



# Cholesteryl esters of $\omega$ -(*O*-acyl)-hydroxy fatty acids in vernix caseosa<sup>S</sup>

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**Abstract** Cholesteryl esters of  $\omega$ -(*O*-acyl)-hydroxy FAs (Chl- $\omega$ OAHFAs) were identified for the first time in vernix caseosa and characterized using chromatography and MS. Chl- $\omega$ OAHFAs were isolated using adsorption chromatography on silica gel and magnesium hydroxide. Their general structure was established using high-resolution and tandem MS of intact lipids, and products of their transesterification and derivatizations. Individual molecular species were characterized using nonaqueous reversed-phase HPLC coupled to atmospheric pressure chemical ionization. The analytes were detected as protonated molecules, and their structures were elucidated in the negative ion mode using controlled thermal decomposition and data-dependent fragmentation. About three hundred molecular species of Chl- $\omega$ OAHFAs were identified in this way. The most abundant Chl- $\omega$ OAHFAs contained 32:1  $\omega$ -hydroxy FA ( $\omega$ -HFA) and 14:0, 15:0, 16:0, 16:1, and 18:1 FAs. The double bond in the 32:1  $\omega$ -HFA was in the *n*-7 and *n*-9 positions. Chl- $\omega$ OAHFAs are estimated to account for approximately 1–2% of vernix caseosa lipids.—Kalužíková, A., V. Vrkoslav, E. Harazim, M. Hoskovec, R. Plavka, M. Buděšínský, Z. Bosáková, and J. Cvačka. Cholesteryl esters of  $\omega$ -(*O*-acyl)-hydroxy fatty acids in vernix caseosa. *J. Lipid Res.* 2017. 58: 1579–1590.

**Supplementary key words** skin lipids • neutral lipids • cholesterol • lipidomics • mass spectrometry

Vernix caseosa is a white cheese-like naturally occurring biofilm that coats the skin of the fetus during the last trimester of gestation and usually remains present on the skin during delivery. It is a highly cellular material consisting of hydrophilic desquamated corneocytes embedded in a lipid matrix. Vernix caseosa is essential for skin development in utero, as well as post-birth adaptation, providing multiple functions (1–3).

Vernix caseosa consists of water (80%), proteins (10%), and a complex mixture of lipids (10%). Although investigations of its lipid composition started more than seventy years ago (4), the entire lipidome of vernix caseosa has not yet been described comprehensively. Lipids exist in intercellular space as free (extractable) components, or they are covalently bound to the cornified envelope. About 90% of the free lipids are nonpolar species like squalene, sterol esters, wax esters, diesters, and triacylglycerols, and the remaining 10% are barrier lipids, mostly cholesterol, free FAs, and ceramides. The lipids covalently linked to the cornified envelope consist of  $\omega$ -hydroxy FAs ( $\omega$ HFAs) and  $\omega$ -hydroxyceramides (5).

Nonpolar diesters form 3–9% of the total vernix caseosa lipids (6, 7). The first report (6) characterized them as esters of an alkane 1,2-diol with two FAs [later described as type II diesters or 1,2-diol diesters (1,2-DDEs)] with a small amount of esters of a HFA with a fatty alcohol and an FA (type I diesters). Four years later, type II diesters were confirmed as the main constituents of the diester fraction (8, 9). The analyses of hydrolysis products also disclosed  $\alpha$ -hydroxy FAs ( $\alpha$ HFAs), sterols, and fatty alcohols, which prompted the authors to hypothesize on the existence of additional types of diesters, type I diesters, and type I diesters with the fatty alcohol having been replaced by a sterol (9). However, the existence of the additional type of diesters in vernix caseosa has never been substantiated. No diester lipids have been detected in adult human sebum or epidermal surface, where triacylglycerols and their

Abbreviations: APCI, atmospheric pressure chemical ionization; Chl- $\omega$ OAHFA, cholesteryl ester of  $\omega$ -(*O*-acyl)-hydroxy FA; CID, collision-induced dissociation; DMF, *N,N*-dimethylformamide; 1,2-DDE, 1,2-diol diester; FAME, FA methyl ester; HFA, hydroxy FA;  $\alpha$ HFA,  $\alpha$ -hydroxy FA;  $\omega$ HFA,  $\omega$ -hydroxy FA; HFAME, hydroxy FA methyl ester;  $\omega$ OAHFA,  $\omega$ -(*O*-acyl)-hydroxy FA; PDC, pyridinium dichromate; RDDE, ring plus double bond equivalent; St- $\omega$ OAHFA, sterol ester of  $\omega$ -(*O*-acyl)-hydroxy FA.

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<sup>S</sup>The online version of this article (available at <http://www.jlr.org>) contains a supplement.

This work was supported by the Czech Science Foundation (Project P206/12/0750) and Charles University in Prague (Project SVV260440).

Manuscript received 25 January 2017 and in revised form 22 May 2017.

Published, *JLR Papers in Press*, June 2, 2017

DOI <https://doi.org/10.1194/jlr.M075333>

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breakdown products (monoacylglycerols, diacylglycerols, and free FAs) are found instead (10, 11). On the contrary, the skin surface lipids of many animals are rich in diesters. Mammals like rabbit, cat, or cow biosynthesize type I diesters, whereas mouse, hamster, guinea-pig, or gerbil produce mostly type II diesters (1,2-DDE) (8, 12–14). A special sort of type II diester, with one FA replaced by isovaleric acid, has been found in the dog (15) and macaque (16). Uropygial (green) glands of birds mostly produce 2,3-DDE (17). The existence of various types of esters in the sebum of humans and animals has been discussed in terms of the physicochemical properties of the lipids and their likely roles in the skin protection and control of the microbiome (10, 11).

Recently, we reported on the analysis of molecular species of 1,2-DDE (18) from vernix caseosa. Diesters were isolated from a total lipid extract using TLC and separated by nonaqueous reversed-phase HPLC. The 1,2-DDE molecular species were identified using tandem MS with atmospheric pressure chemical ionization (APCI). In addition to 1,2-DDE, molecular species of an unknown lipid class appeared in the chromatogram at higher retention times. In this work, we isolated the unknown lipids from a fresh material using multistep chromatography and identified them as cholesteryl esters of  $\omega$ -(*O*-acyl)-hydroxy FAs (Chl- $\omega$ OAHFAs). To the best of our knowledge, these diesters represent a new lipid class for vernix caseosa. A HPLC/APCI-MS<sup>2</sup> method was developed and applied for comprehensive characterization of Chl- $\omega$ OAHFA molecular species in vernix caseosa.

## MATERIALS AND METHODS

### Materials

Vernix caseosa (1–2 g) was collected from healthy newborn subjects delivered at full term (gestation weeks 39–42) immediately after the delivery. The samples were stored in amber glass vials at  $-80^{\circ}\text{C}$ . The study was approved by the Ethics Committee of the General University Hospital, Prague (910/09 S-IV) and the samples were collected with a written informed parental consent.

### Chemicals

LC-MS grade acetonitrile, ethyl acetate, methanol, and propan-2-ol (Sigma-Aldrich) were used as received. Chloroform, hexane, and dichloromethane (Penta, Czech Republic) were distilled in glass from analytical-grade solvents. Rhodamine 6G and silica gel were both from Merck & Co. *N,N*-dimethylformamide (Acros Organics, part of Thermo Fisher Scientific, San Jose, CA), anhydrous magnesium sulfate, magnesium hydroxide, primuline, (*Z*)-octadec-9-enoyl chloride, 16-hydroxyhexadecanoic acid, hydrochloric acid, sodium bicarbonate, pyridine, *N,N*-dicyclohexylcarbodiimide, *N,O*-bis(TMS)acetamide, dichloromethane, 4-(dimethylamino)pyridine, cholesterol, pyridinium dichloroformate, acetyl chloride, and silver carbonate (all from Sigma-Aldrich) were of reagent grade and used as purchased.

### Standard synthesis

Cholesteryl ester of 16-[(9*Z*)-octadec-9-enoyl]oxy]hexadecanoic acid was synthesized by stirring (*Z*)-octadec-9-enoyl chloride

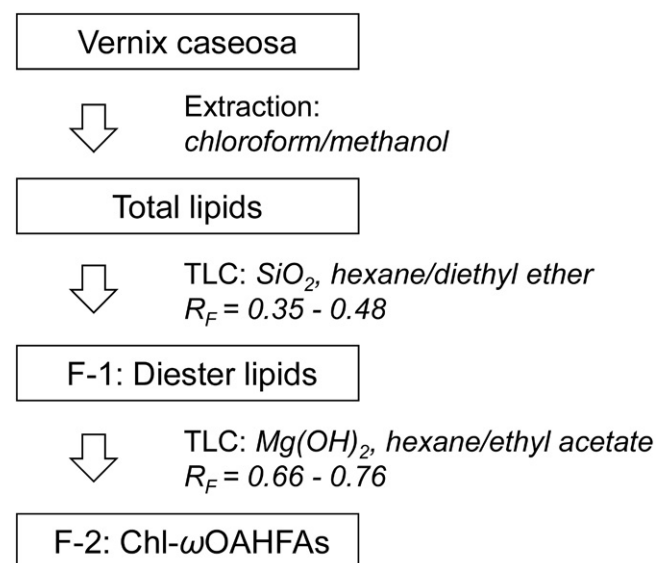
and 16-hydroxyhexadecanoic acid in anhydrous pyridine. The oily product was purified by flash chromatography and treated with cholesterol in dichloromethane in the presence of *N,N*-dicyclohexylcarbodiimide and 4-(dimethylamino)pyridine (19). The final purified product was obtained in 39% overall yield and its structure was verified by NMR. The details on the synthetic procedure and NMR data are given in the supplemental data (Section 1).

### Isolation of total lipids

Lipids were extracted from 20 samples equally representing gender of newborns (10 boys, 10 girls). Each sample was processed separately as follows: The sample (300 mg) was suspended in methanol:chloroform (2:1, by volume; 3 ml) in a conical-bottom glass centrifuge tube, and homogenized using a vortex shaker followed by a 2 min treatment in an ultrasonic bath. Then, chloroform (1 ml) and water (1.8 ml) were added and the suspension was shaken for 1 h. The mixture was centrifuged at 400 *g* for 5 min. The chloroform layer was transferred into a new glass tube, vortexed with 1.8 ml of water and collected after centrifugation. The water layer was re-extracted with 1 ml of chloroform. The chloroform extracts were combined, treated with anhydrous magnesium sulfate to remove the water residues, and filtered through pre-cleaned cotton wool. The lipid extracts from all 20 samples were combined and concentrated on a rotary evaporator ( $37^{\circ}\text{C}$ , 170 mbar) to approximately one-third of the original volume. The rest of the solvent was evaporated under a stream of nitrogen. In total, 6.0 g of vernix caseosa yielded 561.1 mg of total lipids. The lipids were reconstituted in chloroform:methanol (19:1 by vol) at the concentration of 30 mg/ml and stored at  $-20^{\circ}\text{C}$ .

### Fractionation of lipids

Approximately half of the total lipid extract was fractionated in two steps using semi-preparative TLC. In the first step, lipids were separated on glass plates coated with silica gel using hexane:diethyl ether (93:7, by volume) mobile phase. The zones were visualized under UV light after spraying with rhodamine 6G (0.05% in ethanol). Silica gel with diesters ( $R_f = 0.35\text{--}0.48$ ) was scraped off the plates and the lipids were extracted with freshly distilled diethyl ether. The solvent was evaporated under a nitrogen stream. The procedure was used repeatedly (approximately 3 mg of lipids separated in each step) and yielded 27.6 mg of diesters (F-1).



**Fig. 1.** Scheme of the isolation and fractionation procedure.

In the second step, F-1 was reconstituted in chloroform:methanol (2:1, by volume) at a concentration of 30 mg/ml and separated on glass plates coated with magnesium hydroxide (for details on the TLC plate preparation, see supplemental data, Section 2). Hexane:ethyl acetate (99.95:0.05, by volume) was used as a mobile phase. Prior to the separation, a filter paper was inserted into the developing chamber. After the filter paper was fully soaked with the solvent, each plate was developed twice to focus the zones; in the first step to 3/4 of the plate height and then, after air-drying, to the top. After air-drying, the zones were sprayed with 0.05% primuline in ethanol and then visualized under UV light. A synthesized standard [18:1(*n*-9)/16:0-Chl] was used to verify the  $R_f$  of Chl- $\omega$ OAHFAs. The sorbent layer corresponding to  $R_f = 0.66$ – $0.76$  was collected and extracted with diethyl ether. The procedure was used repeatedly to process the whole F-1 (approximately 3 mg separated in each step) and yielded 5.8 mg of Chl- $\omega$ OAHFAs (F-2). The lipids were dissolved in chloroform:methanol (9:1, by volume) at a concentration of 5 mg/ml and stored at  $-20^\circ\text{C}$ . The whole isolation and fractionation procedure is depicted in Fig. 1.

### Transesterification

Chl- $\omega$ OAHFAs were transesterified using an acid catalyst (20). Briefly, F-2 was dissolved in chloroform:methanol (2:3, by volume) in a small glass ampoule. After adding acetyl chloride, the ampoule was sealed and heated at  $70^\circ\text{C}$  for 60 min. The reaction mixture was neutralized with silver carbonate and the organic layer was used for further analyses.

### Trimethylsilylation

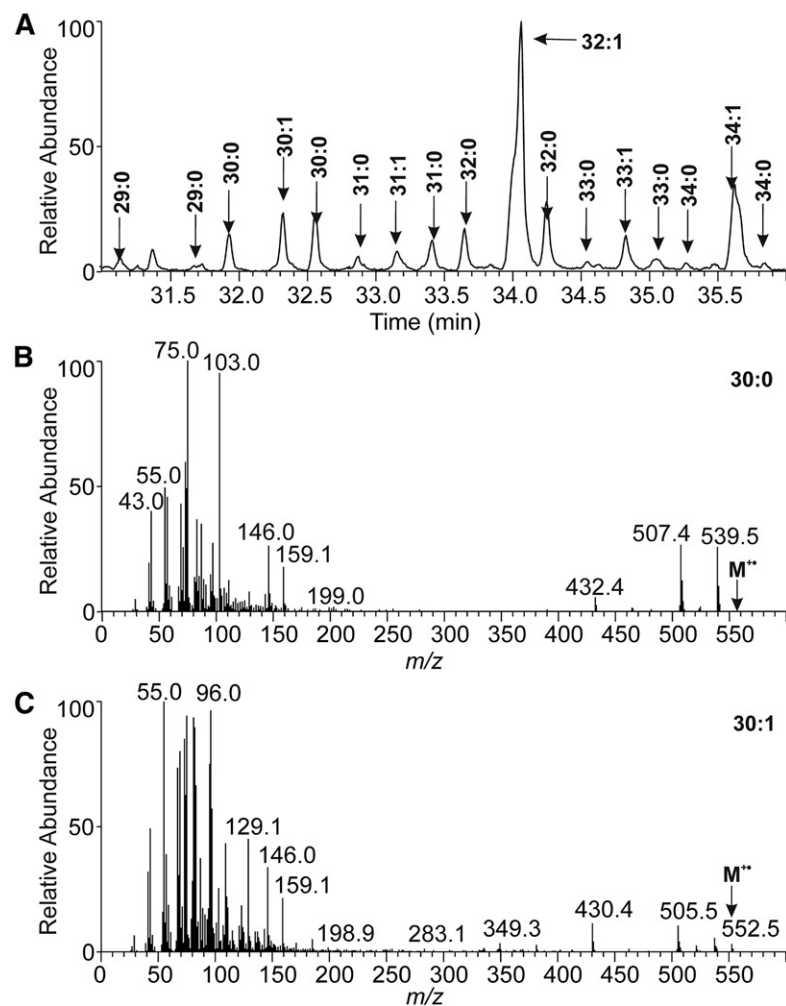
TMS derivatives of HFA were prepared according to a published procedure (21). FA methyl esters (FAMES) were dissolved in dried acetonitrile (0.4 mg/ml) and treated with excess of *N,O*-bis(TMS)acetamide at  $40^\circ\text{C}$  for 10 min. The solvent was evaporated under a stream of nitrogen, the residues were dissolved in chloroform (250  $\mu\text{l}$ ), and the sample was injected onto the GC column.

