(\mathrm{~d}$, $J=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.07(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.05(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 150 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 145.2,141.7,141.6,140.7,140.5,140.0,139.9,139.4,139.1,137.1,136.24,136.23$, 135.7, 129.93, 129.90, 129.8, 129.6, 129.1, 128.9, 128.6, 128.4, 127.4, 127.33, 127.27, 127.0, 126.74, 124.67, 122.5, 73.9, 22.9, 22.8, 21.3, $15.9\left(2 \mathrm{xCH}_{3}\right)$; IR $(\mathrm{KBr}) v_{\max } 3566,3440,3049$, 3028, 2968, 2926, 2866, 1598, 1515, 1488, 1473, 1449, 1431, 1371, 1311, 1290, 1251, 1180, 1141, 1102, 1066, 1021, 1012, 857, 839, 824, 767, 746, $698 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{36} \mathrm{H}_{32} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 503.23454, found: 503.23464; $R_{f}(5 / 1$ hexanes $/ \mathrm{EtOAc})=0.42$ (silica gel plate).

2,3-Diethyl-4-phenyl-1-(p-tolyl)-9H-fluoren-9-ol (9d). With 8d (161 mg, 0.5 mmol )
 following the general procedure C. Column chromatography (5/1 hexanes/EtOAc) yielded $122 \mathrm{mg}(61 \%)$ of the title compound as a light brown solid: mp (decomp) $70-75{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.48$ 7.57 (m, 3H), 7.30-7.41 (m, 7H), 7.09 (dt, $J=7.4,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{dt}, J=$ $7.4,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.93(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.44(\mathrm{~d}, J=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.60-2.66(\mathrm{~m}, 1 \mathrm{H}), 2.55$ $(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.50-2.53(\mathrm{~m}, 1 \mathrm{H}), 2.47(\mathrm{~s}, 3 \mathrm{H}), 1.68(\mathrm{~d}, J=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.03(\mathrm{t}, J=7.5$ $\mathrm{Hz}, 3 \mathrm{H}), 1.02(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 145.2,141.6,141.5,140.6$, $140.3,140.0,139.0,137.1,136.6,136.2,135.7,129.8$, 129.6, 129.5, 129.4, 129.1, 128.8, 128.7, 128.6, 128.4, 127.3, 126.7, 124.6, 122.5, 73.9, 22.9, 22.8, 21.3, 15.9, 15.8; IR (KBr) $v_{\max } 3572,3443,3078,3049,3022,2968$, 2932, 2866, 1598, 1515, 1470, 1446, 1431, 1371, 1308, 1242, 1180, 1141, 1099, 1069, 1036, 1021, 997, 830, 797, 743, 764, $707 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{30} \mathrm{H}_{28} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 427.20324, found: 427.20329; $R_{f}(5 / 1$ hexanes/EtOAc) $=0.44$ (silica gel plate).

4-(4-Chlorophenyl)-2,3-diethyl-1-(p-tolyl)-9H-fluoren-9-ol (9e). With 8e (178 mg, 0.5
 mmol ) following the general procedure C. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 97 mg ( $44 \%$ ) of the title compound as a viscous brown oil: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.51-7.53(\mathrm{~m}, 2 \mathrm{H}), 7.41(\mathrm{~m}, 1 \mathrm{H})$, $7.28-7.35(\mathrm{~m}, 6 \mathrm{H}), 7.12(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.98(\mathrm{dt}, J=7.4,0.7 \mathrm{~Hz}$, $1 \mathrm{H}), 6.05(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.44(\mathrm{~d}, J=4.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.59-2.65(\mathrm{~m}, 1 \mathrm{H})$, $2.53(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.47-2.53(\mathrm{~m}, 1 \mathrm{H}), 2.46(\mathrm{~s}, 3 \mathrm{H}), 1.68(\mathrm{~d}, J=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.02(\mathrm{t}, J=$ $7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.01(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 145.2,141.8,141.5$, $140.3,140.1,139.4,138.8,137.2,136.1,135.6,135.2,133.3,131.0,130.9,129.8,129.6$, 129.10, 129.09, 129.0, 128.6, 128.5, 126.9, 124.8, 122.3, 73.8, 22.81, 22.77, 21.3, 15.9, 15.7; IR (KBr) $v_{\max } 3563,3437,3066,3049,3022,2962,2929,2866,1607,1565,1518,1491$, 1470, 1461, 1425, 1392, 1371, 1248, 1204, 1183, 1141, 1087, 1069, 1018, 997, 857, 821, 803, $761,749,516 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{30} \mathrm{H}_{27} \mathrm{OClNa}(\mathrm{M}+\mathrm{Na})$ calcd: 461.16426, found: 461.16433; $R_{f}(5 / 1$ hexanes $/ E t O A c)=0.43$ (silica gel plate $)$.

2,3-Diethyl-1-(p-tolyl)-4-(4-(trifluoromethyl)phenyl)-9H-fluoren-9-ol (9f). With 8f (195

$\mathrm{mg}, 0.5 \mathrm{mmol}$ ) following the general procedure C . Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $115 \mathrm{mg}(49 \%)$ of the title compound as a viscous brown oil: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.82(\mathrm{~m}, 2 \mathrm{H}), 7.50-7.54$ (m, 2H), $7.41(\mathrm{~m}, 1 \mathrm{H}), 7.29-7.36(\mathrm{~m}, 4 \mathrm{H}), 7.12(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.95$ (dt, $J=7.6,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.89(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.45(\mathrm{~d}, J=4.6 \mathrm{~Hz}, 1 \mathrm{H})$, 2.60-2.66 (m, 1H), 2.51 (q, $J=7.5 \mathrm{~Hz}, 2 \mathrm{H}$ ), 2.47-2.54 (m, 1H), 2.47 (s, 3H), 1.70 (d, $J=4.6$ $\mathrm{Hz}, 1 \mathrm{H}), 1.03(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.01(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $145.2,144.3,141.9,141.2,140.2,140.1,139.6,137.2,136.0,135.4,135.0,130.04,130.01$, 129.9, $129.8\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=32.4 \mathrm{~Hz}\right), 129.53,129.1,128.6,128.5,127.0,125.8\left(\mathrm{q},{ }^{3} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}\right)$, 124.9, 121.6-127.0 (q, $\left.{ }^{1} J_{\mathrm{C}-\mathrm{F}}=272.1 \mathrm{~Hz}\right), 122.1,73.8,22.8,22.7,21.3,15.8,15.7$; IR ( KBr ) $v_{\max } 3566,3428,3075,3046,3016,2968$, 2932, 2872, 1616, 1565, 1509, 1464, 1404, 1374, 1323, 1245, 1171, 1132, 1102, 1069, 1021, $866,827,767 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{31} \mathrm{H}_{27} \mathrm{OF}_{3} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 495.19062, found: 495.19065; $R_{f}(5 / 1$ hexanes/EtOAc) $=0.41$ (silica gel plate).

4-(2,3-Diethyl-9-hydroxy-1-(p-tolyl)-9H-fluoren-4-yl)benzonitrile (9g). With 8g (178 mg,
 0.5 mmol ) following the general procedure C. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 65 mg ( $30 \%$ ) of the title compound as a light brown solid: mp (decomp) 227-232 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.85$ $(\mathrm{m}, 2 \mathrm{H}), 7.50-7.54(\mathrm{~m}, 2 \mathrm{H}), 7.42(\mathrm{~m}, 1 \mathrm{H}), 7.33-7.36(\mathrm{~m}, 3 \mathrm{H}), 7.29(\mathrm{~m}, 1 \mathrm{H})$, $7.13(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.97(\mathrm{dt}, J=7.5,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.88(\mathrm{~d}, J=7.8$ $\mathrm{Hz}, 1 \mathrm{H}), 5.44(\mathrm{~d}, J=4.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.59-2.65(\mathrm{~m}, 1 \mathrm{H}), 2.47-2.53(\mathrm{~m}, 3 \mathrm{H}), 2.47(\mathrm{~s}, 3 \mathrm{H}), 1.70(\mathrm{~d}$, $J=4.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.02(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.00(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 150 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 145.7,145.2,142.0,140.8,140.3,139.9,139.8,137.3,135.8,135.2,134.5,132.7$, 132.6, 130.63, 130.59, 129.9, 129.5, 129.2, 128.53, 128.50, 127.1, 125.0, 121.9, 118.9, 111.5, 73.7, 22.9, 22.7, 21.3, 15.8, 15.7; IR (KBr) $v_{\max } 3476,3090,3075,3049,3028,2971,2926$, 2866, 2235, 1607, 1515, 1473, 1428, 1395, 1371, 1308, 1263, 1192, 1144, 1099, 1069, 1033, , $860,818,761,740,567 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{31} \mathrm{H}_{27} \mathrm{NO}\left(\mathrm{M}^{+}\right)$calcd: 429.2093, found: $429.2095 ; R_{f}(5 / 1$ hexanes $/ E t O A c)=0.31$ (silica gel plate) .

4-(4-Bromophenyl)-2,3-diethyl-1-(p-tolyl)-9H-fluoren-9-ol (9h). With 8h (100 mg, 0.25
 mmol ) following the general procedure C. Column chromatography ( $5 / 1$ hexanes $/ E t O A c$ ) yielded $61 \mathrm{mg}(51 \%)$ of the title compound as a brown solid: mp (decomp) $147-152{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.67(\mathrm{~m}$, $2 \mathrm{H}), 7.41(\mathrm{~m}, 1 \mathrm{H}), 7.33(\mathrm{~m}, 3 \mathrm{H}), 7.24-7.30(\mathrm{~m}, 3 \mathrm{H}), 7.13(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}$, $1 \mathrm{H}), 6.99(\mathrm{dt}, J=7.5,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.05(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.44(\mathrm{bs}, 1 \mathrm{H})$, 2.58-2.67 (m, 1H), 2.52 (q, $J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.48-2.53(\mathrm{~m}, 1 \mathrm{H}), 2.47$ (s, 3H), 1.69 (bs, 1H), $1.02(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.01(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 145.2,141.8$, $141.4,140.2,140.1,139.4,139.3,137.1,136.1,135.5,135.2,132.02,131.96,131.32,131.29$, $129.8,129.6,129.1,128.6,128.5,126.9,124.8,122.3,121.5,73.8,22.81,22.75,21.3,15.84$, 15.75; IR (KBr) $v_{\max } 3563,3047,3025,2962,2929,2869,1601,1512,1488,1467,1431$, 1392, 1371, 1263, 1183, 1099, 1069, 1012, 821, 764, $522 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{30} \mathrm{H}_{27} \mathrm{OBrNa}(\mathrm{M}+\mathrm{Na})$ calcd: 505.11375, found: 505.11368; $R_{f}(5 / 1$ hexanes $/ \mathrm{EtOAc})=0.42$ (silica gel plate).

2,3-Diethyl-4-(naphthalen-2-yl)-1-propyl-9H-fluoren-9-ol (16a). With 15a (162 mg, 0.5
 mmol ) following the general procedure C. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $91 \mathrm{mg}(45 \%)$ of the title compound as a light brown solid: mp (decomp) $93-98{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.99(\mathrm{~m}, 2 \mathrm{H})$, $7.86(\mathrm{~m}, 1 \mathrm{H}), 7.79(\mathrm{~d}, J=8.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.53-7.60(\mathrm{~m}, 3 \mathrm{H}), 7.45(\mathrm{ddd}, J=13.1$, $8.3,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.09(\mathrm{dt}, J=7.4,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.81(\mathrm{dt}, J=7.7,0.9 \mathrm{~Hz}, 1 \mathrm{H})$, $5.86(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.67(\mathrm{~s}, 1 \mathrm{H}), 3.15(\mathrm{~m}, 1 \mathrm{H}), 2.86(\mathrm{~m}, 1 \mathrm{H}), 2.82(\mathrm{dq}, J=7.5,2.2 \mathrm{~Hz}$, $2 \mathrm{H}), 2.46-2.63(\mathrm{~m}, 2 \mathrm{H}), 1.77(\mathrm{~m}, 2 \mathrm{H}), 1.73(\mathrm{~s}, 1 \mathrm{H}), 1.29(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.17(\mathrm{t}, J=7.3$ $\mathrm{Hz}, 3 \mathrm{H}), 0.99(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 146.0, 142.1, 141.1, 140.6, $140.5,140.4,139.1,139.0,138.0,135.9,135.0,133.6,132.5,128.6,128.3,128.19,128.16$, $128.15,128.13,128.08,128.0,127.9,126.7,126.1,125.9,124.48,124.47,122.5,74.6,31.9$, $31.8,25.02,24.96,22.90,22.88,21.8,16.0,15.9,15.1$; IR (KBr) $v_{\max } 3386,3052,2965,2932$, 2866, 1604, 1565, 1470, 1434, 1371, 1302, 1189, 1102, 1069, 1021, 964, 863, 806, 761, 752, $\mathrm{cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{30} \mathrm{H}_{30} \mathrm{O}\left(\mathrm{M}^{+}\right)$calcd: 406.2297, found: 406.2295; $R_{f}(5 / 1$ hexanes $/$ EtOAc $)=0.45($ silica gel plate $)$.

2,3-Diethyl-4-(phenanthren-9-yl)-1-propyl-9H-fluoren-9-ol (16b). With $\mathbf{1 5 b}$ ( $94 \mathrm{mg}, 0.25$

mmol ) following the general procedure C. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $67 \mathrm{mg}(59 \%)$ of the title compound as a light brown solid: mp (decomp) $109-114{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.83(\mathrm{~m}, 2 \mathrm{H})$, $7.88(\mathrm{~m}, 1 \mathrm{H}), 7.71-7.77(\mathrm{~m}, 2 \mathrm{H}), 7.63-7.68(\mathrm{~m}, 2 \mathrm{H}), 7.37-7.60(\mathrm{~m}, 3 \mathrm{H}), 7.02$ $(\mathrm{m}, 1 \mathrm{H}), 6.65(\mathrm{dt}, J=7.8,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.83(\mathrm{dd}, J=7.8,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.70(\mathrm{~d}$, $J=9.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.11-3-31(\mathrm{~m}, 1 \mathrm{H}), 2.91(\mathrm{~m}, 1 \mathrm{H}), 2.83(\mathrm{~m}, 2 \mathrm{H}), 2.55(\mathrm{~m}, 1 \mathrm{H}), 2.27(\mathrm{~m}, 1 \mathrm{H})$, $1.79(\mathrm{~m}, 2 \mathrm{H}), 1.73(\mathrm{~d}, J=9.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.31(\mathrm{dt}, J=7.3,1.0 \mathrm{~Hz}, 3 \mathrm{H}), 1.19(\mathrm{dt}, J=7.2,2.2 \mathrm{~Hz}$, $3 \mathrm{H}), 0.93(\mathrm{dt}, J=7.4,1.2 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 146.0,142.86,142.83$, $141.4,140.8,140.24,140.22,139.29,139.20,136.54,136.52,136.51,136.47,132.58,132.56$, $131.8,131.74,131.73,130.5,130.20,130.18,128.79$, 128.77, 128.68, 128.65, 127.8, 127.7, 126.9, 126.75, 126.73, 126.66, 126.63, 126.56, 126.53, 124.38, 124.35, 122.9, 122.7, 122.5, $122.4,74.65,74.62,32.11,31.61,25.21,24.91,23.3,21.84,21.81,16.23,16.20,16.17,16.12$, 15.13; IR (KBr) $v_{\max } 3321,3072,2959,2932,2866,1607,1494,1464,1452,1374,1251$, 1195, 1099, 1060, 1009, 908, 812, 770, 755, $725 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{34} \mathrm{H}_{32} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 479.23454 , found: $479.23450 ; R_{f}(5 / 1$ hexanes $/ E t O A c)=0.38$ (silica gel plate).

4-(Anthracen-9-yl)-2,3-diethyl-1-propyl-9H-fluoren-9-ol (16c). With 15c (94 mg, 0.25
 mmol ) following the general procedure C. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $55 \mathrm{mg}(48 \%)$ of the title compound as a light brown solid: mp (decomp) $104-109{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.60(\mathrm{~s}, 1 \mathrm{H})$, 8.08 (dd, $J=8.5,3.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.61(\mathrm{dd}, J=8.8,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.41-7.50(\mathrm{~m}$, $4 \mathrm{H}), 7.22-7.28(\mathrm{~m} 2 \mathrm{H}), 6.93(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.47(\mathrm{dt}, J=7.6,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.73(\mathrm{~d}, J$ $=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.91(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.24(\mathrm{~m}, 1 \mathrm{H}), 2.96(\mathrm{~m}, 1 \mathrm{H}), 2.85(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H})$, $2.24(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.86(\mathrm{~m}, 2 \mathrm{H}), 1.78(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.33(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.21$ $(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 0.66(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 145.9,143.3$, 141.7, 141.1, 140.1, 139.4, 137.1, 134.9, 131.6, 130.7, 130.5, 130.4, 128.6, 128.4, 126.64, $126.59,126.5,126.4,125.6,125.4,125.3,124.3,121.8,74.7,31.9,25.1,23.5,21.8,16.2$, 15.5, 15.1; IR (KBr) $v_{\text {max }} 3549,3261,2961,2930,2870,2205,1947,1818,1605,1519,1465$, $1441,1408,1375,1230,1190,1101,1059,1012,973,884,845,762,738 \mathrm{~cm}^{-1} ;$ HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{34} \mathrm{H}_{32} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 479.23454, found: 479.23448; $R_{f}(5 / 1$ hexanes/EtOAc $)=0.39$ (silica gel plate).

2,3-Diethyl-1-propyl-4-(pyren-2-yl)-9H-fluoren-9-ol (16d). With $\mathbf{1 5 d}$ ( $100 \mathrm{mg}, 0.25 \mathrm{mmol}$ )
 following the general procedure C. Column chromatography (5/1 hexanes/EtOAc) yielded 64 mg ( $54 \%$ ) of the title compound as a light brown solid: mp (decomp) $98-103{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.33$ (dd, $J=7.8,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 8.24(\mathrm{~m}, 1 \mathrm{H}), 8.13-8.20(\mathrm{~m}, 3 \mathrm{H}), 8.02(\mathrm{dt}, J=$ $7.6,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.94-8.00(\mathrm{dd}, J=17.0,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.90(\mathrm{dd}, J=9.2$, $6.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.64-7.78(\mathrm{dd}, J=46.1,9.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.98(\mathrm{~m}, 1 \mathrm{H}), 6.53$ (dt, $J=7.6,0.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.75(\mathrm{dd}, J=9.7,4.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.21(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.17-3.32$ $(\mathrm{m}, 1 \mathrm{H}), 2.94(\mathrm{~m}, 1 \mathrm{H}), 2.86(\mathrm{~m}, 2 \mathrm{H}), 2.48(\mathrm{~m}, 1 \mathrm{H}), 2.31(\mathrm{~m}, 1 \mathrm{H}), 1.88(\mathrm{~m}, 2 \mathrm{H}), 1.77(\mathrm{~d}, J=$ $10.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.34(\mathrm{dt}, J=7.3,1.2 \mathrm{~Hz}, 3 \mathrm{H}), 1.26(\mathrm{dt}, J=7.2,1.6 \mathrm{~Hz}, 3 \mathrm{H}), 0.86(\mathrm{t}, J=7.5 \mathrm{~Hz}$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 145.97$, 145.94, 142.85, 142.82, 141.35, 140.77, 140.76, $140.39,139.33,139.29,136.67,136.63,135.63,135.57,133.18,133.16,131.38,131.37$, 131.22, 131.21, 130.75, 130.74, 129.62, 129.55, 128.60, 128.58, 127.97, 127.80, 127.61, $127.57,127.46,127.41,126.55,126.53,126.02,125.40,125.28,125.07,125.05,125.04$, $124.95,124.92,124.38,124.37,122.15,122.11,74.69,74.66,32.05,31.80,25.16,25.00$, 23.2, 21.9, 16.19, 16.17, 15.77, 15.75, 15.16, 15.15; IR (KBr) $v_{\max } 3545,3349,3043,2961$, 2930, 2869, 1697, 1602, 1464, 1374, 1306, 1191, 1178, 1099, 1057, 1022, 944, 847, 760, 743
$\mathrm{cm}^{-1}$; HRMS $(\mathrm{m} / z)$ for $\mathrm{C}_{36} \mathrm{H}_{32} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 503.23454, found: 503.23448; $R_{f}$ (5/1 hexanes $/$ EtOAc $)=0.34$ (silica gel plate) .

### 6.2.3. Synthesis of 9,9 -spirobifluorenes $\mathbf{6 , 1 1}$ and 18

## D: General procedure for Oxidation reaction with PCC (preparation of 5, 10 and

 17). To a solution of pyridinium chlorochromate (PCC, $64 \mathrm{mg}, 0.3 \mathrm{mmol}$ ) and celite ( 64 mg ) in anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ the starting fluorenol $(0.2 \mathrm{mmol})$ was added and the mixture was stirred for 3 h at rt . Afterwards, the residue was filtered through a celite/silica gel plug. Column chromatography of the residue on silica gel yielded products.2,3-Diethyl-1,4-diphenyl-9H-fluoren-9-one (5a). With 4ab (100 mg, 0.26 mmol ) following O Ph the general procedure D. Column chromatography ( $10 / 1$ hexanes/EtOAc) yielded $89 \mathrm{mg}(89 \%)$ of the title compound as a bright yellow solid: mp (decomp) $75-80{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.51-7.57(\mathrm{~m}, 3 \mathrm{H})$, 7.47$7.50(\mathrm{~m}, 2 \mathrm{H}), 7.43-7.46(\mathrm{~m}, 1 \mathrm{H}), 7.38-7.41(\mathrm{~m}, 3 \mathrm{H}), 7.28-7.30(\mathrm{~m}, 2 \mathrm{H}), 7.05(\mathrm{dt}, J=7.4,1.1$ $\mathrm{Hz}, 1 \mathrm{H}), 7.01(\mathrm{dt}, J=7.5,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.79(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.52-2.57(\mathrm{~m}, 4 \mathrm{H}), 1.04(\mathrm{t}, J$ $=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.00(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}){ }^{13} \mathrm{C}$ NMR $\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 193.2,147.8,144.0$, $142.1,140.9,140.2,139.2,137.9,137.3,134.9,133.8,129.3,129.03,128.99,128.7,128.97$, $127.95,127.8,127.2,123.4,122.6,23.4,22.4,15.5,15.4$; IR (KBr) $v_{\max } 3055,3019,2971$, 2932, 2869, 1712, 1598, 1580, 1559, 1467, 1446, 1308, 1192, 1066, 764, $704 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{29} \mathrm{H}_{24} \mathrm{ONa}(\mathrm{M}+\mathrm{Na})$ calcd: 411.17194, found: 411.17219; $R_{f}$ (5/1 hexanes/EtOAc) $=0.55$ (silica gel plate).

2,3-Diethyl-4-phenyl-1-propyl-9H-fluoren-9-one (5b). With 4bb (95 mg, 0.27 mmol )
 following the general procedure D. Column chromatography (10/1 hexanes/EtOAc) yielded $89 \mathrm{mg}(93 \%)$ of the title compound as a bright yellow solid: mp (decomp) $127-132{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.47-$ $7.53(\mathrm{~m}, 4 \mathrm{H}), 7.31(\mathrm{~m}, 2 \mathrm{H}), 7.07(\mathrm{dt}, J=7.4,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.99(\mathrm{dt}, J=7.6,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.74$ (d, $J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.12(\mathrm{~m}, 2 \mathrm{H}), 2.73(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.48(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.63(\mathrm{~m}$, $2 \mathrm{H}), 1.23(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.14(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 0.99(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 150 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 195.0,147.6,144.0,142.8,142.0,140.7,139.5,135.9,135.1,133.7,129.4$, $128.9,128.8,127.7,127.6,123.2,122.5,29.7,24.6,23.4,21.4,15.7,15.3,14.9$; IR (KBr) $v_{\max }$ 3060, 3016, 2965, 2929, 2869, 1703, 1604, 1559, 1464, 1452, 1320, 1186, 1060, 761, $710 \mathrm{~cm}^{-}$ ${ }^{1}$; HRMS $(m / z)$ for $\mathrm{C}_{26} \mathrm{H}_{27} \mathrm{O}(\mathrm{M}+\mathrm{H})$ calcd: 355.20564, found: 355.20576; $R_{f}(5 / 1$ hexanes $/ E t O A c)=0.64($ silica gel plate $)$.

2,3-Diethyl-1-phenyl-4-propyl-9H-fluoren-9-one (5c). With 4cb (70 mg, 0.2 mmol ) following the general procedure D. Column chromatography (10/1 hexanes/EtOAc) yielded $66 \mathrm{mg}(94 \%)$ of the title compound as a bright yellow solid: mp (decomp) $70-75{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.57$ (d, $J$ $=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{~m}, 1 \mathrm{H}), 7.40-7.47(\mathrm{~m}, 4 \mathrm{H}), 7.20-7.22(\mathrm{~m}, 3 \mathrm{H}), 2.96(\mathrm{~m}, 2 \mathrm{H}), 2.79(\mathrm{q}, J=$ $7.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.47(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.75(\mathrm{~m}, 2 \mathrm{H}), 1.27(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}), 1.20(\mathrm{t}, J=7.3$ $\mathrm{Hz}, 3 \mathrm{H}), 0.97(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 193.3, 147.9, 144.3, 142.2, $140.3,139.6,138.3,137.1,135.2,134.2,129.7,128.7,127.90,127.89,127.0,123.7,122.9$, $31.5,23.3,22.5,22.4,15.6,15.5,14.6$; IR (KBr) $v_{\max } 3055,3019,2959,2929,2872,1709$, 1601, 1586, 1562, 1467, 1442, 1305, 1189, 1063, 764, $704 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{26} \mathrm{H}_{27} \mathrm{O}(\mathrm{M}$ + H) calcd: 355.20564, found: $355.20590 ; R_{f}(5 / 1$ hexanes $/ E t O A c)=0.57$ (silica gel plate) .

2,3-Diethyl-1,4-dipropyl-9H-fluoren-9-one (5d). With $\mathbf{4 d b}(61 \mathrm{mg}, 0.20 \mathrm{mmol})$ following the general procedure D. Column chromatography (10/1 hexanes/EtOAc) yielded 55 mg ( $90 \%$ ) of the title compound as a bright yellow yellow solid: mp (decomp) $76-81{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.62(\mathrm{~d}, J=7.2 \mathrm{~Hz}$, $1 \mathrm{H}), 7.51(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.44(\mathrm{dt}, J=7.5,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.23(\mathrm{dt}, J=7.4,0.7 \mathrm{~Hz}, 1 \mathrm{H})$, $3.05(\mathrm{~m}, 2 \mathrm{H}), 2.86(\mathrm{~m}, 2 \mathrm{H}), 2.72(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.67(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.67(\mathrm{~m}, 2 \mathrm{H})$, $1.56(\mathrm{~m}, 2 \mathrm{H}), 1.22(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.18(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.15(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 1.09$ $(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 195.1,147.8,144.3,142.2,141.3,140.8$, 135.5, 135.4, 134.1, 129.4, 127.7, 123.5, 122.8, 31.4, 29.6, 24.5, 23.3, 22.4, 21.5, 15.6, 15.5, $14.8,14.6$; IR (KBr) $v_{\max } 3066,3049,2962,2926,2869,1700,1610,1589,1568,1461,1425$, 1296, 1248, 1192, 1096, 1060, 994, 761, $716 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{23} \mathrm{H}_{29} \mathrm{O}(\mathrm{M}+\mathrm{H})$ calcd: 321.22129 , found: $321.22131 ; R_{f}(5 / 1$ hexanes $/ E t O A c)=0.66$ (silica gel plate) .

2,3-Diethyl-4-(4-methoxyphenyl)-1-(p-tolyl)-9H-fluoren-9-one (10a). With 9a (87 mg, 0.26

mmol ) following the general procedure D . Column chromatography ( $10 / 1$ hexanes/EtOAc) yielded $69 \mathrm{mg}(80 \%)$ of the title compound as a bright yellow solid: mp (decomp) 231-236 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.40$ $(\mathrm{m}, 1 \mathrm{H}), 7.28(\mathrm{~m}, 4 \mathrm{H}), 7.16(\mathrm{~m}, 2 \mathrm{H}), 7.08(\mathrm{~m}, 2 \mathrm{H}), 7.05(\mathrm{~m}, 2 \mathrm{H}), 5.92(\mathrm{~m}$, $1 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 2.55(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.53(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.47(\mathrm{~s}$, $3 \mathrm{H}), 1.03(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.99(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 193.3$, $159.1,148.4,144.1,142.3,141.0,140.6,136.9,136.6,135.0,134.9,133.8,131.4,130.3$,
129.1, 128.7, 128.5, 127.9, 123.4, 122.6, 114.3, 55.3, 23.4, 22.4, 21.4, 15.6, 15.4; IR (KBr) $v_{\max } 3022,3001,2962,2932,2893,2869,2836,1712,1613,1586,1562,1512,1461,1422$, 1302, 1284, 1242, 1192, 1186, 1066, 1030, 842, 824, $779 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{31} \mathrm{H}_{28} \mathrm{O}_{2}$ $\left(\mathrm{M}^{+}\right)$calcd: 432.2089 , found: $432.2091 ; R_{f}(5 / 1$ hexanes $/ \mathrm{EtOAc})=0.39$ (silica gel plate).

2,3-Diethyl-1,4-di-p-tolyl-9H-fluoren-9-one (10b). With 9b ( $83 \mathrm{mg}, 0.2 \mathrm{mmol}$ ) following
 the general procedure D. Column chromatography (10/1 hexanes/EtOAc) yielded $70 \mathrm{mg}(84 \%)$ of the title compound as a bright yellow solid: mp (decomp) $76-81{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.39(\mathrm{~m}, 1 \mathrm{H}), 7.34(\mathrm{~d}, J=$ $7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.28(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.25(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.16(\mathrm{~d}, J=$ $7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.05(\mathrm{dt}, J=7.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.02(\mathrm{dt}, J=7.5,1.5 \mathrm{~Hz}, 1 \mathrm{H})$, $5.87(\mathrm{~m}, 1 \mathrm{H}), 2.55(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.54(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.51(\mathrm{~s}, 3 \mathrm{H}), 2.47(\mathrm{~s}, 3 \mathrm{H})$, $1.03(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.99(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 193.3$, 148.1, 144.1, 142.3, 141.0, 140.3, 137.4, 137.3, 136.6, 136.2, 135.0, 134.9, 133.7, 129.7, 129.11, 129.08, 128.7, 128.5, 127.9, 123.3, 122.6, 23.4, 22.4, $21.4\left(2 \mathrm{xCH}_{3}\right), 15.6,15.4$; IR (KBr) $v_{\max }$ 3078, 3046, 3025, 2965, 2932, 2893, 2872, 1709, 1607, 1586, 1556, 1515, 1467, 1446, 1422, $1374,1299,1195,1063,875,818,779,764,725 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{31} \mathrm{H}_{28} \mathrm{O}(\mathrm{M}+\mathrm{Na})$ calcd: 416.2140 , found: $416.2142 ; R_{f}(5 / 1$ hexanes $/ \mathrm{EtOAc})=0.49$ (silica gel plate).

