ABSTRACT

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Title of thesis: Synthesis and Study of New Unsymmetrical

Azaphthalocyanines bearing azathiocrown ether for

cation binding

Azaphthalocyanines (AzaPc) are planar macrocyclic compounds, which are azaanalogues of synthetic phthalocyanine dyes. AzaPc, due to their system of conjugated
double bonds, show interesting photophysical properties such as fluorescence emission
and absorption in the red area of the spectrum, which is suitable for biological
applications. These properties depend on character of peripheral substituents and central
metal cation. The attachment of alkylamine group enables intramolecular charge
transfer (ICT), that causes efficient quenching of fluorescence (non-fluorescent OFF
state of the sensor), and leads to decrease of fluorescence quantum yield to zero.
Blocking of ICT is used in development of new fluorescent sensors for metal cations, in
which metal cation is coordinated into recognition moiety (e.g. azacrown ether) of the
sensor. It results in a significant increase of fluorescence (fluorescent ON state
of the sensor).

Preparation of new AzaPc sensor started *via* a multi-step procedure resulting in a new azathiocrown (*i.e.* Boc-1,4,7,10-tetrathia-13-azacyclopentadecane), which was deprotected and used in a nucleophilic substitution with 5,6-dichloro-2,3-dicarbonitrile. Desired monosubstituted intermidiate, suitable for subsequent attachment of butoxy group, was not prepared in sufficient purity to be used further to obtain key precursor 5-buthoxy-6-(monoazatetrathia-15-crown-5)-2,3-dicarbonitrile. Another precursor, 5,6-bis(*tert*-butylsulfanyl)pyrazine-2,3-dicarbonitrile was successfully synthesized according to procedure published in literature. The results of this Diploma Thesis will be useful

for a follow-up optimization of reaction steps to get target AzaPc, its sensitivity toward heavy metal cations will be investigated $(Pb^{2+}, Cd^{2+}, Hg^{2+})$.