### Oxidation with pyridinium dichromate in DMF

HFAs were oxidized with pyridinium dichromate (PDC) in *N,N*-dimethylformamide (DMF) (22). The solution of FAMES in chloroform:methanol, 2:3, by volume (100  $\mu\text{l}$ ) was shaken with 5 mg of PDC and 50  $\mu\text{l}$  of DMF for 24 h at laboratory temperature. The reaction mixture was transferred into a small glass column with silica gel, and the reaction products were eluted with chloroform ( $2 \times 200 \mu\text{l}$ ). The sample volume was reduced to 30  $\mu\text{l}$  under a stream of nitrogen.

### GC/EI-MS

The analyses were performed on a 7890N gas chromatograph coupled to a 5975C mass spectrometer, equipped with EI and quadrupole analyzer (Agilent Technologies, Santa Clara, CA). The sample (2  $\mu\text{l}$ ) was injected in the split mode with a split ratio of 5:1. The injector and transfer line temperatures were set to  $350^\circ\text{C}$  and  $340^\circ\text{C}$ , respectively. A DB-5HT fused silica capillary column (15 m  $\times$  250  $\mu\text{m}$ ; a film thickness 0.10  $\mu\text{m}$ ) from Agilent Technologies was used. The carrier gas was helium at a constant



**Fig. 2.** A section of GC/MS chromatogram reconstructed for  $m/z$  75 showing TMS derivatives of HFAMEs from Chl- $\omega$ OAHFAs (A). The EI-MS (70 eV) spectrum of TMS derivative of HFAME 30:0 ( $t_r = 31.9$  min) (B). The EI-MS (70 eV) spectrum of TMS derivative of HFAME 30:1 ( $t_r = 32.3$  min) (C).



flow rate of 1.5 ml/min. The temperature program was set as follows: 100°C (2 min), then 6°C/min to 370°C (3 min). The ion source and quadrupole temperatures were 230°C and 150°C, respectively. EI spectra (70 eV) were recorded from  $m/z$  20 to 700.

### Direct infusion MS and HPLC/APCI-MS

Direct infusion and HPLC/MS experiments were performed using an LTQ Orbitrap XL hybrid FT mass spectrometer equipped with an Ion Max source and controlled by Xcalibur (all Thermo Fisher Scientific). The mass spectrometer was coupled to a HPLC system consisting of a Rheos 2200 quaternary gradient pump (Flux Instruments, Reinach, Switzerland), a PAL HTS autosampler (CTC Analytics, Zwingen, Switzerland), and a DeltaChrom CTC 100 column oven (Watrex, Prague, Czech Republic). Molecular species of Chl- $\omega$ OAHFAs were separated in nonaqueous reversed-phase HPLC and detected by APCI-MS<sup>2</sup> using the conditions as follows: The temperature of the sample tray was set at 10°C. The autosampler injected 5  $\mu$ l of the sample and the injection system was washed with chloroform/acetonitrile (1:1, by volume). Two Nova-Pak C18 stainless-steel columns connected in series (150 and 300 mm  $\times$  3.9 mm, particle size 4  $\mu$ m; Waters, Milford, MA) were placed in a column oven set at 40°C. The mobile phase was prepared from acetonitrile (A) and ethyl acetate (B) using the following linear gradient program: 0 min, 30% of A and 70% of B; 74.5 min, 67.25% of B; 75 min, 67.5% of B; 140 min, 100% of B; 140.5 min, 100% of B; 154 min, 100% of B. The mobile phase flow rate was set as follows: 0–74.5 min, 0.6 ml/min; 75–140 min, 0.15 ml/min; and 140.5–154 min, 0.6 ml/min. The APCI corona discharge current, vaporizer, and heated capillary temperatures were 5  $\mu$ A, 500°C, and 170°C, respectively. The sheath and auxiliary gas (nitrogen) were set at the flow rate of 52 and 22 arbitrary units, respectively. The MS method encompassed three scan events: 1) the Orbitrap full MS scan event in the  $m/z$  300–1,600 range, in the positive ion mode at a resolution of 30,000; 2) the linear ion trap full MS scan event in the  $m/z$  200–1,600 range in the negative ion mode; and 3) the collision-induced dissociation (CID) MS<sup>2</sup> scan event of the first most intense ion from the parent mass list with a normalized collision energy of 21.5% and isolation window 2 Da in the negative ion mode. The CID MS<sup>2</sup> parent mass list was calculated for  $[M - H - \text{Chl} + \text{H}_2\text{O}]^-$  (i.e.,  $[\text{OAHF}]^-$ ) ions of all possible Chl- $\omega$ OAHFAs with the total number of carbons and double bonds in the range of 37–60 and 0–3, respectively. The HPLC/MS<sup>2</sup> data were interpreted manually, with the help of an in-house developed Excel macro.

### Shorthand nomenclature

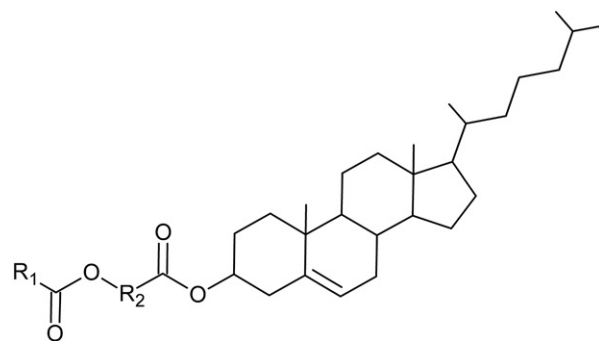
Considering previous literature (23, 24), sterol esters of  $\omega$ -(*O*-acyl)-hydroxy FAs are abbreviated St- $\omega$ OAHFAs. In the case of

cholesterol, Chl- $\omega$ OAHFAs is used. Molecular species within this class are abbreviated using a FA/ $\omega$ HFA-Chl format, which is consistent with the established shorthand nomenclature for  $\omega$ OAHFAs (25) and reflects the chemical structure of Chl- $\omega$ OAHFAs. Thus, for instance, a molecular species identified in this work as 16:0/32:1-Chl corresponds to cholesterol ester of  $\omega$ HFA with 32 carbons and 1 double bond, with the hydroxyl esterified to a saturated FA with 16 carbons. In the ion description, “Chl” is considered a neutral cholesterol molecule. Thus, for instance, protonated  $\omega$ OAHFAs, which is an ion resulting by neutral loss of dehydrated cholesterol from Chl- $\omega$ OAHFAs (M), is  $[M - \text{Chl} + \text{H}_2\text{O}]^+$ .

## RESULTS

### General structure elucidation

As shown in our previous report (18), the unknown lipids eluted in the reversed-phase HPLC at higher retention times than aliphatic 1,2-DDEs. For protonated molecules, high-resolution APCI mass spectra revealed elemental compositions,  $\text{C}_n\text{H}_{2n-x}\text{O}_4$ , where  $x = 11, 13, 15,$  or  $17$ . The existence of four oxygen atoms and similar retention on silica gel as 1,2-DDE pointed out diester lipids. The formulas corresponded to ring plus double bond equivalents (RDBEs) of 6.5–9.5, showing an unusually high degree of unsaturation and/or presence of rings (the RDBE values for 1,2-DDEs ranged from 1.5 to 3.5). Protonated molecules of unknown lipids were found at higher  $m/z$  values ( $m/z$  1,000–1,250) than 1,2-DDE ( $m/z$  800–1,000). The APCI spectra of all molecular species showed neutral loss of 368 Da, typical for sterol-containing compounds. The occurrence of a sterol corresponded well with the high RDBE values. The presence of a sterol in the unknown diesters and its absence in 1,2-DDE made it possible to separate these two diester lipid classes from each other using TLC on magnesium-based sorbents (26, 27). Commercial production of such sorbents was mostly discontinued, which forced us to use common reagent-grade chemicals. Nevertheless, a fraction of unknown sterol-containing diesters with satisfactory purity was obtained. In the next step, the unknown lipids were transesterified and analyzed by high-temperature GC/EI-MS and high-resolution MS. The GC/EI-MS data showed cholesterol and a rich mixture of FAMES (supplemental Fig. S4). A group of FAMES at unusually high retention times was identified as methyl esters of HFAs (HFAMES). Their TMS derivatives (Fig. 2) provided spectra consistent with a hydroxy group at the terminal carbon (28, 29) and revealed saturated and monounsaturated  $\omega$ HFAs with 29–34 carbons. Interestingly, saturated  $\omega$ HFAs eluted in two chromatographically separated peaks, obviously differing by chain branching. The mass spectra of earlier eluting isomers of 30:0 and 32:0 showed significantly more abundant  $m/z$  103, which could indicate methyl branching in the iso position. The chromatographic peaks of TMS derivatives of monounsaturated  $\omega$ HFAMES were distorted, likely because of double bond positional isomers. The original transesterified mixture was also analyzed by high-resolution APCI-MS

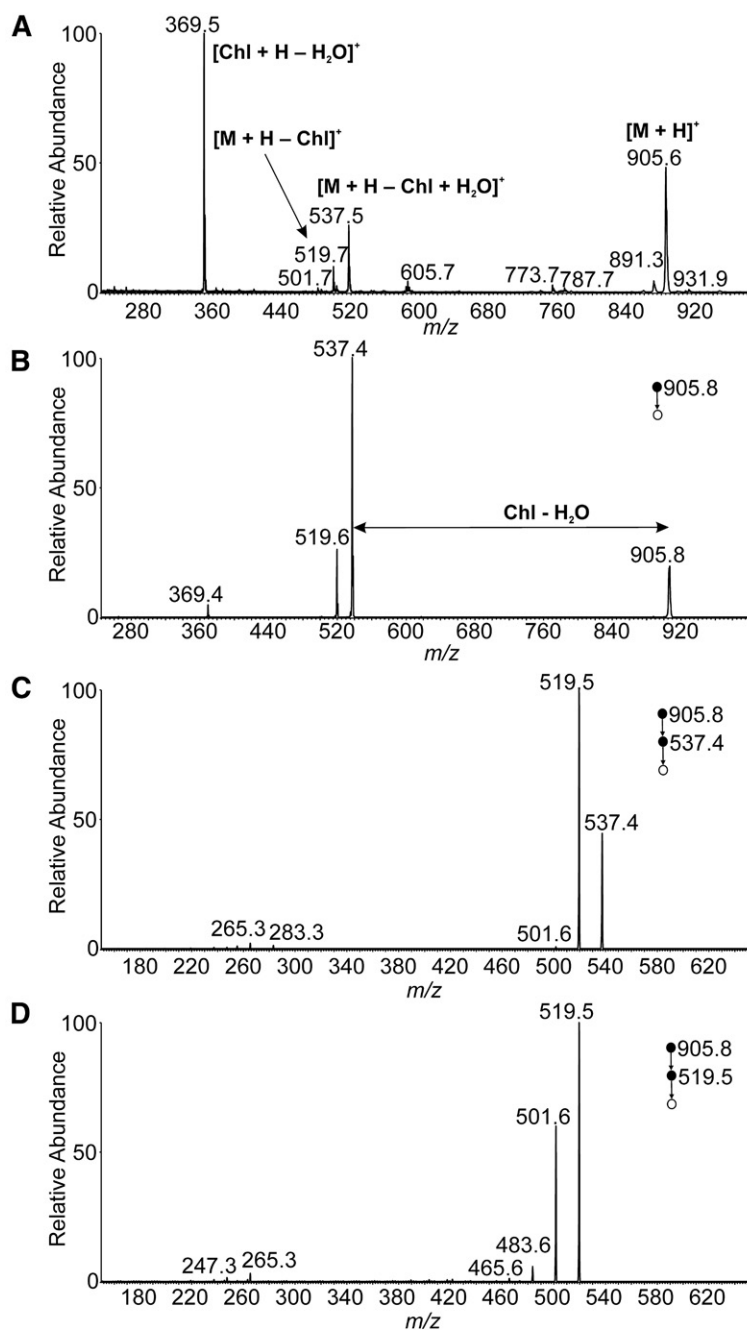


**Fig. 3.** The general structure of Chl- $\omega$ OAHFAs ( $R_1$ , aliphatic chain of FA;  $R_2$ , aliphatic chain of HFA).

(supplemental Fig. S5a). The positive ion spectra displayed abundant  $m/z$  369.3510 (cholesterol;  $[\text{Chl} + \text{H} - \text{H}_2\text{O}]^+$ ) and showed ions consistent with protonated molecules of saturated and monounsaturated hydroxy FAMES (exact masses within 2 ppm error). To confirm the position of the hydroxy group in an independent experiment, oxidation with PDC in DMF was performed. As known from the literature (22), the oxidation of primary alcohol gives carboxyl, whereas the reaction of secondary alcohol provides a carbonyl group. High-resolution electrospray mass spectrum of negative ions showed reaction products with one extra oxygen, i.e., carboxyl derivatives (supplemental Fig. S5b). Therefore, the hydroxy group on the terminal carbons of HFAs was confirmed. Double bond position in unsaturated  $\omega$ HFAs was investigated

using our previously developed method based on gas phase reactions of acetonitrile (30–32). CID MS/MS spectra of the  $[\text{M} + 55]^+$  adduct of  $\omega$ HFAME (32:1) revealed double bonds in two positions,  $n-7$  (more abundant) and  $n-9$  (less abundant), see supplemental Fig. S5c. The existence of two isomers corresponded well with the distorted peaks in GC chromatogram discussed above. The entire analytical strategy used for the lipid class identification is summarized on a flowchart in supplemental Fig. S6.

All the results directed us to a hypothesis of diesters composed of  $\omega$ HFAs having cholesterol attached to the carboxyl and common FA to the hydroxyl. The unknown diester lipids in vernix caseosa were identified as Chl- $\omega$ OAHFAs with the general structure shown in Fig. 3.



**Fig. 4.** APCI mass spectra of 18:1( $n-9$ )/16:0-Chl in the positive ion mode. Full scan spectrum (A). CID MS<sup>2</sup> spectrum of the protonated molecule ( $m/z$  905.8; normalized collision energy 15%) (B). CID MS<sup>3</sup> spectrum of  $[\text{M} + \text{H} - \text{Chl} + \text{H}_2\text{O}]^+$  ( $m/z$  537.4; normalized collision energy 16%) (C). CID MS<sup>3</sup> spectrum of  $[\text{M} + \text{H} - \text{Chl}]^+$  ( $m/z$  519.5; normalized collision energy 16%) (D). The chloroform solution (1 mg/ml) delivered by a syringe pump (5  $\mu\text{l}/\text{min}$ ) was mixed with acetonitrile:ethyl acetate (45:55, by volume) flowing at 150  $\mu\text{l}/\text{min}$ . The mixture was directly infused into the ion source.