4-([1,1'-Biphenyl]-4-yl)-2,3-diethyl-1-(p-tolyl)-9H-fluoren-9-one (10c). With 9c (96 mg,
 0.2 mmol ) following the general procedure D. Column chromatography (10/1 hexanes/EtOAc) yielded 78 mg ( $82 \%$ ) of the title compound as a bright yellow solid: mp (decomp) 194-199 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.81$ $(\mathrm{m}, 2 \mathrm{H}), 7.78(\mathrm{~m}, 2 \mathrm{H}), 7.53(\mathrm{~m}, 2 \mathrm{H}), 7.46(\mathrm{~m}, 2 \mathrm{H}), 7.42(\mathrm{~m}, 2 \mathrm{H}), 7.29(\mathrm{~m}$, $2 \mathrm{H}), 7.18(\mathrm{~m}, 2 \mathrm{H}), 7.06(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.02(\mathrm{dt}, J=7.6,1.4 \mathrm{~Hz}$, $1 \mathrm{H}), 5.96(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.59(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.56(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.47(\mathrm{~s}, 3 \mathrm{H})$, $1.04(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.00(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 193.2$, 147.9, $144.0,142.4,141.2,140.39,140.37,140.3,138.3,136.8,136.7,135.0,134.9,133.8,129.8$, 129.2, 128.9, 128.7, 128.5, 128.0, 127.6, 127.5, 127.0, 123.4, 122.6, 23.4, 22.4, 21.4, 15.6, 15.4; IR (KBr) $v_{\text {max }} 3078,3049,3022,2974,2938,2878,1712,1601,1583,1559,1488,1467$, 1446, 1302, 1230, 1186, 1117, 1063, 1006, 872, 776, 752, $695 \mathrm{~cm}^{-1}$; HRMS ( $m / z$ ) for $\mathrm{C}_{36} \mathrm{H}_{30} \mathrm{O}\left(\mathrm{M}^{+}\right)$calcd: 478.2297, found: 478.2296; $R_{f}(5 / 1$ hexanes $/ \mathrm{EtOAc})=0.47$ (silica gel plate).

2,3-Diethyl-4-phenyl-1-(p-tolyl)-9H-fluoren-9-one (10d). With 9d (81 mg, 0.2 mmol )
 following the general procedure D. Column chromatography (10/1 hexanes/EtOAc) yielded $69 \mathrm{mg}(86 \%)$ of the title compound as a bright yellow solid: mp (decomp) 71-76 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.51-$ 7.57 (m, 3H), 7.38-7.42 (m, 3H), 7.29 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.17(\mathrm{~m}, 2 \mathrm{H}), 7.05$ (dt, $J=7.5,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.00(\mathrm{dt}, J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.79(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.560(\mathrm{q}, J=$ $7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.559(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.47(\mathrm{~s}, 3 \mathrm{H}), 1.04(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.01(\mathrm{t}, J=7.5$ $\mathrm{Hz}, 3 \mathrm{H}$ ) ; ${ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 193.2, 147.7, 144.0, 142.3, 141.1, 140.2, 139.3, $137.2,136.6,134.9,134.9,133.7,129.3,129.1,129.0,128.7,128.5,127.9,127.8,123.4$, 122.5, 23.4, 22.4, 21.4, 15.6, 15.3; IR (KBr) $v_{\max } 3049,3022,2971,2929,2869,1715,1601$, $1559,1512,1464,1440,1305,1186,1060,1009,952,866,764,719,707 \mathrm{~cm}^{-1} ; \operatorname{HRMS}(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{30} \mathrm{H}_{26} \mathrm{O}\left(\mathrm{M}^{+}\right)$calcd: 402.1984, found: $402.1988 ; R_{f}(5 / 1$ hexanes $/ \mathrm{EtOAc})=0.53$ (silica gel plate).

4-(4-Chlorophenyl)-2,3-diethyl-1-(p-tolyl)-9H-fluoren-9-one (10e). With 9e (87 mg, 0.2
 mmol ) following the general procedure D . Column chromatography ( $10 / 1$ hexanes/EtOAc) yielded $75 \mathrm{mg}(86 \%)$ of the title compound as a bright yellow solid: mp (decomp) 193-198 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.54$ $(\mathrm{m}, 2 \mathrm{H}), 7.42(\mathrm{~m}, 1 \mathrm{H}), 7.34(\mathrm{~m}, 2 \mathrm{H}), 7.28(\mathrm{~m}, 2 \mathrm{H}), 7.15(\mathrm{~m}, 2 \mathrm{H}), 7.08(\mathrm{~m}$, $2 \mathrm{H}), 5.91(\mathrm{~m}, 1 \mathrm{H}), 2.53(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.52(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.47$ $(\mathrm{s}, 3 \mathrm{H}), 1.02(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.99(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 193.0, 147.6, 143.7, 142.5, 141.4, 140.2, 137.8, 136.7, 135.7, 134.9, 134.7, 133.9, 133.8, $130.8,129.3,129.2,128.7,128.4,128.1,123.6,122.4,23.3,22.4,21.4,15.6,15.3$; IR (KBr) $v_{\max } 3081,3063,3052,3025,2974,2929,2866,1712,1607,1589,1556,1512,1488,1464$, 1425, 1302, 1198, 1087, 955, 872, 821, 773, $725 \mathrm{~cm}^{-1} ; \operatorname{HRMS}(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{30} \mathrm{H}_{26} \mathrm{OCl}(\mathrm{M}+\mathrm{H})$ calcd: 437.16667 , found: $437.16665 ; R_{f}(5 / 1$ hexanes $/ E t O A c)=0.52$ (silica gel plate) .

2,3-Diethyl-1-(p-tolyl)-4-(4-(trifluoromethyl)phenyl)-9H-fluoren-9-one (10f). With 9f (95
 $\mathrm{mg}, 0.2 \mathrm{mmol}$ ) following the general procedure D . Column chromatography ( $10 / 1$ hexanes/EtOAc) yielded 57 mg ( $61 \%$ ) of the title compound as a bright yellow solid: mp (decomp) $77-82{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.83(\mathrm{~d}$, $J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.54(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.42(\mathrm{~m}, 1 \mathrm{H}), 7.29(\mathrm{~m}, 2 \mathrm{H}), 7.16$ $(\mathrm{m}, 2 \mathrm{H}), 7.08(\mathrm{dt}, J=7.5,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.04(\mathrm{dt}, J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.76$ $(\mathrm{m}, 1 \mathrm{H}), 2.54(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.49(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.47(\mathrm{~s}, 3 \mathrm{H}), 1.02(\mathrm{t}, J=7.5 \mathrm{~Hz}$,
$3 \mathrm{H}), 1.00(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 192.8,147.2,143.5,143.3$, $142.7,141.7,139.9,136.8,135.5,134.9,134.6,133.9,130.2\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=32.6 \mathrm{~Hz}\right), 129.9$, $129.3,128.8,128.4,128.3,126.0\left(\mathrm{q},{ }^{3} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}\right), 121.5-126.9\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=272.2 \mathrm{~Hz}\right), 123.7$, 122.2, 23.4, 22.3, 21.4, 15.5, 15.3; IR (KBr) $v_{\max } 3028,2965,2932,2869,1715,1619,1601$, $1559,1464,1401,1374,1332,1323,1192,1162,1132,1123,1102,1069,1021,872,857$, 824, 770, $719 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{31} \mathrm{H}_{26} \mathrm{OF}_{3}(\mathrm{M}+\mathrm{H})$ calcd: 471.19303, found: 471.19297; $R_{f}(5 / 1$ hexanes/EtOAc $)=0.54$ (silica gel plate) .

4-(2,3-Diethyl-9-oxo-1-(p-tolyl)-9H-fluoren-4-yl)benzonitrile (10g). With 9g (65 mg, 0.15
 mmol ) following the general procedure D . Column chromatography ( $10 / 1$ hexanes/EtOAc) yielded $47 \mathrm{mg}(73 \%)$ of the title compound as a bright yellow solid: mp (decomp) $239-244{ }^{\circ} \mathrm{C}$, ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.87$ (m, 2H), $7.55(\mathrm{~m}, 2 \mathrm{H}), 7.43(\mathrm{~m}, 1 \mathrm{H}), 7.29(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.14(\mathrm{~m}, 2 \mathrm{H})$, $7.10(\mathrm{dt}, J=7.5,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.05(\mathrm{dt}, J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.76(\mathrm{~d}, J=7.6$ $\mathrm{Hz}, 1 \mathrm{H}), 2.54(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.48(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.46(\mathrm{~s}, 3 \mathrm{H}), 1.01(\mathrm{t}, J=7.5 \mathrm{~Hz}$, $3 \mathrm{H}), 0.99(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 192.6,146.8,144.6,143.2$, $142.8,141.9,139.7,136.9,135.0,134.9,134.4,133.9,132.8,130.5,129.3,128.8,128.41$, $128.35,123.8,122.0,118.6,112.1,23.4,22.3,21.4,15.5,15.2$; IR (KBr) $v_{\max } 3120,3081$, 3066, 3046, 3022, 2977, 2926, 2893, 2872, 2229, 1718, 1604, 1559, 1512, 1467, 1422, 1371, $1320,1169,1069,1021,949,872,833,764,716 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{31} \mathrm{H}_{26} \mathrm{ON}(\mathrm{M}+\mathrm{H})$ calcd: 428.20089 , found: $428.20087 ; R_{f}(5 / 1$ hexanes $/ E t O A c)=0.37$ (silica gel plate) .

4-(4-Bromophenyl)-2,3-diethyl-1-(p-tolyl)-9H-fluoren-9-one (10h). With 9h (510 mg, 1.1
 mmol ) following the general procedure D. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $475 \mathrm{mg}(93 \%)$ of the title compound as a bright yellow solid: mp (decomp) $78-83{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.69(\mathrm{~m}$, $2 \mathrm{H}), 7.42(\mathrm{~m}, 1 \mathrm{H}), 7.27-7.30(\mathrm{~m}, 4 \mathrm{H}), 7.15(\mathrm{~m}, 2 \mathrm{H}), 7.08(\mathrm{~m}, 2 \mathrm{H}), 5.92(\mathrm{~m}$, $1 \mathrm{H}), 2.53(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.51(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.46(\mathrm{~s}, 3 \mathrm{H}), 1.02(\mathrm{t}, J$ $=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.99(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 193.0,147.6,143.7$, $142.5,141.4,140.1,138.2,136.8,135.7,134.9,134.7$, 133.9, 132.2, 131.1, 129.2, 128.7, 128.4, 128.2, 123.6, 122.4, 122.0, 23.3, 22.3, 21.4, 15.6, 15.3; IR (KBr) $v_{\max } 3022,2968$, 2932, 2866, 1709, 1604, 1556, 1515, 1485, 1464, 1422, 1371, 1302, 1192, 1120, 1072, 1012, 949, 869, 824, 761, $719 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{30} \mathrm{H}_{26} \mathrm{OBr}(\mathrm{M}+\mathrm{H})$ calcd: 481.11615, found: 481.11621; $R_{f}(5 / 1$ hexanes $/ E t O A c)=0.51$ (silica gel plate) .

2,3-Diethyl-4-(naphthalen-2-yl)-1-propyl-9H-fluoren-9-one (17a). With 16a (81 mg, 0.2
 mmol ) following the general procedure D. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $84 \mathrm{mg}(91 \%)$ of the title compound as a bright yellow solid: mp (decomp) $69-74{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.00(\mathrm{~m}$, $2 \mathrm{H}), 7.87(\mathrm{~m}, 1 \mathrm{H}), 7.81(\mathrm{~s}, 1 \mathrm{H}), 7.56-7.61(\mathrm{~m}, 2 \mathrm{H}), 7.53(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H})$, $7.44(\mathrm{dd}, J=8.3,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.04(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{dt}, J=7.6,1.1$ $\mathrm{Hz}, 1 \mathrm{H}), 5.71(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.15(\mathrm{~m}, 2 \mathrm{H}), 2.75(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.45-2.53(\mathrm{~m}, 2 \mathrm{H})$, $1.65(\mathrm{~m}, 2 \mathrm{H}), 1.25(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.16(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 0.99(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 194.9,147.9,144.0,142.9,142.2,140.7,136.9,135.7,135.1$, $133.8,133.5,132.7,128.8,128.6,128.14,128.10,127.9,127.8,127.7,126.4,126.2,123.2$, 122.5, 29.7, 24.7, 23.5, 21.4, 15.7, 15.4, 14.9; IR (KBr) $v_{\max } 3052,2962,2929,2866,1700$, 1601, 1562, 1467, 1428, 1374, 1299, 1186, 1102, 1063, 1012, 961, 860, 806, 764, 746, 716 $\mathrm{cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{30} \mathrm{H}_{28} \mathrm{O}\left(\mathrm{M}^{+}\right)$calcd: 404.2140, found: 404.2141; $R_{f}$ (5/1 hexanes $/$ EtOAc $)=0.56($ silica gel plate $)$.

2,3-Diethyl-4-(phenanthren-9-yl)-1-propyl-9H-fluoren-9-one (17b). With 16b (100 mg,

0.22 mmol ) following the general procedure D . Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $85 \mathrm{mg}(86 \%)$ of the title compound as a bright yellow solid: mp (decomp) $81-86{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.82(\mathrm{~d}$, $J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.90(\mathrm{dd}, J=7.9,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.77(\mathrm{dt}, J=7.1,1.4 \mathrm{~Hz}, 1 \mathrm{H})$, 7.75 (s, 1H), 7.66-7.71 (m, 2H), 7.59 (dd, $J=8.2,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.45-7.50(\mathrm{~m}$, $2 \mathrm{H}), 6.97(\mathrm{dt}, J=7.5,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.72(\mathrm{dt}, J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.71(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H})$, $3.33(\mathrm{~m}, 1 \mathrm{H}), 3.11(\mathrm{~m}, 1 \mathrm{H}), 2.79(\mathrm{~m}, 2 \mathrm{H}), 2.53(\mathrm{~m}, 1 \mathrm{H}), 2.25(\mathrm{~m}, 1 \mathrm{H}), 1.73(\mathrm{~m}, 2 \mathrm{H}), 1.28(\mathrm{t}, J$ $=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.19(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 0.93(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 195.0,148.6,143.6,143.2,142.6,141.5,135.4,135.0,133.9,133.3,131.6,131.3,130.6$, $130.3,129.2,128.8,127.9,127.7,127.1,126.99,126.95,126.9,126.3,123.2,123.0,122.7$, 122.5, 29.7, 24.7, 23.9, 21.5, 15.84, 15.82, 14; IR (KBr) $v_{\max } 3075,2965,2935,2869,1703$, 1604, 1562, 1491, 1470, 1449, 1377, 1257, 1192, 1060, 761, 746, $722 \mathrm{~cm}^{-1} ;$ HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{34} \mathrm{H}_{31} \mathrm{O}(\mathrm{M}+\mathrm{H})$ calcd: 455.23694 , found: $455.23707 ; R_{f}(5 / 1$ hexanes/EtOAc) $=0.49$ (silica gel plate).

4-(Anthracen-9-yl)-2,3-diethyl-1-propyl-9H-fluoren-9-one (17c). With 16c (100 mg, 0.23
 mmol ) following the general procedure D . Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded $84 \mathrm{mg}(83 \%)$ of the title compound as a bright yellow solid: mp (decomp) $75-80{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.63$ (s, $1 \mathrm{H}), 8.10(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.66(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.46-7.50(\mathrm{~m}, 3 \mathrm{H})$, $7.32(\mathrm{~m} 2 \mathrm{H}), 6.89(\mathrm{dt}, J=7.6,0.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.55(\mathrm{dt}, J=7.6,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.82(\mathrm{~d}, J=7.7 \mathrm{~Hz}$, $1 \mathrm{H}), 3.28(\mathrm{~m}, 2 \mathrm{H}), 2.84(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.25(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.79(\mathrm{~m}, 2 \mathrm{H}), 1.32(\mathrm{t}, J$ $=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.24(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 0.69(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 195.0,149.2,143.34,143.0,142.2,135.0,133.8,133.4,131.6,131.5,130.3,129.5,128.5$, 127.6, 127.2, 126.1, 126.0, 125.5, 123.0, 121.8, 29.8, 24.8, 24.0, 21.5, 15.9, 15.2, 14.9; IR (KBr) $v_{\max } 3049,2959,2926,2886,1703,1601,1562,1464,1404,1293,1245,1222,1183$, 1096, 1054, 1012, 973, 890, 848, 785, 761, 737, $716 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{34} \mathrm{H}_{31} \mathrm{O}(\mathrm{M}+\mathrm{H})$ calcd: 455.23694 , found: $455.23681 ; R_{f}(5 / 1$ hexanes $/ \mathrm{EtOAc})=0.50$ (silica gel plate) .

2,3-Diethyl-1-propyl-4-(pyren-2-yl)-9H-fluoren-9-one (17d). With 16d (96 mg, 0.2 mmol )
 following the general procedure D . Column chromatography (5/1 hexanes/EtOAc) yielded $80 \mathrm{mg}(84 \%)$ of the title compound as a bright yellow solid: mp (decomp) $87-93{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.32$ (d, $J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.25(\mathrm{dd}, J=7.6,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.16-8.22(\mathrm{~m}, 3 \mathrm{H}), 8.04$ (t, $J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.98(\mathrm{~d}, J=3.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.95(\mathrm{~d}, J=4.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.75$ $(\mathrm{d}, J=9.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{~m} \mathrm{1H}), 6.93(\mathrm{dt}, J=7.5,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.58(\mathrm{dt}, J=7.6,1.2 \mathrm{~Hz}, 1 \mathrm{H})$, $5.06(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.31(\mathrm{~m}, 1 \mathrm{H}), 3.18(\mathrm{~m}, 1 \mathrm{H}), 2.81(\mathrm{~m}, 2 \mathrm{H}), 2.48(\mathrm{~m}, 1 \mathrm{H}), 2.29(\mathrm{~m}$, $1 \mathrm{H}), 1.73(\mathrm{~m}, 2 \mathrm{H}), 1.31(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.29(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 0.85(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 195.0,148.6,143.8,143.2,142.6,141.6,135.1,134.3,133.9$, $133.8,131.4,131.2,131.0,129.5,129.2,128.0,127.71,127.68,127.6,127.5,126.2,125.33$, $125.29,125.1,125.0,124.9,124.8,123.2,122.2,29.8,24.8,23.8,21.5,15.9,15.4,14.9$; IR (KBr) $v_{\max } 3040,2963,2931,2870,1702,1603,1565,1465,1423,1375,1297,1187,1057$, 1011, 952, 872, 849, 838, 761, $721 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{36} \mathrm{H}_{31} \mathrm{O}(\mathrm{M}+\mathrm{H})$ calcd: 479.23694, found: $479.23680 ; R_{f}(5 / 1$ hexanes $/ E t O A c)=0.46$ (silica gel plate) .

## E: General procedure for creation of $\mathbf{9 , 9}$ '-spirobifluorene (preparation of $\mathbf{6 , 1 1}$ and

 18). ${ }^{46}$ A solution of 2-bromobiphenyl ( $40 \mu \mathrm{~L}, 0.23 \mathrm{mmol}$ ) in anhydrous THF ( 3 mL ) was cooled down to $-78{ }^{\circ} \mathrm{C}$ and $n$-BuLi $1.6 \mathrm{M}(140 \mu \mathrm{~L}, 0.23 \mathrm{mmol})$ was added dropwise. The resulted solution was stirred for 30 min , followed by the addition of fluorenone ( 0.15 mmol ) in THF ( 3 mL ) and stirred for 3 h first for 15 min at $-78^{\circ} \mathrm{C}$, but during time it was allowed to reach ambient temperature. The reaction mixture was poured in a saturated solution of $\mathrm{NaHCO}_{3}$ solution and extracted with diethyl ether ( 3 x 15 mL ). The combined organic fractions were dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. The crude was purified by column chromatography on silica gel (10/1 hexanes/EtOAc) to yield desired alcohols. The alcohols were dissolved in acetic acid $(10 \mathrm{~mL})$ with a catalytic amount of HCl ( $12 \mathrm{~mol} / \mathrm{L}$ ) and the resulted solution stirred under reflux for 3 h . The reaction mixture was neutralized with $\mathrm{H}_{2} \mathrm{O}$ and $\mathrm{K}_{2} \mathrm{CO}_{3}$ and extracted with diethyl ether ( $3 \times 15 \mathrm{~mL}$ ). Column chromatography of the residue on silica gel provided products.2,3-Diethyl-1,4-diphenyl-9,9'-spirobifluorene (6a). With 5a ( $60 \mathrm{mg}, 0.15 \mathrm{mmol}$ ) following
 the general procedure E . Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $52 \mathrm{mg}(66 \%)$ of the title compound as a colorless solid: mp (decomp) $115-120{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 7.52-7.62 (m, 4H), 7.40-7.46 (m, 1H), 7.40 (m, 1H), 7.28-7.35 (m, 2H), 7.18 (dt, $J=7.3,1.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.05$ (dt, $J=7.5,1.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.80-6.87(\mathrm{~m}, 4 \mathrm{H}), 6.57-6.62(\mathrm{~m}, 2 \mathrm{H})$, 6.39-6.42 (m, 1H), 6.06-6.12 (m, 3H), $2.58(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.28(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.06$ $(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.86(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.7,149.0$, $143.3,142.0,141.8,140.8,140.3,140.0,139.3,137.9,137.1,136.3,133.1,131.3,129.8$, $129.3,128.8,128.0,127.6,127.3,127.0,126.8,126.7,126.6,126.2,124.9,123.8,123.2$, $122.5,119.6,65.5,23.0,22.9,15.8,15.6$; IR (KBr) $v_{\max } 3055,3019,2965,2929,2866,1601$, $1470,1446,1368,1278,1224,1117,1069,1030,752,737 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{41} \mathrm{H}_{32} \mathrm{Na}$ $(\mathrm{M}+\mathrm{Na})$ calcd: 547.23962 , found: $547.23933 ; R_{f}(20 / 1$ hexanes $/ E t O A c)=0.45$ (silica gel plate).

2,3-Diethyl-4-phenyl-1-propyl-9,9'-spirobifluorene (6b). With 5b (87 mg, 0.25 mmol )
 following the general procedure E . Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $88 \mathrm{mg}(74 \%)$ of the title compound as a colorless solid: mp (decomp) 227-232 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$
$7.87(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.56(\mathrm{~m}, 2 \mathrm{H}), 7.51(\mathrm{~m}, 1 \mathrm{H}), 7.46(\mathrm{~m}, 2 \mathrm{H}), 7.39(\mathrm{dt}, J=7.5,0.9 \mathrm{~Hz}$, $2 \mathrm{H}), 7.13(\mathrm{dt}, J=7.5,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.84(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.81(\mathrm{~m}, 2 \mathrm{H}), 6.38(\mathrm{~m}, 1 \mathrm{H}), 5.98$ $(\mathrm{m}, 1 \mathrm{H}), 2.60(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.54(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.47(\mathrm{~m}, 2 \mathrm{H}), 1.11(\mathrm{t}, J=7.4 \mathrm{~Hz}$, $3 \mathrm{H}), 1.02(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.65(\mathrm{~m}, 2 \mathrm{H}), 0.32(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 150 MHz , $\mathrm{CDCl}_{3}$ ) $\delta 149.8,149.6,143.1,141.73,141.70,141.1,140.52,140.46,138.4,137.6,135.1$, 129.9, 128.7, 127.7, 127.4, 127.2, 126.6, 126.4, 124.2, 122.8, 122.5, 119.9, 66.0, 31.1, 23.6, 22.8, 21.7, 16.1, 15.8, 14.8; IR (KBr) $v_{\max } 3055,3016,2962,2932,2869,1598,1470,1449$, $1374,1278,1260,1150,1105,1063,1027,758,740,704 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / z)$ for $\mathrm{C}_{38} \mathrm{H}_{35}(\mathrm{M}+$ H) calcd: 591.27333, found: 591.27349; $R_{f}(20 / 1$ hexanes $/ E t O A c)=0.48$ (silica gel plate).

2,3-Diethyl-1-phenyl-4-propyl-9,9'-spirobifluorene (6c). With 5c (66 mg, 0.2 mmol )
 following the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided 67 mg ( $68 \%$ ) of the title compound as a colorless solid: mp (decomp) $106-111{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.90(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.30-7.34(\mathrm{~m}, 3 \mathrm{H}), 7.18(\mathrm{t}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.04(\mathrm{t}, J$ $=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.99(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.76(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.57$ (t, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.49(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.04(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.22(\mathrm{~m}, 2 \mathrm{H}), 2.86(\mathrm{q}, J$ $=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.26(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 1.95(\mathrm{~m}, 2 \mathrm{H}), 1.32(\mathrm{t}, J=7.7 \mathrm{~Hz}, 3 \mathrm{H}), 1.31(\mathrm{t}, J=7.5$ $\mathrm{Hz}, 3 \mathrm{H}), 0.84(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 150.0,149.2,143.7,142.1$, $141.9,140.4,139.9,138.1,137.8,137.4,135.7,129.5,127.2,126.9,126.7,126.4,126.0$, $124.7,123.8,123.6,122.8,119.6,65.5,32.0,23.5,23.2,22.0,16.0,15.6,14.8$; IR (KBr) $v_{\max }$ 3058, 3034, 2962, 2932, 2872, 1598, 1473, 1449, 1368, 1266, 1236, 1108, 1072, 1027, 758, $731,704 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{38} \mathrm{H}_{35}(\mathrm{M}+\mathrm{H})$ calcd: 491.27333, found: 491.27334; $R_{f}(20 / 1$ hexanes $/ E t O A c)=0.52($ silica gel plate $)$.

2,3-Diethyl-1,4-dipropyl-9,9'-spirobifluorene ( $\mathbf{6 d}$ ). With $\mathbf{5 d}$ ( $50 \mathrm{mg}, 0.15 \mathrm{mmol}$ ) following
 the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided 57 mg ( $83 \%$ ) of the title compound as a colorless solid: mp (decomp) $101-106{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.86$ $(\mathrm{m}, 2 \mathrm{H}), 7.82(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.37(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.28(\mathrm{dt}, J=$ $7.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.11(\mathrm{dt}, J=7.5,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 6.97(\mathrm{dt}, J=7.5,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.78(\mathrm{~m}, 2 \mathrm{H})$, $6.46(\mathrm{~m}, 1 \mathrm{H}), 3.11(\mathrm{~m}, 2 \mathrm{H}), 2.80(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.57(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 1.86(\mathrm{~m}, 2 \mathrm{H})$, $1.70(\mathrm{~m}, 2 \mathrm{H}), 1.28(\mathrm{t}, J=7.7 \mathrm{~Hz}, 3 \mathrm{H}), 1.25(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.09(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.60$ $(\mathrm{m}, 2 \mathrm{H}), 0.29(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 149.69, 148.97, 143.25,
$141.95,141.78,140.80,140.27,140.04,139.31,137.91,137.11,136.34,133.09,131.26$, $129.79,129.36,129.32,128.78,127.95,127.58$, 127.33, 126.97, 126.80, 126.74, 126.60, $126.15,124.89,123.80,123.22,122.53,119.63,65.51,22.99,22.92,15.77$, 15.61; IR (KBr) $v_{\text {max }} 3058,3037,3016,2953,2929,2896,2869,1598,1476,1446,1365,1284,1219,1114$, 1069, 1030, 758, $740 \mathrm{~cm}^{-1}$; $\operatorname{HRMS}(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{35} \mathrm{H}_{37}(\mathrm{M}+\mathrm{H})$ calcd: 457.2895, found: $457.2895 ; R_{f}(20 / 1$ hexanes $/ E t O A c)=0.54$ (silica gel plate) .

2,3-Diethyl-4-(4-methoxyphenyl)-1-(p-tolyl)-9,9'-spirobifluorene (11a). With 10 a ( 65 mg ,
 0.15 mmol ) following the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $71 \mathrm{mg}(83 \%)$ of the title compound as a colorless solid: mp (decomp) $260-265{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 600 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.43(\mathrm{~m}, 2 \mathrm{H}), 7.30(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.18(\mathrm{dt}, J=7.4,1.0$ $\mathrm{Hz}, 2 \mathrm{H}), 7.14(\mathrm{~m}, 2 \mathrm{H}), 7.04(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.90(\mathrm{dt}, J=7.3,1.1$ $\mathrm{Hz}, 1 \mathrm{H}), 6.84(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.40(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.37$ (d, $J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.19(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.95(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 2.59(\mathrm{q}, J$ $=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.31(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.18(\mathrm{~s}, 3 \mathrm{H}), 1.06(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.87(\mathrm{t}, J=7.4$ $\mathrm{Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 158.8,149.7,149.1,143.7,142.01,141.99,140.5$, $140.3,139.3,138.3,135.8,134.2,134.1,133.1,130.8,129.1,126.9,126.8,126.74,126.70$, $126.5,123.8,123.2,122.6,119.4,114.1,65.5,55.3,23.1,22.9,21.1,15.8,15.7$; IR (KBr) $v_{\max }$ 3063, 3040, 3022, 2956, 2932, 2869, 2839, 1613, 1574, 1512, 1476, 1467, 1440, 1425, 1371, $1290,1245,1177,1111,1063,1033,824,815,770,737 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{43} \mathrm{H}_{37} \mathrm{O}(\mathrm{M}+$ H) calcd: 569.28389 , found: $569.28369 ; R_{f}(20 / 1$ hexanes $/ E t O A c)=0.36$ (silica gel plate).

2,3-Diethyl-1,4-di-p-tolyl-9,9'-spirobifluorene (11b). With 11b ( $62 \mathrm{mg}, 0.15 \mathrm{mmol}$ )
 following the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided 61 mg ( $74 \%$ ) of the title compound as a colorless solid: mp (decomp) $226-231{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 600 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.40(\mathrm{~s}, 4 \mathrm{H}), 7.29(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.18(\mathrm{dt}, J=7.5,1.0$ $\mathrm{Hz}, 2 \mathrm{H}), 7.04(\mathrm{dt}, J=7.5,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.88(\mathrm{dt}, J=7.6,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.83$ (dt, $J=7.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.39(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.37(\mathrm{~d}, J=7.7$ $\mathrm{Hz}, 2 \mathrm{H}), 6.14(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.95(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.57(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.55(\mathrm{~s}$, $3 \mathrm{H}), 2.30(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.16(\mathrm{~s}, 3 \mathrm{H}), 1.06(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.86(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.7$, 149.1, 143.7, 142.0, 140.3, 140.2, 139.2, 138.0, 137.7, $136.8,136.2,134.2,134.0,129.6,129.4,129.1,126.9,126.8,126.7,126.5,123.8,123.1$,
$122.6,119.4,65.5,23.1,22.9,21.5,21.1,15.8,15.7$; IR (KBr) $v_{\text {max }} 3060,3043,3016,2977$, 2959, 2929, 2869, 1515, 1476, 1449, 1428, 1371, 1281, 1239, 1111, 1060, 1021, 863, 824, $755,749,734 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{43} \mathrm{H}_{37}(\mathrm{M}+\mathrm{H})$ calcd: 553.28898, found: 553.28902; $R_{f}$ $(20 / 1$ hexanes $/ \mathrm{EtOAc})=0.50($ silica gel plate $)$.

4-([1,1'-Biphenyl]-4-yl)-2,3-diethyl-1-(p-tolyl)-9,9'-spirobifluorene (11c). With 10c (72
 $\mathrm{mg}, 0.15 \mathrm{mmol}$ ) following the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $58 \mathrm{mg}(63 \%)$ of the title compound as a colorless solid: mp (decomp) $>270{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 600 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.86(\mathrm{~m}, 2 \mathrm{H}), 7.83(\mathrm{~m}, 2 \mathrm{H}), 7.60(\mathrm{~m}, 2 \mathrm{H}), 7.54(\mathrm{t}, J=7.5 \mathrm{~Hz}$, $2 \mathrm{H}), 7.42(\mathrm{~m}, 2 \mathrm{H}), 7.30(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.19(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}, 2 \mathrm{H})$, $7.05(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.87(\mathrm{dt}, J=7 ., 1.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.84(\mathrm{dt}, J=7.4,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.81$ (d, $J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.40(\mathrm{~m}, 1 \mathrm{H}), 6.38(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.22(\mathrm{~m}, 1 \mathrm{H}), 5.96(\mathrm{~d}, J=7.9 \mathrm{~Hz}$, $2 \mathrm{H}), 2.62(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.32(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.18(\mathrm{~s}, 3 \mathrm{H}), 1.09(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H})$, $0.88(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.7$, 149.1, 143.8, 142.0, 141.8, $140.8,140.5,140.1,139.9,139.8,139.5,137.9,135.8,134.11,134.10,130.3,129.1,128.9$, $127.4,127.3,127.1,126.9,126.78,126.77,126.73,126.6,123.8,123.2,122.6,119.4,65.5$, 29.7, 23.1, 23.0, 21.1, 15.9, 15.7; IR (KBr) $v_{\max } 3078,3058,3043,3016,2968,2929,2869$, $1601,1518,1491,1440,1386,1368,1111,1060,1006,848,824,755,737 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{48} \mathrm{H}_{39}(\mathrm{M}+\mathrm{H})$ calcd: 615.30463 , found: $615.30474 ; R_{f}(20 / 1$ hexanes $/ \mathrm{EtOAc})=0.40$ (silica gel plate).