## Optimization of HPLC/APCI-MS

The mass spectra of a synthetic standard were studied with the aim to develop an HPLC/MS method for comprehensive characterization of Chl- $\omega$ OAHFA molecular species. The APCI spectrum of 18:1(*n*-9)/16:0-Chl in the positive ion mode (Fig. 4A) showed protonated molecule ( $m/z$  905.6), protonated and dehydrated cholesterol [Chl + H - H<sub>2</sub>O]<sup>+</sup> ( $m/z$  369.5), and fragments consistent with a neutral loss of dehydrated cholesterol ( $m/z$  537.5) and cholesterol ( $m/z$  519.7). The CID MS/MS spectrum of protonated molecule (Fig. 4B) resembled the full scan spectrum, providing identical fragments. Structural information on the FAs and/or HFAs was searched in the MS<sup>3</sup> spectra of [M + H - Chl + H<sub>2</sub>O]<sup>+</sup> (Fig. 4C) and [M + H - Chl]<sup>+</sup> (Fig. 4D). Unluckily, the fragmentation channels proceeded almost exclusively via elimination of water. Thus, positive ion mode turned out to be useless for structure elucidation of Chl- $\omega$ OAHFAs. The standard did not provide [M - H]<sup>-</sup> in the negative ion mode because of the absence of groups prone to deprotonation. Fortunately, elevated temperature in the ion source induced thermal degradation to OAHFA that easily deprotonated in the corona discharge (Fig. 5A). The process was not very efficient under the conditions used and required optimization of the ion source parameters. The ion source temperature and the mobile phase flow rate, as well as ion optics voltages, were tuned to maximize signal of deprotonated OAHFA (supplemental data, Section 6). Fragmentation of deprotonated  $\omega$ OAHFAs is known to give structural information on FAs and HFAs; the MS/MS spectrum of 18:1(*n*-9)/16:0 (Fig. 5B) showed the same fragments as in previously published work (33).

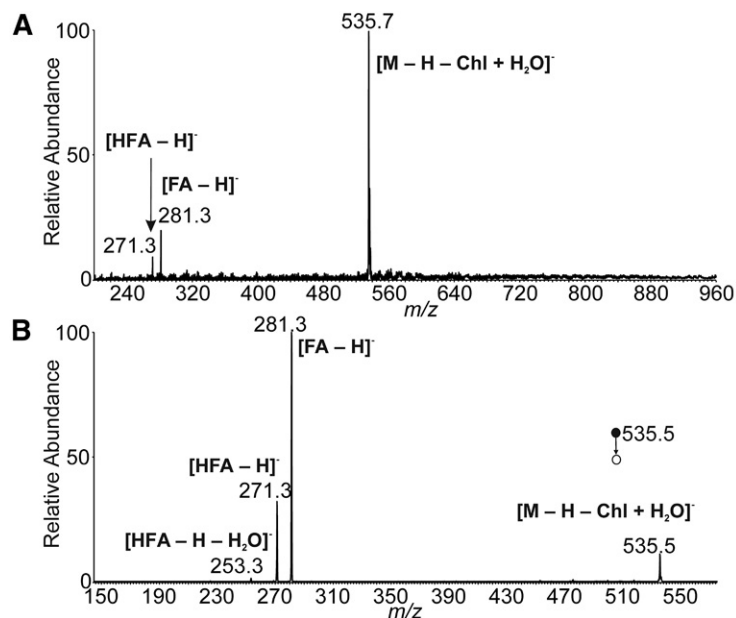
The chromatography was optimized using the lipid sample from vernix caseosa, with the aim to achieve the highest possible resolution for molecular species in a reasonable time. Based on our previous experience with 1,2-DDEs (18), a nonaqueous reversed-phase system with the two Nova-Pak C<sub>18</sub> columns connected in series with a total

length of 45 cm was developed. Various binary mobile phases containing methanol, propan-2-ol, acetonitrile, ethyl acetate, and acetone were studied. Finally, we ended up with separation conditions similar to those used for 1,2-DDE (18), i.e., a linear increase of ethyl acetate in acetonitrile in 140 min. The flow rate of the mobile phase in the elution window of Chl- $\omega$ OAHFAs was reduced from 0.6 to 0.15 ml/min to increase detection sensitivity.

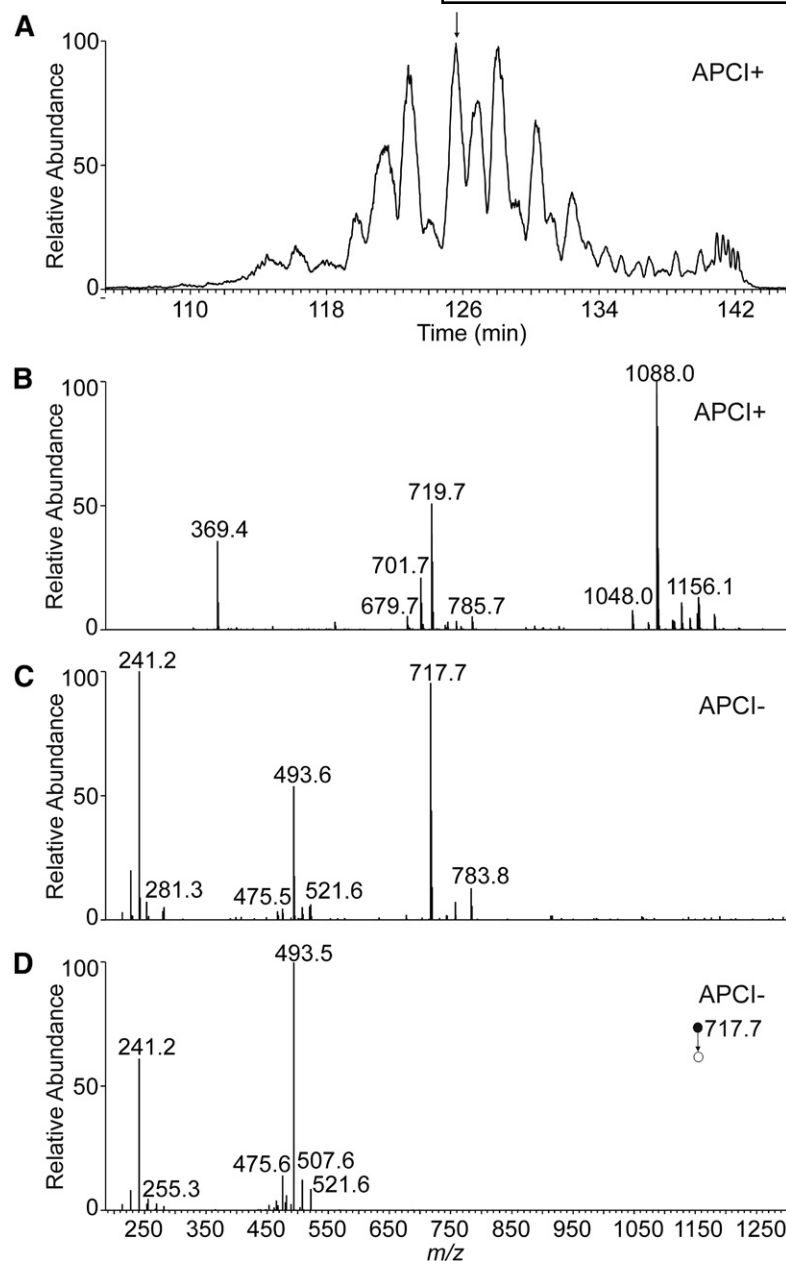
## Chl- $\omega$ OAHFAs in vernix caseosa

The base peak chromatogram of Chl- $\omega$ OAHFAs showed many overlapping peaks (Fig. 6A). The extracted chromatograms displayed up to several peaks for each  $m/z$  value, obviously representing isomers. The elution order followed the equivalent carbon number concept (34); retention increased with the length of the chain and decreased with the number of double bonds. As shown in Fig. 7, the molecular species appeared on a band rising almost linearly with the retention time (at higher retention times the curve rose exponentially as the flow rate increased from 150  $\mu$ l/min back to 600  $\mu$ l/min).

The molecular species of Chl- $\omega$ OAHFAs were detected and identified using APCI-tandem MS with data-dependent scanning. The high-resolution full-scan mass spectra of positively charged ions (Fig. 6B) were used for determining the total number of carbons and double bonds in fatty chains and served for confirmation of the expected elemental composition. The typical mass errors were in the range of 0.5–1.5 ppm. The ion-trap full-scan mass spectra in the negative ion mode (showing deprotonated  $\omega$ OAHFAs, Fig. 6C) further confirmed the total number of carbons and double bonds in fatty chains and served for data-dependent selection of precursors for MS/MS. The ion-trap MS<sup>2</sup> spectra in the negative ion mode (Fig. 6D) revealed FAs and  $\omega$ HFAs. The deprotonated  $\omega$ HFA ions ([HFA - H]<sup>-</sup>) were always accompanied by less abundant water loss peaks ([HFA - H - H<sub>2</sub>O]<sup>-</sup>), which made it possible to distinguish them from deprotonated FA ions.



**Fig. 5.** APCI mass spectra of 18:1(*n*-9)/16:0-Chl in the negative ion mode. Full scan spectrum (A). CID MS<sup>2</sup> spectrum of [M - H - Chl + H<sub>2</sub>O]<sup>-</sup> ( $m/z$  535.5; normalized collision energy 15%) (B). The chloroform solution (1 mg/ml) delivered by a syringe pump (5  $\mu$ l/min) was mixed with acetonitrile:ethyl acetate (45:55, by volume) flowing at 150  $\mu$ l/min. The mixture was directly infused into the ion source.



**Fig. 6.** Positive ion base-peak chromatogram ( $m/z$  1,000–1,300) of Chl- $\omega$ OAHFAs isolated from vernix caseosa (A) and the APCI mass spectra used for the structure elucidation of a species with  $t_R = 126.0$  min (B–D). Full scan spectrum in the positive ion mode (B). Full scan spectrum in the negative ion mode (C). CID  $MS^2$  spectrum of  $m/z$  717.7 in the negative ion mode (normalized collision energy 21.5%) (D).

In total, 295 molecular species of Chl- $\omega$ OAHFAs were fully characterized in 59 chromatographic peaks, and Chl- $\omega$ OAHFAs in an additional 11 peaks were characterized by the total number of carbons and double bonds. The list of the 50 most abundant species is given in **Table 1** (for all identified Chl- $\omega$ OAHFAs see supplemental Table S1). The relative proportions of identified molecular species were estimated from peak areas integrated in the chromatograms reconstructed for  $[M + H]^+$  and relative intensities of deprotonated FA ions in  $MS^2$  spectra. It is important to note that response factors of lipids depend on the number of double bonds and carbon chain length (35, 36). As neither standards nor response factors were available for quantification, the relative proportions must be considered merely as an estimate.

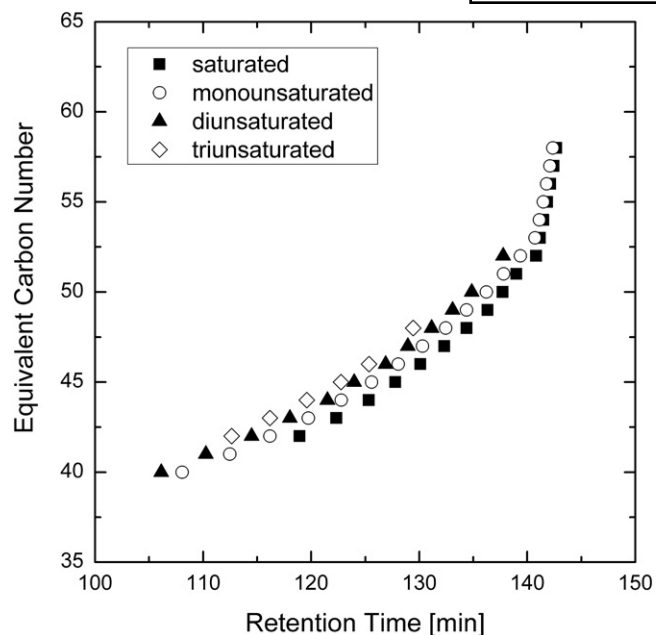
The most abundant molecular species were composed of cholesterol,  $\omega$ HFA 32:1, and FAs commonly found in skin

lipids, namely FA 15:0, FA 14:0, FA 18:1, FA 16:1, and FA 16:0 (15:0/32:1-Chl, 14:0/32:1-Chl, 18:1/32:1-Chl, 16:1/32:1-Chl, and 16:0/32:1-Chl). Relative proportions of FAs and HFAs in Chl- $\omega$ OAHFAs calculated from HPLC/APCI- $MS^2$  are shown in **Fig. 8**. Chl- $\omega$ OAHFAs detected in vernix caseosa appeared to contain 34 FAs with 12–28 carbons (up to two double bonds) and 32  $\omega$ HFAs with 24–38 carbons (up to three double bonds). All saturated Chl- $\omega$ OAHFAs corresponded to 11% of the total integrated signal, whereas monounsaturated, diunsaturated, and triunsaturated species accounted for 55, 30, and 4%, respectively.

## DISCUSSION

Indirect evidence of cholesterol-containing diesters in vernix caseosa appeared in 1969 when a subclass of type I





**Fig. 7.** Plot of calculated equivalent carbon number (ECN) values versus retention times for the Chl- $\omega$ OAHFAs identified in vernix caseosa (ECN = CN-2DB where CN is the total number of carbons and DB is the total number of double bonds in the  $\omega$ OAHFA part of the molecules).

diesters [also called “type III diester waxes” (11)], composed of  $\alpha$ HFAs esterified on the hydroxyl group with an unsubstituted FA and on the carboxyl group with a sterol, was proposed (9). However, a diester lipid class with a sterol moiety has never been substantiated in vernix caseosa. Our work brings evidence that the sterol-containing diesters are, in fact, Chl- $\omega$ OAHFAs comprised of long-chain  $\omega$ HFAs.

$\omega$ HFAs form 2–3% of free (unesterified) FAs in vernix caseosa (5). Although they have been identified in hydrolysates of sterol esters, wax esters, diol diesters, and triacylglycerols (5), intact neutral lipids containing  $\omega$ HFAs have not yet been reported. The detection of  $\omega$ HFAs in the diol diester fraction (5) was very likely because of Chl- $\omega$ OAHFAs co-isolated with diol diesters. As regards more polar lipids,  $\omega$ HFAs are found in three ceramide subclasses, EOS (Cer 1), EOH (Cer 4), and EOP (Cer 9), which together form about 1/4 of all vernix caseosa ceramides (5, 37, 38). These ceramides are composed of long-chain  $\omega$ HFAs linked by an amide bond with a sphingoid base (sphingosine in EOS, 6-hydroxy-sphingosine in EOH, and dihydro-sphingosine in EOP), with the  $\omega$ -hydroxyl group esterified with a FA. The  $\omega$ HFAs in EOS have long aliphatic chains with 28–32 carbons (37). Vernix caseosa ceramides are derived from the fetal epidermis and they represent the key barrier lipids (38). As in the case of stratum corneum,  $\omega$ HFAs and ceramides with  $\omega$ HFAs are also covalently linked (esterified through the  $\omega$ -hydroxyl group) to the cornified cell envelope (39). Whereas covalently bound  $\omega$ -hydroxyceramides in stratum corneum mostly contain 30:0, 32:1, and 34:1 chains (40, 41),  $\omega$ -hydroxyeicosanoic acid predominates in vernix caseosa; C30-C34  $\omega$ HFAs are present as minor components (5).