2,3-Diethyl-4-phenyl-1-(p-tolyl)-9,9'-spirobifluorene (11d). With 10 d ( $60 \mathrm{mg}, 0.15 \mathrm{mmol}$ )
 following the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $56 \mathrm{mg}(73 \%)$ of the title compound as a colorless solid: mp (decomp) 238-243 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 600 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.60(\mathrm{~m}, 2 \mathrm{H}), 7.54(\mathrm{~m}, 3 \mathrm{H}), 7.30(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.18$ (dt, $J=7.4,1.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.04 (dt, $J=7.4,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.86(\mathrm{dt}, J=7.4,1.3$ $\mathrm{Hz}, 1 \mathrm{H}), 6.83(\mathrm{dt}, J=7.4,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.39(\mathrm{~m}, 1 \mathrm{H}), 6.37(\mathrm{~d}, J=7.6$ $\mathrm{Hz}, 2 \mathrm{H}), 6.07(\mathrm{~m}, 1 \mathrm{H}), 5.96(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.58(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.31(\mathrm{q}, J=7.4 \mathrm{~Hz}$, $2 \mathrm{H}), 2.18(\mathrm{~s}, 3 \mathrm{H}), 1.06(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.87(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 150 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 149.7,149.1,143.7,142.0,141.8,140.9,140.4,139.9,139.4,137.8,136.2,134.13$, $134.08,129.8,129.1,128.8,127.3,126.9,126.8,126.72,126.70,126.5,123.8,123.2,122.5$, $119.4,65.5,23.0,22.9,21.1,15.8,15.7$; IR (KBr) $v_{\max } 3052,3013,2968,2926,2869,1601$,

1488, 1446, 1428, 1350, 1251, 1198, 1183, 1126, 1018, 860, 815, 758, 749, $701 \mathrm{~cm}^{-1}$;HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{42} \mathrm{H}_{35}(\mathrm{M}+\mathrm{H})$ calcd: 539.27333, found: 539.27337; $R_{f}(20 / 1$ hexanes/EtOAc $)=$ 0.44 (silica gel plate).

4-(4-Chlorophenyl)-2,3-diethyl-1-(p-tolyl)-9,9'-spirobifluorene (11e). With 10e (65 mg,
 $0.15 \mathrm{mmol})$ following the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided 72 mg ( $82 \%$ ) of the title compound as a colorless solid: mp (decomp) 247-252 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 600 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.58(\mathrm{~m}, 2 \mathrm{H}), 7.48(\mathrm{~m}, 2 \mathrm{H}), 7.29(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.18$ (dt, $J=7.4,1.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.04 (dt, $J=7.4,1.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.92 (dt, $J=7.3,1.1$ $\mathrm{Hz}, 1 \mathrm{H}), 6.86(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.77(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.41(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.37$ $(\mathrm{d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.17(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.94(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.54(\mathrm{q}, J=7.4 \mathrm{~Hz}$, $2 \mathrm{H}), 2.30(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.18(\mathrm{~s}, 3 \mathrm{H}), 1.04(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.86(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.8,148.9,143.9,142.0,141.5,140.6,140.0,139.8,139.3$, 137.7, 134.7, 134.2, 134.0, 133.3, 131.3, 129.1, 129.0, 126.9, 126.82, 126.80, 126.78, 126.75, $123.8,123.4,122.3,119.4,65.5,23.0,22.9,21.1,15.7,15.6$; IR (KBr) $v_{\max } 3060,3040,3016$, 2971, 2965, 2929, 2866, 1598, 1512, 1494, 1476, 1443, 1368, 1281, 1223, 1090, 1060, 1015, 842, 821, 758, $740 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{42} \mathrm{H}_{33} \mathrm{Cl}(\mathrm{M}+\mathrm{H})$ calcd: 573.23436, found: $573.23444 ; R_{f}(20 / 1$ hexanes $/ E t O A c)=0.57$ (silica gel plate) .

2,3-Diethyl-1-(p-tolyl)-4-(4-(trifluoromethyl)phenyl)-9,9'-spirobifluorene (11f). With $\mathbf{1 0 f}$

$(47 \mathrm{mg}, \quad 0.1 \mathrm{mmol})$ following the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $55 \mathrm{mg}(90 \%)$ of the title compound as a colorless solid: mp (decomp) 202$207{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.88(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.68(\mathrm{~d}, J=$ $7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.30(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.19(\mathrm{dt}, J=7.5,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.05(\mathrm{dt}$, $J=7.5,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.89(\mathrm{dt}, J=7.4,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.85(\mathrm{dt}, J=7.4,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=$ $7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.42(\mathrm{~m}, 1 \mathrm{H}), 6.38(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.03(\mathrm{~m}, 1 \mathrm{H}), 5.95(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H})$, $2.53(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.31(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.18(\mathrm{~s}, 3 \mathrm{H}), 1.04(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.87$ (t, $J=7.4 \mathrm{~Hz}, 3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.8,148.8,144.9,144.1,142.0,141.3$, $140.7,140.0,139.7,137.4,134.6,134.2,133.9,126.9-133.1$, (q, $\left.{ }^{1} J_{\mathrm{C}-\mathrm{F}}=275.6 \mathrm{~Hz}\right), 130.36$, $129.6,\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=32.6 \mathrm{~Hz}\right), 129.4,129.0,128.7,127.9,127.6,127.0,126.8,125.7\left(\mathrm{q},{ }^{3} J_{\mathrm{C}-\mathrm{F}}=\right.$ $3.5 \mathrm{~Hz}), 123.8,123.4,122.1,119.4,65.5,23.0,22.9,21.1,15.71,15.65$; IR (KBr) $v_{\max } 3060$, 3037, 3013, 2968, 2932, 2869, 1613, 1512, 1473, 1452, 1374, 1320, 1165, 1129, 1108, 1069,

1021, 851, 758, 743, $728 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{43} \mathrm{H}_{34} \mathrm{~F}_{3}(\mathrm{M}+\mathrm{H})$ calcd: 607.26071, found: $607.26063 ; R_{f}(20 / 1$ hexanes $/ \mathrm{EtOAc})=0.46$ (silica gel plate) .

4-(4-Bromophenyl)-2,3-diethyl-1-(p-tolyl)-9,9'-spirobifluorene (11h). With 10h (495 mg,
 1.03 mmol ) following the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $466 \mathrm{mg}(74 \%)$ of the title compound as a slight yellow solid: mp (decomp) 111-116 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.74(\mathrm{~m}, 2 \mathrm{H}), 7.42(\mathrm{~m}, 2 \mathrm{H}), 7.30(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H})$, $7.19(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.05(\mathrm{dt}, J=7.5,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 6.93(\mathrm{dt}, J=7.4$, $1.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{dt}, J=7.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.78(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.41(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H})$, $6.38(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.19(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.94(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.55(\mathrm{q}, J=7.4$ $\mathrm{Hz}, 2 \mathrm{H}), 2.30(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.18(\mathrm{~s}, 3 \mathrm{H}), 1.05(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.86(\mathrm{t}, J=7.4 \mathrm{~Hz}$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.7,148.9,144.0,142.0,141.5,140.6,139.9,139.81$, $139.79,137.6,134.7,134.2,133.9,132.0,131.7,129.0$, $126.9,126.84,126.80,126.78$, 126.76, 123.8, 123.4, 122.3, 121.4, 119.40, 65.47, 23.0, 22.9, 21.1, 15.74, 15.65; IR (KBr) $v_{\max } 3066,3013,2959,2926,2866,1512,1491,1449,1425,1371,1156,1111,1069,1003$, 839, 824, 758, 752, 737, $522 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{42} \mathrm{H}_{34} \mathrm{Br}(\mathrm{M}+\mathrm{H})$ calcd: 617.18384, found: $617.18402 ; R_{f}(20 / 1$ hexanes $/ E t O A c)=0.53$ (silica gel plate) .

2,3-Diethyl-4-(naphthalen-2-yl)-1-propyl-9,9'-spirobifluorene (18a). With 17 a ( 84 mg ,
 0.21 mmol ) following the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $64 \mathrm{mg}(57 \%)$ of the title compound as a colorless solid: mp (decomp) $154-159{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.06(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.02(\mathrm{~m}, 1 \mathrm{H}), 7.98(\mathrm{~s}, 1 \mathrm{H}), 7.93(\mathrm{~m}$, $1 \mathrm{H}), 7.88(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.58-7.63(\mathrm{~m}, 3 \mathrm{H}), 7.39(\mathrm{dt}, J=7.5,0.8 \mathrm{~Hz}$, $2 \mathrm{H}), 7.16(\mathrm{~m}, 2 \mathrm{H}), 6.90(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.79(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.70(\mathrm{dt}, J=7.8,1.2$ $\mathrm{Hz}, 1 \mathrm{H}), 6.39(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.98(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.64(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.60(\mathrm{~m}$, $1 \mathrm{H}), 2.52(\mathrm{~m}, 1 \mathrm{H}), 1.81(\mathrm{~m}, 2 \mathrm{H}), 1.15(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.02(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.69(\mathrm{~m}$, $2 \mathrm{H}), 0.35(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.8,149.6,143.2,141.72$, $141.69,141.65,140.8,140.6,138.5,138.4,137.8,134.8,133.7,132.6,128.6,128.4,128.3$, $128.2,127.9,127.7,127.4,126.7,126.4,126.1,125.9,124.3,124.2,122.8,122.5,120.0,66.0$, 31.1, 23.7, 22.9, 21.7, 16.1, 15.9, 14.8; IR (KBr) $v_{\max } 3055,3019,2974,2959,2950,2929$, $2869,1604,1506,1470,1446,1392,1374,1281,905,860,833,806,755,746 \mathrm{~cm}^{-1}$; HRMS
$(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{42} \mathrm{H}_{37}(\mathrm{M}+\mathrm{H})$ calcd: 541.28898, found: 541.28907; $R_{f}(20 / 1$ hexanes/EtOAc $)=$ 0.46 (silica gel plate).

2,3-Diethyl-4-(phenanthren-9-yl)-1-propyl-9,9'-spirobifluorene (18b). With 17b ( 65 mg ,
 0.15 mmol ) following the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $81 \mathrm{mg}(91 \%)$ of the title compound as a colorless solid: mp (decomp) $>270^{\circ} \mathrm{C}{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.86(\mathrm{dd}, J=8.3,2.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.96(\mathrm{dd}, J=7.8,1.2 \mathrm{~Hz}, 1 \mathrm{H})$, $7.91(\mathrm{~s}, 1 \mathrm{H}), 7.88(\mathrm{dd}, J=7.6,0.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.78(\mathrm{dt}, J=7.0,1.4 \mathrm{~Hz}, 1 \mathrm{H})$, 7.65-7.72 (m, 3H), $7.51(\mathrm{~m}, 1 \mathrm{H}), 7.39(\mathrm{ddt}, J=7.5,2.8,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.18(\mathrm{~m}, 2 \mathrm{H}), 6.95(\mathrm{~d}, J$ $=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.83(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.70(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.53(\mathrm{dt}, J=7.9,1.2$ $\mathrm{Hz}, 1 \mathrm{H}), 6.35(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.97(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.66(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.60(\mathrm{~m}$, $1 \mathrm{H}), 2.32(\mathrm{~m}, 1 \mathrm{H}), 1.85(\mathrm{~m}, 2 \mathrm{H}), 1.16(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.93(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.73(\mathrm{~m}$, $2 \mathrm{H}), 0.36(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.8,149.7,149.5,143.4$, 141.7, 141.6, 141.3, 141.1, 139.1, 138.0, 137.0, 132.4, 132.1, 131.9, 130.6, 130.3, 128.9, $127.9,127.8,127.7,127.5,127.4,127.0,126.78,126.76,126.7,126.6,126.4,124.4,124.0$, $122.9,122.8,122.7,122.5,120.0,66.1,31.2,23.7,23.2,21.7,16.23,16.18,14.8$; IR ( KBr ) $v_{\max } 3071,3015,2965,2931,2869,1599,1472,1447,1426,1373,1281,1154,1059,891$, 866, 757, 745, $725 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{46} \mathrm{H}_{38}(\mathrm{M}+\mathrm{H})$ calcd: 591.30463, found: $591.30468 ; R_{f}(20 / 1$ hexanes $/ \mathrm{EtOAc})=0.38$ (silica gel plate) .

4-(Anthracen-9-yl)-2,3-diethyl-1-propyl-9,9'-spirobifluorene (18c). With $\mathbf{1 7 c}$ ( $68 \mathrm{mg}, 0.15$

mmol ) following the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $45 \mathrm{mg}(51 \%)$ of the title compound as a light brown solid: mp (decomp) $251-256{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.66(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.88(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.74(\mathrm{~d}, J=$ $8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.64(\mathrm{~m}, 2 \mathrm{H}), 7.39-7.44(\mathrm{~m}, 4 \mathrm{H}), 7.28(\mathrm{~m}, 1 \mathrm{H}), 7.19$ (dt, $J=7.7$, $1.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.87(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.64(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.40(\mathrm{dt}, J=7.6,1.2 \mathrm{~Hz}$, $1 \mathrm{H}), 6.31(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.10(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.66(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.22(\mathrm{q}, J=$ $7.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.88(\mathrm{~m}, 2 \mathrm{H}), 1.16(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.78(\mathrm{~m}, 2 \mathrm{H}), 0.64(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H})$, $0.38(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.8,149.7,149.5,143.4,141.8$, $141.6,141.3,141.1,139.2,138.1,137.0,132.4,132.1,132.0,130.6,130.3,128.9,127.9$, $127.82,127.76,127.51,127.46,127.0,126.81,126.78,126.7,126.6,126.4,124.4,124.0$, $122.9,122.8,122.7,122.5,120.0,68.1,31.2,23.8,23.2,21.7,16.3,16.2,14.8$; IR (KBr) $v_{\max }$

3064, 2963, 2871, 1525, 1468, 1446, 1372, 1345, 1262, 1241, 1155, 1055, 1027, 938, 755, $737 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{46} \mathrm{H}_{39}(\mathrm{M}+\mathrm{H})$ calcd: 591.30463, found: 591.30469; $R_{f}(20 / 1$ hexanes $/$ EtOAc $)=0.43($ silica gel plate $)$.

2,3-Diethyl-1-propyl-4-(pyren-2-yl)-9,9'-spirobifluorene (18d). With 17d (72 mg, 0.15

mmol ) following the general procedure E. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $54 \mathrm{mg}(59 \%)$ of the title compound as a colorless solid: mp (decomp) $159-164{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.39(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.21-8.27(\mathrm{~m}, 3 \mathrm{H}), 8.17(\mathrm{~d}$, $J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.13(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.04(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.99(\mathrm{~d}$, $J=9.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.89(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.83(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.43(\mathrm{t}$, $J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.21(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.18(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.96(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H})$, $6.90(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.66(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.39(\mathrm{t}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.34(\mathrm{~d}, J=7.5 \mathrm{~Hz}$, $1 \mathrm{H}), 5.32(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.68(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.51(\mathrm{~m}, 1 \mathrm{H}), 2.36(\mathrm{~m}, 1 \mathrm{H}), 1.87(\mathrm{~m}$, 2 H ), 1.19 (t, $J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.85$ (t, $J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.78$ (m, 2H), 0.38 (t, $J=7.1 \mathrm{~Hz}, 3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.7,149.6,149.5,143.4,141.8,141.7,141.6,141.4,141.1$, $139.3,138.1,136.2,133.0,131.4,131.3,130.8,129.9,128.2,127.83,127.82,127.76,127.71$, $127.65,127.50,127.46,127.37,126.7,126.4,126.0,125.4,125.2,125.11,125.07,125.0$, $124.4,124.1,122.7,122.2,120.03,120.00,66.1,31.2,23.7,23.2,21.8,16.3,15.7,14.8$; IR (KBr) $v_{\max } 3039,2963,2929,2870,1601,1584,1470,1446,1424,1374,1282,1242,1177$, 1154, 1056, 848, 820, 756, 740, $726 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / z)$ for $\mathrm{C}_{48} \mathrm{H}_{39}(\mathrm{M}+\mathrm{H})$ calcd: 615.30463, found: 615.30465; $R_{f}(20 / 1$ hexanes/EtOAc $)=0.37$ (silica gel plate) .

### 6.2.4. Synthesis of 9,9 '-spirobifluorenes 13,14 , and carborane-14a

F: General procedure for Suzuki reaction (preparation of 13). Starting compound 11h ( 0.1 mmol ), boronic acid compound ( 0.12 mmol ), $\mathrm{Pd}(\mathrm{OAc})_{2}(1.3 \mathrm{mg}, 5 \mathrm{~mol} \%)$ and $\mathrm{K}_{2} \mathrm{CO}_{3}(17 \mathrm{mg}, 0.12 \mathrm{mmol})$ were dissolved in a microwave vial ethanol ( 5 mL ) and the reaction was stirred 3 h at $80^{\circ} \mathrm{C}$. The reaction mixture was cooled down, quenched with water and extracted with diethyl ether. The organic fraction was dried over $\mathrm{Mg}_{2} \mathrm{SO}_{4}$, filtered off and concentrated on vacuum evaporator. Column chromatography of the residue on silica gel (hexanes/EtOAc) provided products.

2,3-Diethyl-4-(4'-methoxy-[1,1'-biphenyl]-4-yl)-1-(p-tolyl)-9,9'-spirobifluorene
(13a).


With $\mathbf{1 1 h}(61 \mathrm{mg}, 0.1 \mathrm{mmol})$ and 4-methoxyphenylboronic acid following the general procedure F. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided 40 mg ( $62 \%$ ) of the title compound as a colorless solid: mp (decomp) $138-143{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.82$ (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.76(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.58(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.31(\mathrm{~d}$, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.20(\mathrm{dt}, J=7.3,0.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.08(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.06$ (dt, $J=7.4,0.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.88(\mathrm{dt}, J=7.3,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.82-6.85(\mathrm{~m}, 3 \mathrm{H}), 6.39-6-43(\mathrm{~m}, 3 \mathrm{H})$, $6.25(\mathrm{dd}, J=6.9,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.97(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 2.63(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H})$, $2.33(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.19(\mathrm{~s}, 3 \mathrm{H}), 1.10(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.89(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 159.3,149.7,149.1,143.8,142.0,141.9,140.5,140.2,139.5$, $139.4,139.3,137.9,135.9,134.2,134.1,133.4,130.2,129.1,128.1,127.0,126.9,126.81$, $126.76,126.6,123.9,123.3,122.6,119.4,114.4,65.6,55.5,23.1,23.0,21.1,15.9,15.7$; IR (KBr) $v_{\max } 3069,3037,3013,2953,2836,1604,1500,1470,1437,1278,1251,1086,1039$, 1009, 827, 809, $743 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{49} \mathrm{H}_{41} \mathrm{O}(\mathrm{M}+\mathrm{H})$ calcd: 645.31519, found: $645.31521 ; R_{f}(20 / 1$ hexanes $/ \mathrm{EtOAc})=0.30$ (silica gel plate) .

## 2,3-Diethyl-1-(p-tolyl)-4-(4'-(trifluoromethyl)-[1,1'-biphenyl]-4-yl)-9,9'-spirobifluorene


(13b). With $\mathbf{1 1 h}(61 \mathrm{mg}, 0.1 \mathrm{mmol})$ following the general procedure F . Column chromatography of the residue on silica gel (20/1 hexanes/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided 39 mg ( $58 \%$ ) of the title compound as a colorless solid: mp (decomp) 157-162 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.92(\mathrm{~d}, \mathrm{~J}=8.1 \mathrm{~Hz}$, $2 \mathrm{H}), 7.87(\mathrm{~d}, \mathrm{~J}=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.79(\mathrm{~d}, \mathrm{~J}=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.65(\mathrm{~d}, \mathrm{~J}=8.2 \mathrm{~Hz}$, 2 H ), 7.30 (d, J = $7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.19 (dt, J = 7.4, $0.9 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.06 (dt, J = 7.5, $1.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.87(\mathrm{dt}, \mathrm{J}=7.3,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.84(\mathrm{dt}, \mathrm{J}=7.4,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.80$ $(\mathrm{d}, \mathrm{J}=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.43(\mathrm{~m}, 1 \mathrm{H}), 6.38(\mathrm{~d}, \mathrm{~J}=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.22(\mathrm{~m}, 1 \mathrm{H}), 5.96(\mathrm{~d}, \mathrm{~J}=7.9 \mathrm{~Hz}$, 2H), 2.61 ( $q, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), $2.33(\mathrm{q}, \mathrm{J}=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.18(\mathrm{~s}, 3 \mathrm{H}), 1.09(\mathrm{t}, \mathrm{J}=7.4 \mathrm{~Hz}, 3 \mathrm{H})$, $0.88(\mathrm{t}, \mathrm{J}=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.8,149.0,144.3,143.9,142.0$, 141.7, 141.0, 140.6, 140.0, 139.7, 138.4, 137.7, 135.4, 134.2, 134.0, 130.6, 129.8, 129.1, $127.5,127.4,126.8\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=16.8 \mathrm{~Hz}\right), 125.8,\left(\mathrm{q},{ }^{3} J_{\mathrm{C}-\mathrm{F}}=3.6 \mathrm{~Hz}\right), 123.2-126.5,\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=\right.$ 160.1 Hz ) $123.8,123.3,122.4,119.4,65.5,23.1,23.0,21.1,15.9,15.7$; IR (KBr) $v_{\max } 3063$, 3037, 3019, 2965, 2929, 2872, 1622, 1515, 1470, 1446, 1371, 1326, 1165, 1129, 1069, 1006, 836, 824, 758, 746, $737 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{49} \mathrm{H}_{38} \mathrm{~F}_{3}(\mathrm{M}+\mathrm{H})$ calcd: 683.29201, found: $683.29207 ; R_{f}(20 / 1$ hexanes $/ E t O A c)=0.52$ (silica gel plate) .

## 2,3-Diethyl-4-(4-((4-methoxyphenyl)ethynyl)phenyl)-1-(p-tolyl)-9,9'-spirobifluorene


(14a). Starting compound $\mathbf{1 1 h}(61 \mathrm{mg}, 0.1 \mathrm{mmol}), \mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}$ catalyst ( $8.5 \mathrm{mg}, 5 \mathrm{~mol} \%$ ) and $\mathrm{CuI}(0.01 \mathrm{mmol}, 10 \mathrm{~mol} \%)$ were dissolved in a microwave vial in diisopropylamine ( 5 mL ), afterwards 4-ethynylanisole $(29 \mu \mathrm{~L}, 0.2 \mathrm{mmol})$ was added and the reaction was stirred at $180^{\circ} \mathrm{C}$ in the microwave reactor for 1 h . The reaction mixture was cooled down, filtered off with celite/silica and washed with diethyl ether. The organic fraction was concentrated on vacuum evaporator. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $36 \mathrm{mg}(54 \%)$ of the title compound as a colorless solid: mp (decomp) 131-136 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.76(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.56 (d, $J=8.7 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.52 (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.30(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.19(\mathrm{dt}, J=7.4,0.6$ $\mathrm{Hz}, 2 \mathrm{H}), 7.05(\mathrm{dt}, J=7.5,0.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.93(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.91(\mathrm{dt}, J=7.5,0.9 \mathrm{~Hz}, 1 \mathrm{H})$, $6.85(\mathrm{dt}, J=7.4,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.41(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.37(\mathrm{~d}, J=$ $7.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.24(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.95(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.87(\mathrm{~s}, 3 \mathrm{H}), 2.58(\mathrm{q}, J=7.4$ $\mathrm{Hz}, 2 \mathrm{H}), 2.31(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.18(\mathrm{~s}, 3 \mathrm{H}), 1.05(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.87(\mathrm{t}, J=7.4 \mathrm{~Hz}$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ) $\delta$ 159.7, 149.7, 149.0, 143.9, 142.0, 141.6, 140.8, 140.5,
139.9, 139.7, 137.6, 135.5, 134.2, 134.1, 133.1, 131.9, 130.0, 129.1, 127.0, 126.9, 126.81, $126.78,126.7,123.8,123.3,122.6,122.5,119.4,115.4,114.1,89.8,88.2,65.5,55.4,23.1$, 22.9, 21.1, 15.8, 15.7; IR (KBr) $v_{\max } 3040,2971,2872,2836,2211,1604,1565,1518,1446$, 1371, 1287, 1251, 1171, 1105, 1030, 839, 755, $743 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / z)$ for $\mathrm{C}_{51} \mathrm{H}_{41} \mathrm{O}(\mathrm{M}+\mathrm{H})$ calcd: 669.31519 , found: $669.31522 ; R_{f}(20 / 1$ hexanes $/ E t O A c)=0.25$ (silica gel plate).

2,3-Diethyl-4-(4-(phenylethynyl)phenyl)-1-(p-tolyl)-9,9'-spirobifluorene (14b). Starting
 compound 11h ( $122 \mathrm{mg}, 0.25 \mathrm{mmol}$ ), PEPPSI-IPr catalyst ( $8.5 \mathrm{mg}, 5 \mathrm{~mol} \%$ ) and $\mathrm{CuI}(0.01 \mathrm{mmol}, 10 \mathrm{~mol} \%)$ were dissolved in a microwave vial in triethylamine $(2 \mathrm{~mL})$ and THF ( 3 mL ), afterwards phenylacetylene ( $55 \mu \mathrm{~L}$, 0.5 mmol ) was added and the reaction was stirred 3 h at $80^{\circ} \mathrm{C}$. The reaction mixture was cooled down, filtered off with celite/silica and washed with diethyl ether. The organic fraction was concentrated on vacuum evaporator. Column chromatography of the residue on silica gel (20/1 hexanes/EtOAc) provided $116 \mathrm{mg}(72 \%)$ of the title compound as a colorless solid: mp (decomp) $167-172{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.78(\mathrm{~m}, 2 \mathrm{H}), 7.62(\mathrm{~m}, 2 \mathrm{H}), 7.53(\mathrm{~m}, 2 \mathrm{H}), 7.35-7.43(\mathrm{~m}, 3 \mathrm{H})$, 7.29 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.18$ (dt, $J=7.4,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.04(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.91$ (dt, $J=7.3,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.40(\mathrm{~d}, J=7.5$ $\mathrm{Hz}, 1 \mathrm{H}), 6.37(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.22(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.94(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.56(\mathrm{q}, J$ $=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.30(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.17(\mathrm{~s}, 3 \mathrm{H}), 1.05(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.86(\mathrm{t}, J=7.4$ $\mathrm{Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 149.7, 149.0, 143.9, 142.0, 141.6, 141.2, 140.5, 139.9, 139.7, 137.6, 135.4, 134.2, 134.0, 132.5, 132.1, 131.7, 130.0, 129.2, 129.0, 128.43, 128.36, 126.94, 126.86, 126.80, 126.77, 126.7, 123.8, 123.31, 123.27, 122.5, 122.2, 119.4, 89.8, 89.5, 29.7, 23.0, 22.9, 21.1, 15.8, 15.7; IR (KBr) $v_{\max } 3060,2969,2931,2870,2251$, $2230,1596,1515,1487,1446,1373,1281,1238,1179,1158,1068,1019,913,848,823,755$, $742,689 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{50} \mathrm{H}_{38} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 661.28657, found: 661.28601; $R_{f}$ $(20 / 1$ hexanes $/$ EtOAc $)=0.46($ silica gel plate $)$.

## 2,3-Diethyl-1-(p-tolyl)-4-(4-((4-(trifluoromethyl)phenyl)ethynyl)phenyl)-9,9'-


spirobifluorene (14c). Starting compound $\mathbf{1 1 h}(61 \mathrm{mg}, 0.1 \mathrm{mmol})$, $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(3.5 \mathrm{mg}, 5 \mathrm{~mol} \%)$ and $\mathrm{CuI}(0.01 \mathrm{mmol}, 10 \mathrm{~mol} \%)$ were dissolved in a microwave vial in DMF ( 3 mL ) and $\mathrm{NEt}_{3}(2 \mathrm{~mL})$, afterwards 1-ethynyl-trifluortoluene ( $65 \mu \mathrm{~L}, 0.15 \mathrm{mmol}$ ) was added and the reaction was stirred over night at $80{ }^{\circ} \mathrm{C}$. The reaction mixture was cooled down, filtered off with celite/silica and washed with diethyl ether. The organic fraction was concentrated on vacuum evaporator. Column chromatography of the residue on silica gel ( $10 / 1$ hexanes/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided 40 mg ( $57 \%$ ) of the title compound as a light brown solid: mp (decomp) $151-156{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $7.80(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.72(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.66(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.56(\mathrm{~d}, J=8.3$ $\mathrm{Hz}, 2 \mathrm{H}), 7.30(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.19(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.05(\mathrm{dt}, J=7.5,1.1 \mathrm{~Hz}, 2 \mathrm{H})$, $6.91(\mathrm{dt}, J=7.4,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{dt}, J=7.4,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.41(\mathrm{~m}$, $1 \mathrm{H}), 6.38(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.21(\mathrm{~m}, 1 \mathrm{H}), 5.95(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.56(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H})$, $2.31(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.18(\mathrm{~s}, 3 \mathrm{H}), 1.06(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.87(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.8,148.9,144.0,142.0,141.8,141.5,140.6,139.81,139.78$, $137.5,135.3,134.2,134.0,132.2,131.9,130.1,129.0,127.0\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=32.6 \mathrm{~Hz}\right), 126.8,125.3$ $\left(\mathrm{q},{ }^{3} J_{\mathrm{C}-\mathrm{F}}=3.6 \mathrm{~Hz}\right), 123.8,123.4,122.4,121.5,121.3-126.8,\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=272.1 \mathrm{~Hz}\right), 119.4,91.9$, 88.4, 65.5, 23.0, 22.9, 21.1, 15.74, 15.66; IR (KBr) $v_{\max } 3069,3037,2968,2932,2875,2220$, $1610,1518,1476,1449,1374,1323,1171,1132,1108,1069,1015,845,824,758,743 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{51} \mathrm{H}_{38} \mathrm{~F}_{3}(\mathrm{M}+\mathrm{H})$ calcd: 707.29201, found: 707.29211; $R_{f}$ (20/1 hexanes $/ E t O A c)=0.51($ silica gel plate $)$.

1-[2,3-Diethyl-4-phenyl)-1-(p-tolyl)-9,9'-spirobifluorenyl]-2-(4-methoxyphenyl)-1,2-
 dicarba-closo-dicarborane (carborane-14a). Starting compound $14 \mathbf{a}\left(67 \mathrm{mg}, 0.1 \mathrm{mmol}\right.$ and $6,9-\left(\mathrm{Me}_{2} \mathrm{~S}\right)_{2} \mathrm{~B}_{10} \mathrm{H}_{12}(70 \mathrm{mg}, 0.15 \mathrm{mmol})$ were dissolved in a microwave vial under argon atmosphere in toluene ( 5 mL ) and the reaction was stirred at $120^{\circ} \mathrm{C}$ for 16 h . The reaction mixture was cooled down, quenched with MeOH and concentrated on vacuum evaporator. Column chromatography of the residue on silica gel ( $2 / 1$ hexanes/EtOAc) provided $47 \mathrm{mg}(60 \%)$ of the title compound as a colorless solid: mp (decomp) $149-154{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.70(\mathrm{~d}, J=8.5 \mathrm{~Hz}$, $2 \mathrm{H}), 7.53(\mathrm{~m}, 2 \mathrm{H}), 7.35(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.29(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.18(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}$, $2 \mathrm{H}), 7.02(\mathrm{dt}, J=7.5,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 6.82(\mathrm{dt}, J=7.5,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.76(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 2 \mathrm{H})$,
$6.72(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.66(\mathrm{dt}, J=7.9,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.36-6-39(\mathrm{~m}, 3 \mathrm{H}), 5.91(\mathrm{~d}, J=8.0 \mathrm{~Hz}$, $2 \mathrm{H}), 5.69(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.74(\mathrm{~s}, 3 \mathrm{H}), 2.35(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.26(\mathrm{q}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H})$, $2.17(\mathrm{~s}, 3 \mathrm{H}), 0.90(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.83(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H})$; additional: ${ }^{1} \mathrm{H}-{ }^{11} \mathrm{~B}$ NMR ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.41(\mathrm{bs}, 2 \mathrm{H}), 2.70(\mathrm{bs}, 2 \mathrm{H}), 2.61(\mathrm{bs}, 4 \mathrm{H}), 2.41(\mathrm{bs}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 160.9,149.8,148.8,143.9,143.1,142.0,141.2,140.6,139.9,139.7,137.1$, $134.3,134.2,133.8,132.3,131.2,130.0,129.9,128.9,126.9,126.8,126.7,126.6,123.7$, $123.4,123.2,122.0,119.4,113.6,86.1,85.6,65.4,55.2,23.0,22.7,21.1,15.60$. 15.58; IR (KBr) $v_{\max } 3219,3063,2966,2932,2870,2592,1607,1514,1462,1447,1374,1301,1260$, 1184, 1075, 1032, 836, 824, 756, 744, $730 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{51} \mathrm{H}_{50} \mathrm{OB}_{10} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 811.46844, found: 811.46871 , for $\mathrm{C}_{51} \mathrm{H}_{50} \mathrm{O}^{10} \mathrm{BB}_{9} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 810.47207, found: 810.47278, for $\mathrm{C}_{51} \mathrm{H}_{50} \mathrm{O}^{10} \mathrm{~B}_{2} \mathrm{~B}_{8} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 809.47571, found: 809.47651, for $\mathrm{C}_{51} \mathrm{H}_{50} \mathrm{O}^{10} \mathrm{~B}_{3} \mathrm{~B}_{7} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 808.47934, found: 808.48029; $R_{f}(20 / 1$ hexanes/EtOAc $)=$ 0.27 (silica gel plate).

### 6.2.5. X-ray data of 9,9 '-spirobifluorenes

Crystallographic data for 6a, 6c, 11a, 11c, 11e, 13a, 14b, carborane-14a and 18b were collected on Nonius KappaCCD diffractometer equipped with Bruker APEX-II CCD detector by monochromatized $\operatorname{MoK} \alpha$ radiation $(\lambda=0.71073 \AA)$ at a temperature either 120 K or 150 (2) K. The structures were solved by direct methods (SHELXS) ${ }^{69}$ and refined by full matrix least squares based on $F^{2}$ (SHELXL97). The absorption corrections were carried on using multi-scan method. The hydrogen atoms were found on difference Fourier map and were recalculated into idealized positions. All hydrogen atoms were refined as fixed (riding model) with assigned temperature factors $\mathrm{H}_{\mathrm{iso}}(\mathrm{H})=1.2 \mathrm{U}_{\mathrm{eq}}$ (pivot atom) or $1.5 \mathrm{U}_{\mathrm{eq}}$ for methyl moiety. The crystallographic data are summarized in Table Exp1-4.

Table Exp1. X-ray crystallographic data 9,9'-spirobifluorenes 6a and 6c

|  | 6 a | 6 c |
| :---: | :---: | :---: |
| Formula | $\mathrm{C}_{41} \mathrm{H}_{32}$ | $\mathrm{C}_{38} \mathrm{H}_{34}$ |
| M.w. | 524.67 | 490.65 |
| Crystal system | orthorhombic | triclinic |
| Space group [No.] | Pca2 ${ }_{1}$ (No 29) | $P-1$ (No 2) |
| $a[\AA]$ | 29.4235 (18) | 11.0066 (2) |
| $b$ [ $\AA$ ] | 11.4496 (7) | 11.0845 (2) |
| $c[\AA]$ | 17.2284 (14) | 11.5394 (2) |
| $\alpha{ }^{\circ}{ }^{\circ}$ |  | 81.815 (1) |
| $\beta\left[{ }^{\circ}\right]$ |  | 87.544 (1) |
| $\gamma\left[{ }^{\circ}\right]$ |  | 80.909 |
| Z | 8 | 2 |
| $\mathrm{V}\left[\AA^{3}\right]$ | 5804.0 (7) | 1375.69 (4) |
| $\mathrm{D}_{x}\left[\mathrm{~g} \mathrm{~cm}^{-3}\right]$ | 1.201 | 1.184 |
| Crystal size [mm] | $0.74 \times 0.38 \times 0.29$ | $0.56 \times 0.54 \times 0.34$ |
| Crystal color, shape | a colorless prism | a colorless prism |
| $\mu\left[\mathrm{mm}^{-1}\right]$ | 0.07 | 0.07 |
| $T_{\text {min }}, T_{\text {max }}$ | 0.952, 0.980 | 0.964, 0.978 |
| Measured reflections | 37907 | 20792 |
| Independent diffractions ( $R_{\text {int }}{ }^{\text {a }}$ ) | 9397 (0.049) | 6299 (0.018) |
|  | 8103 | 5492 |
| No. of parameters | 743 | 346 |
| $R^{b}$ | 0.072 | 0.041 |
| $w R\left(F^{2}\right)$ for all data | 0.184 | 0.110 |
| $\mathrm{GOF}^{\text {c }}$ | 1.06 | 1.05 |
| Residual electron density [ $\mathrm{e} / \AA^{3}$ ] | 0.62, -0.290 | 0.26, -0.24 |
| CCDC no. | 1403002 | 1403003 |
|  |  |  |
| ${ }^{b} R(F)=\Sigma\| \| F_{\mathrm{o}}\left\|-\left\|F_{\mathrm{c}}\right\|\right\| / \Sigma\left\|F_{\mathrm{o}}\right\|, w R\left(F^{2}\right)=\left[\Sigma\left(w\left(F_{\mathrm{o}}^{2}-F_{\mathrm{c}}^{2}\right)^{2}\right) /\left(\Sigma w\left(F_{\mathrm{o}}^{2}\right)^{2}\right)\right]^{1 / 2} .$ |  | $\left.\left.w\left(F_{\mathrm{o}}{ }^{2}-F_{\mathrm{c}}{ }^{2}\right)^{2}\right) /\left(\Sigma w\left(F_{\mathrm{o}}{ }^{2}\right)^{2}\right)\right]^{1 / 2}$. |



Figure Exp1. Platon plot of 6a showing displacement ellipsoids at a 30\% probability level


Figure Exp2. Platon plot of $\mathbf{6 c}$ showing displacement ellipsoids at a $30 \%$ probability level

Table Exp2. X-ray crystallographic data for 9,9'-spirobifluorenes 11a, 11c and 11e

|  | 11a | 11c | 11e |
| :---: | :---: | :---: | :---: |
| Formula | $\mathrm{C}_{43} \mathrm{H}_{36} \mathrm{O}$ | $\mathrm{C}_{48} \mathrm{H}_{38}$ | $\mathrm{C}_{42} \mathrm{H}_{33} \mathrm{Cl}$ |
| M.w. | 568.72 | 614.78 | 573.13 |
| Crystal system | monoclinic | triclinic | monoclinic |
| Space group [No.] | $P 2_{1} / \mathrm{n}$ (No14) | P1 (No. 2) | $P 2_{1} / n$ (No 14) |
| $a[\mathrm{~A}]$ | 9.4308 (2) | 8.3330 (2) | 35.6573 (8) |
| $b$ [ $\AA$ ] | 17.6455 (3) | 11.6908 (3) | 9.6469 (2) |
| $c$ [ $\AA$ ] | 18.5101 (3) | 18.3693 (4) | 22.6588 (4) |
| $a\left[{ }^{\circ}\right]$ |  | 88.570 (1) |  |
| $\beta\left[{ }^{\circ}{ }^{\text {] }}\right.$ | 92.764 (1) | 86.248 (1) | 128.8580 (5) |
| $\gamma\left[{ }^{\circ}\right]$ |  | 72.826 (1) |  |
| Z | 4 | 2 | 8 |
| $\mathrm{V}\left[{ }^{3}{ }^{3}\right]$ | 3076.70 (10) | 1706.04 (7) | 6069.4 (2) |
| $\mathrm{D}_{x}\left[\mathrm{~g} \mathrm{~cm}^{-3}\right]$ | 1.228 | 1.197 | 1.254 |
| Crystal size [mm] | $0.22 \times 0.58 \times 0.60$ | $0.14 \times 0.39 \times 0.65$ | $0.57 \times 0.56 \times 0.41$ |
| Crystal color, shape | a colorless prism | a colorless prism | a colorless prism |
| $\mu\left[\mathrm{mm}^{-1}\right]$ | 0.071 | 0.067 | 0.156 |
| $T_{\text {min }}, T_{\text {max }}$ | 0.89, 0.98 | 0.93, 0.99 | 0.83, 0.94 |
| Measured reflections | 21163 | 18051 | 16117 |
| Independent diffractions ( $R_{\text {int }}{ }^{\text {a }}$ ) | 7059 (0.019) | 7428 (0.024) | 6626 (0.018) |
| Observed diffract. [I>2 ${ }^{\text {(I) }}$ ] | 5820 | 5525 | 5486 |
| No. of parameters | 401 | 436 | 391 |
| $R^{b}$ | 0.041 | 0.0471 | 0.042 |
| $w R\left(F^{2}\right)$ for all data | 0.118 | 0.1366 | 0.111 |
| $\mathrm{GOF}^{\text {c }}$ | 1.00 | 1.06 | 0.97 |
| Residual electron density [ $\left.\mathrm{e} / \AA^{3}{ }^{3}\right]$ | 0.28, -0.25 | 0.25, -0.22 | 0.61, -0.30 |
| CCDC no. | 1542107 | 1542108 | 1542109 |
| $\begin{aligned} & { }^{a} R_{\text {int }}=\Sigma\left\|F_{\mathrm{o}}^{2}-F_{\mathrm{o}, \text { mean }}{ }^{2}\right\| \Sigma F_{\mathrm{o}}{ }^{2} \\ & { }^{b} R(F)=\Sigma\| \| F_{\mathrm{o}}\left\|-\left\|F_{\mathrm{c}}\right\|\right\| / \Sigma \mid \\ & {\left[\Sigma\left(w\left(F_{\mathrm{o}}{ }^{2}-F_{\mathrm{c}}^{2}\right)^{2}\right) /\left(N_{\text {diffrs }}-N_{\text {params }}\right)\right]^{1}} \end{aligned}$ | $\mid, w R\left(F^{2}\right)=[\Sigma$ | $\left.v\left(F_{\mathrm{o}}^{2}-F_{\mathrm{c}}^{2}\right)^{2}\right) /(\Sigma w($ | $\left.\left.\left.F_{0}^{2}\right)^{2}\right)\right]^{1 / 2} . \quad{ }^{c} \text { GOF }=$ |



Figure Exp3. Platon plot of 11a showing displacement ellipsoids at a $50 \%$ probability level


Figure Exp4. Platon plot of 11c showing displacement ellipsoids at a $50 \%$ probability level


Figure Exp5. Platon plot of 11e showing displacement ellipsoids at a $50 \%$ probability level

Table Exp3. X-ray crystallographic data for 9, ${ }^{9}$ '-spirobifluorenes 13a and 14b

|  | 13a | 14b |
| :---: | :---: | :---: |
| Formula | $\mathrm{C}_{49} \mathrm{H}_{40} \mathrm{O}$ | $\mathrm{C}_{50} \mathrm{H}_{38}, \mathrm{CHCl}_{3}$ |
| M.w. | 644.81 | 758.17 |
| Crystal system | triclinic | triclinic |
| Space group [No.] | P1 (No 2) | P1 (No 2) |
| $a[\AA]$ | 8.8276 (3) | 9.9228(4) |
| $b[\AA]$ | 11.6527 (3) | 12.9083(6) |
| $c[\AA]$ | 18.1740 (5) | 16.2089(7) |
| $a\left[{ }^{\circ}\right]$ | 91.5340(10) | 75.028(2) |
| $\beta\left[{ }^{\circ}\right]$ | 90.2770(10) | 85.120(2) |
| $\gamma\left[{ }^{\circ}\right]$ | 107.9720(10) | 71.846(2) |
| Z | 2 | 2 |
| $\mathrm{V}\left[\AA^{3}\right]$ | 1777.45(9) | 1905.80(15) |
| $\mathrm{D}_{x}\left[\mathrm{~g} \mathrm{~cm}^{-3}\right]$ | 1.205 | 1.321 |
| Crystal size [mm] | $0.24 \times 0.16 \times 0.15$ | $0.15 \times 0.17 \times 0.27$ |
| Crystal color, shape | a colorless prism | a colorless prism |
| $\mu\left[\mathrm{mm}^{-1}\right]$ | 0.533 | 0.278 |
| $T_{\text {min }}, T_{\text {max }}$ | 0.86, 0.96 | 0.93, 0.96 |
| Measured reflections | 34479 | 42155 |
| Independent diffractions ( $R_{\text {int }}{ }^{\text {a }}$ ) | 6986 (0.029) | 8754 (0.067) |
| Observed diffract. [I>2 $\sigma(\mathrm{I}$ )] | 5694 | 7212 |
| No. of parameters | 455 | 490 |
| $R^{\text {b }}$ | 0.043 | 0.042 |
| $w R\left(F^{2}\right)$ for all data | 0.110 | 0.124 |
| GOF ${ }^{\text {c }}$ | 1.04 | 1.05 |
| Residual electron density [ $\left.\mathrm{e} / \AA^{3}{ }^{3}\right]$ | 0.19, -0.21 | 0.43, -0.68 |
| CCDC no. | 1542110 | 1542111 |
| ${ }^{a} R_{\text {int }}=\Sigma\left\|F_{\mathrm{o}}{ }^{2}-F_{\mathrm{o}, \text { mean }}{ }^{2}\right\| / \Sigma F_{\mathrm{o}}{ }^{2}$ |  |  |
| $\begin{aligned} & { }^{b} R(F)=\Sigma\| \| F_{\mathrm{o}}\left\|-\left\|F_{\mathrm{c}}\right\|\right\| / \Sigma \mid \\ & {\left[\Sigma\left(w\left(F_{\mathrm{o}}^{2}-F_{\mathrm{c}}{ }^{2}\right)^{2}\right) /\left(N_{\text {diffrs }}-N_{\text {params }}\right)\right]} \end{aligned}$ | $F_{\text {o }} \mid, w R\left(F^{2}\right)=[\Sigma$ | $\left.\left.w\left(F_{\mathrm{o}}{ }^{2}-F_{\mathrm{c}}^{2}\right)^{2}\right) /\left(\Sigma w\left(F_{\mathrm{o}}{ }^{2}\right)^{2}\right)\right]^{1 / 2}$. |



Figure Exp6. Platon plot of 13a show displacement ellipsoids at a 50\% probability level


Figure Exp7. Platon plot of $\mathbf{1 4 b}$ show displacement ellipsoids at a $50 \%$ probability level

Table Exp4. X-ray crystallographic data for 9, ${ }^{\prime}$-spirobifluorenes $\mathbf{1 8 b}$ and carborane-14a

|  | 18b | carborane-14a |
| :---: | :---: | :---: |
| Formula | $\mathrm{C}_{46} \mathrm{H}_{38}$ | $\mathrm{C}_{51} \mathrm{H}_{50} \mathrm{~B}_{10} \mathrm{O}$ |
| M.w. | 590.76 | 787.01 |
| Crystal system | monoclinic | monoclinic |
| Space group [No.] | Pl (21/c1) | P21/c |
| $a[\AA]$ | 16.8419(5) | 16.9235(4) |
| $b$ [ $\AA$ ] | 8.6097(2) | 9.1410(2) |
| $c[\AA]$ | 23.3066(6) | 28.7305(7) |
| $a\left[{ }^{\circ}\right]$ |  |  |
| $\beta\left[{ }^{\circ}\right]$ | 98.9400(10) | 100.7280(10) |
| $\gamma\left[{ }^{\circ}\right]$ |  |  |
| Z | 4 | 4 |
| $\mathrm{V}\left[\AA^{3}\right]$ | 3338.49(15) | 4366.86(18) |
| $\mathrm{D}_{x}\left[\mathrm{~g} \mathrm{~cm}^{-3}\right]$ | 1.175 | 1.197 |
| Crystal size [mm] | $0.22 \times 0.13 \times 0.04$ | $0.35 \times 0.20 \times 0.12$ |
| Crystal color, shape | a colorless bar | a colorless plate |
| $\mu\left[\mathrm{mm}^{-1}\right]$ | 0.498 | 0.489 |
| $T_{\text {min }}, T_{\text {max }}$ | 0.82, 0.98 | 0.85, 0.94 |
| Measured reflections | 25788 | 48496 |
| Independent diffractions ( $R_{\text {int }}{ }^{\text {a }}$ ) | 6307 (0.051) | 8591 (0.034) |
| Observed diffract. [I>2 $\sigma(\mathrm{I})$ ] | 5047 | 7272 |
| No. of parameters | 418 | 603 |
| $R^{\text {b }}$ | 0.057 | 0.045 |
| $w R\left(F^{2}\right)$ for all data | 0.137 | 0.121 |
| $\mathrm{GOF}^{\text {c }}$ | 1.057 | 1.03 |
| Residual electron density [ $\left.\mathrm{e} / \AA^{3}{ }^{3}\right]$ | 0.43, -0.27 | 0.22, -0.24 |
| CCDC no. | 1542112 | 1542113 |
| ${ }^{\bar{a}} R_{\text {int }}=\Sigma\left\|F_{\mathrm{o}}{ }^{2}-F_{\text {o,mean }}{ }^{2}\right\| / \Sigma F_{0}{ }^{2}$ |  |  |
| ${ }^{b} R(F)=\Sigma\| \| F_{\mathrm{o}}\left\|-\left\|F_{\mathrm{c}}\right\|\right\| / \Sigma\left\|F_{\mathrm{o}}\right\|, w R\left(F^{2}\right)=\left[\Sigma\left(w\left(F_{\mathrm{o}}{ }^{2}-F_{\mathrm{c}}{ }^{2}\right)^{2}\right) /\left(\Sigma w\left(F_{\mathrm{o}}^{2}\right)^{2}\right)\right]^{1 / 2}$. |  |  |



Figure Exp8. Platon plot of $\mathbf{1 8 b}$ showing displacement ellipsoids at a $50 \%$ probability level


Figure Exp9. Platon plot of carborane-14a show displacement ellipsoids at a $50 \%$ probability level

### 6.2.6. Photophysical analysis of $\mathbf{9 , 9}$ '-spirobifluorenes

## Chemicals.

9,10-Diphenylanthracene (DPA) was purchased from Sigma-Aldrich.

The UV/Vis absorption spectra were recorded using Unicam 340 spectrometer. Corrected steady-state emission spectra were recorded on an Aminco Bowman (AB2) spectrometer.

Quantum yields of fluorescence of a sample $\left(\Phi_{\mathrm{s}}\right)$ were calculated relative to diphenyleneanthracene DPA ( $\Phi_{\text {ref }}=0.97$ in cyclohexane) as a reference. $\Phi_{\mathrm{s}}$ was determined according to the following equation. ${ }^{70}$

$$
\begin{equation*}
\Phi_{\mathrm{s}}=\Phi_{\mathrm{ref} \times} \times \mathrm{A}_{\mathrm{s}} / \mathrm{A}_{\mathrm{ref}} \tag{1}
\end{equation*}
$$

where $A_{s}$ and $A_{\text {ref }}$ represent integrated areas of the emission (in arbitrary units) of a sample and the reference, respectively.

Micromolar solutions of the reference (DPA) and the samples were prepared in cyclohexane. The absorption of the reference and the samples were the same at the selected excitation wavelength ( $\mathrm{A} \leq 0.1$ ). Quantum yields were calculated from 5 independent measuring. The average value is reported.

The emission quantum yields ( $\Phi_{\mathrm{s}}$ ) of the derivatives were determined in cyclohexane by standard procedures with DPA as reference (Figures Exp10-13 and tables Exp5-8).

## UV/Vis absorption and emission spectra of 6a-d.



6a


6b | $\begin{array}{r}-D P A \\ -6 a \\ -6 b \\ -6 c\end{array}$ |
| :---: |
| $-6 d$ |



6c


6d



Figure Exp10. Absorption spectra of $\mathbf{6 a - d}$ in cyclohexane (left). Normalized and corrected emission spectra of $10^{-6} \mathrm{M}$ samples in cyclohexane. Emission after excitation at $\lambda=240-243$ nm (right).

Table Exp5. Photophysical properties of 6a-d derivatives in cyclodexane

| Sample | $\lambda_{\text {abs }} / \mathbf{n m}\left(\varepsilon / 10^{4} \mathbf{~ m o l}^{-1} \mathbf{d m}^{\mathbf{3}} \mathbf{c m}^{-1}\right)$ | $\lambda_{\text {exc }} / \mathbf{n m}$ | $\lambda_{\text {em }} / \mathbf{n m}$ | $\Phi_{\text {s }}{ }^{\text {a }}$ |
| :---: | :---: | :---: | :---: | :---: |
| 6 a | 217 (4.8) | 241 | 333 | 0.62 |
|  | 283 (1.6) |  |  |  |
|  | 300 (0.9) |  |  |  |
|  | 312 (1.2) |  |  |  |
| 6b | 218 (5.0) | 241 | 315, 328 | 0.72 |
|  | 233 (5.1) |  |  |  |
|  | 271 (2.3) |  |  |  |
|  | 282 (2.0) |  |  |  |
|  | 312 (1.5) |  |  |  |
| 6 c | 219 (4.4) | 242 | 315, 328 | 0.87 |
|  | 260 (1.9) |  |  |  |
|  | 282 (1.7) |  |  |  |
|  | 312 (1.3) |  |  |  |
| 6d | 218 (5.1) | 243 | 330 | 0.48 |
|  | 232 (5.3) |  |  |  |
|  | 270 (2.2) |  |  |  |
|  | 282 (1.9) |  |  |  |
|  | 312 (1.4) |  |  |  |

${ }^{\text {a }}$ Quantum yields were calculated relative to DPA ( $\Phi_{\mathrm{r}}=0.97$ ).

## UV/Vis absorption and emission spectra of 11a-g.



11a


11d


11b



11e


11c



Figure Exp11. Absorption spectra of 11a-g in cyclohexane (left). Normalized and corrected emission spectra of $10^{-6} \mathrm{M}$ samples in cyclohexane. Emission after excitation at $\lambda=240-243$ nm (right).

Table Exp6. Photophysical properties of 11a-g derivatives in cyclohexane

| Sample | $\lambda_{\text {abs }} / \mathbf{n m}\left(\varepsilon / 10^{4} \mathrm{~mol}^{-1} \mathbf{d m}^{\mathbf{3}} \mathrm{cm}^{-1}\right)$ | $\lambda_{\text {exc }} / \mathbf{n m}$ | $\lambda_{\text {em }} / \mathbf{n m}$ | $\Phi_{\text {s }}{ }^{\text {a }}$ |
| :---: | :---: | :---: | :---: | :---: |
| 11a | 262 (sh) | 241 | 330 | 0.50 |
|  | 269 (sh) |  |  |  |
|  | 300 (sh) |  |  |  |
|  | 313 (1.6) |  |  |  |
| 11b | 262 (sh) | 241 | 331 | 0.61 |
|  | 284 (sh) |  |  |  |
|  | 300 (sh) |  |  |  |
|  | 313 (1.6) |  |  |  |
| 11c | 250 (6.0) | 243 | 367 | 0.87 |
|  | 255 (sh) |  |  |  |
|  | 300 (sh) |  |  |  |
|  | 313 (1.7) |  |  |  |
| 11d | 261 (sh) | 242 | 333 | 0.66 |
|  | 284 (sh) |  |  |  |
|  | 300 (sh) |  |  |  |
|  | 313 (1.9) |  |  |  |
| 11e | 262 (sh) | 241 | 344 | 0.61 |
|  | 284 (sh) |  |  |  |
|  | 300 (sh) |  |  |  |
|  | 313 (1.5) |  |  |  |
| 11f | 263 (sh) | 240 | 361 | 0.64 |
|  | 274 (sh) |  |  |  |
|  | 300 (sh) |  |  |  |
|  | 313 (1.6) |  |  |  |

${ }^{\text {a }}$ Quantum yields were calculated relative to DPA ( $\Phi_{\mathrm{r}}=0.97$ ).

## UV/Vis absorption and emission spectra of 13 and 14.






Figure Exp12. Absorption spectra of 13 and 14 in cyclohexane (left). Normalized and corrected emission spectra of $10^{-6} \mathrm{M}$ samples in cyclohexane. Emission after excitation at $\lambda=$ 248-255 nm (right).

Table Exp7. Photophysical properties of $\mathbf{1 3}$ and $\mathbf{1 4}$ derivatives in cyclohexane

| Sample | $\lambda_{\text {abs }} / \mathbf{n m}\left(\varepsilon / 10^{5} \mathbf{m o l}^{-1} \mathbf{d m}^{\mathbf{3}} \mathbf{c m}^{-1}\right)$ | $\lambda_{\text {exc }} / \mathbf{n m}$ | $\lambda_{\text {em }} / \mathrm{nm}$ | $\boldsymbol{\Phi}_{\text {s }}{ }^{\text {a }}$ |
| :---: | :---: | :---: | :---: | :---: |
| 13a | 216 (1.1) | 254 | 366 | 0.65 |
|  | 264 (0.9) |  |  |  |
|  | 284 (sh) |  |  |  |
|  | 313 (sh) |  |  |  |
| 13b | 218 (1.1) | 251 | 369 | 0.99 |
|  | 250 (sh) |  |  |  |
|  | 261 (sh) |  |  |  |
|  | 300 (sh) |  |  |  |
| 14a | 220 (1.4) | 253 | 367 | 0.82 |
|  | 285 (sh) |  |  |  |
|  | 296 (sh) |  |  |  |
|  | 313 (sh) |  |  |  |
| 14b | 220 (1.6) | 254 | 369 | 0.79 |
|  | 305 (sh) |  |  |  |
|  | 312 (sh) |  |  |  |
|  | 328 (sh) |  |  |  |
| 14c | 220 (1.0) | 248 | 370 | 1.00 |
|  | 266 (sh) |  |  |  |
|  | 291 (sh) |  |  |  |
|  | 300 (sh) |  |  |  |
|  | 312 (sh) |  |  |  |

[^1]
## UV/Vis absorption and emission spectra of 18a-d.



18a


18b


18c


18d



Figure Exp13. Absorption spectra of 18a-d in cyclohexane (left). Normalized and corrected emission spectra of $10^{-6} \mathrm{M}$ samples in cyclohexane. Emission after excitation at $\lambda=269 \mathrm{~nm}$ (right).

Table Exp8. Photophysical properties of 18a-d derivatives in cyclohexane

| Sample | $\lambda_{\text {abs }} / \mathbf{n m}\left(\varepsilon / 10^{5} \mathrm{~mol}^{-1} \mathrm{dm}^{\mathbf{3}} \mathrm{cm}^{-1}\right)$ | $\lambda_{\text {exc }} / \mathbf{n m}$ | $\lambda_{\text {em }} / \mathrm{nm}$ | $\boldsymbol{\Phi}_{\mathrm{s}}{ }^{\text {a }}$ |
| :---: | :---: | :---: | :---: | :---: |
| 18a | 229 (1.2) | - | - | 0.00 |
|  | 270 (sh) |  |  |  |
|  | 282 (sh) |  |  |  |
|  | 299 (sh) |  |  |  |
|  | 312 (0.2) |  |  |  |
| 18b | 215 (sh) | 267 | 353 | 0.12 |
|  | 228 (sh) |  | 370 |  |
|  | 231 (sh) |  | 391 |  |
|  | 300 (sh) |  | 413 |  |
|  | 312 (0.1) |  |  |  |
| 18c | 219 (sh) | 270 | 392 | 0.80 |
|  | 260 (1.1) |  | 405 |  |
|  | 312 (0.1) |  | 429 |  |
|  | 380 (0.1) |  | 454 |  |
|  | 401 (0.1) |  | 485 |  |
| 18d | 237 (sh) | 267 | 378 | 0.11 |
|  | 246 (1.0) |  | 389 |  |
|  | 278 (0.7) |  | 398 |  |
|  | 328 (0.3) |  |  |  |
|  | 345 (0.5) |  |  |  |

${ }^{\text {a }}$ Quantum yields were calculated relative to DPA ( $\Phi_{\mathrm{r}}=0.97$ ).

### 6.3. Synthesis of dispiroindeno $[2,1-c]$ fluorenes

### 6.3.1. Synthesis of starting material

## A: General procedure for Sonogashira reaction ${ }^{60}$ (preparation of 19, 30 and 37).

 Starting halo-arylaldehyde ( 2.0 mmol ), $\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}(69 \mathrm{mg}, 0.1 \mathrm{mmol}, 5 \mathrm{~mol} \%)$ and $\mathrm{CuI}(38$ $\mathrm{mg}, 0.2 \mathrm{mmol}, 10 \mathrm{~mol} \%$ ) were dissolved in a Schlenk flask with reflux condensor in triethylamine ( 7 mL ) and THF ( 7 mL ), afterwards ethynyl-arylaldehyde ( 2.4 mmol ) was added and the reaction was stirred 3 h under reflux. The reaction mixture was cooled down, filtered off with celite/silica and washed with diethyl ether. The organic fraction was concentrated on vacuum evaporator. Column chromatography of the residue on silica gel (hexanes/EtOAc) provided products.2,2'-(Ethyne-1,2-diyl)dibenzaldehyde (19). With 2-ethynylbenzaldehyde ( $650 \mathrm{mg}, 5.0$
 mmol ) and 2-bromobenzaldehyde ( $580 \mu \mathrm{~L}, 5.0 \mathrm{mmol}$ ) following the general procedure A. Column chromatography ( $5 / 1$ hexanes/EtOAc) yielded 1.24 g ( $96 \%$ ) of the title compound as a light brown solid: mp (decomp) 113-118 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 10.59(\mathrm{~d}, J=0.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.94$ (ddd, $J$ $=7.8,1.3,0.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.67(\mathrm{ddd}, J=7.7,1.3,0.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.61(\mathrm{dt}, J=7.4,1.4 \mathrm{~Hz}, 2 \mathrm{H})$, $7.50(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 190.9,136.0,133.8,133.5,129.3,128.0,125.3$, 91.6; IR (KBr) $v_{\max } 3366,3135,3083,3067,3025,2998,2833,2747,1988,1951,1869,1697$, $1636,1593,1567,1484,1454,1390,1291,1263,1194,1159,1094,954,865,814,762,642$, $525,434 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{O}_{2}\left(\mathrm{M}^{+}\right)$calcd: 234.0681, found: 234.0679; $R_{f}(2 / 1$ hexanes $/ E t O A c$ ) $=0.38$ (silica gel plate). The spectral data were in accordance with previously published results. ${ }^{71}$

2-((2-Formyl-4-(trifluoromethyl)phenyl)ethynyl)-5-methoxybenzaldehyde (30). With 2-
 ethynyl-5-methoxybenzaldehyde ${ }^{57}(235 \mathrm{mg}, 1.4 \mathrm{mmol})$ and 2-bromo-5-(trifluoromethyl)benzaldehyde ${ }^{58}$ ( $225 \mathrm{mg}, 0.9 \mathrm{mmol}$ ) following the general procedure A. Column chromatography (5/1 hexanes/EtOAc) yielded 305 mg as a mixture of the title compound with inseparable side product in the ratio of 1:0.4 ( $217 \mathrm{mg}, 73 \%$ ) as a light brown oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 10.62(\mathrm{~s}, 1 \mathrm{H}), 10.56(\mathrm{~s}, 1 \mathrm{H}), 8.22(\mathrm{~s}, 1 \mathrm{H}), 7.84(\mathrm{dd}, J=8.2,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.81(\mathrm{~s}, 1 \mathrm{H}), 7.64$ $(\mathrm{d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.47(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.18(\mathrm{dd}, J=8.6,2.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$

NMR (100 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 190.5,189.7,160.9,137.8,135.8,135.2,134.7,133.9,125.1$ (q, $\left.{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.7 \mathrm{~Hz}\right), 121.7,120.6-126.4\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=382.4 \mathrm{~Hz}\right), 121.5,111.4,110.5,105.0,94.5$, 90.4, 89.0, 78.9, 55.8; IR (KBr) $v_{\max } 3363,3083$, 3064, 2953, 2865, 2761, 2206, 1697, 1600, 1557, 1505, 1489, 1433, 1392, 1330, 1286, 1252, 1228, 1190, 1122, 1094, 1068, 1026, 936, 879, 834, 730, $633 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{18} \mathrm{H}_{11} \mathrm{O}_{3} \mathrm{~F}_{3}\left(\mathrm{M}^{+}\right)$calcd: 332.0660, found: 332.0662; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.44$ (silica gel plate) .