Ester lipids containing  $\omega$ HFAs are important constituents of meibum, which is a secretion of holocrine meibomian glands in the eyelids of humans and most animals. Meibum lipids form the outermost layer of the tear film that protects the ocular surface from desiccating and bacterial infection (42). The existence of long-chain  $\omega$ HFAs in meibum is known from the early eighties when they were found in steer and human samples (43, 44). Mostly monounsaturated straight-chain  $\omega$ HFAs with 30–36 carbons formed approximately 10% of the total acids. Soon after, St- $\omega$ OAHFAs ( $\omega$ Type I-St) were isolated and identified in steer and human meibum. It was the first report on these diesters in animal samples (45). In addition to the  $\omega$ HFAs, meibomian St- $\omega$ OAHFAs appeared to contain predominantly FAs with 18:1 and 16:1 chains and cholesterol (60%) or lathosterol (35%). St- $\omega$ OAHFAs formed approximately 5% of steer meibomian lipids. The interest in St-OAHFAs increased again after discovering  $\omega$ OAHFAs in meibum (33). The  $\omega$ OAHFAs are likely either precursors or degradation products of St- $\omega$ OAHFAs, and they are considered to be one of the amphiphilic compounds that separate and stabilize the interfacial layer between the very hydrophobic lipids of meibum and the aqueous layer of the tear film (46). They are linked to dry eye disease and might represent biomarkers of the disease progression (47). The complex mixture of  $\omega$ OAHFAs in human meibum contains molecular species composed mostly of monounsaturated FAs with 18 and 16 carbons and 28:1–34:1  $\omega$ HFAs (46, 48). Recently, 61  $\omega$ OAHFAs with 34–56 carbon atoms and one to seven double bonds in human meibum have been reported (25). The occurrence of St- $\omega$ OAHFAs in meibum was reconfirmed by the analysis of their intact molecules (49). The  $\omega$ OAHFA moieties of St- $\omega$ OAHFAs and  $\omega$ OAHFAs in meibum are closely related (50). Their abundances are also similar, with each lipid class accounting for 3–5% of meibomian lipids (24, 47). St- $\omega$ OAHFAs have also been found in meibum of several animal species, including dogs, mice, and rabbits (23). Very recently,  $\omega$ OAHFAs similar to the meibomian ones have been detected in equine sperm and the essential role of these amphiphilic lipids in sperm function was hypothesized (51). It is worth noting that mammals also biosynthesize isomeric OAHFAs derived from HFAs with nonterminal hydroxyl. These recently discovered endogenous lipids (also known as fatty acyl esters of hydroxy FAs) exhibit anti-inflammatory and anti-diabetic effects (52, 53).

In this work, Chl- $\omega$ OAHFAs comprised approximately 20% of the nonpolar diester fraction. As diesters form 3–9% of the total lipids (6, 7), Chl- $\omega$ OAHFAs are estimated to account for approximately 1–2% of vernix caseosa lipids, which is less than in human meibum (3–5%) (24, 45, 47). Chl- $\omega$ OAHFAs were detected and comprehensively characterized from a pooled sample representing 10 male and 10 female full-term babies. Moreover, some experiments were also performed with samples from individual subjects (data not shown). Chl- $\omega$ OAHFAs were detected in all samples at a similar level, suggesting that these esters are commonly present in vernix caseosa and that they are not related to fetus gender. Our experiments showed that the sterol



TABLE 1. List of 50 abundant Chl- $\omega$ OAHFAs identified in vernix caseosa

$t_R$ (min)	$m/z$ of $[M + H]^+$	$m/z$ of $[\omega\text{OAHFA}]^-$	Relative Peak Area <sup>a</sup> (%)	Identification <sup>b</sup>
112.64	1,098.02	727.66	0.53	16:1/32:2-Chl
114.50	1,072.01	701.65	1.45	16:1/30:1-Chl
114.50	1,072.01	701.65	0.57	14:1/32:1-Chl
116.18	1,045.99	675.63	1.46	14:0/30:1-Chl
118.04	1,086.02	715.66	0.68	16:1/31:1-Chl
118.04	1,086.02	715.66	0.49	15:0/32:2-Chl
119.60	1,126.05	755.69	0.75	18:2/32:1-Chl
119.74	1,060.01	689.65	2.53	<b>15:0/30:1-Chl</b>
119.74	1,060.01	689.65	1.41	13:0/32:1-Chl
119.74	1,060.01	689.65	1.29	14:0/31:1-Chl
121.51	1,100.04	729.68	5.33	<b>16:1/32:1-Chl</b>
121.51	1,100.04	729.68	1.76	<b>18:1/30:1-Chl</b>
121.51	1,100.04	729.68	0.53	14:0/34:2-Chl
122.78	1,074.02	703.66	5.85	<b>14:0/32:1-Chl</b>
122.78	1,074.02	703.66	0.94	16:0/30:1-Chl
122.78	1,074.02	703.66	0.62	15:0/31:1-Chl
122.78	1,074.02	703.66	0.50	16:1/30:0-Chl
123.97	1,114.05	743.69	1.27	17:1/32:1-Chl
123.97	1,114.05	743.69	0.72	16:1/33:1-Chl
123.97	1,114.05	743.69	0.64	15:0/34:2-Chl
123.97	1,114.05	743.69	0.60	18:1/31:1-Chl
124.24	1,074.02	703.66	1.17	14:0/32:1-Chl
125.33	1,048.01	677.65	0.81	14:0/30:0-Chl
125.58	1,088.04	717.68	6.14	<b>15:0/32:1-Chl</b>
125.58	1,088.04	717.68	0.81	14:0/33:1-Chl
126.41	1,088.04	717.68	0.90	15:0/32:1-Chl
126.89	1,128.07	757.71	5.70	<b>18:1/32:1-Chl</b>
126.89	1,128.07	757.71	1.97	<b>16:1/34:1-Chl</b>
127.77	1,062.02	691.66	0.80	15:0/30:0-Chl
128.06	1,102.05	731.69	4.33	<b>16:0/32:1-Chl</b>
128.06	1,102.05	731.69	2.38	<b>14:0/34:1-Chl</b>
128.06	1,102.05	731.69	0.78	15:0/33:1-Chl
128.06	1,102.05	731.69	0.74	18:1/30:0-Chl
128.94	1,102.05	731.69	1.71	<b>16:0/32:1-Chl</b>
128.94	1,102.05	731.69	0.62	18:1/30:0-Chl
128.94	1,102.05	731.69	0.52	14:0/34:1-Chl
128.94	1,142.08	771.72	0.67	18:1/33:1-Chl
128.94	1,142.08	771.72	0.56	17:1/34:1-Chl
130.10	1,076.04	705.68	1.02	14:0/32:0-Chl
130.10	1,076.04	705.68	0.58	16:0/30:0-Chl
130.30	1,116.07	745.71	0.48	18:1/31:0-Chl
130.30	1,116.07	745.71	2.36	<b>15:0/34:1-Chl</b>
130.30	1,116.07	745.71	1.74	<b>17:0/32:1-Chl</b>
131.14	1,156.10	785.74	2.37	<b>18:1/34:1-Chl</b>
132.32	1,090.05	719.69	0.96	15:0/32:0-Chl
132.44	1,130.08	759.72	1.77	<b>16:0/34:1-Chl</b>
132.44	1,130.08	759.72	0.77	18:1/32:0-Chl
133.34	1,130.08	759.72	0.52	16:0/34:1-Chl
134.38	1,144.10	773.74	0.70	17:0/34:1-Chl
134.38	1,104.07	733.71	0.75	16:0/32:0-Chl

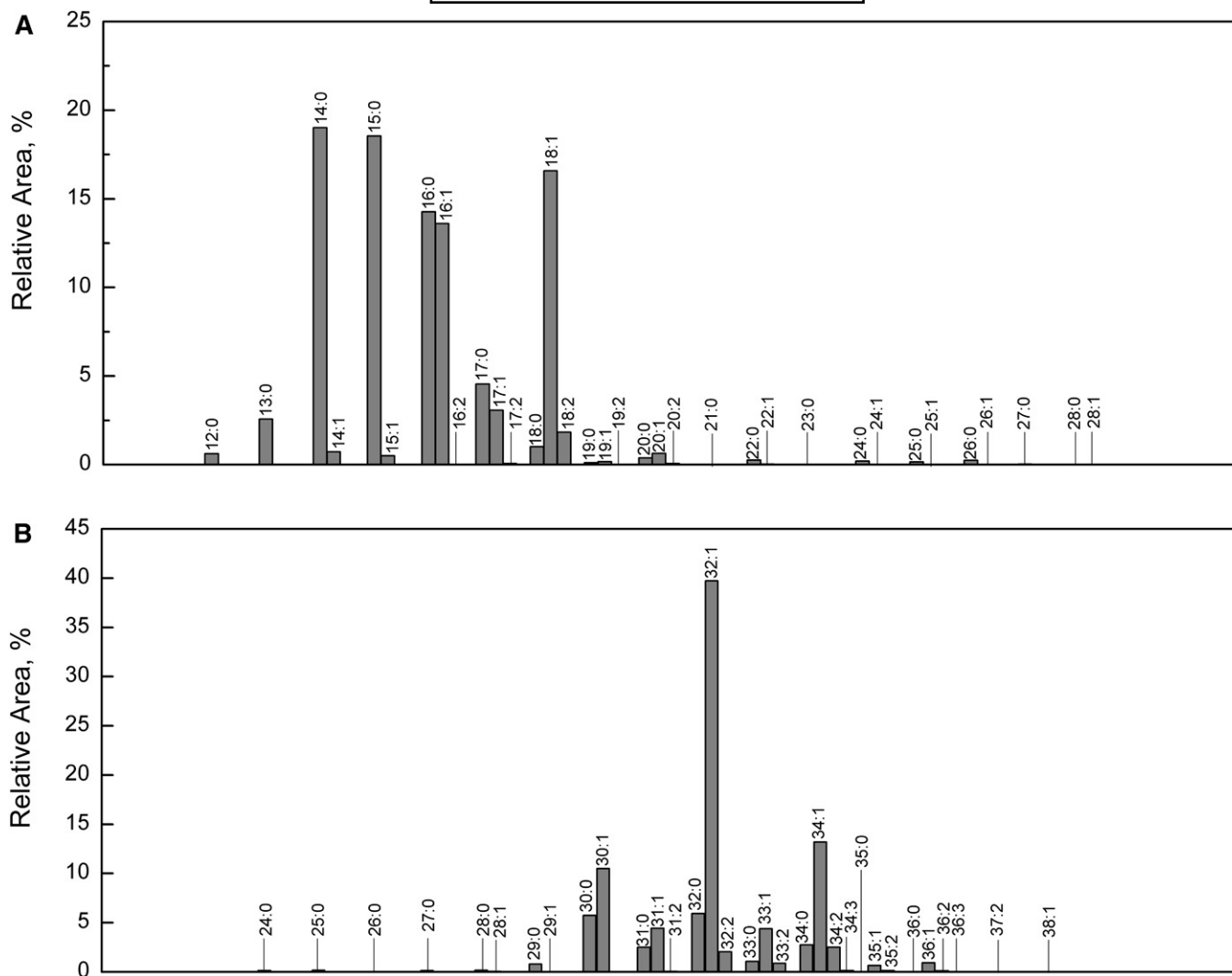
A full list of all 306 Chl- $\omega$ OAHFAs can be found in the supplemental Table S1.

<sup>a</sup>The values of relative peak area were calculated in two steps. In the first step, full scan trace in the positive ion mode was used to create reconstructed ion chromatograms for individual  $m/z$  values ( $[M + H]^+$ ). The peak areas were integrated and put in relative values. In the second step, relative proportions of the isomers within each peak were established from relative intensities of the deprotonated FA ions in the negative ion MS/MS spectra.

<sup>b</sup>Molecular species with the relative peak area higher than 1.5% are bolded.

moiety in Chl- $\omega$ OAHFAs is represented by cholesterol; la-thosterol, reported in addition to cholesterol in meibomian St- $\omega$ OAHFAs (45), has been detected in this work only in traces.  $\omega$ HFA in Chl- $\omega$ OAHFAs from vernix caseosa resembled those existing in meibomian St- $\omega$ OAHFAs and  $\omega$ OAHFAs. The most abundant  $\omega$ HFA in vernix caseosa (32:1, 34:1, and 30:1) were also the most abundant in St- $\omega$ OAHFAs (44, 45) and  $\omega$ OAHFAs (25) from meibum. High proportions of these  $\omega$ HFA have also been reported for equine sperm (51). Saturated  $\omega$ HFA of meibum are both straight chain and methyl branched (44), which is

also in agreement with our findings. The level of similarity can be illustrated by *i*-30:0 and *i*-32:0  $\omega$ HFA identified in vernix caseosa based on the mass spectra of their TMS derivatives (Fig. 2). The earlier work showed that meibomian even-carbon  $\omega$ HFA are *iso*-methyl branched, whereas odd-carbon  $\omega$ HFA are predominantly of the *anteiso* type (44). As regards double bond position in unsaturated  $\omega$ HFA in meibum, the available data are scarce. The action of  $\Delta$ 9-desaturases was proposed (44) and *n*-9 and *n*-7 double bonds were assumed (46). We clearly demonstrated that the double bonds in 32:1  $\omega$ HFA in vernix caseosa are in




**Fig. 8.** Histogram showing relative proportions of FAs (A) and  $\omega$ HFAs (B) in Chl- $\omega$ OAHFAs from vernix caseosa. The proportions were calculated using HPLC/MS data given in supplemental Table S1 (the extended version of Table 1).

both positions, *n*-7 (more abundant) and *n*-9 (less abundant). As regards FAs, the largest signals provided Chl- $\omega$ OAHFAs with 14:0, 15:0, 16:0, 16:1, and 18:1 chains, which is not surprising for a sample of vernix caseosa origin (54, 55). The distribution of FAs in Chl- $\omega$ OAHFAs (Fig. 8A) was similar to the overall representation of FAs in the vernix caseosa lipidome reported earlier (56) (compare supplemental Fig. S8). The higher levels of 16:1 and 18:1 observed for Chl- $\omega$ OAHFAs has to be interpreted with caution because, as discussed above, APCI-MS is somewhat more sensitive to unsaturated lipids (35, 36). Nevertheless, the results indicate that Chl- $\omega$ OAHFAs are biosynthesized from a pool of all FAs, likely with a certain preference for 16:1 and 18:1 chains. The meibomian St- $\omega$ OAHFAs and  $\omega$ OAHFAs tend to incorporate fatty acyls with 16 and 18 carbons more exclusively (24, 25, 45, 47). Meibomian  $\omega$ OAHFAs have been reported to comprise a relatively large proportion of FA 18:2 (24), which is not the case of vernix caseosa Chl- $\omega$ OAHFAs. Nevertheless, 18:2 was the most abundant diunsaturated FA in Chl- $\omega$ OAHFAs from vernix caseosa. Considering the intact molecules, this study

showed the ratio of unsaturated and saturated Chl- $\omega$ OAHFAs as 89%:10%, which is in a good agreement with meibum ( $91 \pm 3\%:9 \pm 1\%$ ) (57). The main species in vernix caseosa are represented by 15:0/32:1-Chl, 14:0/32:1-Chl, 18:1/32:1-Chl, 16:1/32:1-Chl, and 16:0/32:1-Chl, which reflects the higher incorporation of saturated shorter chain FAs than in meibum.

The biological role of Chl- $\omega$ OAHFAs in vernix caseosa remains to be clarified. The diesters are structurally related to cholesteryl esters, important constituents of neutral lipids in vernix caseosa (6). Chl- $\omega$ OAHFAs might be a storage or inactivated form of  $\omega$ OAHFAs, similarly to cholesteryl esters of very long chain FAs and  $\omega$ OAHFAs in meibum (46, 58). However, it is not known whether nonesterified  $\omega$ OAHFAs exist in vernix caseosa (or eventually in amniotic fluid) or not. One can imagine that the amphiphilic properties of  $\omega$ OAHFAs might be important for proper functions and/or cohesion of vernix caseosa. It seems reasonable to assume that Chl- $\omega$ OAHFAs are products of the sebaceous gland, taking into account that they are produced by a meibomian gland (a type of sebaceous gland).