1,1'-(Ethyne-1,2-diyl)bis(2-naphthaldehyde) (37). With 1-ethynyl-2-naphthaldehyde ${ }^{71}$ (350
 $\mathrm{mg}, 1.5 \mathrm{mmol}$ ) and 1-bromo-2-naphthaldehyde ( $260 \mathrm{mg}, 1.4 \mathrm{mmol}$ ) following the general procedure A. Filtration of the crude mixture with $5 / 1$ hexanes/EtOAc yielded $490 \mathrm{mg}(96 \%)$ of the title compound as a light yellow solid: mp (decomp) $126-131{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $10.98(\mathrm{~s}, 1 \mathrm{H}), 10.97(\mathrm{~s}, 1 \mathrm{H}), 8.66(\mathrm{~m}, 2 \mathrm{H}), 8.06(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 8.00(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 2 \mathrm{H})$, $7.96(\mathrm{~m}, 2 \mathrm{H}), 7.72-7.75(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 191.2,135.8,134.9,133.3$, 130.0, 129.6, 128.8, 128.3, 127.0, 125.8, 122.5, 95.4; IR (KBr) $v_{\max } 3342,3074,2838,2238$, $1693,1617,1589,1461,1434,1397,1371,1336,1248,1231,867,822,780,745 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{24} \mathrm{H}_{14} \mathrm{O}_{2}\left(\mathrm{M}^{+}\right)$calcd: 334.0994, found: 334.0995; $R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=$ 0.46 (silica gel plate). The spectral data were in accordance with previously published results. ${ }^{73}$

## B: General procedure for alkynylation reaction ${ }^{60}$ (preparation of 20, 24, 25 and 38).

 $n$-BuLi 1.6M ( $5.3 \mathrm{mmol}, 3 \mathrm{eq}$ ) was added dropwise to a solution of alkyne compound ( 5.3 mmol, 3 eq ) in anhydrous THF ( 15 mL ), at $-78^{\circ} \mathrm{C}$. After 30 min of stirring, dialdehyde compound ( 1.78 mmol ) in THF ( 5 mL ) was added and the reaction mixture was stirred for 5 $\min$ at $-78{ }^{\circ} \mathrm{C}$, afterwards it was allowed to warm up to ambient temperature and the reaction was stirred 3 h . Then $\mathrm{NH}_{4} \mathrm{Cl}$ aq was used to quench the reaction mixture, it was extracted with diethyl ether ( $3 \times 15 \mathrm{~mL}$ ), the combined organic fractions were washed with a saturated solution of $\mathrm{NaHCO}_{3}$ and brine, the organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$, filtered, and concentrated under reduced pressure. Column chromatography of the residue on silica gel (hexanes/EtOAc) provided products.

With 19 ( $234 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) and 4-ethynylanisole ( $390 \mu \mathrm{~L}, 3.0$ mmol) following the general procedure $B$. Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 470 mg ( $94 \%$ ) of the title compound as a light brown solid: mp (decomp) 57-62 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.83(\mathrm{dd}, J=7.7,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.76(\mathrm{dd}, J=7.6,1.5 \mathrm{~Hz}$, $1 \mathrm{H}), 7.60(\mathrm{dd}, J=7.3,1.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.35-7-41(\mathrm{~m}, 6 \mathrm{H}), 7.32(\mathrm{dt}, J=7.5,1.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.77(\mathrm{~m}$, $4 \mathrm{H}), 6.21(\mathrm{~s}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 1 \mathrm{H}), 3.75(\mathrm{~s}, 3 \mathrm{H}), 3.74(\mathrm{~s}, 3 \mathrm{H}) 3.70(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 159.58,159.56,142.3,142.2,133.2,132.7,132.6,128.9,128.8,128.19,128.17$, 127.2, 126.9, 121.7, 121.6, 114.5, 114.4, 113.8, 92.2, 86.9, 86.81, 86.80, 86.79, 63.7, 63.6, 55.1; IR (KBr) $v_{\max } 3372,3065,2958,2836,2226,1605,1569,1509,1490,1443,1291,1249$, 1173, 1106, 1032, 963, 832, 761, $536 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{34} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 521.17233 found: 521.17222; $R_{f}(1 / 1$ hexanes $/ E t O A c)=0.38$ (silica gel plate) .

1,1'-(Ethyne-1,2-diylbis(2,1-phenylene))bis(3-(p-tolyl)prop-2-yn-1-ol) (20b). With 19 (106
 $\mathrm{mg}, 0.45 \mathrm{mmol}$ ) and 1-ethynyl-4-p-toluene ( $170 \mu \mathrm{~L}, 1.35 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 187 mg ( $89 \%$ ) of the title compound as a red solid: mp (decomp) $152-157{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.85$ (dd, $J=7.7,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.78(\mathrm{dd}, J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.62(\mathrm{~m}, 2 \mathrm{H}), 7.34-7-44(\mathrm{~m}, 8 \mathrm{H})$, 7.10 (m, 4H), 6.20 (s, 1H), 6.19 (s, 1H), 3.51 (s, 1H), 3.26 ( $\mathrm{s}, 1 \mathrm{H}), 2.33$ (s, 3H), 2.32 ( $\mathrm{s}, 3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.2,142.1,138.71,138.66,132.8,132.7,131.7,129.02$, 129.00, 128.97, 128.39, 128.37, 127.3, 127.1, 121.83, 121.80, 119.3, 92.31, 92.26, 87.37, 87.31, 87.30, 87.2, 63.95, 63.88, 21.47, 21.46; IR (KBr) $v_{\max } 3324,3061,3028,2920,2860$, 2232, 2193, 1635, 1601, 1509, 1489, 1448, 1377, 1307, 1270, 1181, 1024, 964, 816, 760, 526 $\mathrm{cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{34} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 489.18250 found: 489.18242; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.31($ silica gel plate $)$.

1,1'-(Ethyne-1,2-diylbis(2,1-phenylene))bis(3-phenylprop-2-yn-1-ol) (20c). With 19 (468

 $\mathrm{mg}, 2.0 \mathrm{mmol}$ ) and phenylacetylene ( $660 \mu \mathrm{~L}, 6.0 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 730 mg ( $84 \%$ ) of the title compound as a light brown solid: mp (decomp) 177-182 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.83(\mathrm{~d}, J=7.5 \mathrm{~Hz}$, $1 \mathrm{H}), 7.76(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.60(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.31-7-46(\mathrm{~m}, 10 \mathrm{H}), 7.25(\mathrm{~m}, 4 \mathrm{H})$,
$6.20(\mathrm{~s}, 2 \mathrm{H}), 3.79(\mathrm{~s}, 1 \mathrm{H}), 3.50(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.1,132.8,131.8$, $129.0,128.5,128.4,128.2,127.3,127.0,122.4,121.8,92.3,88.1,87.04,86.97,63.82,63.75$; IR (KBr) $v_{\max } 3276,3055,2921,2869,2231,1598,1489,1451,1282,1182,1105,1037,970$, 913, 752, 690, $525 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{32} \mathrm{H}_{22} \mathrm{O}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 461.15120, found: 461.15127; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.24$ (silica gel plate $)$.

## 1,1'-(Ethyne-1,2-diylbis(2,1-phenylene))bis(3-(4-bromophenyl)prop-2-yn-1-ol)



With 19 ( $234 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) and 1-bromo-4-ethynylbenzene ( 543 $\mathrm{mg}, 3.0 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 455 mg ( $77 \%$ ) of the title compound as a light brown solid: mp (decomp) $181-186{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.73-7.87(\mathrm{~m}, 2 \mathrm{H}), 7.62(\mathrm{~m}, 2 \mathrm{H}), 7.35-7.48(\mathrm{~m}, 10 \mathrm{H}), 7.28-7-33$ $(\mathrm{m}, 4 \mathrm{H}), 6.21(\mathrm{~m}, 1 \mathrm{H}), 6.18(\mathrm{~m}, 1 \mathrm{H}), 3.44(\mathrm{bs}, 1 \mathrm{H}), 3.19(\mathrm{bs}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 142.0,132.9,132.8,131.8,129.1,129.0,128.6,128.5,128.47,128.4,128.3,127.3$, 127.1, 122.4, 121.8, 92.3, 88.0, 87.2, 63.94, 63.88; IR (KBr) $v_{\max } 3286,3064,2883,2229$, $1587,1570,1487,1451,1394,1283,1181,1071,1027,1013,968,822,754,690,524 \mathrm{~cm}^{-1}$; HRMS $\left(\mathrm{m} / \mathrm{z}\right.$ ) for $\mathrm{C}_{32} \mathrm{H}_{20} \mathrm{O}_{2} \mathrm{Br}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 616.97223 found: 616.97233; $R_{f}(2 / 1$ hexanes $/$ EtOAc $)=0.29($ silica gel plate $)$.

## 1,1'-(Ethyne-1,2-diylbis(2,1-phenylene))bis(3-(4-(trifluoromethyl)phenyl)prop-2-yn-1-ol)


(20e). With 19 ( $234 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) and 1-ethynyl-4(trifluoromethyl)benzene ( $490 \mu \mathrm{~L}, 3.0 \mathrm{mmol}$ ) following the general procedure B. Column chromatography (2/1 hexanes/EtOAc) yielded 483 mg ( $86 \%$ ) of the title compound as a light brown solid: mp (decomp) $149-154{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.74$ (dd, $J=7.8$, $1.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.63(\mathrm{dd}, J=7.3,1.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.52-7.57(\mathrm{~m}, 8 \mathrm{H}), 7.42(\mathrm{dt}, J=7.5,1.6 \mathrm{~Hz}, 2 \mathrm{H})$, $7.38(\mathrm{dd}, J=7.5,1.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.21(\mathrm{~s}, 2 \mathrm{H}), 3.65(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 141.6, 133.0, 132.0, 129.2, 128.7, 127.3, $125.2\left(\mathrm{q}^{2}{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}\right), 121.8,120.6-126.4\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}\right.$ $=188.4 \mathrm{~Hz}), 92.3,90.4,85.6,63.9$; IR (KBr) $v_{\max } 3281,3089,2945$, 2927, 2861, 2230, 1615, 1489, 1453, 1405, 1328, 1161, 1118, 1070, 1027, 1018, 969, 843, $756 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{34} \mathrm{H}_{20} \mathrm{O}_{2} \mathrm{~F}_{6} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 597.12597 found: 597.12606; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.38$ (silica gel plate).

1,1'-(Ethyne-1,2-diylbis(2,1-phenylene))bis(3-(thiophen-3-yl)prop-2-yn-1-ol) (20f). With
 19 ( $234 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) and 3-ethynyl-thiophene ( $300 \mu \mathrm{~L}, 3.0 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 398 mg ( $88 \%$ ) of the title compound as a light brown solid: mp (decomp) $57-62{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.81$ (dd, $J=7.7,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.75$ (dd, $J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.61(\mathrm{dd}, J=7.5,1.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.47$ $(\mathrm{m}, 2 \mathrm{H}), 7.38-7.43(\mathrm{~m}, 2 \mathrm{H}), 7.35(\mathrm{dt}, J=7.5,1.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.24(\mathrm{t}, J=3.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{t}, J=$ $3.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.13 (ddd, $J=5.0,2.2,1.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.19 (s, 1H), 6.18 ( $\mathrm{s}, 1 \mathrm{H}), 3.62(\mathrm{~s}, 1 \mathrm{H}), 3.36$ $(\mathrm{s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 162.4,142.0,141.9,132.9,132.7,129.9,129.40$, $129.38,129.04,129.00,128.42,128.40,127.2,127.0,125.29,125.26,121.73,121.68,121.4$, $92.28,92.27,87.70,87.67,82.3,82.2,63.9,63.8$; IR (KBr) $v_{\max } 3527,3334,3106,3061$, 3022, 2869, 2231, 1599, 1520, 1489, 1449, 1358, 1270, 1177, 1027, 986, 864, 783, 760, 626 $\mathrm{cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{28} \mathrm{H}_{18} \mathrm{O}_{2} \mathrm{~S}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 473.06404 found: 473.06415; $R_{f}(2 / 1$ hexanes $/ E t O A c$ ) $=0.24$ (silica gel plate) .

1,1'-(Ethyne-1,2-diylbis(2,1-phenylene))bis(hex-2-yn-1-ol) (20g). With 19 (234 mg, 1.0
 mmol ) and 1-pentyne ( $300 \mu \mathrm{~L}, 3.0 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 351 mg ( $95 \%$ ) of the title compound as a light brown oil: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.77$ (dd, $J=7.7,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.72$ (dd, $J=7.7,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.57(\mathrm{~m}, 2 \mathrm{H}), 7.39$ (dt, $J=7.5,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.37(\mathrm{dt}, J=7.5,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.32(\mathrm{~m}, 2 \mathrm{H}), 5.94(\mathrm{~m}, 2 \mathrm{H}), 3.42(\mathrm{~s}$, $1 \mathrm{H}), 3.21(\mathrm{~s}, 1 \mathrm{H}), 2.27(\mathrm{~m}, 4 \mathrm{H}), 1.57(\mathrm{~m}, 4 \mathrm{H}), 0.99(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 0.98(\mathrm{t}, J=7.4 \mathrm{~Hz}$, 3 H ) ${ }^{13}{ }^{\mathrm{C}}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.7,142.5,132.6,132.5,128.8,128.13,128.10,127.0$, $126.8,121.70,121.68,92.1,92.0,88.0,87.9,79.3,79.2,63.5,63.4,22.0,20.84,20.82,13.53$, 13.51; IR (KBr) $v_{\max } 3558,3063,3022,2962,2933,2899,2871,2833,2277,2226,1598$, $1565,1489,1449,1379,1328,1276,1189,1133,1033,992,807,758,630 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{26} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 393.18250 found: $393.18269 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.29$ (silica gel plate).

1,1'-(Ethyne-1,2-diylbis(2,1-phenylene))bis(3-(trimethylsilyl)prop-2-yn-1-ol) (20h). With
 19 ( $234 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) and ethynyltrimethylsilane ( $430 \mu \mathrm{~L}, 3.0 \mathrm{mmol}$ ) following the general procedure B. Column chromatography (2/1 hexanes/EtOAc) yielded 361 mg ( $84 \%$ ) of the title compound as a light brown oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.77(\mathrm{dd}, J=7.7,1.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.70(\mathrm{dd}, J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.58(\mathrm{~m}, 2 \mathrm{H}), 7.39(\mathrm{~m}, 2 \mathrm{H}), 7.34(\mathrm{~m}, 2 \mathrm{H}), 3.55(\mathrm{~s}, 1 \mathrm{H}), 3.25(\mathrm{~s}$, $1 \mathrm{H}), 0.21(\mathrm{~s}, 9 \mathrm{H}), 0.19(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 141.7,141.6,132.8,132.6$, 128.92, 128.87, 128.4, 128.3, 127.3, 127.0, 121.9, 121.8, 104.2, 104.1, 92.2, 92.1, 63.8, 63.6, $-0.20,-0.21$; IR (KBr) $v_{\max } 3346,3058,3025,2959,2898,2173,1598,1490,1449,1408$, 1374, 1250, 1188, 1038, 983, 844, $759 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{O}_{2} \mathrm{NaSi}_{2}(\mathrm{M}+\mathrm{Na})$ calcd: 453.16765 found: $453.16768 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.30$ (silica gel plate) .

1,1'-(Ethyne-1,2-diylbis(2,1-phenylene))bis(3-(naphthalen-1-yl)prop-2-yn-1-ol)


With 19 ( $234 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) and 1-ethynylnaphthalene ( $430 \mu \mathrm{~L}, 3.0$ $\mathrm{mmol})$ following the general procedure B . Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 458 mg ( $85 \%$ ) of the title compound as a light brown solid: mp (decomp) $86-91{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.30$ $(\mathrm{m}, 1 \mathrm{H}), 8.25(\mathrm{~m}, 1 \mathrm{H}), 7.91(\mathrm{dd}, J=7.7,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.85(\mathrm{dd}, J=7.6,1.4$ $\mathrm{Hz}, 1 \mathrm{H}), 7.76-7.81(\mathrm{~m}, 4 \mathrm{H}), 7.68(\mathrm{~m}, 2 \mathrm{H}), 7.66(\mathrm{dt}, J=8.5,1.2 \mathrm{~Hz}, 2 \mathrm{H})$, 7.32-7-48 (m, 10H), $6.39(\mathrm{~s}, 1 \mathrm{H}), 6.37(\mathrm{~s}, 1 \mathrm{H}), 3.65(\mathrm{~s}, 1 \mathrm{H}), 3.42(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.2,142.1,133.2,133.03,132.96,132.9,130.8,129.13,129.12,129.0$, 128.49, 128.46, 128.17, 128.15, 127.3, 127.1, 126.8, 126.4, 126.3, 126.12, 126.08, 125.1, $121.8,121.7,120.98,119.95,92.9,92.5,92.4,85.2,85.1,64.2$; IR (KBr) $v_{\max } 3371,3058$, 2954, 2869, 2219, 1586, 1489, 1450, 1396, 1284, 1187, 1098, 1054, 1016, 950, 800, 773, 760 $\mathrm{cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{40} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 561.18250, found: 561.18256; $R_{f}(2 / 1$ hexanes $/$ EtOAc $)=0.28($ silica gel plate $)$.

1-(2-Iodophenyl)-3-(4-(trifluoromethyl)phenyl)prop-2-yn-1-ol (24). With 2-iodo
 benzaldehyde ( $116 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and 1-ethynyl-4-(trifluoromethyl)benzene ( $82 \mu \mathrm{~L}, 0.5 \mathrm{mmol}$ ) following the general procedure B . Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 195 mg ( $96 \%$ ) of the title compound as a colorless liquid: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.87$ (dd, $J=7.9$, $1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.80(\mathrm{dd}, J=7.8,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.58(\mathrm{~s}, 4 \mathrm{H}), 7.43(\mathrm{dt}, J=7.7,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.06$ (dt, $J=7.6,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.90(\mathrm{~d}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.59(\mathrm{~d}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100
$\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.1,139.8,132.0,130.3,128.8,128.1,122.6-130.6\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=365.7 \mathrm{~Hz}\right)$, $125.2\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}\right), 97.9,90.2,85.5,69.0$; IR (KBr) $v_{\max } 3285,3180,3061,2887,2229$, 1614, 1497, 1464, 1436, 1405, 1326, 1279, 1170, 1137, 1114, 1070, 1053, 1012, 970, 840, $752,731 \mathrm{~cm}^{-1} ;$ HRMS $(m / z)$ for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{OF}_{3} \mathrm{INa}(\mathrm{M}+\mathrm{Na})$ calcd: 424.96206, found: 424.96208; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.52($ silica gel plate $)$.

1-(2-Ethynylphenyl)-3-(4-methoxyphenyl)prop-2-yn-1-ol (25). With 2-ethynyl
 benzaldehyde ( $260 \mathrm{mg}, 2.0 \mathrm{mmol}$ ) and 4-ethynylanisole ( $260 \mu \mathrm{~L}, 2.0$ mmol ) following the general procedure B . Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 436 mg ( $83 \%$ ) of the title compound as a light brown liquid: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.78(\mathrm{dd}, J=7.8,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.53(\mathrm{dd}, J$ $=7.6,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.39-7.44(\mathrm{~m}, 3 \mathrm{H}), 7.31(\mathrm{dt}, J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{~m}, 2 \mathrm{H}), 6.09(\mathrm{~s}$, $1 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.41(\mathrm{~s}, 1 \mathrm{H}), 2.77(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 159.8$, 143.1, 133.3, 133.2, 132.8, 129.4, 129.2, 128.2, 127.4, 127.3, 126.9, 120.4, 114.5, 113.9, 86.7, 82.7, 81.9, 81.1, 63.8, 55.2; IR (KBr) $v_{\max } 3427,3288,3066,2935,2837,2227,1606,1509,1479$, 1446, 1291, 1248, 1174, 1107, 1029, 964, 832, 761, 663, $536 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{18} \mathrm{H}_{14} \mathrm{O}_{2}$ $\left(\mathrm{M}^{+}\right)$calcd: 262.0994 , found: $262.0999 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.35$ (silica gel plate).

## 1-(2-((2-(1-Hydroxy-3-(4-(trifluoromethyl)phenyl)prop-2-yn-1-yl)phenyl)ethynyl)-

 phenyl)-3-(4-methoxyphenyl)prop-2-yn-1-ol (26). With 24 ( $200 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $25(131 \mathrm{mg}, 0.5 \mathrm{mmol})$ following the general procedure A. Column chromatography (2/1 hexanes/EtOAc) yielded 244 mg ( $91 \%$ ) of the title compound as mixture as a light brown oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (major) $\delta 7.80(\mathrm{dd}, J=7.7,1.1 \mathrm{~Hz}$, $1 \mathrm{H}), 7.73(\mathrm{dd}, J=7.8,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.62(\mathrm{dt}, J=6.9,1.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.52(\mathrm{~m}, 4 \mathrm{H}), 7.34-7.44(\mathrm{~m}$, $6 \mathrm{H}), 6.77(\mathrm{~m}, 2 \mathrm{H}), 6.22(\mathrm{~s}, 1 \mathrm{H}), 6.18(\mathrm{~s}, 1 \mathrm{H}), 3.95(\mathrm{~s}, 1 \mathrm{H}), 3.76(\mathrm{~s}, 3 \mathrm{H}), 3.54(\mathrm{~s}, 1 \mathrm{H})$, (minor) $\delta 7.83(\mathrm{dd}, J=7.7,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.76(\mathrm{dd}, J=7.7,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.62(\mathrm{dt}, J=6.9,1.3 \mathrm{~Hz}, 2 \mathrm{H})$, $7.52(\mathrm{~m}, 4 \mathrm{H}), 7.34-7.44(\mathrm{~m}, 6 \mathrm{H}), 6.79(\mathrm{~m}, 2 \mathrm{H}), 6.22(\mathrm{~s}, 1 \mathrm{H}), 6.18(\mathrm{~s}, 1 \mathrm{H}), 3.77(\mathrm{~s}, 3 \mathrm{H}), 3.63$ (s, 1H), $3.29(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (major) $\delta 159.8,142.1,141.8,133.3$, 132.92, 132.86, 132.0, 129.1, 128.6, 128.4, 127.3, 127.0, 119.8-126.2 (q, ${ }^{1} J_{\mathrm{C}-\mathrm{F}}=272.4 \mathrm{~Hz}$ ), $125.2\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}\right), 121.7,113.9,92.4,92.1,90.7,87.2,86.6,85.4,64.0,63.8,55.2$, (minor) $\delta 159.8,142.1,141.7,133.3,132.8,132.7,132.0,129.1,128.5,128.4,127.3,127.0$, $119.8-126.2\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=272.4 \mathrm{~Hz}\right), 125.2\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}\right), 121.8,114.3,92.5,92.1,90.7$, 87.2, 86.6, 85.5, 63.9, 63.7, 55.2; IR (KBr) $v_{\max } 3348,3064,2928,2871,2234,1606,1509$,

1490, 1443, 1323, 1291, 1249, 1171, 1125, 1068, 1032, 1018, 965, 833, $760 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{34} \mathrm{H}_{23} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 559.14915, found: 559.14928; $R_{f}(2 / 1$ hexanes $/$ EtOAc $)=0.28($ silica gel plate $)$.

## 1-(2-((2-(1-Hydroxy-3-(p-tolyl)prop-2-yn-1-yl)-4-(trifluoromethyl)phenyl)ethynyl)-5-

 methoxyphenyl)-3-(p-tolyl)prop-2-yn-1-ol (31). With 30 (270 $\mathrm{mg}, 0.8 \mathrm{mmol}$ ) and 1-ethynyl-4-p-toluene ( $310 \mu \mathrm{~L}, 2.4 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 345 mg ( $75 \%$ ) of the title compound as mixture as a light brown solid: mp (decomp) $72-77{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.51$ (s, $1 \mathrm{H}), 7.49(\mathrm{~s}, 1 \mathrm{H}), 7.31-7.39(\mathrm{~m}, 6 \mathrm{H}), 7.07-7.12(\mathrm{~m}, 4 \mathrm{H}), 6.85(\mathrm{dt}, J=8.5,2.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.15$ (d, $J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.04(\mathrm{~d}, J=5.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.86(\mathrm{~s}, 3 \mathrm{H}), 3.76(\mathrm{~s}, 1 \mathrm{H}), 2.69(\mathrm{~s}, 1 \mathrm{H}) 2.33(\mathrm{~s}$, $6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 160.6,159.8,145.7,143.6,138.7,135.1,134.0,133.9$, 131.7, 131.7, 129.0, 127.9-143.5 (q, $\left.{ }^{1} J_{\mathrm{C}-\mathrm{F}}=254.8 \mathrm{~Hz}\right), 119.3,114.3,114.1,114.0\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=\right.$ $4.7 \mathrm{~Hz}), 113.9,112.9,112.6,112.5,112.1,90.9,87.3,87.2,86.9,79.9,77.9,63.9,63.4,55.5$, 21.5; IR (KBr) $v_{\max } 3372,3031,3004,2923,2836,2210,1605,1509,1487,1429,1329,1293$, 1272, 1168, 1126, 1095, 1028, 973, 816, $527 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{36} \mathrm{H}_{28} \mathrm{O}_{3} \mathrm{~F}_{3}(\mathrm{M}+\mathrm{H})$ calcd: 565.19851, found: 565.19818; $R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=0.34$ (silica gel plate).

## 1,1'-(Ethyne-1,2-diylbis(naphthalene-1,2-diyl))bis(3-(4-methoxyphenyl)prop-2-yn-1-ol)


(38a). With 37 ( $250 \mathrm{mg}, 0.75 \mathrm{mmol}$ ) and 4-ethynylanisole ( $290 \mu \mathrm{~L}$, 2.25 mmol ) following the general procedure B. Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 423 mg ( $94 \%$ ) of the title compound as a light brown solid: mp (decomp) $97-102{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.66(\mathrm{~m}, 2 \mathrm{H}), 8.00(\mathrm{dd}, J=8.6,4.7 \mathrm{~Hz}$, $2 \mathrm{H}), 7.92(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.88(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.56-7.63$ $(\mathrm{m}, 4 \mathrm{H}), 7.40(\mathrm{~m}, 4 \mathrm{H}), 6.78(\mathrm{~m}, 4 \mathrm{H}), 6.62(\mathrm{~s}, 2 \mathrm{H}), 3.76(\mathrm{~s}, 6 \mathrm{H}), 3.47(\mathrm{~s}, 1 \mathrm{H}), 3.28(\mathrm{~s}, 1 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 159.7,141.4,141.3,133.4,133.3,133.0,129.5,128.3,127.4$, $126.8,126.6,124.3,124.2,118.9,118.8,114.4,114.4,113.88,113.82,95.6,95.5,87.3,87.2$, 87.1, 87.0, 64.14, 64.10, 55.2; IR (KBr) $v_{\max } 3418,3055,3004,2956,2929,2836,2223,1605$, $1569,1509,1463,1440,1291,1249,1173,1107,1033,979,829,802,762 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{42} \mathrm{H}_{30} \mathrm{O}_{4} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 621.20363 , found: $621.20367 ; R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=$ 0.16 (silica gel plate).

1,1'-(Ethyne-1,2-diylbis(naphthalene-1,2-diyl))bis(3-phenylprop-2-yn-1-ol) (38b). With
 $37(84 \mathrm{mg}, 0.25 \mathrm{mmol})$ and phenylacetylene ( $83 \mu \mathrm{~L}, 0.75 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 119 mg ( $88 \%$ ) of the title compound as a light brown solid: mp (decomp) $93-98{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $8.68(\mathrm{~m}, 2 \mathrm{H}), 8.03(\mathrm{dd}, J=8.6,3.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.95(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H})$, $7.89(\mathrm{~m}, 2 \mathrm{H}), 7.54-7.61(\mathrm{~m}, 4 \mathrm{H}), 7.45(\mathrm{~m}, 4 \mathrm{H}), 7.22-7.32(\mathrm{~m}, 6 \mathrm{H}), 6.63(\mathrm{~s}, 1 \mathrm{H}), 6.62(\mathrm{~s}, 1 \mathrm{H})$, $3.47(\mathrm{~s}, 1 \mathrm{H}), 3.28(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 141.2,141.1,133.4,133.3,133.03$, $133.02,131.8,129.6,128.5,128.32,128.26,128.22,128.19,127.5,126.8,126.6,124.3$, $124.2,122.4,122.3,119.0,118.9,95.6,95.5,88.6,88.5,87.26,87.1,64.10,64.08 . ;$ IR (KBr) $v_{\max } 3371,3053,2926,2856,2224,1591,1571,1484,1443,1401,1260,1218,1173,1032$, 980, 918, 824, 755, $693 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{40} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 561.18250, found: $561.18256 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.24$ (silica gel plate) .

## 1,1'-(Ethyne-1,2-diylbis(naphthalene-1,2-diyl))bis(3-(4-(trifluoromethyl)phenyl)prop-2-


yn-1-ol) (38c). With 37 ( $84 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) and 1-ethynyl-4-(trifluoromethyl)-benzene ( $122 \mu \mathrm{~L}, 0.75 \mathrm{mmol}$ ) following the general procedure B. Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded $128 \mathrm{mg}(76 \%)$ of the title compound as a light brown solid: mp (decomp) $98-103{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.63(\mathrm{~m}$, $2 \mathrm{H}), 7.86-7.94(\mathrm{~m}, 6 \mathrm{H}), 7.54-7.62(\mathrm{~m}, 4 \mathrm{H}), 7.39-7.42(\mathrm{~m}, 4 \mathrm{H}), 7.44-$ $7.49(\mathrm{~m}, 4 \mathrm{H}), 6.63(\mathrm{~s}, 1 \mathrm{H}), 6.61(\mathrm{~s}, 1 \mathrm{H}), 3.83(\mathrm{~s}, 1 \mathrm{H}), 3.73(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 140.7,140.6,133.4,133.3,133.1,133.0,132.0,131.9,129.8,129.7,128.4,127.62$, 127.60, 127.04, 127.00, 126.5, 126.4, $126.0\left(\mathrm{q},{ }^{3} J_{\mathrm{C}-\mathrm{F}}=1.4 \mathrm{~Hz}\right), 125.0\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}\right)$, $122.9-129.8\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=261.6 \mathrm{~Hz}\right), 124.2,124.1,119.0,118.9,95.6,95.5,90.9,90.8,85.8$, 85.7, 64.1, 64.0; IR (KBr) $v_{\max } 3381,3053,2935,2856,2099,1612,1408,1318,1249,1218$, 1166, 1125, 1073, 1021, 983, 835, $762 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{42} \mathrm{H}_{24} \mathrm{O}_{2} \mathrm{~F}_{6} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 697.15727 , found: $697.15738 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.25$ (silica gel plate).

### 6.3.2. Cyclotrimerization reaction for 21, 27, 32 and 39

C: General procedure for cyclotrimerization with Wilkinson's catalyst $\mathbf{R h C l}\left(\mathbf{P P h}_{3}\right)_{3}$ (Preparation of 21, 27, 32 and 39). A dry microwave vial was charged with starting triyndiol ( 0.25 mmol ), and dissolved under argon atmosphere in THF ( 5 mL ). After addition of Wilkinson's catalyst ( $9 \mathrm{mg}, 0.0075 \mathrm{mmol}, 3 \mathrm{~mol} \%$ ) and the additive $\mathrm{Ag}_{2} \mathrm{CO}_{3}(4 \mathrm{mg}, 0.015$ $\mathrm{mmol}, 6 \mathrm{~mol} \%$ ) the reaction mixture was sealed and heated up to $180^{\circ} \mathrm{C}$ for 1.5 h in a microwave reactor. The reaction mixture was cooled down to room temperature and the solvent was evaporated under reduced pressure. Column chromatography of the residue on silica gel yielded products.

6,7-Bis(4-methoxyphenyl)-5,8-dihydroindeno[2,1-c]fluorene-5,8-diol (21a). With 20a (125 MeO chromatography ( $1 / 1$ hexanes/EtOAc) yielded 114 mg ( $92 \%$ ) of the title compound as a light brown solid: mp (decomp) $>260{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 400 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.37(\mathrm{dd}, J=7.7,3.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.60(\mathrm{~m}, 2 \mathrm{H}), 7.49(\mathrm{t}, J=$ $7.6 \mathrm{~Hz}, 3 \mathrm{H}), 7.37(\mathrm{~m}, 3 \mathrm{H}), 7.14(\mathrm{~m}, 2 \mathrm{H}), 6.80(\mathrm{~m}, 4 \mathrm{H}), 5.88(\mathrm{~s}, 1 \mathrm{H}), 5.59$ $(\mathrm{s}, 1 \mathrm{H}) 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 1.90(\mathrm{~s}, 1 \mathrm{H}), 1.78(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $158.6,147.1,145.8,139.9,137.3,129.93,129.89,128.9,128.8,127.9,125.5,125.4,123.51$, $123.47,114.1,74.3,74.1,55.2,55.1$; IR (KBr) $v_{\max } 3554,3008,2933,2837,1608,1517$, 1462, 1432, 1289, 1247, 1180, 1107, 1031, 837, 763, $736 \mathrm{~cm}^{-1} ;$ HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{34} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{Na}$ $(\mathrm{M}+\mathrm{Na})$ calcd: 521.17233 , found: $521.17249 ; R_{f}(1 / 1$ hexanes/EtOAc) $=0.43$ (silica gel plate).

6,7-Di-p-tolyl-5,8-dihydroindeno[2,1-c]fluorene-5,8-diol (21b). With 20b (45 mg, 0.1
 mmol ) following the general procedure C. Column chromatography ( $2 / 1 / 1$ hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $45 \mathrm{mg}(96 \%)$ of the title compound as a light brown solid: mp (decomp) 254-259 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 8.37 (d, $J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 8.23-8.32(\mathrm{~m}, 2 \mathrm{H}), 7.59(\mathrm{~m}, 3 \mathrm{H}), 7.49(\mathrm{~m}, 3 \mathrm{H})$, 7.34-7.43 (m, 4H), 7.02 (d, $J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.88(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.85$ $(\mathrm{d}, J=2.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.33(\mathrm{~s}, 6 \mathrm{H}), 1.74(\mathrm{~d}, J=3.9 \mathrm{~Hz}, 2 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $146.9,145.8$, 139.9, 137.4, 136.9, 135.0, 134.7, 129.4, 129.3, 128.8, 128.2, 127.8, 125.4, 123.5, 74.3, 21.3; IR (KBr) $v_{\max } 3575,3550,3518,3460,3046,3023,2947,2921,2868,1605$, $1515,1464,1431,1328,1247,1184,1105,1071,1022,933,830,763,735 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$
for $\mathrm{C}_{34} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 489.18250 , found: 489.18242; $R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=$ 0.40 (silica gel plate).

6,7-Diphenyl-5,8-dihydroindeno[2,1-c]fluorene-5,8-diol (21c). With 20c (109 mg, 0.25
 $\mathrm{mmol})$ following the general procedure C. Column chromatography ( $2 / 1 / 1$ hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $102 \mathrm{mg}(93 \%)$ of the title compound as a light brown solid: $\mathrm{mp}($ decomp $)>260{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 8.42 (dd, $J=7.8,3.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.64(\mathrm{~m}, 2 \mathrm{H}), 7.53(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}), 7.40$ (ddt, $J=7.4,2.0 .0 .9 \mathrm{~Hz}, 3 \mathrm{H}), 7.29-7.31(\mathrm{~m}, 2 \mathrm{H}), 7.17-7.25(\mathrm{~m}, 6 \mathrm{H}), 5.90(\mathrm{~s}, 1 \mathrm{H}), 5.62(\mathrm{~s}$, 1 H ), 1.72-1.87 (bs, 2H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 146.8,146.1,145.81,145.77,139.8$, 139.7, 138.6, 137.8, 137.7, 137.4, 135.1, 134.9, 128.9, 128.9, 128.0, 127.3, 127.2, 125.5, $125.4,123.54,123.49,74.3,74.1$; IR (KBr) $v_{\max } 3579,3554,3423,3050,3021,2945,2865$, 1604, 1464, 1441, 1422, 1330, 1179, 1101, 1067, 1029, 848, 766, 742, 714, $704 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{34} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 461.15120, found: 461.15131; $R_{f}(2 / 1$ hexanes/EtOAc) $=0.26$ (silica gel plate).

6,7-Bis(4-bromophenyl)-5,8-dihydroindeno[2,1-c]fluorene-5,8-diol (21d). With 20d (148
 $\mathrm{mg}, \quad 0.25 \mathrm{mmol}$ ) following the general procedure C. Column chromatography ( $2 / 1 / 1$ hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $124 \mathrm{mg}(84 \%)$ of the title compound as a light brown solid: mp (decomp) 222-227 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.37$ (dd, $J=7.7,2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.59 (d, $J=7.3$ $\mathrm{Hz}, 2 \mathrm{H}), 7.50(\mathrm{t}, J=7.5 \mathrm{~Hz}, 4 \mathrm{H}), 7.39(\mathrm{~m}, 4 \mathrm{H}), 7.28-7-33(\mathrm{~m}, 2 \mathrm{H}), 6.78(\mathrm{~m}$, $2 \mathrm{H}), 5.84(\mathrm{~m}, 1 \mathrm{H}), 5.78(\mathrm{~s}, 1 \mathrm{H}), 1.62(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 146.7, 145.9, $145.8,139.6,139.5,137.5,137.4,136.9,136.6,136.4,135.5,135.4,135.2,132.1,131.7$, $130.4,129.1,129.0,128.9,128.2,128.1,127.5,125.4,123.6,121.6,74.2,74.1$; IR (KBr) $v_{\max }$ 3566, 3444, 3050, 2868, 1605, 1489, 1465, 1427, 1392, 1330, 1247, 1179, 1103, 1073, 1026, 1012, 833, 761, 736, $729 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{32} \mathrm{H}_{22} \mathrm{O}_{2} \mathrm{Br}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 616.97223, found: $616.97181 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.38$ (silica gel plate $)$.

6,7-Bis(4-(trifluoromethyl)phenyl)-5,8-dihydroindeno[2,1-c]fluorene-5,8-diol (21e). With
 $\mathbf{2 0 e}(143 \mathrm{mg}, 0.25 \mathrm{mmol})$ following the general procedure C. Column chromatography ( $2 / 1 / 1$ hexanes/EtOAc/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $138 \mathrm{mg}(96 \%)$ of the title compound as a light brown solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $\left.400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.39(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.64-7.72(\mathrm{~m}, 2 \mathrm{H})$, $7.61(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 4 \mathrm{H}), 7.53(\mathrm{~m}, 4 \mathrm{H}), 7.41(\mathrm{t}, J=7.3 \mathrm{~Hz}, 4 \mathrm{H}), 7.02$ (bs, 2H), 5.79 (s, 2H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 146.7, 146.5, 145.9, 145.8, 139.7, 139.5, $137.5,136.7,136.4,135.5,129.1,122.6-129.6\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=272.1 \mathrm{~Hz}\right), 128.3,128.1,127.4$, $125.5\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=2.9 \mathrm{~Hz}\right), 123.5,121.5,74.2$; IR ( KBr ) $v_{\max } 3303,3040,2945,2923,1617$, 1467, 1406, 1328, 1164, 1120, 1069, 1021, 849, 762, $741 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{34} \mathrm{H}_{20} \mathrm{O}_{2} \mathrm{~F}_{6} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 597.12597 found: 597.12610; $R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=0.44$ (silica gel plate).

6,7-Di(thiophen-3-yl)-5,8-dihydroindeno[2,1-c]fluorene-5,8-diol (21f). With $20 f(113 \mathrm{mg}$,

0.25 mmol ) following the general procedure C . Column chromatography (2/1/1 hexanes/EtOAc/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $108 \mathrm{mg}(95 \%)$ of the title compound as a light brown solid: mp (decomp) 231-236 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.33(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.62(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.50(\mathrm{t}, J$ $=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.39(\mathrm{ddt}, J=7.4,2.8,0.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.31(\mathrm{dd}, J=7.9,3.0 \mathrm{~Hz}$, $1 \mathrm{H}), 7.27(\mathrm{~m}, 1 \mathrm{H}), 7.23(\mathrm{dd}, J=2.9,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.16(\mathrm{dd}, J=2.9,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{dd}, J=$ $4.9,1.2 \mathrm{~Hz}, 2 \mathrm{H}), 5.88(\mathrm{~s}, 1 \mathrm{H}), 5.64(\mathrm{~s}, 1 \mathrm{H}), 1.91(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 147.2, 146.7, 145.8, 145.7, 139.7, 139.6, 137.9, 137.8, 135.3, 135.1, 133.9, 132.8, 129.2, 129.0, 128.9, 128.6, 128.1, 126.1, 125.5, 125.41, 125.37, 124.2, 123.6, 123.54, 123.49, 74.3, 74.1; IR (KBr) $v_{\max } 3546,3393,3103,3029,2869,1605,1466,1440,1353,1264,1192,1070$, 1024, 966, 856, 838, 802, 761, 732, $624 \mathrm{~cm}^{-1}$; HRMS ( $m / z$ ) for $\mathrm{C}_{28} \mathrm{H}_{18} \mathrm{O}_{2} \mathrm{~S}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 473.06404 found: $473.06409 ; R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=0.30$ (silica gel plate) .

6,7-Dipropyl-5,8-dihydroindeno[2,1-c]fluorene-5,8-diol (21g). With $\mathbf{2 0 g}$ ( $92 \mathrm{mg}, 0.25$ mmol ) following the general procedure C. Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded 68 mg ( $73 \%$ ) of the title compound as a light brown solid: mp (decomp) 204-209 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.25$ (d, $J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.67(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.45(\mathrm{t}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.35$ $(\mathrm{t}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 5.69(\mathrm{~s}, 1 \mathrm{H}), 5.67(\mathrm{~s}, 1 \mathrm{H}), 3.11(\mathrm{~m}, 2 \mathrm{H}), 2.80(\mathrm{~m}, 2 \mathrm{H}), 1.69(\mathrm{~m}, 4 \mathrm{H}), 1.57$ ( $\mathrm{s}, 2 \mathrm{H}$ ) , $1.11(\mathrm{t}, J=7.2 \mathrm{~Hz}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 146.3, 145.9, 140.1, 139.7,
128.9, 128.1, 127.5, 125.2, 123.1, 74.7, 31.1, 24.9, 15.0; IR (KBr) $v_{\max } 3530,3366,3050$, 2957, 2927, 2870, 1607, 1586, 1465, 1431, 1374, 1294, 1258, 1201, 1086, 1020, 933, 842, 763, 731, $724 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{26} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 393.18250, found: 393.18264; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.39$ (silica gel plate $)$.

6,7-Bis(trimethylsilyl)-5,8-dihydroindeno[2,1-c]fluorene-5,8-diol (major) (21h-TMS), 6TMS ${ }^{\text {TMS }} \quad$ (Trimethylsilyl)-5,8-dihydroindeno[2,1-c]fluorene-5,8-diol $\quad$ (minor) (21h). With $20 \mathrm{~h}(108 \mathrm{mg}, 0.25 \mathrm{mmol})$ following the general procedure C . Column chromatography ( $2 / 1$ hexanes/EtOAc) yielded $85 \mathrm{mg}(80 \%)$ of the title compound as mixture as a light brown oil: ${ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$,
 $\left.\mathrm{CDCl}_{3}\right) \delta 8.31(\mathrm{dd}, J=7.9,2.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.70(\mathrm{~m}, 2 \mathrm{H}), 7.46(\mathrm{~m}, 2 \mathrm{H}), 7.38$ (dt, $J=7.4,0.9 \mathrm{~Hz}, 2 \mathrm{H}), 5.72(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.64(\mathrm{~m}, 1 \mathrm{H}), 1.70(\mathrm{~s}$, $1 \mathrm{H}), 1.57(\mathrm{~s}, 1 \mathrm{H}), 0.47(\mathrm{~s}, 9 \mathrm{H}), 0.24(\mathrm{~s}, 9 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ (major) 152.9, 147.4, 147.0, 146.3, 139.9, 139.8, 136.4, 130.1, 128.9, 128.5, 127.8, 127.7, $125.3,125.18,123.55,123.4,75.4,75.1,1.0,0.1 ; \delta$ (minor) 153.7, 147.9, 147.2, 146.6, 140.0, $139.9,135.8,134.7,129.8,128.8,128.5,127.7,125.2,125.0,123.6,75.6,75.3,1.0,0.1$; IR $(\mathrm{KBr}) v_{\max } 3540,3393,3067,3046,2954,2896,1606,1465,1407,1380,1248,1198,1117$, 1039, 933, 843, 760, $736 \mathrm{~cm}^{-1}$; $\operatorname{HRMS}(m / z)$ for $\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{O}_{2} \mathrm{NaSi}_{2}(\mathrm{M}+\mathrm{Na})$ (major) calcd: 453.16765, found: 453.16774, HRMS $(m / z)$ for $\mathrm{C}_{23} \mathrm{H}_{22} \mathrm{O}_{2} \mathrm{NaSi}(\mathrm{M}+\mathrm{Na})$ (minor) calcd: 381.12813, found: 381.12830; $R_{f}\left(2 / 1\right.$ hexanes/EtOAc) (major) $=0.44$ (silica gel plate), $R_{f}(2 / 1$ hexanes $/$ EtOAc $)($ minor $)=0.26($ silica gel plate $)$.

6,7-Di(naphthalen-1-yl)-5,8-dihydroindeno[2,1-c]fluorene-5,8-diol (21i). With 20i (135
 $\mathrm{mg}, 0.25 \mathrm{mmol}$ ) following the general procedure C. Column chromatography ( $2 / 1 / 1$ hexanes/EtOAc/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded 112 mg ( $83 \%$ ) of the title compound as a light brown solid: mp (decomp) $217-222{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.49(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.63(\mathrm{~d}, J=8.1$ $\mathrm{Hz}, 4 \mathrm{H}$ ), 7.58 (t, $J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.50-7.54(\mathrm{~m}, 6 \mathrm{H}), 7.34-7.39(\mathrm{~m}, 4 \mathrm{H}), 7.25(\mathrm{t}, J=7.2 \mathrm{~Hz}$, $2 \mathrm{H}), 7.14(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 5.32(\mathrm{~s}, 2 \mathrm{H}), 1.65(\mathrm{bs}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $147.0,145.8,139.8,137.0,134.1,133.6,130.9,129.3,128.9,128.2,128.1,128.0,125.9$, 125.7, 125.6, 124.4, 123.6, 74.29; IR (KBr) $v_{\max } 3553,3422,3053,2953,2869,1593,1507$, $1465,1390,1330,1189,1040,1019,801,780,767,733 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / z)$ for $\mathrm{C}_{40} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{Na}$ $(\mathrm{M}+\mathrm{Na})$ calcd: 561.18250 , found: $561.18255 ; R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=0.36$ (silica gel plate).

## 6-(4-Methoxyphenyl)-7-(4-(trifluoromethyl)phenyl)-5,8-dihydroindeno[2,1-c]fluorene-



5,8-diol (27). With 26 ( $134 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) following the general procedure C. Column chromatography ( $2 / 1 / 1$ hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded 113 mg ( $84 \%$ ) of the title compound as a light brown solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.37$ (d, $J=7.8 \mathrm{~Hz}$, $2 \mathrm{H}), 7.61(\mathrm{t}, J=7.9 \mathrm{~Hz}, 4 \mathrm{H}), 7.51(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.37-7.41(\mathrm{~m}$, $6 \mathrm{H}), 5.86(\mathrm{~s}, 1 \mathrm{H}), 5.81(\mathrm{~s}, 1 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 1.78(\mathrm{~s}, 1 \mathrm{H}), 1.69(\mathrm{~s}, 1 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 158.8,146.5,145.9,145.8,139.6,136.6,135.1,122.9-129.2\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=272.1 \mathrm{~Hz}\right)$, $129.1,128.9,128.2,125.5,125.4,123.6\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.9 \mathrm{~Hz}\right), 74.3,74.2,55.2$; IR (KBr) $v_{\max }$ 3566, 3221, 3040, 2952, 2921, 2861, 1609, 1513, 1463, 1434, 1405, 1323, 1249, 1163, 1125, 1107, 1065, 1032, 840, 764, $736 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{34} \mathrm{H}_{23} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 559.14915 , found: $559.14932 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.34$ (silica gel plate) .

## 3-Methoxy-6,7-di-p-tolyl-10-(trifluoromethyl)-5,8-dihydroindeno[2,1-c]fluorene-5,8-diol

 (32). With 31 ( $134 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) following the general procedure C. Column chromatography ( $2 / 1 / 1$ hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $84 \mathrm{mg}(60 \%)$ of the title compound as a light brown solid: mp (decomp) 141-146 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.40$ (d, $J=8.2$ $\mathrm{Hz}, 1 \mathrm{H}), 8.20(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.82(\mathrm{~s}, 1 \mathrm{H}), 7.72(\mathrm{~d}, J=8.2 \mathrm{~Hz}$, $1 \mathrm{H}), 7.34-7.41(\mathrm{~m}, 3 \mathrm{H}), 7.16(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.02(\mathrm{dd}, J=8.6,2.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.00(\mathrm{~m}, 2 \mathrm{H})$, $6.76(\mathrm{~m}, 2 \mathrm{H}), 5.89(\mathrm{~s}, 1 \mathrm{H}), 5.81(\mathrm{~s}, 1 \mathrm{H}), 3.88(\mathrm{~s}, 3 \mathrm{H}), 2.33(\mathrm{~s}, 6 \mathrm{H}), 1.80(\mathrm{~s}, 1 \mathrm{H}), 1.76(\mathrm{~s}, 1 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 160.0,148.1,147.4,146.8,146.1,143.5,138.4,137.0,136.5$, $135.7,134.5,134.4,132.4,132.1,130.2,129.5,129.1,128.4,126.1,126.0,125.6-133.6$ (q, $\left.{ }^{1} J_{\mathrm{C}-\mathrm{F}}=194.6 \mathrm{~Hz}\right), 124.3,123.1,122.3\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}\right), 114.6,110.9,74.2,74.0,55.5,21.2$; IR (KBr) $v_{\max } 3567,3458,3019,2923,2871,1611,1512,1484,1434,1327,1300,1252$, 1166, 1144, 1123, 1177, 1034, 833, 818, $527 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{36} \mathrm{H}_{28} \mathrm{O}_{3} \mathrm{~F}_{3}(\mathrm{M}+\mathrm{H})$ calcd: 565.19851, found: $565.19819 ; R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=0.46$ (silica gel plate).

## 8,9-Bis(4-methoxyphenyl)-7,10-dihydrobenzo[c]benzo[6,7]indeno[1,2-g]fluorene-7,10-


diol (39a). With 38a ( $150 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) following the general procedure C . Column chromatography ( $2 / 1 / 1$ hexanes/EtOAc/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $145 \mathrm{mg}(96 \%)$ of the title compound as diastereoisomers as a light brown solid: mp (decomp) $178-183{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.89(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.82(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.79(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H})$, $7.39(\mathrm{dd}, J=8.4,0.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.18(\mathrm{~m}, 4 \mathrm{H}), 6.87(\mathrm{~m}, 4 \mathrm{H}), 6.53(\mathrm{ddd}, J=8.3,6.8,1.3 \mathrm{~Hz}$, $2 \mathrm{H}), 5.84(\mathrm{~s}, 2 \mathrm{H}), 3.81(\mathrm{~s}, 6 \mathrm{H}), 2.05(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 158.5,145.3$, $143.6,138.0,137.4,134.6,134.3,129.6,129.2,129.1,127.9,127.7,125.6,124.5,124.2$, $122.2,113.9,74.4,55.2$; $\mathrm{IR}(\mathrm{KBr}) v_{\max } 3548,3053,3001,2954,2835,1608,1511,1462$, $1362,1287,1247,1176,1033,832,800,744 \mathrm{~cm}^{-1} ; \operatorname{HRMS}(\mathrm{m} / z)$ for $\mathrm{C}_{42} \mathrm{H}_{30} \mathrm{O}_{4} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 621.20363 , found: $621.20369 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.22$ (silica gel plate) .

8,9-Diphenyl-7,10-dihydrobenzo[c]benzo[6,7]indeno[1,2-g]fluorene-7,10-diol (39b). With
 38b ( $107 \mathrm{mg}, 0.2 \mathrm{mmol}$ ) following the general procedure C. Column chromatography ( $2 / 1 / 1$ hexanes/ $\mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $103 \mathrm{mg}(96 \%)$ of the title compound as diastereoisomers as a light brown solid: mp (decomp) $171-176{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.58(\mathrm{~s}, 1 \mathrm{H}), 9.02(\mathrm{~d}, J=8.3$ $\mathrm{Hz}, 1 \mathrm{H}), 8.95(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 8.00(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.93(\mathrm{~m}, 2 \mathrm{H})$, $7.87(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.68-7.85(\mathrm{~m}, 5 \mathrm{H}), 7.66(\mathrm{~s}, 1 \mathrm{H}), 7.60-7.64(\mathrm{~m}, 4 \mathrm{H}), 7.53-7.58(\mathrm{~m}$, $4 \mathrm{H}), 6.26(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.84(\mathrm{~s}, 1 \mathrm{H}), 1.88(\mathrm{bs}, 1 \mathrm{H}), 1.77(\mathrm{~d}, J=4.6 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 130.8,129.6,129.4,129.3,129.2,128.7,128.6,128.1,128.0,127.3$, $126.9,126.8,126.7,126.0,125.5,124.5,124.3,124.1,123.0,122.7,121.9,116.2,75.5,74.1$; IR (KBr) $\nu_{\max } 3580,3426,3055,2951,1608,1498,1464,1441,1423,1331,1176,1067$, 1029, 766, 742, $714 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{40} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 561.18250, found: $561.18267 ; R_{f}(2 / 1$ hexanes/EtOAc $)=0.30($ silica gel plate $)$.

## 8,9-Bis(4-(trifluoromethyl)phenyl)-7,10-dihydrobenzo[c]benzo[6,7]indeno[1,2-g]


fluorene-7,10-diol (39c). With 38c ( $100 \mathrm{mg}, 0.15 \mathrm{mmol}$ ) following the general procedure C. Column chromatography (2/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $93 \mathrm{mg}(94 \%)$ of the title compound as diastereoisomers as a light brown solid: mp (decomp) 189-194 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 7.92(\mathrm{~m}, 2 \mathrm{H}), 7.84(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.67-$ $7.79(\mathrm{~m}, 4 \mathrm{H}), 7.57(\mathrm{~m}, 4 \mathrm{H}), 7.48(\mathrm{~m}, 2 \mathrm{H}), 7.40(\mathrm{~m}, 2 \mathrm{H}), 7.21(\mathrm{~m}, 2 \mathrm{H}), 6.65(\mathrm{~m}, 1 \mathrm{H}), 6.56(\mathrm{~m}$,
$1 \mathrm{H}), 5.73(\mathrm{~s}, 2 \mathrm{H}), 1.96(\mathrm{~s}, 1 \mathrm{H}), 1.59(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.0,145.1$, $144.2,143.9,142.0,141.0,137.4,136.1,135.4,134.4,134.1,131.6,130.1,130.0,129.7$, $129.4,129.2,128.2,128.1,127.5,127.2,126.9,125.9,125.3\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=4.7 \mathrm{~Hz}\right), 124.7,124.6$, 124.0, 122.6, 122.1, 121.8, 119.6-126.5 (q, ${ }^{1} J_{\mathrm{C}-\mathrm{F}}=305.7 \mathrm{~Hz}$ ), 116.6, $75.5,74.2$; IR $(\mathrm{KBr}) v_{\max }$ 3478, 3053, 3018, 2929, 2846, 1626, 1598, 1446, 1301, 1270, 1232, 1177, 1049, 1028, 990, 824, 769, $696 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{42} \mathrm{H}_{24} \mathrm{O}_{2} \mathrm{~F}_{6} \mathrm{Na}(\mathrm{M}+\mathrm{Na})$ calcd: 697.15727, found: $697.15750 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.38$ (silica gel plate $)$.

### 6.3.3. Synthesis of Dispiroindeno[2,1-c]fluorenes 23, 29, 34 and 41

D: General procedure for Oxidation reaction with PCC (preparation of 22, 28, 33 and 40). To a solution of pyridinium chlorochromate (PCC, $64 \mathrm{mg}, 0.3 \mathrm{mmol}$ ) and celite ( 64 mg ) in anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ the starting indeno[2,1-c]fluorene-5,8-diol ( 0.2 mmol ) was added and the mixture was stirred for 3 h at rt . Afterwards the residue was filtered through a celite/silica gel plug. Column chromatography of the residue on silica gel yielded products.

6,7-Bis(4-methoxyphenyl)indeno[2,1-c]fluorene-5,8-dione (22a). With 21a (230 mg, 0.46
 mmol ) following the general procedure D. Column chromatography ( $2 / 1 / 1$ hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $210 \mathrm{mg}(94 \%)$ of the title compound as a bright yellow solid: mp (decomp) 245-250 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 8.21(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.69(\mathrm{dd}, J=7.3,0.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.66(\mathrm{dt}$, $J=7.6,1.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.41(\mathrm{dt}, J=7.5,0.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.90(\mathrm{~m}, 4 \mathrm{H}), 6.77(\mathrm{~m}$, $4 \mathrm{H}), 3.79(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 191.8, 158.8, 143.3, 142.7, 138.0, 136.2, $135.5,134.8,130.9,129.6,127.2,124.8,123.7,113.0,55.1$; IR (KBr) $v_{\max } 3073,3046,3007$, 2959, 2837, 1714, 1609, 1517, 1466, 1426, 1302, 1288, 1246, 1200, 1179, 1030, 941, 835, $754,721, \mathrm{~cm}^{-1} ;$ HRMS $(m / z)$ for $\mathrm{C}_{34} \mathrm{H}_{23} \mathrm{O}_{4}(\mathrm{M}+\mathrm{H})$ calcd: 495.15909, found: 495.15921; $R_{f}$ $(2 / 1$ hexanes $/ E t O A c)=0.47$ (silica gel plate) .

6,7-Di-p-tolylindeno[2,1-c]fluorene-5,8-dione (22b). With 21b (140 mg, 0.3 mmol )
 following the general procedure D. Column chromatography (2/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $125 \mathrm{mg}(90 \%)$ of the title compound as a bright yellow solid: mp (decomp) $167-172{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $8.21(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.68(\mathrm{dd}, J=7.3,0.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.65(\mathrm{dt}, J=7.6,1.3$ $\mathrm{Hz}, 2 \mathrm{H}), 7.41(\mathrm{dt}, J=7.5,0.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.02(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 4 \mathrm{H}), 6.88(\mathrm{~m}, 4 \mathrm{H})$, $2.31(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 191.7,143.4,142.7,138.0,136.9,136.2,135.4$, $134.8,132.0,129.6,129.3,128.2,124.8,123.7,21.4$; IR (KBr) $v_{\max } 3069,3043,3010,2919$, 2863, 1734, 1711, 1605, 1578, 1517, 1465, 1423, 1377, 1304, 1237, 1193, 1096, 1069, 1022, 931, 851, 826, 765, 753, 722, $665 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{34} \mathrm{H}_{23} \mathrm{O}_{2}(\mathrm{M}+\mathrm{H})$ calcd: 463.16926, found: $463.16927 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.41$ (silica gel plate) .

6,7-Diphenylindeno[2,1-c]fluorene-5,8-dione (22c). With 21c (285 mg, 0.65 mmol )
 following the general procedure D . Column chromatography (2/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $257 \mathrm{mg}(92 \%)$ of the title compound as a bright yellow solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $8.20(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.69(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.65(\mathrm{dt}, J=7.6,1.2 \mathrm{~Hz}$, $2 \mathrm{H}), 7.41(\mathrm{dt}, J=7.5,0.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.20-7.22(\mathrm{~m}, 6 \mathrm{H}), 6.99-7.02(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 191.6, 143.0, 142.7, 138.1, 136.1, 135.3, 135.0, 134.8, 129.7, 129.4, 127.4, 127.3, 124.8, 123.8; IR (KBr) $v_{\max } 3055,2921,1711,1607,1571,1466,1417,1379,1302$, $1198,1076,1027,936,830,765,752,716,702 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{32} \mathrm{H}_{19} \mathrm{O}_{2}(\mathrm{M}+\mathrm{H})$ calcd: 435.13796 , found: $435.13789 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.36$ (silica gel plate).

6,7-Bis(4-bromophenyl)indeno[2,1-c]fluorene-5,8-dione (22d). With 21d (297 mg, 0.5
 mmol ) following the general procedure D. Column chromatography ( $2 / 1 / 1$ hexanes/ $\mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $285 \mathrm{mg}(96 \%)$ of the title compound as a bright yellow solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $8.22(\mathrm{~m}, 2 \mathrm{H}), 7.70(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.68(\mathrm{dt}, J=7.6,1.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.32-$ $7.37(\mathrm{~m}, 4 \mathrm{H}), 7.25(\mathrm{~m}, 2 \mathrm{H}), 6.98(\mathrm{~m}, 2 \mathrm{H}), 6.84(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 191.4,142.5,141.3,138.6,136.3,135.2,135.1,133.6,131.1,130.9,130.0,127.6$, 125.0, 123.9, 122.0; IR (KBr) $v_{\max } 3055,3041,3010,1712,1606,1580,1492,1466,1423$, 1393, 1298, 1238, 1196, 1099, 1074, 1012, 933, 835, 756, $719 \mathrm{~cm}^{-1}$; HRMS ( $m / z$ ) for $\mathrm{C}_{32} \mathrm{H}_{17} \mathrm{O}_{2} \mathrm{Br}_{2}(\mathrm{M}+\mathrm{H})$ calcd: 590.95898, found: 590.95892; $R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=0.40$ (silica gel plate).