Sebaceous glands in the developing skin are a presumed source of structurally related cholesteryl esters in vernix caseosa (2). On the other hand, OAHFA structural motives in certain ceramides (5, 37, 38) cannot be overlooked, and the epidermal origin of Chl- $\omega$ OAHFAs should be considered as well.

We finally wish to briefly comment on the analytical methodology used in this work. Although positive ion mode of APCI-MS provided intense signals of protonated Chl- $\omega$ OAHFAs, it has turned out to be unusable for detailed structural elucidation because of virtually missing fragments related to FAs and HFAs. Negative ion mode is unlikely to be efficient for diester-type structures lacking easily deprotonatable groups. Although deprotonated molecules of St-OAHFAs have been reported (49), we found the negative ion mode to be insensitive. On-line thermal degradation of Chl- $\omega$ OAHFAs to  $\omega$ OAHFAs in the APCI source worked well and made it possible to detect about 300 molecular species of Chl- $\omega$ OAHFAs, by far the highest number of reported Chl- $\omega$ OAHFAs to date. 

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# Analysis of (*O*-acyl) alpha- and omega-hydroxy fatty acids in vernix caseosa by high-performance liquid chromatography-Orbitrap mass spectrometry

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Received: 2 July 2019 / Revised: 6 December 2019 / Accepted: 11 December 2019  
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## Abstract

Fatty acid esters of long-chain hydroxy fatty acids or (*O*-acyl)-hydroxy fatty acids (OAHFAs) were identified for the first time in vernix caseosa and characterized using chromatography and mass spectrometry. OAHFAs were isolated from the total lipid extract by a two-step semipreparative TLC. The general structure of OAHFAs was established using high-resolution and tandem mass spectrometry of intact lipids and their transesterification and derivatization products. Two isomeric lipid classes were identified: *O*-acyl esters of  $\omega$ -hydroxy fatty acids ( $\omega$ OAHFA) and *O*-acyl esters of  $\alpha$ -hydroxy fatty acids ( $\alpha$ OAHFAs). To the best of our knowledge,  $\alpha$ OAHFAs have never been detected in any biological sample before. Chromatographic separation and identification of OAHFAs species were achieved using non-aqueous reversed-phase HPLC coupled to electrospray ionization hybrid linear ion trap-Orbitrap mass spectrometry. The lipid species were detected as deprotonated molecules, and their structures were elucidated using data-dependent fragmentation in the negative ion mode. More than 400 OAHFAs were identified in this way. The most abundant  $\omega$ OAHFAs species were 28:0/ $\omega$ -18:2, 29:0/ $\omega$ -18:2, 30:0/ $\omega$ -18:2, 32:0/ $\omega$ -18:2, and 30:0/ $\omega$ -18:3, while  $\alpha$ OAHFAs comprised saturated species 21:0/ $\alpha$ -24:0, 22:0/ $\alpha$ -24:0, 23:0/ $\alpha$ -24:0, 24:0/ $\alpha$ -24:0, and 26:0/ $\alpha$ -24:0. OAHFAs were estimated to account for approximately 0.04% of vernix caseosa lipids.

**Keywords** Skin lipids · Amphiphilic lipids · Lipidomics · Mass spectrometry · Vernix caseosa

## Abbreviations

CID Collision-induced dissociation  
ECN Equivalent carbon number

FAHFA Fatty acid esters of hydroxy fatty acid  
FAME Fatty acid methyl ester  
HFA Hydroxy fatty acid  
HFAME Hydroxy fatty acid methyl ester  
OAHFA (*O*-Acyl)-hydroxy fatty acid  
RDBE Ring and double bond equivalent

Published in the topical collection *Current Progress in Lipidomics* with guest editors Michal Holčápek, Gerhard Liebisch, and Kim Ekroos.

**Electronic supplementary material** The online version of this article (<https://doi.org/10.1007/s00216-019-02348-2>) contains supplementary material, which is available to authorized users.

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## Introduction

Vernix caseosa is a complex biofilm, which provides antimicrobial protection to the fetus [1, 2] and is characterized by skin cleansing, antioxidant, wound healing, and moisturizing properties [3–6]. It is composed of water-containing corneocytes (80%) embedded in the overlying matrix of lipids (10%) and proteins (10%) [7]. Vernix caseosa has long been thought to be uniquely human; however, recently it was discovered coating the back of California sea lion fetus [8].

Chromatography and mass spectrometry are the main methods of studying the lipid composition of vernix caseosa. They have shown that vernix caseosa lipodome is an extremely

complex mixture of various lipids. The first report [9] on the lipid composition of vernix caseosa dates back to 1939 and since that time there have been many studies on its composition [10–14]. Vernix caseosa lipids consist mainly of free lipids where neutral species including sterol esters, wax esters, diacylated diols, and triacylglycerols predominate. More polar components, such as free cholesterol, free fatty acids, and ceramides represent the barrier lipids accounting for approximately 10% of the total free lipids. The bound lipids of vernix caseosa mainly consist of fatty acids (FAs),  $\omega$ -hydroxy fatty acids ( $\omega$ HFAs), and  $\omega$ -hydroxyceramides [15]. When compared to internal tissues, vernix caseosa lipids are characterized by the presence of unusual fatty acyls. The aliphatic chains can be extremely long, methyl-branched, with an odd number of carbon atoms and double bonds in unusual positions [16]. Besides them, hydroxylated fatty acyls are often found with the hydroxyl group mostly located on the terminal ( $\omega$ HFAs) or the second ( $\alpha$ HFAs or 2-HFAs) carbon of the chain [15, 17].

A new class of polar lipids, (*O*-acyl)- $\omega$ -hydroxy fatty acids ( $\omega$ OAHFAs), has been described over the past decade [18–23]. These lipids consist of a long chain HFA (longer than 24 carbons) with the hydroxyl at the terminal ( $\omega$ ) carbon to which a fatty acid is attached via an ester bond. To date,  $\omega$ OAHFAs have been detected in human skin [18], meibum [19–21], equine sperm [22], and amniotic fluid [23], and as structural motifs in more complex lipids [14, 20, 24, 25]. Free  $\omega$ OAHFAs are important amphiphilic lipids in human meibum because they stabilize the tear film and thus prevent drying of the ocular surface [26]. They can serve as biomarkers of the dry eye disease as their concentration decreases with the severity of the disease [27]. The meibomian  $\omega$ OAHFAs are mostly composed of monounsaturated FAs with 18, 16 carbons, and long-chain monounsaturated  $\omega$ HFAs with 28–34 carbons [18–23]. Similar  $\omega$ OAHFA chains constitute Ch1- $\omega$ OAHFAs in meibum and vernix caseosa [14, 19], but intact (free)  $\omega$ OAHFAs have never been described in vernix caseosa. Many of  $\omega$ OAHFAs known from human meibum have also been detected in equine sperm and amniotic fluid. Equine sperm cells contain  $\omega$ OAHFAs in their heads, where they possibly serve a surfactant role important during the process of fertilization [22]. In equine amniotic fluid, they may provide an interface for the absorption of lipophilic nutrients through the fetal skin and aid fetal lung development [23]. Isomeric OAHFAs with the ester bond linkage at various (non-terminal) positions of the hydroxy fatty acid chain (commonly referred to as fatty acid esters of hydroxy fatty acid, FAHFAs) are lately discovered endogenous lipids found in adipose tissue and serum [28]. They are released by adipocytes in white adipose tissue and have been demonstrated to act as signaling molecules associated with insulin secretion and glucose metabolism, which can be potentially used for diabetes 2-type treatment [28, 29]. Very

recently, FAHFAs have been shown to incorporate into triacylglycerols [30].

In this work, we analyzed OAHFAs in vernix caseosa. Due to the high complexity of the sample, it was essential to perform a multi-stage sample pretreatment to obtain a fraction enriched with the desired analytes. Since OAHFAs may have different positions of the ester bond on the hydroxy acid chain, it was necessary to determine what particular type(s) of OAHFAs exist in vernix caseosa. Derivatization reactions and mass spectrometry methods were used for this purpose. Finally, OAHFA molecular species were characterized in detail by high-performance liquid chromatography coupled-electrospray ionization high-resolution tandem mass spectrometry (HPLC/ESI-MS/MS). Using this approach, two OAHFA subclasses were identified in vernix caseosa, namely, (*O*-acyl)- $\omega$ -hydroxy fatty acids ( $\omega$ OAHFAs) and (*O*-acyl)- $\alpha$ -hydroxy fatty acids ( $\alpha$ OAHFAs). To the best of our knowledge,  $\alpha$ OAHFAs represent a newly discovered lipid subclass that has not been detected in any biological sample yet. Both  $\alpha$ OAHFAs and  $\omega$ OAHFAs are newly described lipids for vernix caseosa.

## Materials and methods

### Material

With written, informed parental consent, we collected vernix caseosa (1–2 g) from healthy newborn subjects delivered at full-term (gestation week 38–41) immediately after the delivery. The samples were stored in amber glass vials at  $-80\text{ }^{\circ}\text{C}$  until extraction.

### Chemicals

LC-MS grade acetonitrile, methanol, toluene, and propan-2-ol (Sigma-Aldrich, St. Luis, MO, USA) were used as received. Chloroform and hexane (Penta, Czech Republic) were distilled in glass from analytical-grade solvents. Silica gel Kieselgel 60G (Merck & Co., Kenilworth, NJ, USA), *N,N*-dimethylformamide (Acros Organics, part of Thermo Fisher Scientific, USA), (*Z*)-10-(hexadec-9-enoyloxy)octadecanoic acid (palmitoleic acid-10-hydroxystearic acid abbreviated 10-POHSA; Cayman Chemical, Ann Arbor, MI, USA), 20-hydroxy eicosanoic acid, 22-hydroxy docosanoic acid (Larodan AB, Solna, Sweden), 16-hydroxy hexadecanoic acid, (*Z*)-octadec-9-enoyl chloride, palmitoyl chloride, stearoyl chloride, sodium chloride, hydrochloric acid, sodium bicarbonate, anhydrous magnesium sulfate, *N,O*-bis(trimethylsilyl)acetamide, acetic acid, silver carbonate, acetyl chloride, and trimethylsilyldiazomethane (all from Sigma-Aldrich, St. Luis, MO, USA) were of reagent grade and used as purchased. OAHFA standards 18:1/ $\alpha$ -16:0, 18:1/ $\omega$ -16:0,



18:0/ $\alpha$ -20:0, and 16:0/ $\omega$ -22:0 were synthesized as outlined in Electronic Supplementary Material (ESM).

### Isolation of total lipids

Sixteen samples of vernix caseosa equally representing gender (8 from newborn boys and 8 from newborn girls) were used for lipid extraction with methanol:chloroform (2:1, by vol.). Each sample (300 mg) was processed separately according to our previous work [14]. The chloroform extracts were combined to generate a pooled sample and treated with anhydrous magnesium sulfate to remove residual water. The extract was filtered and evaporated to near dryness using a rotary evaporator. The rest of the solvent was evaporated under a stream of nitrogen. In total, 4.8 g of vernix caseosa yielded 435.2 mg of total lipids. The lipids were reconstituted in chloroform:methanol (19:1, by vol.) at the concentration of 30 mg/ml and stored at  $-20\text{ }^{\circ}\text{C}$ .

### Fractionation of lipids

Approximately 1/4 (ca. 100 mg of lipids) of the total lipid extract was used to isolate the OAHFAs lipid fraction. For this purpose, TLC glass plates (60 mm  $\times$  120 mm) were coated with silica gel (Kieselgel 60G) in our laboratory. In the first step, polar lipids were separated from nonpolar lipids on pre-cleaned TLC plates using hexane:diethyl ether:acetic acid (80:20:1, by vol.) mobile phase. The zones were visualized under ultraviolet light after spraying with primuline (5 mg in 100 ml of acetone:water (80:20, by vol.)). Silica gel corresponding to  $R_F = 0.12\text{--}0.59$  was scraped off the plates and the lipids were extracted with chloroform:methanol (1:1, by vol.) mixture. The TLC was performed repeatedly with approximately 5 mg of lipids separated in each step until the whole 1/4 of the total lipid extract was processed. The polar lipid extracts from the individual TLC runs were combined and the solvent was evaporated under a stream of nitrogen. The yield of polar lipids was 20.6 mg (F-1). In the next step, F-1 was reconstituted in chloroform:methanol (2:1, by vol.) at a concentration of 30 mg/ml and spiked with 32  $\mu\text{l}$  of 10-POHSA (0.5 mg/ml) for later quantification. The sample was applied on TLC plates (60 mm  $\times$  120 mm) coated with silica gel (Kieselgel 60G) and the lipid classes were separated using hexane:diethyl ether:acetic acid (70:30:2, by vol.) mobile phase. After air-drying, the zones were sprayed with primuline and then visualized under ultraviolet light. The standard of 18:1/ $\omega$ -16:0 was used to mark the  $R_F$  of OAHFAs. Silica gel corresponding to  $R_F = 0.47\text{--}0.53$  was scraped off the plates and extracted with chloroform:methanol (1:1, by vol.) mixture. The procedure was used repeatedly to process all F-1 and yielded 1.3 mg of a lipid fraction (F-2) with OAHFAs. The F-2 was dissolved in 300  $\mu\text{l}$  of chloroform:methanol (2:1, by vol.) and stored at  $-20\text{ }^{\circ}\text{C}$  until the HPLC/MS analysis.

The extraction of total lipids, isolation of OAHFA, and subsequent structure elucidation steps are depicted in ESM (Fig. S1).

### Transesterification

The OAHFAs structure elucidation encompassed an acidic transesterification [31] producing a mixture of fatty acid methyl esters (FAMES) and free HFAs (referred to as F-2A). Briefly, lipids (F-2) were dissolved in chloroform:methanol (2:3, by vol.) in a small glass ampoule. After adding acetyl chloride, the ampoule was sealed and placed in a dry bath for 1 h at  $70\text{ }^{\circ}\text{C}$ . The reaction mixture was neutralized by adding silver carbonate. The reaction mixture was separated from solid precipitate (AgCl) and used for GC/MS analysis. For the next experiments, the solvent was evaporated under a stream of nitrogen and the F-2A lipids were re-dissolved in toluene:methanol (3:2, by vol.).

### Methylation with trimethylsilyldiazomethane

Methyl esters of HFAs were prepared according to a published procedure [32]. To a stirred solution of F-2A in 5 ml of toluene:methanol (3:2, by vol.), an etheric solution (2 mol/l) of trimethylsilyldiazomethane (TMSCHN<sub>2</sub>) was added dropwise until the yellow color persisted. The reaction mixture was stirred for 30 min followed by adding approximately three drops of concentrated acetic acid. The excess solvent was evaporated under a stream of nitrogen. The reaction mixture (F-2B) containing FAMES and methyl esters of HFAs (HFAMES) was used for further analysis.

### Trimethylsilylation

Trimethylsilylation of HFAMES was performed according to a procedure reported by Carvahlo [33]. F-2B was dissolved in dried acetonitrile. After adding an excess of *N,O*-bis(trimethylsilyl)acetamide, the reaction mixture was placed in a dry bath for 10 min at  $40\text{ }^{\circ}\text{C}$ . The excess solvent was evaporated under a stream of nitrogen and the sample was injected into the GC column.

### GC/EI-MS

All GC/EI-MS analyses were performed using a 7890N gas chromatograph equipped with a 5975C quadrupole mass spectrometer and a DB-5HT fused silica capillary column (15 m  $\times$  250  $\mu\text{m}$ , film thickness 0.10  $\mu\text{m}$ ), all from Agilent Technologies, Santa Clara, CA, USA. The injector and transfer line temperatures were set to  $350\text{ }^{\circ}\text{C}$  and  $340\text{ }^{\circ}\text{C}$ , respectively. Injections of 3  $\mu\text{l}$  were made using a splitless injection mode. The carrier gas was helium at a constant flow rate of 1.5 ml/min. The oven temperature program started at  $100\text{ }^{\circ}\text{C}$

(held for 2 min) followed by a linear increase to 370 °C at 6 °C/min. The temperatures of the ion source and quadrupole were set at 230 °C and 150 °C, respectively. EI spectra (70 eV) were collected in the  $m/z$  range 50–600.

## HPLC/ESI-MS

HPLC/MS analyses were conducted on a liquid chromatograph consisting of a 10390 Rheos Allegro UHPLC pump (Flux Instruments, Basel-Stadt, Switzerland), an HTS9 PAL autosampler (GenTech Scientific, Inc., Arcade, NY, USA) and a Jetstream 2 Plus column thermostat (ABL&E-JASCO Magyarország, Budapest, Hungary). The HPLC system was coupled to an LTQ Orbitrap XL hybrid FT mass spectrometer with electrospray ionization and controlled by Xcalibur (all Thermo Fisher Scientific, Waltham, MA, USA).

The chromatography of the molecular species of OAHFAs was performed using non-aqueous reversed-phase HPLC and ESI-MS<sup>2</sup> detection with the following conditions: two Nova-Pak C18 stainless-steel columns connected in series of a total length 45 cm (150 and 300 mm × 3.9 mm, particle size 4 μm; Waters, Milford, MA, USA) were used at a column temperature of 45 °C. The sample tray temperature was set at 15 °C, the autosampler injected 10 μl of the sample, and the injector system was washed with chloroform:methanol (1:1, by vol.). The analytes were eluted under isocratic conditions using a mobile phase prepared from propan-2-ol and methanol (1:1, by vol.) with the addition of 0.05% ammonium acetate. The flow rate was maintained at 250 μl/min during the entire separation. The detection was conducted in the negative ion mode with the capillary temperature 300 °C, the source voltage 4.7 kV, capillary voltage – 47.5 V, and the tube lens offset – 57.5 V. Nitrogen served both as the sheath and auxiliary gas at a flow rate of 38 and 9 arbitrary units, respectively. The MS method included two scan events: (1) the Orbitrap full MS scan in the  $m/z$  150–1500 range and (2) the Orbitrap collision-induced dissociation (CID) MS<sup>2</sup> scan of the first most intense ion from the parent mass list with a normalized collision energy of 25% and isolation window 1.5 Da. The parent mass list was calculated for  $[M - H]^-$  ions of all possible OAHFAs with the total number of carbons and double bonds in the range of 30–60 and 0–3, respectively. The exact masses were measured by the Orbitrap at a resolution of 60,000 and the mass errors were within the range of 0.5–1.5 ppm. The HPLC/MS<sup>2</sup> data were interpreted manually. Due to the presence of interfering compounds in the sample, the quantitative analysis of OAHFAs required the addition of an internal standard (10-POHSA). Peak areas for each OAHFA were obtained by integration of the appropriate reconstructed ion chromatograms for  $[M - H]^-$  and quantified by comparison to the reconstituted ion peak of an internal standard. The relative proportions of the individual molecular species with the same  $m/z$  value were estimated from the peak

areas of  $[M - H]^-$  ions and the relative intensities of deprotonated HFA ions in MS<sup>2</sup> spectra.

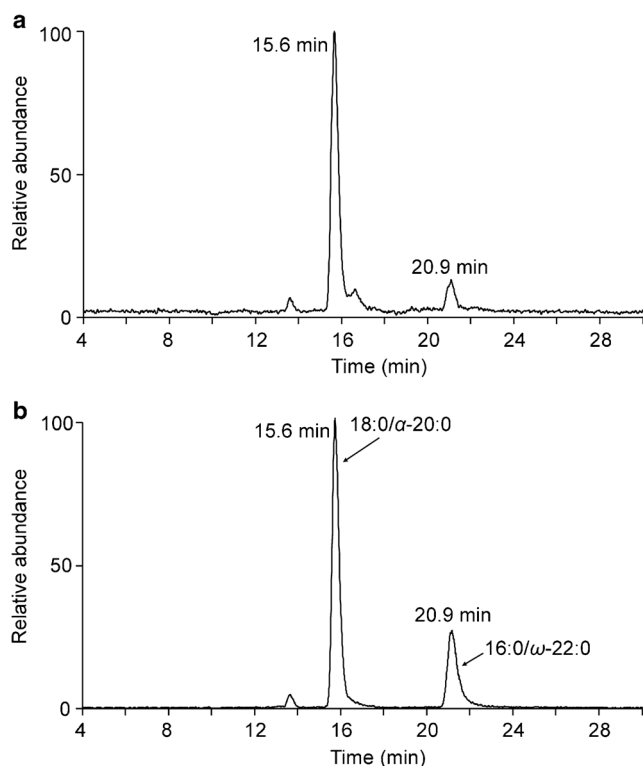
## Shorthand nomenclature

Considering previous literature [34, 35], the molecular species of OAHFAs are described as  $X_1:Y_1/X_2:Y_2$ , where  $X_1$  is the total number of carbons of the FA and  $X_2$  is the total number of carbons of an HFA whereas  $Y_1$  and  $Y_2$  represents the number of double bonds in the FA and HFA respectively. In species, where ester position is known, the  $X_1:Y_1/z-X_2:Y_2$  abbreviation is used;  $z$  represents the position of the ester linkage relative to a carboxylic acid. Thus, for instance, a molecular species identified in this work as 18:1/ $\omega$ -30:0 is an ester of an FA with 18 carbons and one double bond with saturated 30-carbon HFA having the ester linkage on the terminal carbon of the HFA chain.

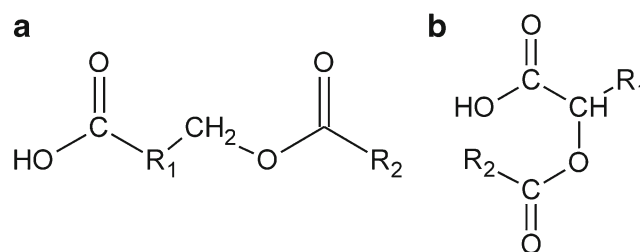
## Results and discussion

The lipidome of vernix caseosa is one of the most complex lipid mixtures, which makes the characterization of its minor components challenging. Multiple separation steps are usually required to prepare the sample. In this work, two-step semipreparative TLC on silica gel was repeatedly used to obtain enough sample for the analysis. In the first step, polar lipids were separated from nonpolar lipids, which exist in large excess in vernix caseosa. The second step was aimed at isolating OAHFAs from a complex mixture of polar lipids. Careful optimization of the HPLC/MS conditions was required for the comprehensive characterizing of OAHFAs in vernix caseosa. Molecular species of OAHFAs were separated on two Nova-Pak C18 columns connected in series. Even though the isolation procedure consisted of several steps, the F-2 appeared to contain also unwanted lipids identified as free FAs. Therefore, the HPLC conditions were optimized with the aim to separate OAHFAs from contaminating lipids and achieve good chromatographic resolution of OAHFA molecular species. Finally, isocratic separation with a mobile phase containing propan-2-ol, methanol and ammonium acetate was used. The mass spectra indicated that OAHFAs eluted in the 28–35 min range. The high-resolution ESI spectra of the species eluting within this range revealed elemental compositions  $C_nH_{2n-x}O_4^-$ , where  $x = 3, 5, 7$  or  $9$ , depending on the degree of unsaturation, and the MS/MS spectra were in agreement with the published data [19]. In order to determine the position of the ester group, OAHFAs were hydrolyzed and derivatized. For these experiments, OAHFAs were isolated from F-2 using HPLC. Lipids from the 28–35 min range were collected, transesterified and analyzed by high-temperature

GC/EI-MS. As expected, the transesterified sample appeared to be a mixture of FAMES and free HFAs. They were identified based on their mass spectra and retention, using NIST Mass Spectral Library and literature data [36]. Then, the reaction mixture was treated with TMSCHN<sub>2</sub> to produce HFAMEs, which were in the next step trimethylsilylated with *N,O*-bis(trimethylsilyl)acetamide. Most of the TMS derivatives provided spectra consistent with the hydroxy group at the terminal carbon [37, 38]. However, besides the  $\omega$ TMS derivatives, the spectra also revealed the presence of  $\alpha$ TMS derivatives (see ESM Fig. S2). Therefore, in addition to  $\omega$ OAHFAs, a new subclass of fatty acid esters of hydroxy fatty acids,  $\alpha$ OAHFAs, was likely present in vernix caseosa. To confirm the hypothesis, standards of OAHFAs were synthesized and their retention times compared to those of naturally occurring species in vernix caseosa. Retention times of synthetic standards 16:0/ $\omega$ -22:1 and 18:0/ $\alpha$ -20:1 were 15.6 min and 20.9 min, respectively. They were in excellent agreement with the retention times of 38:0 OAHFAs observed in the chromatogram of vernix caseosa (Fig. 1). All these results led us to the conclusion that both  $\omega$ OAHFAs and  $\alpha$ OAHFAs are indeed present in vernix caseosa. The general structures of these lipids are shown in Fig. 2.



**Fig. 1** Chromatograms reconstructed for  $m/z$  593.5 showing saturated OAHFA species with 38 carbons in their chains. Sample of OAHFAs isolated from vernix caseosa (a), and the same sample spiked with 2-(stearoyloxy)icosanoic acid (18:0/ $\alpha$ -20:0) and 22-(palmitoyloxy)docosanoic acid (16:0/ $\omega$ -22:0) (b)



**Fig. 2** General structure of  $\omega$ OAHFAs (a) and  $\alpha$ OAHFAs (b)

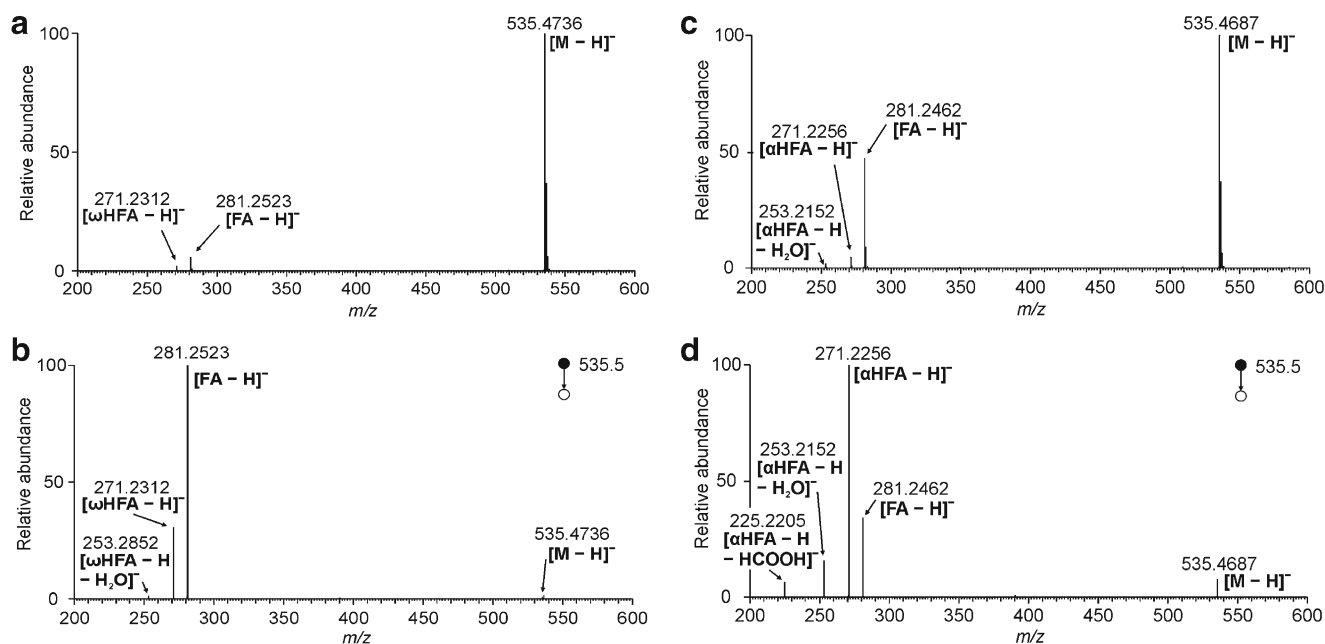
### Mass spectrometry of OAHFA standards

Mass spectra of  $\omega$ OAHFAs have been studied previously and thus their fragmentation is well known [19, 34]. We were interested if one can use MS/MS for distinguishing omega form alpha isomers. For this purpose, mass spectra of synthetic standards were investigated. The first pair of standards was oleic acid esters of hydroxy palmitic acids (Fig. 3). The full-scan Orbitrap ESI spectra of both 18:1/ $\omega$ -16:0 and 18:1/ $\alpha$ -16:0 in the negative ion mode showed a strong signal of deprotonated molecules  $[M - H]^-$  accompanied by fragments consistent with deprotonated FA and deprotonated HFAs. The intensities of these fragments were notably more abundant in alpha isomer, likely because of the proximity of the ester group to the negative charge site. The same fragments, i.e., deprotonated oleic acid ( $[FA - H]^-$ ) and deprotonated hydroxy palmitic acids ( $[HFA - H]^-$ ), were detected also in the CID MS/MS spectra generated from  $[M - H]^-$ . In addition to these fragments, a water loss peak ( $[HFA - H - H_2O]^-$ ) was present in the spectra of both isomers. Whereas  $[FA - H]^-$  was the base peak in the spectrum of 18:1/ $\omega$ -16:0, the most abundant ion in the spectrum of 18:1/ $\alpha$ -16:0 was  $[HFA - H]^-$ . Nevertheless, the main difference between the  $\omega$ - and  $\alpha$ -isomer was a fragment consistent with formic acid loss from deprotonated hydroxy fatty acid ( $[HFA - H - HCOOH]^-$ ); it was present in the MS/MS spectrum of 18:1/ $\alpha$ -16:0, but not in 18:1/ $\omega$ -16:0. Clearly, this fragment made it possible to differentiate between these two isomers. The second pair of synthetic standards were saturated OAHFAs with 38 carbons (16:0/ $\omega$ -22:0 and 18:0/ $\alpha$ -20:0). As expected, their mass spectra provided analogous fragmentation to OAHFA isomers discussed above, including the  $[HFA - H - HCOOH]^-$  ion present only in the  $\alpha$ -isomer (see ESM Fig. S3). The mechanism of HCOOH elimination from deprotonated alpha hydroxy acids is known and involves the carboxylate group, the hydroxyl hydrogen, and one beta hydrogen [39, 40].