6,7-Bis(4-(trifluoromethyl)phenyl)indeno[2,1-c]fluorene-5,8-dione (22e). With 21e (400
 $\mathrm{mg}, 0.7 \mathrm{mmol}$ ) following the general procedure D. Column chromatography (2/1/1 hexanes/EtOAc/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $363 \mathrm{mg}(91 \%)$ of the title compound as a bright yellow solid: mp (decomp) $>260{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.23(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.72(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.70(\mathrm{dt}$, $J=7.6,1.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.48(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.46(\mathrm{dt}, J=7.3,0.5 \mathrm{~Hz}, 2 \mathrm{H})$, $7.11(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 191.3$, 142.5, 140.8, 138.8, 138.5, $136.3,135.2,135.1,130.1,129.9,122.6-129.6\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=272.1 \mathrm{~Hz}\right) 125.33,125.05,124.6(\mathrm{q}$, ${ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}$ ), 124.0; IR (KBr) $v_{\max } 3067,3040,2922,1712,1606,1467,1409,1328,1171$, 1127, 1111, 1067, 1020, 935, 850, 758, $721 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{34} \mathrm{H}_{17} \mathrm{O}_{2} \mathrm{~F}_{6}(\mathrm{M}+\mathrm{H})$ calcd: 571.11273 found: $571.11255 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.34$ (silica gel plate) .

6,7-Di(thiophen-3-yl)indeno[2,1-c]fluorene-5,8-dione (22f). With 21 f ( $315 \mathrm{mg}, 0.7 \mathrm{mmol}$ )
 following the general procedure D . Column chromatography (2/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $284 \mathrm{mg}(91 \%)$ of the title compound as a light orange solid: mp (decomp) $>260{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.18$ $(\mathrm{d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.71(\mathrm{dd}, J=7.3,0.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.65(\mathrm{dt}, J=7.6,1.2 \mathrm{~Hz}$, $2 \mathrm{H}), 7.42(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.22(\mathrm{dd}, J=4.9,2.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.90(\mathrm{dd}, J=2.9,1.2 \mathrm{~Hz}, 2 \mathrm{H})$ $6.78(\mathrm{dd}, J=4.9,1.2 \mathrm{~Hz}, 2 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 191.4,142.6,138.5,138.2$, $136.5,135.3,134.9,134.5,129.7,129.0,124.8,124.7,124.1,123.8$; IR (KBr) $v_{\max } 3114$, 3085, 3049, 3013, 1709, 1607, 1583, 1465, 1437, 1349, 1314, 1275, 1204, 1189, 1071, 953, 832, 808, 763, 745, 718, $663 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{28} \mathrm{H}_{15} \mathrm{O}_{2} \mathrm{~S}_{2}(\mathrm{M}+\mathrm{H})$ calcd: 447.05080 found: $447.05076 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.34$ (silica gel plate) .

6,7-Dipropylindeno[2,1-c]fluorene-5,8-dione (22g). With $\mathbf{2 1 g}$ ( $130 \mathrm{mg}, 0.35 \mathrm{mmol}$ )
 following the general procedure D . Column chromatography (2/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $124 \mathrm{mg}(96 \%)$ of the title compound as a bright yellow solid: mp (decomp) 181-186 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $8.07(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.74(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.59(\mathrm{dt}, J=7.6,1.1 \mathrm{~Hz}$, $2 \mathrm{H}), 7.38(\mathrm{t}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.12(\mathrm{~m}, 4 \mathrm{H}), 1.54(\mathrm{~m}, 4 \mathrm{H}), 0.99(\mathrm{t}, J=7.3 \mathrm{~Hz}$, $6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 194.6, 146.0, 143.0, 140.1, 135.9, 135.4, 134.7, 129.1, 124.5, 123.5, 28.4, 24.4, 14.6; IR (KBr) $v_{\max } 3082,3052,2970,2926,2868,1701,1608,1588$, $1466,1419,1313,1293,1253,1209,1157,1113,1086,1016,960,923,837,756,719 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{26} \mathrm{H}_{22} \mathrm{O}_{2}\left(\mathrm{M}^{+}\right)$calcd: 366.1620 , found: $366.1625 ; R_{f}(2 / 1$ hexanes/EtOAc $)=$ 0.51 (silica gel plate).

6-(Trimethylsilyl)indeno[2,1-c]fluorene-5,8-dione (22h). With 21h (59 mg, 0.16 mmol ) TMS following the general procedure D . Column chromatography (2/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $47 \mathrm{mg}(83 \%)$ of the title compound as a bright yellow solid: mp (decomp) $260{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.08$ (m, 2H), 7.82 (s, 1H), 7.78 (ddd, $J=7.4,1.2,0.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.63(\mathrm{~m}, 2 \mathrm{H}), 7.41$ (ddt, $J=7.5,2.0,0.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), $0.41(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 193.9,192.9$, 145.6, 143.7, 143.6, 142.6, 139.1, 139.0, 138.3, 135.1, 135.0, 134.9, 134.7, 130.8, 129.7, $129.5,125.0,124.7,124.2,123.9,-1.30$; IR (KBr) $v_{\max } 3082,3005,2953,2896,1714,1611$, $1580,1471,1413,1323,1246,1207,1096,1084,975,929,867,843,824,758,716 \mathrm{~cm}^{-1}$;

HRMS $(m / z)$ for $\mathrm{C}_{23} \mathrm{H}_{19} \mathrm{O}_{2} \mathrm{Si}(\mathrm{M}+\mathrm{H})$ calcd: 355.11488, found: 355.11493; $R_{f}(2 / 1$ hexanes $/$ EtOAc $)=0.48($ silica gel plate $)$.

6,7-Di(naphthalen-1-yl)indeno[2,1-c]fluorene-5,8-dione (22i). With 21i (271 mg, 0.50
 mmol ) following the general procedure D . Column chromatography (2/1/1 hexanes/EtOAc/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $258 \mathrm{mg}(96 \%)$ of the title compound as a bright yellow solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.32(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.76(\mathrm{dd}, J=7.7,1.6 \mathrm{~Hz}, 2 \mathrm{H})$, 7.76 (dt, $J=7.6,1.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.55-7-67 (m, 6H), 7.37-7.45 (m, 4H), 7.29 $(\mathrm{t}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.23(\mathrm{dd}, J=7.1,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.16(\mathrm{dt}, J=8.1,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.04(\mathrm{~m}, 1 \mathrm{H})$, $6.97(\mathrm{dd}, J=8.2,7.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.84(\mathrm{dd}, J=7.1,1.1 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $191.0,142.90$, $142.85,141.9,141.8,138.5,137.73,137.71,135.30,135.25,134.9,133.4$, $132.9,132.8,132.7,132.5,131.6,129.8,128.4,128.0,127.9,127.8,127.5,125.9,125.4$, $125.3,125.24,125.21,125.0,124.7,124.2,123.98,123.95$; IR (KBr) $v_{\max } 3052,2919,1712$, $1608,1507,1467,1375,1317,1296,1202,1003,935,799,778,759,718 \mathrm{~cm}^{-1} ;$ HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{40} \mathrm{H}_{23} \mathrm{O}_{2}(\mathrm{M}+\mathrm{H})$ calcd: 535.16926, found: 535.16938; $R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=0.39$ (silica gel plate).

6-(4-Methoxyphenyl)-7-(4-(trifluoromethyl)phenyl)indeno[2,1-c]fluorene-5,8-dione (28).


With $27(188 \mathrm{mg}, 0.35 \mathrm{mmol})$ following the general procedure D . Column chromatography ( $2 / 1 / 1$ hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $149 \mathrm{mg}(80 \%)$ of the title compound as a bright yellow solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.19$ (d, $J=7.7 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.63-7.72 (m, 4H), $7.47(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.43(\mathrm{dt}, J=7.5,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.41(\mathrm{dt}, J=7.5,0.7$ $\mathrm{Hz}, 1 \mathrm{H}), 7.13(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.87(\mathrm{~m}, 2 \mathrm{H}), 6.73(\mathrm{~m}, 2 \mathrm{H}), 3.78(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 191.6,191.5,159.0,142.7,142.6,142.4,141.5,139.1,138.8,137.9,136.4$, 136.1, 135.3, 135.1, 135.0, 125.6-136.2 (q, ${ }^{1} J_{\mathrm{C}-\mathrm{F}}=366.4 \mathrm{~Hz}$ ), 134.9, 130.8, 130.0, 129.9, $129.8,129.1,126.4,124.9,124.4\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}\right), 123.9,123.8,113.1,55.1$; IR (KBr) $v_{\max }$ 3073, 3004, 2952, 2952, 2905, 2836, 1712, 1608, 1514, 1467, 1425, 1325, 1302, 1247, 1166, 1121, 1074, 1035, 1020, 932, 837, 757, $719 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{34} \mathrm{H}_{20} \mathrm{O}_{3} \mathrm{~F}_{3}(\mathrm{M}+\mathrm{H})$ calcd: 533.13591 , found: $533.13596 ; R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=0.38$ (silica gel plate) .

3-Methoxy-6,7-di-p-tolyl-10-(trifluoromethyl)indeno[2,1-c]fluorene-5,8-dione (33). With
 32 ( $188 \mathrm{mg}, 0.35 \mathrm{mmol}$ ) following the general procedure D . Column chromatography ( $2 / 1 / 1$ hexanes/ $\mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $68 \mathrm{mg}(87 \%)$ of the title compound as a bright yellow solid: mp (decomp) 109-114 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.26(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.03(\mathrm{~d}, J$ $8.4,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.04(\mathrm{dd}, J=7.8,1.8 \mathrm{~Hz}, 4 \mathrm{H}), 6.90(\mathrm{dd}, J=8.1,1.8 \mathrm{~Hz}, 4 \mathrm{H}), 3.88(\mathrm{~s}, 3 \mathrm{H})$, 2.34 (s, 6H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 191.2, 190.2, 161.2, 145.8, 144.3, 143.0, 139.2, $137.5,137.1,137.0$, 136.7, 136.1, 135.6, 135.2, 134.6, 131.7, 131.6, 131.5, 131.4, 129.3, 129.2, 128.3, 128.2, 124.8, 123.5, $121.4\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}\right), 120.6,119.4-127.5\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=\right.$ 272.5 Hz ), 109.5, 55.7, 21.4; IR (KBr) $v_{\max } 3022,3001,2924,2869,1713,1622,1608,1487$, 1434, 1329, 1302, 1246, 1169, 1130, 1071, 1021, 914, 835, 791, $754 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{36} \mathrm{H}_{24} \mathrm{O}_{3} \mathrm{~F}_{3}(\mathrm{M}+\mathrm{H})$ calcd: 561.16721, found: 561.16724; $R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=0.54$ (silica gel plate).

## 8,9-Bis(4-methoxyphenyl)benzo[c]benzo[6,7]indeno[1,2-g]fluorene-7,10-dione



With 39a ( $272 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) following the general procedure D . Column chromatography ( $2 / 1 / 1$ hexanes/EtOAc/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded 229 mg ( $86 \%$ ) of the title compound as a red solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.91(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.84(\mathrm{~d}, J=8.3 \mathrm{~Hz}$, $2 \mathrm{H}), 7.81(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.32(\mathrm{~m}, 2 \mathrm{H}), 7.28(\mathrm{~d}, J=9.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.14$ (d, $J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.87(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.84(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H})$, 6.67 (ddd, $J=8.2,6.9,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.81(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 191.4$, $158.9,145.2,142.3,138.5,137.8,137.2,134.2,131.7,130.8,128.7,128.6,128.4,127.0$, 126.7, 125.7, 119.7, 113,1, 55.1; IR (KBr) $v_{\max } 3052,2930,2833,1704,1609,1516,1441$, 1291, 1249, 1178, 1109, 1035, 901, 835, 800, $777 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{42} \mathrm{H}_{27} \mathrm{O}_{4}(\mathrm{M}+\mathrm{H})$ calcd: 595.19039, found: 595.19049; $R_{f}(2 / 1$ hexanes $/ \mathrm{EtOAc})=0.32$ (silica gel plate $)$.

8,9-Diphenylbenzo[c]benzo[6,7]indeno[1,2-g]fluorene-7,10-dione (40b). With 39b (103
 $\mathrm{mg}, \quad 0.19 \mathrm{mmol}$ ) following the general procedure D . Column chromatography ( $2 / 1 / 1$ hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded $96 \mathrm{mg}(95 \%)$ of the title compound as a red solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.92(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.85(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.81(\mathrm{~d}, J=8.2$ $\mathrm{Hz}, 2 \mathrm{H}$ ), 7.68 (m, 1H), 7.57 (m, 1H), 7.33 (ddd, $J=8.1,6.8,1.1 \mathrm{~Hz}, 2 \mathrm{H})$, $7.30(\mathrm{~m}, 2 \mathrm{H}), 7.21-7.28(\mathrm{~m}, 6 \mathrm{H}), 7.00(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.69(\mathrm{ddd}, J=8.3,7.8,1.2 \mathrm{~Hz}$, $2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 191.2, 145.1, 142.1, 138.7, 137.7, 137.2, 134.8, 134.1, $130.9,130.2,130.1,129.9,129.6,129.2,128.9,128.7,128.6,128.4,128.3,128.1,128.0$, $127.9,127.8$, 127.7, 127.5, 127.4, 127.3, 127.1, 126.6, 125.7, 125.5, 125.0, 123.4, 119.9, 119.7, 117.0; IR (KBr) $v_{\max } 3054,2919,1710,1617,1570,1464,1419,1375,1304,1197$, 1074, 1025, 936, 830, 766, 752, 716, $701 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{40} \mathrm{H}_{23} \mathrm{O}_{2}(\mathrm{M}+\mathrm{H})$ calcd: 535.16926, found: 535.16935; $R_{f}(2 / 1$ hexanes/EtOAc $)=0.40$ (silica gel plate $)$.

## 8,9-Bis(4-(trifluoromethyl)phenyl)benzo[c]benzo[6,7]indeno[1,2-g]fluorene-7,10-dione


(40c). With 39c ( $93 \mathrm{mg}, 0.14 \mathrm{mmol}$ ) following the general procedure D . Column chromatography ( $2 / 1 / 1$ hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded 85 mg ( $92 \%$ ) of the title compound as a red solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.95(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.87(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.81$ (d, $J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.53(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.49(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.37$ (ddd, $J=8.1,6.9,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.30(\mathrm{~m}, 4 \mathrm{H}), 7.12(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.67$ (ddd, $J=8.3,6.9,1.2 \mathrm{~Hz}, 2 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 190.8,145.0,139.7,139.2$, 138.2, 138.0, 137.4, 133.9, 131.4, 130.6, 130.2, 130.01, 129.96, 129.7, 129.6, 129.0, 128.9, $128.5,128.4,126.6,126.0,117.5-125.5\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=236.3 \mathrm{~Hz}\right), 125.4,124.7\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.5 \mathrm{~Hz}\right)$, 119.7; IR (KBr) $v_{\max } 3072,3041,2920,1711,1605,1469,1404,1329,1170,1127,1067$, 1019, 934, 855, 757, $721 \mathrm{~cm}^{-1}$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{42} \mathrm{H}_{21} \mathrm{O}_{2} \mathrm{~F}_{6}(\mathrm{M}+\mathrm{H})$ calcd: 671.14403, found: $671.14394 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.42$ (silica gel plate $)$.

E: General procedure for creation of dispiroindeno[2,1-c]fluorenes (preparation of 23, 29, 34 and 41). ${ }^{46}$ A solution of 2-bromobiphenyl ( $40 \mu \mathrm{~L}, 0.23 \mathrm{mmol}$ ) in anhydrous THF $(3 \mathrm{~mL})$ was cooled down to $-78{ }^{\circ} \mathrm{C}$ and $n$-BuLi $1.6 \mathrm{M}(140 \mu \mathrm{~L}, 0.23 \mathrm{mmol})$ was added dropwise. The resulted solution was stirred for 30 min , followed by the addition of fluorenone ( 0.07 mmol ) in THF ( 3 mL ) and stirred for 3 h first for 15 min at $-78^{\circ} \mathrm{C}$, but during time it was allowed to reach ambient temperature. The reaction mixture was poured in a saturated solution of $\mathrm{NaHCO}_{3}$ solution and extracted with diethyl ether ( $3 \times 15 \mathrm{~mL}$ ). The combined organic fractions were dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. The crude was purified by column chromatography on silica gel (10/1 hexanes/EtOAc) to yield desired alcohols. The alcohols were dissolved in acetic acid ( 10 mL ) with a catalytic amount of HCl ( $12 \mathrm{~mol} / \mathrm{L}$ ) and the resulted solution stirred under reflux for 3 h . The reaction mixture was neutralized with $\mathrm{H}_{2} \mathrm{O}$ and $\mathrm{K}_{2} \mathrm{CO}_{3}$ and extracted with diethyl ether ( $3 \times 15 \mathrm{~mL}$ ). Column chromatography of the residue on silica gel provided products.

## 6',7'-bis(4-methoxyphenyl)dispirofluorene-9,5'-indeno[2,1-c]fluorene-8',9' 'fluorene


(23a). With 22a ( $98 \mathrm{mg}, 0.20 \mathrm{mmol}$ ) following the general procedure E . Column chromatography of the residue on silica gel (5/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided $87 \mathrm{mg}(57 \%)$ of the title compound as a colorless solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 8.76 (d, $J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.48(\mathrm{dt}, J=7.9,1.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.31(\mathrm{~d}, J=7.4$ $\mathrm{Hz}, 4 \mathrm{H}), 7.15(\mathrm{dt}, J=7.4,1.1 \mathrm{~Hz}, 4 \mathrm{H}), 7.13(\mathrm{dt}, J=7.5,0.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.06(\mathrm{dt}, J=7.5,1.2 \mathrm{~Hz}$, $4 \mathrm{H}), 6.91$ (d, $J=7.4 \mathrm{~Hz}, 4 \mathrm{H}), 6.59$ (dd, $J=7.6,0.6 \mathrm{~Hz}, 2 \mathrm{H}), 5.73(\mathrm{~m}, 4 \mathrm{H}), 5.57(\mathrm{~m}, 4 \mathrm{H}), 3.48$ (s, 6H) ${ }^{13}{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 156.2,150.5,148.9,147.1,142.0,141.4,139.6,136.4$, $130.5,128.8,127.5,127.1,127.0,126.9,124.0,123.7,123.4,119.7,111.1,66.2,54.9$; IR (KBr) $v_{\max } 3040,2953,2931,2827,1612,1516,1465,1447,1418,1283,1243,1174,1106$, 1036, 833, 765, 754, 740, $732 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{58} \mathrm{H}_{38} \mathrm{O}_{2}(\mathrm{M}+\mathrm{H})$ calcd: 767.29446, found: 767.29447; $R_{f}(1 / 1$ hexanes $/ E t O A c)=0.66$ (silica gel plate) .

## 6',7'-Di-p-tolyldispirofluorene-9,5'-indeno[2,1-c]fluorene-8',9'-fluorene (23b). With 22b

 $(116 \mathrm{mg}, 0.25 \mathrm{mmol})$ following the general procedure E. Column chromatography of the residue on silica gel (5/1/1 hexanes/ $\mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided $99 \mathrm{mg}(56 \%)$ of the title compound as a colorless solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$
$8.76(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.47(\mathrm{dt}, J=7.8,1.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.25-7.27(\mathrm{~m}, 4 \mathrm{H}), 7.15(\mathrm{dt}, J=7.4$, $1.1 \mathrm{~Hz}, 4 \mathrm{H}), 7.12(\mathrm{dt}, J=7.3,0.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.05(\mathrm{dt}, J=7.5,1.2 \mathrm{~Hz}, 4 \mathrm{H}), 6.90(\mathrm{~d}, J=7.5 \mathrm{~Hz}$, $4 \mathrm{H}), 6.58(\mathrm{dd}, J=7.6,0.6 \mathrm{~Hz}, 2 \mathrm{H}), 5.96(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 4 \mathrm{H}), 5.55(\mathrm{~m}, 4 \mathrm{H}), 1.88(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 150.6,148.9,146.9,142.0,141.4,139.7,136.4,133.3,133.1$, 129.3, 127.5, 127.1, 127.0, 126.8, 126.0, 124.0, 123.8, 123.4, 119.5, 66.2, 20.9; IR (KBr) $v_{\max }$ 3042, 2917, 2866, 1600, 1510, 1446, 1415, 1367, 1282, 1244, 1155, 1109, 1020, 826, 810, $754,741,730 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{58} \mathrm{H}_{39}(\mathrm{M}+\mathrm{H})$ calcd: 735.30463, found: 735.30470; $R_{f}$ $(2 / 1$ hexanes $/ E t O A c)=0.56($ silica gel plate $)$.

6',7'-Diphenyldispirofluorene-9,5'-indeno[2,1-c]fluorene-8',9'-fluorene (23c). With 22c
 $(109 \mathrm{mg}, 0.25 \mathrm{mmol})$ following the general procedure E. Column chromatography of the residue on silica gel (5/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided $108 \mathrm{mg}(61 \%)$ of the title compound as a colorless solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 8.78$ (d, $J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.49(\mathrm{dt}, J=7.8,1.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.24-7.27(\mathrm{~m}, 4 \mathrm{H}), 7.14(\mathrm{~m}, 6 \mathrm{H})$, $7.06(\mathrm{dt}, J=7.5,1.2 \mathrm{~Hz}, 4 \mathrm{H}), 6.93(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 4 \mathrm{H}), 6.60(\mathrm{dd}, J=7.6,0.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.42$ $(\mathrm{m}, 2 \mathrm{H}), 6.18(\mathrm{~m}, 4 \mathrm{H}), 5.67(\mathrm{dd}, J=8.2,1.2 \mathrm{~Hz}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 150.5$, $148.7,146.4,142.0,141.3,139.5,136.6,135.9,129.6,127.6,127.2,127.1,126.9,125.3$, 124.3, 124.1, 123.8, 123.4, 119.7, 66.2; IR (KBr) $v_{\max } 3058,2953,2927,2866,1601,1463$, $1445,1412,1367,1155,1071,1029,823,753,741,702 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{56} \mathrm{H}_{35}(\mathrm{M}+$ H) calcd: 707.27333 , found: $707.27344 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.58$ (silica gel plate).

## 6',7'-Bis(4-bromophenyl)dispirofluorene-9,5'-indeno[2,1-c]fluorene-8',9''-fluorene (23d).



With 22 d ( $59 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) following the general procedure E . Column chromatography of the residue on silica gel (5/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided $58 \mathrm{mg}(67 \%)$ of the title compound as a colorless solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 8.77 (d, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.50(\mathrm{~m}, 2 \mathrm{H}), 7.34$ (dd, $J=7.5,0.7 \mathrm{~Hz}, 2 \mathrm{H})$, $7.25(\mathrm{~m}, 2 \mathrm{H}), 7.21(\mathrm{dt}, J=7.5,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.20(\mathrm{dt}, J=7.4,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.13-7.18(\mathrm{~m}, 2 \mathrm{H})$, $7.07(\mathrm{t}, J=7.5 \mathrm{~Hz}, 4 \mathrm{H}), 6.91(\mathrm{~m}, 4 \mathrm{H}), 6.62(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.32(\mathrm{~m}, 2 \mathrm{H}), 6.28(\mathrm{~m}, 1 \mathrm{H})$, $6.21(\mathrm{~m}, 1 \mathrm{H}), 5.66(\mathrm{~m}, 2 \mathrm{H}), 5.53(\mathrm{~m}, 2 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 150.5,150.2,148.6$, $148.5,142.0,141.9,141.1,131.1,131.0,129.6,129.4,128.6,128.3,127.8,127.2,127.0$, $125.5,125.3,124.5,124.0,123.7,123.5,119.8$; IR (KBr) $v_{\max } 3041,2932,2839,1601,1485$, $1446,1414,1386,1282,1155,1072,1011,833,756,747,738 \mathrm{~cm}^{-1} ; \operatorname{HRMS}(m / z)$ for
$\mathrm{C}_{56} \mathrm{H}_{32} \mathrm{Br}_{2}(\mathrm{M}+\mathrm{H})$ calcd: 863.09435, found: 863.09459; $R_{f}(2 / 1$ hexanes/EtOAc $)=0.62$ (silica gel plate).

6',7'-Bis(4-(trifluoromethyl)phenyl)dispirofluorene-9,5'-indeno[2,1-c]fluorene-8',9'-
 fluorene (23e). With 22e ( $57 \mathrm{mg}, 0.10 \mathrm{mmol}$ ) following the general procedure E. Column chromatography of the residue on silica gel (5/1/1 hexanes/ $\mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided $69 \mathrm{mg}(82 \%)$ of the title compound as a colorless solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $8.80(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.53(\mathrm{dt}, J=7.7,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.25(\mathrm{~d}, J=7.3$ $\mathrm{Hz}, 4 \mathrm{H}), 7.17-7.19(\mathrm{~m}, 6 \mathrm{H}), 7.09(\mathrm{dt}, J=7.5,1.2 \mathrm{~Hz}, 4 \mathrm{H}), 6.94(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 4 \mathrm{H}), 6.64$ (dd, $J=7.6,0.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.46(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H}), 5.80(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 4 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 150.1,148.1,146.8,141.9,140.9,139.3,137.3,137.2,129.7,122.0-129.1\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=\right.$ $271.9 \mathrm{~Hz}), 128.1,127.4,127.3,124.2,123.8,123.6,122.4\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.8 \mathrm{~Hz}\right), 119.9,66.0$; IR (KBr) $v_{\max } 3064,3023,2932,2839,1617,1448,1404,1328,1168,1121,1109,1065,1020$, 850, 841, 756, 745, $736 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{58} \mathrm{H}_{33} \mathrm{~F}_{6}(\mathrm{M}+\mathrm{H})$ calcd: 843.24810, found: 843.24886; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.60($ silica gel plate $)$.

6',7'-Di(thiophen-3-yl)dispirofluorene-9,5'-indeno[2,1-c]fluorene-8',9'-fluorene
With $22 f(45 \mathrm{mg}, 0.1 \mathrm{mmol})$ following the general procedure E. Column chromatography of the residue on silica gel (5/1/1 hexanes/EtOAc/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided $38 \mathrm{mg}(53 \%)$ of the title compound as a colorless solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $8.77(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.50(\mathrm{dt}, J=7.6,1.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.40(\mathrm{~m}, 2 \mathrm{H}), 7.24(\mathrm{~m}, 2 \mathrm{H}), 7.15(\mathrm{t}, J$ $=7.4 \mathrm{~Hz}, 4 \mathrm{H}), 6.85-7.03(\mathrm{~m}, 8 \mathrm{H}), 6.63(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.19(\mathrm{~s}, 2 \mathrm{H}), 5.39(\mathrm{~m}, 4 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 150.4,148.2,147.3,142.0,141.2,136.9,136.1,135.1,128.7$, $127.8,127.2,127.1,124.1,123.5,123.0,122.5,121.2,119.8,66.2$; IR (KBr) $v_{\max } 3062,2949$, 2871, 1598, 1464, 1447, 1349, 1281, 1225, 1154, 1069, 1005, 857, 803, 750, 739, 724, 390, $368 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{52} \mathrm{H}_{31} \mathrm{~S}_{2}(\mathrm{M}+\mathrm{H})$ calcd: 719.18617, found: 719.18622; $R_{f}(2 / 1$ hexanes $/$ EtOAc $)=0.56($ silica gel plate $)$.

6',7'-Dipropyldispirofluorene-9,5'-indeno[2,1-c]fluorene-8',9'-fluorene (23g). With 22g
 $(37 \mathrm{mg}, 0.1 \mathrm{mmol})$ following the general procedure E. Column chromatography of the residue on silica gel (5/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided $38 \mathrm{mg}(59 \%)$ of the title compound as a colorless solid: mp (decomp) 218-223 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 8.67(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.87(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 4 \mathrm{H}), 7.42(\mathrm{dt}, J=7.9,1.2 \mathrm{~Hz}, 2 \mathrm{H})$, 7.39 (dt, $J=7.5,1.1 \mathrm{~Hz}, 4 \mathrm{H}), 7.15(\mathrm{dt}, J=7.5,1.1 \mathrm{~Hz}, 4 \mathrm{H}), 7.09(\mathrm{dt}, J=7.6,0.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.92$ (d, $J=7.6 \mathrm{~Hz}, 4 \mathrm{H}), 6.58(\mathrm{dd}, J=7.6,0.7 \mathrm{~Hz}, 2 \mathrm{H}), 1.65(\mathrm{~m}, 4 \mathrm{H}), 0.45(\mathrm{~m}, 4 \mathrm{H}), 0.21(\mathrm{t}, J=7.1$ $\mathrm{Hz}, 6 \mathrm{H}$ ) ; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 150.3$, 149.4, 146.7, 141.6, 141.4, 138.8, 135.7, $127.8,127.5,127.1,126.9,124.2,123.6,123.0,120.0,66.6,30.3,23.8,14.5$; IR (KBr) $v_{\max }$ $3063,3034,3016,2958,2926,2868,1595,1577,1466,1447,1413,1373,1281,1153,1084$, 1027, 752, $739 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{50} \mathrm{H}_{39}(\mathrm{M}+\mathrm{H})$ calcd: 639.30463, found: 639.30476; $R_{f}$ $(2 / 1$ hexanes $/ E t O A c)=0.59($ silica gel plate $)$.

Dispirofluorene-9,5'-indeno[2,1-c]fluorene-8',9''-fluoren-6'-yltrimethylsilane (23h). With
 $\mathbf{2 2 h}(42 \mathrm{mg}, 0.12 \mathrm{mmol})$ following the general procedure E. Column chromatography of the residue on silica gel (5/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided $39 \mathrm{mg}(52 \%)$ of the title compound as a colorless solid: mp (decomp) 254-259 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 8.43(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.32(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.86(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.74(\mathrm{dd}$, $J=7.3,0.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.65(\mathrm{dt}, J=7.6,1.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.44-7.52(\mathrm{~m}, 4 \mathrm{H}), 7.37-7.42(\mathrm{~m}, 4 \mathrm{H})$, 7.20 (dt, $J=7.5,1.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.14 (dt, $J=7.5,1.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.00(\mathrm{~m}, 1 \mathrm{H}), 6.78$ (m, 3H), 6.71 (s, 1H), 0.18 ( $\mathrm{s}, 9 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 155.6,149.9,148.4,144.9,141.7,140.4$, $140.2,138.4,134.7,134.5,130.1,128.81,128.75,128.0,127.9,127.8,124.5,124.3,124.1$, 123.9, 123.6, 120.2, -1.1; IR (KBr) $v_{\max } 3065,2956,2930,2861,1604,1549,1465,1449$, 1413, 1347, 1285, 1245, 1151, 1081, 982, 932, 868, 842, 755, 734, $705 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{473} \mathrm{H}_{35} \mathrm{Si}(\mathrm{M}+\mathrm{H})$ calcd: 626.24235 , found: $626.24239 ; R_{f}(2 / 1$ hexanes/EtOAc $)=0.64$ (silica gel plate).
$6^{\prime}, 7^{\prime}-\mathrm{Di}($ naphthalen-1-yl)dispiro[fluorene-9,5'-indeno[2,1-c]fluorene-8',9'-fluorene]

(23i). With $\mathbf{2 2 i}$ ( $134 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) following the general procedure E . Column chromatography of the residue on silica gel (10/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided $109 \mathrm{mg}(54 \%)$ of the title compound as a colorless solid: mp (decomp) $>260^{\circ} \mathrm{C}$; Due to lack of solubility in commom deuterated solvents only weak signals for ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR were detectable: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.85(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.52(\mathrm{dt}, J=7.9,0.6$ Hz, 2H), 7.06-7.23 (m, 12H), 6.85-6.93 (m, 6H), 6.80 (d, $J=7.7 \mathrm{~Hz}, 4 \mathrm{H}), 6.67$ (d, $J=7.6 \mathrm{~Hz}$, $2 \mathrm{H}), 6.58(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.52(\mathrm{dt}, J=7.5,0.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.38(\mathrm{dt}, J=7.5,1.0 \mathrm{~Hz}, 2 \mathrm{H})$, 6.02 (t, $J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 5.59(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 150.7$, $149.8,147.6,145.6,142.3,141.3,140.7,137.4,137.4,132.9,131.6,131.4,127.8,127.3$, $127.2,127.1,127.0,126.7,126.5,126.2,125.9,124.1,123.8,123.7,123.5,123.0,119.9$, 118.4, 76.79, 66.3, 60.3; IR (KBr) $v_{\max } 3059,3038,3010,2959,1601,1508,1447,1379$, 1364, 1282, 1158, 1030, 1005, 943, 800, 778, 752, 739, $727 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{64} \mathrm{H}_{39}(\mathrm{M}$ + H) calcd: 807.30463, found: $807.30505 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.65$ (silica gel plate).