### OAHFAs in vernix caseosa

HPLC/ESI-MS/MS was employed to characterize OAHFAs in the vernix caseosa sample. Lower flow rate than the optimum value was used (0.25 ml/min) to slow down the elution, which provided us with extra time for fragmenting as many



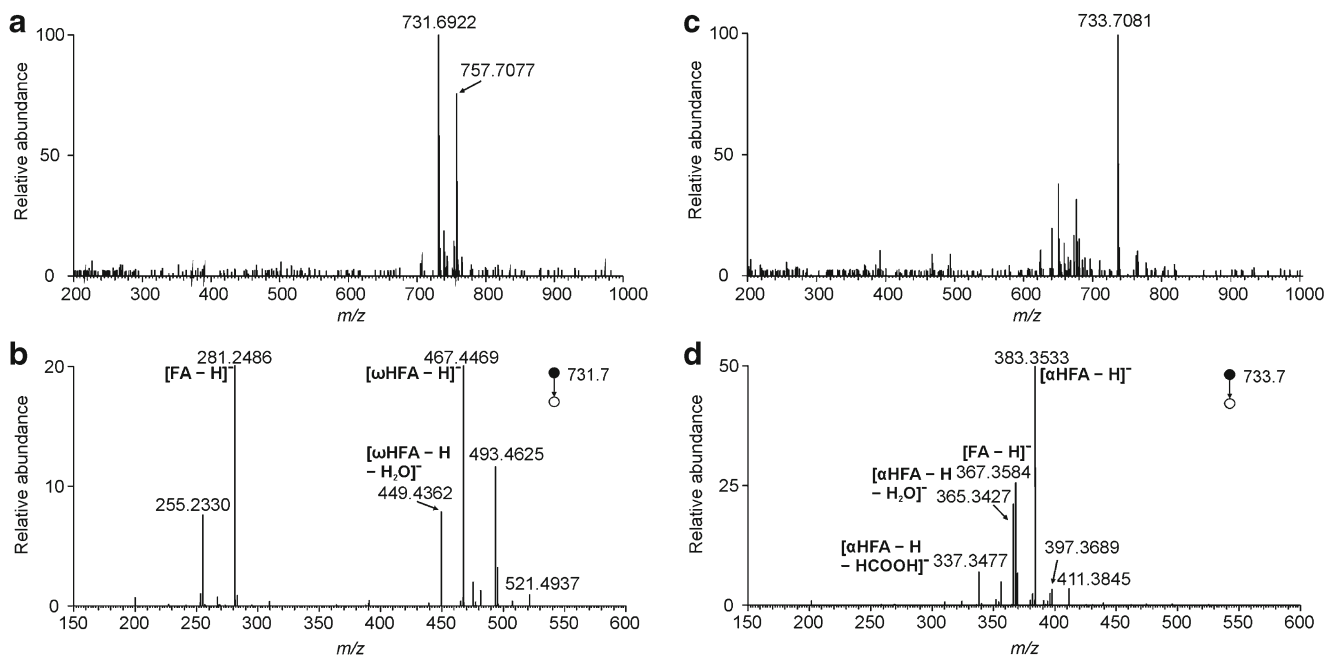


**Fig. 3** ESI mass spectra of 18:1/ $\omega$ -16:0 and 18:1/ $\alpha$ -16:0 standards in the negative ion mode taken from HPLC/MS/MS data. Full-scan spectrum of 18:1/ $\omega$ -16:0 (a), CID MS<sup>2</sup> spectrum of 18:1/ $\omega$ -16:0 [M-H]<sup>-</sup> ( $m/z$

535.5; normalized collision energy 25%) (b), full-scan spectrum of 18:1/ $\alpha$ -16:0 (c), and CID MS<sup>2</sup> spectrum of 18:1/ $\alpha$ -16:0 [M-H]<sup>-</sup> ( $m/z$  535.5; normalized collision energy 25%) (d)

precursors as possible within a chromatographic peak. Representative spectra of selected OAHFAs are shown in Fig. 4. The high-resolution full scan provided exact masses of deprotonated molecules, which served for determining the total number of carbons and double bonds in OAHFAs. Molecular species were characterized in MS<sup>2</sup> spectra using

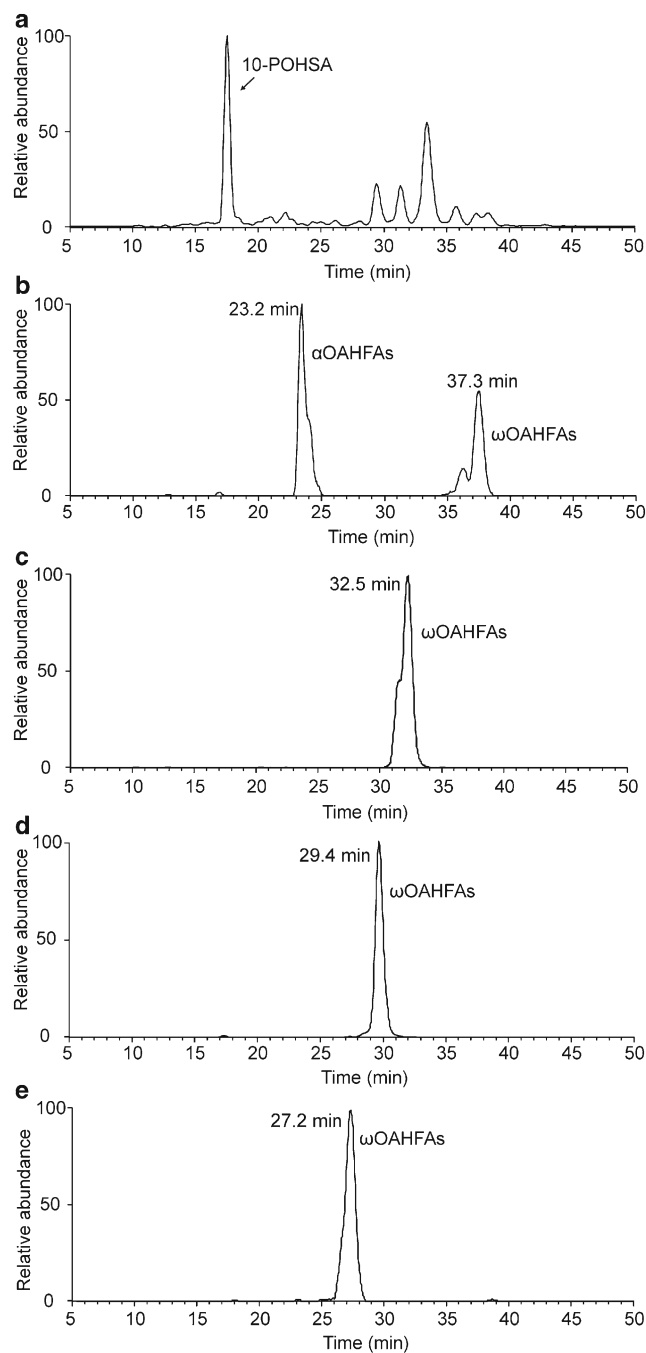
[FA-H]<sup>-</sup> and [HFA-H]<sup>-</sup> fragments, which were easily distinguished from each other based on their exact masses. The position of the ester linkage in OAHFAs (alpha or omega) was deduced from the presence or absence of [HFA-H-HCOOH]<sup>-</sup> ions and from retention data (see below). In the MS<sup>2</sup> spectra of  $\alpha$ OAHFAs, the [HFA-H]<sup>-</sup> ions were



**Fig. 4** Negative ion mode ESI mass spectra of  $\omega$ OAHFAs with  $t_R = 37.1$  min (a, b) and  $\alpha$ OAHFAs with  $t_R = 25.9$  min (c, d) from vernix caseosa. Full-scan spectrum taken in 37.1 min (a) and CID MS<sup>2</sup> spectrum of  $m/z$  731.69 (5x magnified) recorded at the normalized collision energy

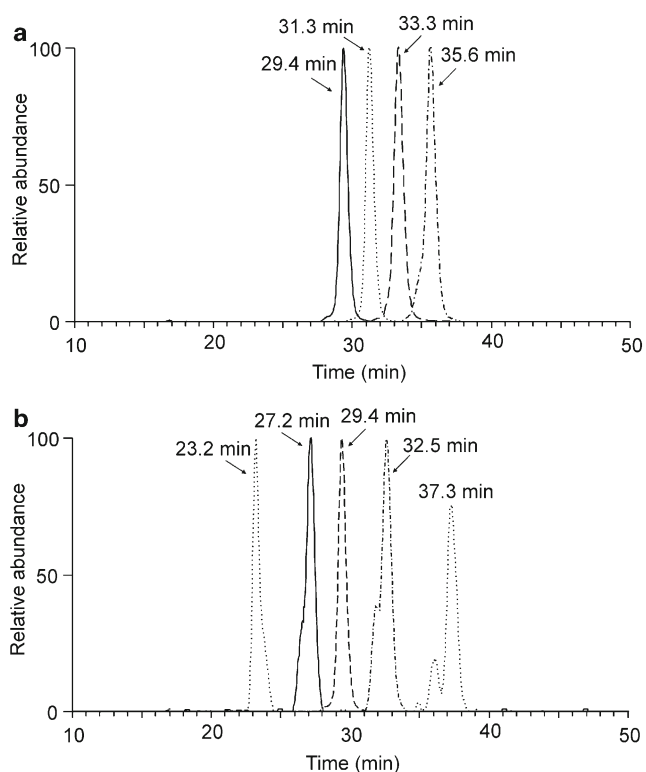
25% (b); the most abundant signals were consistent with 18:1/ $\omega$ -30:0. Full-scan spectrum taken in 25.9 min (c) and CID MS<sup>2</sup> spectrum of  $m/z$  733.70 ( $\times 2$  magnified) recorded at the normalized collision energy 25% (d); the most abundant signals were consistent with 24:0/ $\alpha$ -24:0

significantly more abundant than  $[FA - H]^-$  fragments. In Fig. 4a, two signals at  $m/z$  731.6922 and 757.7077 represent OAHFA 48:1 and OAHFA 50:2, respectively. The CID spectrum of the first peak showed deprotonated HFA 30:0 at  $m/z$  467.4469, its dehydration product at  $m/z$  449.4362, no formic acid loss peak, and FA 18:1 at  $m/z$  281.2486, which indicated OAHFA 18:1/ $\omega$ -30:0. There were also signals of deprotonated HFA 32:1 at  $m/z$  493.4625 and deprotonated FA 16:0 at  $m/z$  255.2330 from OAHFA 16:0/ $\omega$ -32:1. In Fig. 4c,  $m/z$  733.7081 (OAHFA 48:0) was interpreted as OAHFA 24:0/ $\alpha$ -24:0 based on the deprotonated HFA 24:0 at  $m/z$  383.3533, dehydrated deprotonated HFA 24:0 at  $m/z$  365.3427, formic acid loss from deprotonated HFA 24:0 at  $m/z$  337.3477, and deprotonated FA 24:0 at  $m/z$  367.3584. The chromatographic data of OAHFAs were complex and revealed a high number of species, often in co-eluting peaks (Fig. 5). Isobaric OAHFAs were observed in reconstructed chromatograms for  $[M - H]^-$   $m/z$  values. Saturated OAHFAs provided up to four isomeric peaks arranged in two distributions (Fig. 5b). The species with shorter retention times were interpreted as  $\alpha$ OAHFAs because of the presence of  $[HFA - H - HCOOH]^-$  fragments. The peaks at higher retention times corresponded to  $\omega$ OAHFAs. Within both distributions, typically two peaks existed which were considered to be methyl branching variants. This is consistent with the fact that methyl branched species accounted for about 30% of all saturated FAMES in the transesterified OAHFAs sample measured by GC/MS. Methyl branching of aliphatic chains is very common in skin-related lipids [41] and chromatographic separation of methyl branching variants of various lipids in similar reversed-phase systems was observed previously [13, 42]. Based on our experience with other lipids, we hypothesize that the more retained species were straight-chain lipids, while those with shorter retention were methyl-branched. Unfortunately, no methyl branched OAHFA standards were available to prove the hypothesis. Unsaturated OAHFAs provided only one distribution of closely eluting peaks. They were considered omega isomers because no  $[HFA - H - HCOOH]^-$  fragments were detected. Similarly to saturated OAHFAs, partially separated peaks were explained by methyl branching and possibly also by a retention time variability of isobaric species having the carbon atoms differently distributed between FA and HFA chains. Within each unsaturation group, the retention time of OAHFAs increased with the total number of carbon atoms. Each double bond decreased retention, in agreement with the equivalent carbon number (ECN) concept (Fig. 6) [43]. The ECN vs. retention time plot (Fig. 7) clearly showed two bands, confirming the existence of two different lipid classes, i.e.,  $\alpha$ OAHFAs and  $\omega$ OAHFAs. The band corresponding to  $\alpha$ OAHFAs is shifted to higher ECN values, which is in agreement with the lower retention of  $\alpha$ OAHFAs in the reversed-phase HPLC. The ECN vs. retention time plot was used to confirm correctness of the structures deduced from mass



**Fig. 5** Base peak chromatograms of OAHFAs isolated from vernix caseosa with internal standard added (a) and reconstructed chromatograms for  $m/z$  705.6766 showing saturated OAHFAs with 46 carbons (b),  $m/z$  703.6615 showing monounsaturated OAHFAs with 46 carbons (c),  $m/z$  701.6459 showing diunsaturated OAHFAs with 46 carbons (d), and  $m/z$  699.6299 showing triunsaturated OAHFAs with 46 carbons (e)

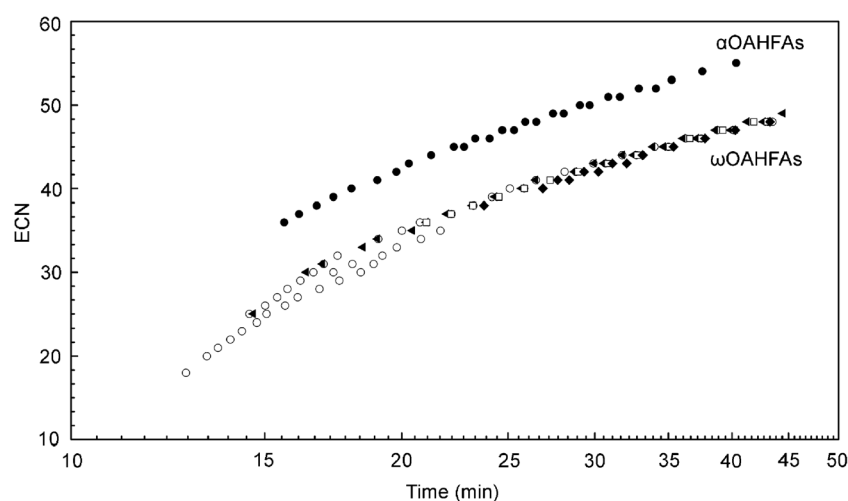
spectra. The ECN value calculated for a molecular species was checked against the ECN value read in the graph for the corresponding retention time, depending on the ester linkage position.



**Fig. 6** Reconstructed chromatograms representing diunsaturated OAHFAs from vernix caseosa with 46 (solid), 47 (dotted), 48 (dashed), and 49 carbons (dot-dash) (a). Reconstructed chromatograms representing saturated (dotted), monounsaturated (dot-dash), diunsaturated (dashed) and triunsaturated (solid) OAHFAs from vernix caseosa with 46 carbons (b)

In total, we identified 250  $\omega$ OAHFA molecular species in 47 chromatographic peaks and 104  $\alpha$ OAHFA molecular species in 18 chromatographic peaks. In addition, another 63 chromatographic peaks gave 58  $\omega$ OAHFA molecular species

and 5  $\alpha$ OAHFA molecular species, which were only partially identified by the total number of carbons and double bonds because the parent ion intensities were not high enough to be fragmented in the data-dependent scan. A list of the 30 most abundant species (i.e., those with the highest intensity of deprotonated molecule in the full-scan spectra) is given in Table 1 and the complete list of all identified OAHFAs can be found in ESM (Table S1). The molecular species of OAHFAs were quantified using internal standard 10-POHSA. A single-point calibration method was used for quantification, considering equal response factors of all OAHFAs including the internal standard. The concentrations of OAHFAs were calculated based on the full-scan extracted ion peak areas relative to that of the internal standard 10-POHSA in the negative ion mode. The OAHFA species with the same mass eluting within the same peak were quantified based on the relative peak intensities of the HFA fragments in the fragmentation spectra. The composition of OAHFAs was also expressed in relative values as peak areas divided by the total (summed) peak area of all OAHFAs in the sample. Both these quantifications are to some extent biased because the response factors within a lipid class are known to depend on the number of double bonds and carbon chain length [44, 45]. Therefore, the quantitative data included in Table 1 and ESM Table S1 are only a rough estimate of the real concentrations and proportions of OAHFAs. The total concentrations of  $\alpha$ OAHFAs and  $\omega$ OAHFAs in the sample were approximately 0.01 mg/ml and 0.13 mg/ml, respectively, which corresponds to 0.003 mg and 0.04 mg in 103.2 mg of vernix caseosa lipids. Therefore, it is estimated that vernix caseosa lipids contained 0.003% of  $\alpha$ OAHFAs and 0.04% of  $\omega$ OAHFAs. The concentration of  $\omega$ OAHFAs in vernix caseosa lipids was significantly lower than in meibum, where these species represent about



**Fig. 7** Plot of calculated equivalent carbon number (ECN) values versus retention times for OAHFAs identified in vernix caseosa (ECN = CN - 2 dB where CN is the total number of carbons and DB is the total number of double bonds in OAHFAs). Black circles, saturated  $\alpha$ OAHFAs

species; white circles, saturated  $\omega$ OAHFAs species; black triangles, monounsaturated  $\omega$ OAHFAs species; white squares, diunsaturated  $\omega$ OAHFAs species; and black rhombuses, triunsaturated  $\omega$ OAHFAs species



**Table 1** List of the 30 most abundant OAHFAs identified in vernix caseosa using HPLC/MS<sup>2</sup>. A full list of all 417 OAHFAs can be found in the ESM, Table S1

<i>t</i> <sub>R</sub> min	<i>m/z</i> [M - H] <sup>-</sup>	Conc. in total lipids ppm	Rel. peak area %	Identification
18.01	621.58	1.92	4.5	16:0/α-24:0
21.28	677.65	1.57	3.6	20:0/α-24:0
23.23	645.58	1.56	3.7	18:2/ω-24:0
25.85	673.61	3.88	9.0	18:2/ω-26:0
25.91	733.70	3.05	7.1	24:0/α-24:0
26.88	699.63	1.53	3.6	18:3/ω-28:0
27.30	687.63	2.62	6.1	18:2/ω-27:0
28.96	701.65	2.14	5.0	16:1/ω-30:1
28.96	701.65	32.46	75.6	18:2/ω-28:0
29.06	761.74	1.82	4.3	26:0/α-24:0
29.33	727.66	7.30	17.0	18:2/ω-30:1
30.24	727.66	5.35	12.5	18:3/ω-30:0
30.82	715.66	32.57	76.0	18:2/ω-29:0
31.11	741.68	4.03	9.4	18:2/ω-31:1
32.53	703.66	1.65	3.8	16:1/ω-30:0
32.53	703.66	2.71	6.3	18:1/ω-28:0
32.82	729.68	3.50	8.2	16:1/ω-32:1
32.82	729.68	106.15	247.4	18:2/ω-30:0
32.82	729.68	5.83	13.6	18:1/ω-30:1
33.15	755.69	24.57	57.3	18:2/ω-32:1
34.60	717.68	2.37	5.5	18:1/ω-29:0
35.05	743.69	2.18	5.1	17:1/ω-32:1
35.05	743.69	20.86	48.6	18:2/ω-31:0
35.37	769.71	2.91	6.8	18:2/ω-33:1
36.59	757.71	4.38	10.2	16:1/ω-34:1
36.59	757.71	8.08	18.8	18:1/ω-32:1
37.11	731.69	8.87	20.7	18:1/ω-30:0
37.50	757.71	14.40	33.6	18:2/ω-32:0
37.78	783.72	10.71	24.9	18:2/ω-34:1
41.88	785.74	2.68	6.3	18:1/ω-34:1

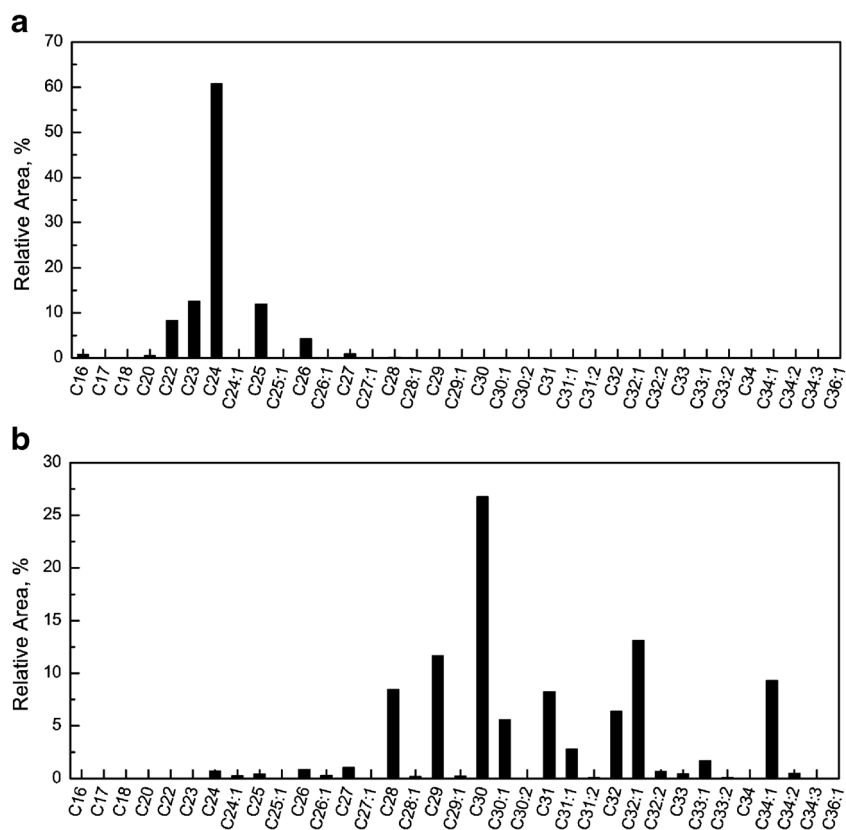
3–3.5% of the total lipids [46]. All saturated ωOAHFAs corresponded to 9% of the total integrated signal, whereas monounsaturated, diunsaturated, and triunsaturated ωOAHFA species accounted for 10, 60, and 15%, respectively. The saturated αOAHFAs corresponded to the remaining 6% of the integrated signal. The α-isomers and ω-isomers corresponded to 6 and 94% of OAHFAs, respectively.

Vernix caseosa ωOAHFAs appeared to be mostly diunsaturated species with ωHFAs 30:0, 29:0, 28:0, 32:1, 34:1, and FA 18:2 identified as linoleic acid (retention time and mass spectra matched with methyl linoleate in GC/MS). Among the most abundant ωOAHFAs were 28:0/ω-18:2, 29:0/ω-18:2, 30:0/ω-18:2, 32:0/ω-18:2, and 30:0/ω-18:3. The composition of αOAHFAs differed significantly from the composition of ωOAHFAs. The abundant αOAHFAs contained αHFA 24:0 and a wide range of odd and even-chained saturated FA with 20–26 carbons. The main species were 21:0/α-24:0, 22:0/α-24:0, 23:0/α-24:0, 24:0/α-24:0, and 26:0/α-24:0. The distributions of ωHFAs and αHFAs in OAHFAs are shown in Fig. 8 and the distributions of FAs in ωOAHFAs and αOAHFAs are provided in Fig. 9. While ωOAHFAs were composed of 32 different ωHFAs with 16–36 carbons (up to 3 double bonds) and 28 different FAs with 12–24 (up to three double bonds), αOAHFAs were

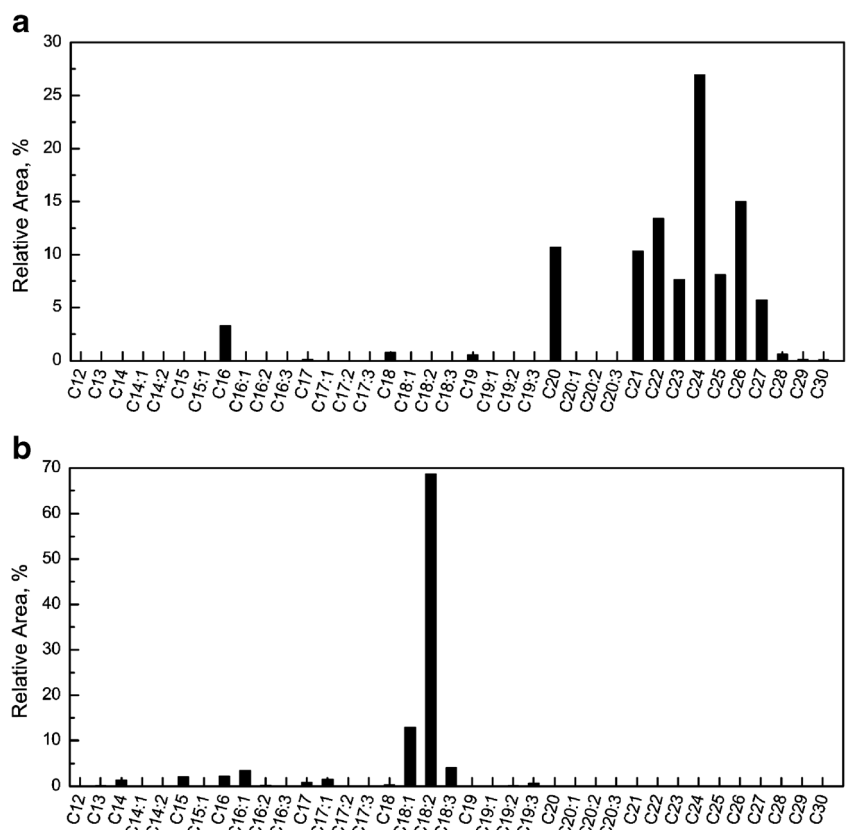
composed of 12 saturated αHFAs with 16–29 carbons and 16 saturated FAs with 15–30 carbons.

The existence of ωOAHFAs and αOAHFAs in vernix caseosa raises questions about their origin and biological role. By comparing the composition of the acyl chains with other lipids, one can estimate the biosynthetic relationships between them. The ωOAHFA motifs exist in the skin and vernix caseosa omega-*O*-acylceramides, where they are linked by an amide bond to sphingoid base (to sphingosine in EOS (Cer1), 6-hydroxysphingosine in EOH (Cer4), and phytosphingosine in EOP (Cer9)) [47]. The ωOAHFA moiety consists of ultra-long ωHFA chains esterified primarily to linoleic acid. The *O*-acylceramides are essential for the formation of the epidermal barrier, which protects the skin from environmental factors and trans-epidermal water loss [48]. The altered content of *O*-acylceramides in the skin may result in atopic dermatitis [49] or lamellar ichthyosis [50]. In vernix caseosa, *O*-acylceramides constitute about 25% of all ceramides [15]. The ωOAHFA motifs also exist in Chl-ωOAHFA diesters [14, 21, 51]. These lipids are composed of ωOAHFAs linked by an ester bond to cholesterol or lathosterol, and they account for approximately 1–2% of vernix caseosa lipids. The most abundant species are composed of cholesterol, ωHFA 32:1, and common skin FAs like 14:0, 15:0, 16:0, 16:1 or 18:1. The different composition of the

**Fig. 8** Histograms showing relative proportions of  $\alpha$ HFAs in  $\alpha$ OAHFAs (a) and  $\omega$ HFAs in  $\omega$ OAHFAs (b) from vernix caseosa calculated using HPLC/MS<sup>2</sup> data (see ESM Table S1, the extended version of Table 1)



**Fig. 9** Histograms showing relative proportions of FAs in  $\alpha$ OAHFAs (a) and  $\omega$ OAHFAs (b) from vernix caseosa calculated using HPLC/MS<sup>2</sup> data (see ESM Table S1, the extended version of Table 1)



OAHFA moieties in Chl- $\omega$ OAHFAs and *O*-acylceramides stems from their different origins. While ceramides are considered products of epidermal development, Chl- $\omega$ OAHFAs are of sebaceous origin [7]. The existence of  $\alpha$ OAHFA motif in vernix caseosa lipids remains elusive. There are pieces of evidence that  $\alpha$ OAHFA subunits exist in *O*-acylceramides and diester waxes, but no conclusive proof has been given so far. In the EOS (Cer 1) fraction, about 8% of  $\alpha$ HFA was reported [25]. It suggests the existence of EAS ceramides consisting of a sphingoid base *N*-linked to  $\alpha$ -hydroxy FAs and esterified with another FA. EAS ceramides were indeed isolated and reported; however, their structure was not clarified [25]. The most frequently occurring  $\omega$ HFA chains in vernix caseosa  $\omega$ OAHFAs were saturated (30:0, 29:0, 28:0, 31:0) and monounsaturated (32:1, 34:1), while species with monounsaturated  $\omega$ HFAs 30:1, 32:1, and 34:1 prevail in meibum [27, 34]. Higher levels of monounsaturated  $\omega$ HFAs in  $\omega$ OAHFAs were observed in equine sperm and amniotic fluid [22, 23], as well as in Chl- $\omega$ OAHFAs in vernix caseosa [14]. The  $\omega$ HFA chains in vernix caseosa  $\omega$ OAHFAs were closest to  $\omega$ HFAs detected in *O*-acylceramides EOS [25]. The structural similarity to EOS is further manifested by the dominant presence of linoleic acid [25]. Linoleic acid can also be found in  $\omega$ OAHFAs from meibum or equine sperm and amniotic fluid, but monounsaturated chains like 16:1 and 18:1 are significantly more abundant [23, 27, 34]. Therefore,  $\omega$ OAHFAs can be biosynthetically related to *O*-acylceramides EOS. As regards intact (free)  $\alpha$ OAHFAs, they have not been identified in any biological sample yet. The  $\alpha$ OAHFA motif exists in Type I diesters waxes ( $\alpha$ HFAs esterified by FA on its hydroxyl group and by fatty alcohol on its carboxyl group) found in animal skin. These lipids contain short-chain saturated  $\alpha$ HFA with 14–18 carbons and almost all FA chains are saturated, having 8–32 carbon atoms, both odd and even, with relatively large amounts of FAs 14:0 and 16:0 [52]. The  $\alpha$ OAHFA moieties in the Type I diesters waxes from animal skin are thus not similar to  $\alpha$ OAHFAs detected in vernix caseosa.

In this work, OAHFAs were characterized from a pooled sample of vernix caseosa equally representing the gender of newborns. Additional experiments were performed with samples sorted by sex of the newborns, but no differences were found. Therefore, the OAHFAs composition in vernix caseosa is likely unrelated to gender.

The function of OAHFAs in vernix caseosa remains to be clarified. As amphiphiles, OAHFAs may act as surfactants contributing to the cohesiveness of vernix caseosa. They may provide interphase between nonpolar lipids of vernix caseosa and the aqueous environment of the amniotic fluid.

**Acknowledgments** This work was supported by the Grant Agency of Charles University in Prague (Project No. 1182216), the Charles University in Prague (Project SVV260440), and from European Regional Development Fund; OP RDE; Project: “Chemical biology for drugging undruggable targets (ChemBioDrug)” (No. CZ.02.1.01/0.0/0.0/16\_019/0000729).

## Compliance with ethical standards

**Conflict of interest** The authors declare that they have no conflict of interest.

**Ethical approval** The study was approved by the Ethics Committee of the General University Hospital in Prague (910/09 S-IV).

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