## 6'-(4-Methoxyphenyl)-7'-(4-(trifluoromethyl)phenyl)dispirofluorene-9,5'-indeno[2,1-c]-


fluorene-8',9'-fluorene (29). With 28 ( $53 \mathrm{mg}, 0.10 \mathrm{mmol}$ ) following the general procedure E. Column chromatography of the residue on silica gel ( $5 / 1 / 1$ hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided $69 \mathrm{mg}(82 \%)$ of the title compound as a colorless solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.78(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.51(\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H})$, 7.32 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.25(\mathrm{~d}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.14-7.20(\mathrm{~m}, 6 \mathrm{H}), 7.08(\mathrm{t}, J=7.5 \mathrm{~Hz}, 4 \mathrm{H})$, $6.92(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 4 \mathrm{H}), 6.62(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.44(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 5.81(\mathrm{~d}, J=7.9$ $\mathrm{Hz}, 2 \mathrm{H}), 5.75(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}) 5.58(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}) 3.48(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 156.5,150.5,150.1,148.6,148.4,147.4,146.5,142.0,141.9,141.2,140.0,138.6$, $138.3,137.0,136.7,130.4,129.8,128.0,127.82,127.80$, 127.31, 127.28, 127.2, 127.12, 127.07, 124.1, 123.8, 123.5-131.7 (q, $\left.{ }^{1} J_{\mathrm{C}-\mathrm{F}}=271.9 \mathrm{~Hz}\right), 123.7,123.53,123.49,122.2\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}\right.$ $=3.8 \mathrm{~Hz}), 119.8,119.7,111.3,66.2,66.0,54.9$; IR (KBr) $v_{\max } 3062$, 3037, 2951, 2831, 1614, $1510,1465,1448,1418,1323,1244,1165,1119,1107,1070,1019,846,829,756,738 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{58} \mathrm{H}_{36} \mathrm{OF}_{3}(\mathrm{M}+\mathrm{H})$ calcd: 805.27128, found: 805.27179; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.64($ silica gel plate $)$.

## 3'-Methoxy-6',7'-di-p-tolyl-10'-(trifluoromethyl)dispirofluorene-9,5'-indeno[2,1-c]


fluorene-8',9'-fluorene (34). With $33(30 \mathrm{mg}, 0.05 \mathrm{mmol})$ following the general procedure E. Column chromatography of the residue on silica gel ( $5 / 1 / 1$ hexanes/EtOAc/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided $35 \mathrm{mg}(84 \%)$ of the title compound as a colorless solid: mp (decomp) $>260{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.79(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 8.58(\mathrm{~d}, J=8.7 \mathrm{~Hz}$, $1 \mathrm{H}), 7.71$ (dd, $J=8.2,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.25(\mathrm{~m}, 5 \mathrm{H}), 7.13-7.19(\mathrm{~m}, 4 \mathrm{H}), 7.06(\mathrm{~m}, 4 \mathrm{H}), 6.90(\mathrm{~d}, J$ $=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.88(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.79(\mathrm{~s}, 1 \mathrm{H}), 6.10(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.96(\mathrm{~d}, J=7.9$ $\mathrm{Hz}, 4 \mathrm{H}$ ), 5.52 (dd, $J=8.0,2.5 \mathrm{~Hz}, 4 \mathrm{H}$ ), $3.64(\mathrm{~s}, 3 \mathrm{H}), 1.88(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ) $\delta 159.5,152.9,151.0,148.7,147.8,147.5,147.0,145.1,142.1,142.0,137.0-141.7$ (q, $\left.{ }^{1} J_{\mathrm{C}-\mathrm{F}}=267.7 \mathrm{~Hz}\right), 133.8,133.7,133.4,132.8,132.7,129.2,129.1,127.1,127.0,126.9,126.0$, $125.9,124.5,124.1,123.8,123.7,123.6,123.0,120.8\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.2 \mathrm{~Hz}\right), 119.7,119.5,113.2$, 109.3, 66.2, 66.1, 55.4, 29.7, 20.9; IR (KBr) $v_{\max } 3067,3039,3018,2920,2834,1605,1518$, 1481, 1447, 1369, 1329, 1266, 1242, 1166, 1126, 1078, 1033, 814, 740, $729 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{60} \mathrm{H}_{39} \mathrm{OF}_{3}(\mathrm{M}+\mathrm{Na})$ calcd: 855.28452, found: 855.28494; $R_{f}(2 / 1$ hexanes/EtOAc $)=$ 0.68 (silica gel plate).

8',9'-Bis(4-methoxyphenyl)dispiro[fluorene-9,7'-benzo[c]benzo[6,7]indeno[1,2-g]-

fluorene-10',9'-fluorene] (41a). With 40 a ( $59 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) following the general procedure E. Column chromatography of the residue on silica gel ( $5 / 1 / 1$ hexanes/EtOAc/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided 67 mg (78\%) of the title compound as a slight orange solid: mp (decomp) $>260$ ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.87(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.80(\mathrm{~d}, J=$ $8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.63(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.52(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.33(\mathrm{~d}, J$ $=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.30(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 4 \mathrm{H}), 7.22(\mathrm{~m}, 4 \mathrm{H}), 7.06(\mathrm{t}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.92(\mathrm{t}, J=7.5$ $\mathrm{Hz}, 2 \mathrm{H}), 6.75(\mathrm{~m}, 4 \mathrm{H}), 6.71(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 5.89(\mathrm{dd}, J=8.4,2.7 \mathrm{~Hz}, 2 \mathrm{H}), 5.75(\mathrm{dd}, J=$ 8.3, 2.4 Hz, 2H), $5.68(\mathrm{~m}, 4 \mathrm{H}), 3.51(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 156.4,148.4$, $148.1,147.6,147.3,142.7,142.0,138.9,137.8,136.2,133.3,131.4,129.8,129.6,128.9$, $128.8,128.1,127.34,127.30,127.2,127.0,126.9,125.0,124.4,124.0,123.8,121.3,120.2$, $119.5,111.5,111.1,66.9,54.9$; IR (KBr) $v_{\max } 3052,3052,2832,1611,1516,1448,1364$, 1285, 1244, 1175, 1107, 1035, 815, $739 \mathrm{~cm}^{-1}$; HRMS $(m / z)$ for $\mathrm{C}_{66} \mathrm{H}_{43} \mathrm{O}_{2}(\mathrm{M}+\mathrm{H})$ calcd: 867.32576 , found: $867.32593 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.52$ (silica gel plate $)$.

fluorene] (41b). With 40 b ( $53 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) following the general procedure E. Column chromatography of the residue on silica gel (5/1/1 hexanes $/ \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided $61 \mathrm{mg}(76 \%)$ of the title compound as a slight orange solid: mp (decomp) $>260{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.87(\mathrm{dd}, J=8.5,0.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.80(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.63$ (d, $J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.44(\mathrm{~m}, 2 \mathrm{H}), 7.30-7.34(\mathrm{~m}, 4 \mathrm{H}), 7.20-7.24(\mathrm{~m}, 6 \mathrm{H}), 7.04(\mathrm{dt}, J=7.5,1.0$ $\mathrm{Hz}, 2 \mathrm{H}), 6.91(\mathrm{dt}, J=7.6,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 6.67-6.77(\mathrm{~m}, 6 \mathrm{H}), 6.48(\mathrm{~m}, 2 \mathrm{H}), 6.33(\mathrm{~m}, 2 \mathrm{H}), 6.17$ (m, 2H), 5.76 (dt, $J=6.9,1.2 \mathrm{~Hz}, 4 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 148.5,148.0,147.5$, 146.7, 146.6, 142.7, 141.9, 138.8, 137.7, 136.4, 135.9, 133.4, 130.3, 129.64, 129.61, 129.0, $128.9,128.1,127.4,127.3,127.0$, $126.9,125.6,125.3$, $125.1,124.5,124.4,124.0,123.8$, 121.3, 120.2, 119.9, 119.6, 67.8.; IR (KBr) $v_{\max } 3049,3018,2956,2922,2853,1602,1515$, 1446, 1280, 1239, 1215, 1177, 1149, 1032, 810, 741, $696 \mathrm{~cm}^{-1} ;$ HRMS $(\mathrm{m} / \mathrm{z})$ for $\mathrm{C}_{64} \mathrm{H}_{39}(\mathrm{M}+$ H) calcd: 807.30463 , found: $807.30471 ; R_{f}(2 / 1$ hexanes $/ E t O A c)=0.56$ (silica gel plate) .

## 8',9'-Bis(4-(trifluoromethyl)phenyl)dispiro[fluorene-9,7'-benzo[c]benzo[6,7]indeno[1,2-

 $g]$-fluorene-10',9'-fluorene] (41c). With 40c (67 mg, 0.1 mmol ) following the general procedure E. Column chromatography of the residue on silica gel ( $5 / 1 / 1$ hexanes/EtOAc/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) provided 74 mg (79\%) of the title compound as a slight orange solid: mp (decomp) $>260$ ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.84(\mathrm{dd}, J=8.4,4.6 \mathrm{~Hz}, 4 \mathrm{H}), 7.68$ (d, $J=8.3 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.45(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.36(\mathrm{dt}, J=7.5,1.0 \mathrm{~Hz}$, $2 \mathrm{H}), 7.30(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.21-7.28(\mathrm{~m}, 6 \mathrm{H}), 7.09(\mathrm{dt}, J=7.5,0.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.93(\mathrm{dt}, J=$ $7.5,0.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.76(\mathrm{~m}, 4 \mathrm{H}), 6.70(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.61(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.45(\mathrm{~d}, J=$ $7.9 \mathrm{~Hz}, 2 \mathrm{H}$ ), $5.90(\mathrm{t}, J=7.2 \mathrm{~Hz}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 148.2,147.3,146.9$, $146.8,142.6,141.8,139.3,138.4,137.0,135.4,133.4,130.4,129.7,129.5,129.1,128.9$, $128.2,128.1,127.8,127.5,127.4,127.2,127.1,127.0$, 126.8 , 125.3, 125.2, 124.6, 124.0, $123.9,122.6\left(\mathrm{q},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=3.4 \mathrm{~Hz}\right), 121.2,120.4,119.8,66.6$; IR (KBr) $v_{\max } 3056,2922,2866$, 1619, 1446, 1405, 1325, 1246, 1166, 1125, 1079, 1018, 810, $738 \mathrm{~cm}^{-1}$; HRMS ( $\mathrm{m} / \mathrm{z}$ ) for $\mathrm{C}_{66} \mathrm{H}_{37} \mathrm{~F}_{6}(\mathrm{M}+\mathrm{H})$ calcd: 943.27940, found: 943.27954; $R_{f}(2 / 1$ hexanes $/ E t O A c)=0.58$ (silica gel plate).

### 6.3.4. X-ray analysis of dispiroindeno[2,1-c]fluorenes

Crystallographic data was collected on Bruker D8 VENTURE Kappa Duo PHOTON100 by $\mathrm{I} \mu \mathrm{S}$ micro-focus sealed tube at a temperature 150 K . The structures were solved by direct methods (XP) and refined by full matrix least squares based on $F^{2}$ (SHELXL2014). ${ }^{69}$ The hydrogen atoms on carbons were fixed into idealised positions (riding model) and assigned temperature factors either $\mathrm{H}_{\mathrm{iso}}(\mathrm{H})=1.2 \mathrm{U}_{\mathrm{eq}}$ (pivot atom) or 1.5 for methyl moiety. PLATON/ SQUEEZE (Spek, A. L., Acta Crystallographica Section D-Biological Crystallography $\mathbf{2 0 0 9}, 65,148$ ) procedure was used to correct the diffraction data of $\mathbf{4 1}$ for the presence of the disordered solvents.

Table Exp9. X-ray crystallographic data for indenofluorenes 23b, 40a, and 41a

|  | 23b | 40a | 41a |
| :---: | :---: | :---: | :---: |
| Formula | $\mathrm{C}_{58} \mathrm{H}_{38}$ | $\mathrm{C}_{42} \mathrm{H}_{26} \mathrm{O}_{2}$ | $\mathrm{C}_{66} \mathrm{H}_{42} \mathrm{O}_{2}$ |
| M.w. | 734.88 | 594.63 | 866.99 |
| $\lambda$ [ ${ }^{\text {a }}$ ] | 0.71073 | 0.71073 | 1.54178 |
| Crystal system | monoclinic | triclinic | orthorhombic |
| Space group [No.] | $P 2_{1} / n$ (No. 14) | P-1 (No. 2) | Pna2 ${ }_{1}$ (No 33) |
| $a[\AA]$ | 13.4315(4) | 9.7309(3) | 16.3016(5) |
| $b[\AA]$ | 13.7634(3) | 12.1645(4) | 13.3191(4) |
| $c[\AA]$ | 20.9116(5) | 14.1640(5) | 21.3258 (6) |
| $a\left[{ }^{\circ}\right]$ |  | 65.993(1) |  |
| $\beta$ [ $\left.{ }^{\circ}\right]$ | 98.4780(10) | 86.855(1) |  |
| $\gamma\left[{ }^{\circ}\right]$ |  | 86.512(1) |  |
| Z | 4 | 2 | 4 |
| $\mathrm{V}\left[\AA^{3}\right]$ | 3823.54(17) | 1527.91(9) | 4630.3(2) |
| $\mathrm{D}_{x}\left[\mathrm{~g} \mathrm{~cm}^{-3}\right]$ | 1.277 | 1.293 | 1.244 |
| Crystal size [mm] | $0.09 \times 0.20 \times 0.19$ | $0.09 \times 0.27 \times 0.28$ | $0.03 \times 0.17 \times 0.52$ |
| Crystal color, shape | a colorless prism | a red plate | a colorless bar |
| $\mu\left[\mathrm{mm}^{-1}\right]$ | 0.072 | 0.082 | 0.567 |
| $T_{\text {min }}, T_{\text {max }}$ | 0.89, 0.99 | 0.96, 0.98 | 0.88, 0.98 |
| Measured reflections | 41342 | 33305 | 41883 |
| Independent diffract. ( $R_{\text {int }}{ }^{\text {a }}$ ) | 7545 (0.034) | 7039 (0.024) | 8955 (0.029) |
| Observed diffract. [I>2 $\sigma(\mathrm{I})$ ] | 6313 | 6085 | 8711 |
| No. of parameters | 525 | 417 | 615 |
| $R^{\text {b }}$ | 0.044 | 0.041 | 0.029 |
| $w R\left(F^{2}\right)$ for all data | 0.120 | 0.111 | 0.071 |
| $\mathrm{GOF}^{\text {c }}$ | 1.04 | 1.03 | 1.05 |
| Residual electron density [ $\mathrm{e} / \mathrm{A}^{3}$ ] | 0.41, -0.24 | 0.27, -0.31 | 0.25, -0.18 |
| CCDC no. | 1548067 | 1548069 | 1548068 |
| $\begin{aligned} & { }^{{ }^{a} R_{\text {int }}=\Sigma\left\|F_{\mathrm{o}}{ }^{2}-F_{\mathrm{o}, \text { mean }}{ }^{2}\right\| / \Sigma F_{\mathrm{o}}^{2}} \\ & { }^{b} R(F)=\Sigma\| \| F_{\mathrm{o}}\left\|-\left\|F_{\mathrm{c}}\right\|\right\| / \Sigma \\ & {\left[\Sigma\left(w\left(F_{\mathrm{o}}{ }^{2}-F_{\mathrm{c}}^{2}\right)^{2}\right) /\left(N_{\text {diffrs }}-N_{\text {params }}\right)\right.} \end{aligned}$ | $\left\|F_{\mathrm{o}}\right\|, w R\left(F^{2}\right)=$ | $\left(w\left(F_{\mathrm{o}}^{2}-F_{\mathrm{c}}^{2}\right)^{2}\right) /($ | $\left.\left.w\left(F_{\mathrm{o}}^{2}\right)^{2}\right)\right]^{1 / 2} . \quad{ }^{c} \mathrm{GC}$ |



Figure Exp14. Platon plot of 23b showing displacement ellipsoids at a $50 \%$ probability level


Figure Exp15. PLATON drawing of 40a showing displacement ellipsoids on the 50\% probability level


Figure Exp16. PLATON drawing of 41a showing displacement ellipsoids on the $50 \%$ probability level

### 6.3.5. Photophysical analysis of dispiroindeno[2,1-c]fluorenes

## Chemicals.

9,10-Diphenylanthracene (DPA) was purchased from Sigma-Aldrich.

The UV/Vis absorption spectra were recorded using Unicam 340 spectrometer. Corrected steady-state emission spectra were recorded on an Aminco Bowman (AB2) spectrometer.

Quantum yields of fluorescence of a sample $\left(\Phi_{\mathrm{s}}\right)$ were calculated relative to DPA ( $\Phi_{\text {ref }}$ $=0.97$ in cyclohexane) as a reference. $\Phi_{\mathrm{s}}$ was determined according to the following equation: ${ }^{70}$

$$
\begin{equation*}
\Phi_{\mathrm{s}}=\Phi_{\mathrm{ref} \times} \times \mathrm{A}_{\mathrm{s}} / \mathrm{A}_{\mathrm{ref}} \tag{1}
\end{equation*}
$$

where $A_{s}$ and $A_{\text {ref }}$ represent integrated areas of the emission (in arbitrary units) of a sample and the reference, respectively.

Micromolar solutions of the reference (DPA) and the samples were prepared in cyclohexane. The absorption of the reference and the samples were the same at the selected excitation wavelength ( $\mathrm{A} \leq 0.1$ ). Quantum yields were calculated from 5 independent measurments. The average value is reported.

The emission quantum yields ( $\Phi_{\mathrm{s}}$ ) of the derivatives were determined in cyclohexane by standard procedures with DPA as reference (see Figures Exp17-19 and Tables Exp10-12).

## UV/Vis absorption and emission spectra of 23a-i.



23a


23b


23c


23d


23e


23f


23g


23h


23i



Figure Exp17. Absorption spectra of $10^{-6} \mathrm{M} \mathrm{DPA}$ and 23a-i in cyclohexane (left). Normalized and corrected emission spectra of $10^{-6} \mathrm{M}$ samples in cyclohexane. Excitation $\lambda=$ 264-271 nm (right).

Table Exp10. Photophysical properties of 23a-i derivatives in cyclohexane

| Sample | $\lambda_{\text {abs }} / \mathrm{nm}\left(\varepsilon / 10^{5} \mathrm{~mol}^{-1} \mathrm{dm}^{3} \mathrm{~cm}^{-1}\right)$ | $\lambda_{\text {exc }} / \mathrm{nm}$ | $\lambda_{\mathrm{em}} / \mathrm{nm}$ | $\Phi_{\mathrm{s}}{ }^{\text {a }}$ |
| :--- | :--- | :--- | :--- | :--- |
| 23a | $213(0.2), 240(\mathrm{sh}), 265(0.1), 612(0.08)$ | 264 | 370,380 | 0.31 |
| 23b | $207(0.1), 240(\mathrm{sh}), 267(0.07), 312(0.05)$ | 266 | 366 | 0.09 |
| 23c | $225(0.1), 239(\mathrm{sh}), 266(0.07), 311(0.05)$ | 266 | 366 | 0.67 |
| 23d | $217(0.1), 223(\mathrm{sh}), 266(0.05), 312(0.04)$ | 269 | 365 | 0.33 |
| 23e | $226(0.07), 239(\mathrm{sh}), 268(0.04), 311(0.03)$ | 269 | 364 | 0.70 |
| 23f | $226(0.07), 238(\mathrm{sh}), 268(0.03), 312(0.02)$ | 271 | 366,380 | 0.20 |
| 23g | $237(0.09), 258(\mathrm{sh}), 277(\mathrm{sh}), 311(0.05)$ | 269 | 351,359 | 0.63 |
| 23h | $214(0.2), 231(\mathrm{sh}), 282(\mathrm{sh}), 296(0.1)$ | 266 | - | 0.00 |
| 23i | $217(0.4), 242(\mathrm{sh}), 268(0.1), 312(0.09)$ | 264 | 369 | 0.28 |

[^2]
## UV/Vis absorption and emission spectra of 29 and 34.



29


34



Figure Exp18. Absorption spectra of $10^{-6} \mathrm{M}$ DPA, 29 and 34 in cyclohexane (left). Normalized and corrected emission spectra of $10^{-6} \mathrm{M}$ samples in cyclohexane. Excitation $\lambda=$ 268-272 nm (right).

Table Exp11. Photophysical properties of 29 and 34 in cyclohexane

| Sample | $\lambda_{\text {abs }} / \mathrm{nm}\left(\varepsilon / 10^{5} \mathrm{~mol}^{-1} \mathrm{dm}^{3} \mathrm{~cm}^{-1}\right)$ | $\lambda_{\text {exc }} / \mathrm{nm}$ | $\lambda_{\text {em }} / \mathrm{nm}$ | $\Phi_{\mathrm{s}}{ }^{\text {a }}$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathbf{2 9}$ | $227(0.05), 238($ sh $), 266($ sh $), 312(0.03)$ | 272 | 370 | 0.54 |
| $\mathbf{3 4}$ | $225(0.09), 236($ sh $), 274(0.06), 311(0.04)$ | 268 | 389 | 0.60 |

[^3]
## UV/Vis absorption and emission spectra of 41a-c.





Figure Exp19. Absorption spectra of $10^{-6} \mathrm{M}$ DPA and samples 41a-c in cyclohexane (left). Normalized and corrected emission spectra of $10^{-6} \mathrm{M}$ samples in cyclohexane. Excitation $\lambda=$ 266-269 nm (right).

Table Exp12. Photophysical properties of 41a-c derivatives in cyclohexane

| Sample | $\lambda_{\text {abs }} / \mathrm{nm}\left(\varepsilon / 10^{5} \mathrm{~mol}^{-1} \mathrm{dm}^{3} \mathrm{~cm}^{-1}\right)$ | $\lambda_{\text {exc }} / \mathrm{nm}$ | $\lambda_{\text {em }} / \mathrm{nm}$ | $\Phi_{\mathrm{s}}{ }^{\text {a }}$ |
| :--- | :--- | :--- | :--- | :--- |
| 41a | $217(0.1), 240(\mathrm{sh}), 315(0.02)$ | 269 | 418,428 | 0.75 |
| 41b | $217(0.09), 225(\mathrm{sh}), 245(0.09), 314(0.02)$ | 269 | 398,419 | 0.87 |
| 41c | $229(\mathrm{sh}), 242(0.1), 314(0.03)$ | 269 | 408,423 | 0.88 |

[^4]
## 7. Author's publications

Author's Publications relevant to This Thesis

1. A $[2+2+2]$ Cyclotrimerization Approach to Selectively Substituted Fluorenes and Fluorenols, and Their Conversion to 9,9'- Spirobifluorenes
R. P. Kaiser, F. Hessler, J. Mosinger, I. Císařová, M. Kotora; Chem. Eur. J., 2015, 21, 13577.
2. Synthesis of selectively 4-substituted 9,9'-spirobifluorenes and modulation of their photophysical properties
R. P. Kaiser, J. Mosinger, I. Císařová, M. Kotora; Org. Biomol. Chem., 2017, 15, 6913.

## Other Publications

1. Selective Borylation of [4]-helicene
D. Nečas, R. P. Kaiser, J. Ulč; Eur. J. Org. Chem., 2016, 21, 5647.
2. Direct regioselective $\mathbf{C}-\mathbf{H}$ borylation of [5]-helicene
R. P. Kaiser, J. Ulč, I. Císařová, D. Nečas; $R S C$ $A d v$., 2018, 8,580

## 8. Abstrakt

Disertační práce je rozdělena do dvou částí zabývajících se syntézou a analýzou strukturně příbuzných (i) $9,9^{\prime}$-spirobifluorenů (SBFs) a (ii) dispiroindeno[2,1-c]fluorenů (DSIFs).

Disertační práce je rozdělena do dvou částí zaměřených na syntézu a fotofyzikální analýzu dvou skupin strukturně prŕbuzných látek: (i) 9,9 '-spirobifluorenů (SBFs) a (ii) dispiroindeno[2,1-c]fluorenů (DS-IFs).
(i) V první části byla vyvinuta nová pětikroková syntéza SBFs , kde klíčovým krokem byla Rh-komplexy katalyzovaná $[2+2+2]$ cyklotrimerizace symetricky či nesymetricky substituovaných diyn alkoholů s alkyny. Optimalizací této reakce (katalyzátor, teplota, čas, rozpouštědlo atd.) byly nalezeny ideální podmínky pro přípravu 1,2,3,4-substituovaných fluorenolů (klíčové intermediáty) a jako nejefektivnější se ukázal být Wilkinsonův katalyzátor $\left(\mathrm{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}\right)$. Tyto různě substituované fluorenoly byly následně ve dvou krocích převedeny na 19 nových SBFs, u kterých byl zkoumán vliv substituentu (elektronakceptorní, elektrondonorní) v poloze 4 na jejich spektroskopické vlastnosti. Získaná data ukazují emisní maxima $\lambda_{\text {em }}$ v rozsahu 315 až 389 nm s excelentními kvantovými výtěžky $\Phi_{\mathrm{s}}$ (až 1 ).
(ii) V druhé části byla tato nová metoda pro přípravu SBFs aplikována na přípravu DS-IFs. Rh-katalyzovaná $[2+2+2]$ cyklotrimerizace triyndiolů poskytla indeno[2,1c]fluorendioly ve vysokých výtěžcích. Tyto byly opět ve dvou krocích převedeny na odpovídající DS-IFs. Celkem bylo připraveno 9 symetricky substituovaných DS-IFs (pozice 6 a 7) a 2 nesymetricky substituované DS-IFs (pozice 3 a 10). Pro potvrzení syntetického potenciálu této reakční sekvence byly připraveny také 3 [7]helikální DS-IFs. Spektroskopická analýza všech produktů poskytla emisní maxima v rozsahu 351 až 428 nm s excelentními kvantovými výtěžky $\Phi_{\mathrm{s}}$ (až 0.8 ). [7]helikální DS-IFs mají navíc nejvyšší dosud namě̌̌ené hodnoty kvantových výtěžků $\Phi_{\mathrm{s}}$ pro tento tip látek.

## 9. Abstract

This work is divided into two parts that are focused on the synthesis and assessment of photophysical properties of two structurally related compounds: (i) $9,9^{\prime}$-spirobifluorenes (SBFs) and (ii) dispiroindeno[2,1-c]fluorenes (DS-IFs).
(i) In the first part, a five-step synthetic approach to SBFs was developed and as the crucial step was used an intermolecular [2+2+2] cyclotrimerization of symmetrically or unsymmetrically substituted diynols with alkynes catalyzed by Rh-complexes. Catalyst screening showed that Wilkinson's catalyst $\left(\operatorname{RhCl}\left(\mathrm{PPh}_{3}\right)_{3}\right)$ had the highest efficiency in yielding 1,2,3,4-substituted fluorenols - the key intermediates. The fluorenols were then converted into SBFs bearing various electron-donating and -withdrawing groups, aromatic substituents, and $\pi$-extended aromatic hydrocarbon moieties (PAHs). Altogether 19 different SBFs were prepared and their photophysical properties screened. The fluorescent emission maxima $\lambda_{\text {em }}$ were in the range of 315-389 nm with excellent quantum yields $\Phi_{\mathrm{s}}$ (up to 1.00 ). As far as the substituent effect is concerned the presence of electron-withdrawing substituents on the SBF scaffold results in the red-shift of the emission maxima.
(ii) In the second part, a similar synthetic strategy was applied for synthesis DS-IFs. In this instance was utilized again the Rh-complex catalyzed intramolecular [2+2+2] cyclotrimerization of triyndiols, which provided indeno[2,1-c]fluorenr-5,8-diols in high yields. These were then converted into the corresponding DS-IFs. In total, 9 symmetrically substituted DS-IFs at positions 6 and 7 and 2 unsymmetrically substituted DS-IFs at positions 3 and 10 were synthesized. Furthermore, to demonstrate the potential of this synthetic approach, three [7]-helical DS-IFs were prepared. Photophysical properties of all compounds were evaluated as well. Their fluorescent emission maxima $\lambda_{\mathrm{em}}$ were in the range of 351-428 nm with excellent quantum yields $\Phi_{\mathrm{s}}$ (up to 0.80 ). [7]-helical DS-IFs have the highest measured quantum yields among the class of helicene like compounds recorded thus far.

## 10. List of Abbreviations

| DMAP | $N, N^{\prime}$-Dimethylamino pyridine |
| :--- | :--- |
| DMA | $N, N^{\prime}$-Dimethylacetamide |
| DMF | $N, N^{\prime}$-Dimethylformamide |
| DPA | 9,10 -Diphenylanthracene |
| Et | Ethyl |
| eq | Equivalent |
| Fc | Ferrocenyl |
| h | Hour |
| HRMS | High Resolution Mass Spectroscopy |
| Hz | Hertz |
| IR | Infrared Spectroscopy |
| $J$ | Coupling constant |
| LDA | Lithium diisopropylamide |
| Me | Methyl |
| mp | Melting Point |
| MS | Molecular sieve |
| NMR | Nuclear Magnetic Resonance |
| $p-$ Tol | para-Tolyl |
| PCC | Pyridinium chlorochromate |
| PEPPSI ${ }^{\text {TM }}$ | Pyridine-enhanced precatalyst preparation stabilization and initiation |
|  | ([1,3-Bis(2,6-Diisopropylphenyl)imidazol-2-ylidene](3-chloropyridyl) |
|  | palladium(II) dichloride) |
| Ph | Phenyl |
| ppm | Parts per million |
| $n$-Pr | $n$-Propyl |
| THF | Tetrahydrofuran |
| TLC | Thin layer chromatography |
| UV | Ultraviolet |
|  |  |

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[^0]:    ${ }^{\text {a }}$ Quantum yields were calculated
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[^3]:    ${ }^{\text {a }}$ Quantum yields were calculated relative to DPA ( $\Phi_{\mathrm{r}}=0.97$ ).

[^4]:    ${ }^{\text {a }}$ Quantum yields were calculated relative to DPA ( $\Phi_{\mathrm{r}}=0.97$ ).

